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(19) **United States**(12) **Patent Application Publication**  
**Xia et al.**(10) **Pub. No.: US 2013/0140549 A1**(43) **Pub. Date: Jun. 6, 2013**(54) **BICARBAZOLE COMPOUNDS FOR OLEDs****Publication Classification**(75) Inventors: **Chuanjun Xia**, Lawrenceville, NJ (US);  
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**Ming-Cheng Kuo**, Taichung County (TW)(51) **Int. Cl.**  
**H01L 51/00** (2006.01)  
(52) **U.S. Cl.**  
CPC ..... **H01L 51/0067** (2013.01)  
USPC ..... **257/40; 544/212**(73) Assignee: **UNIVERSAL DISPLAY CORPORATION**, Ewing, NJ (US)(57) **ABSTRACT**(21) Appl. No.: **13/816,407**(22) PCT Filed: **Aug. 20, 2010**(86) PCT No.: **PCT/US2010/046218**§ 371 (c)(1),  
(2), (4) Date: **Feb. 11, 2013**

Novel organic compounds comprising a bicarbazole core are provided. In particular, the compounds has a 3,3'-bicarbazole core substituted at the 9-position with a triazine or pyrimidine. The compounds may be used in organic light emitting devices to provide devices having improved efficiency and improved lifetime.

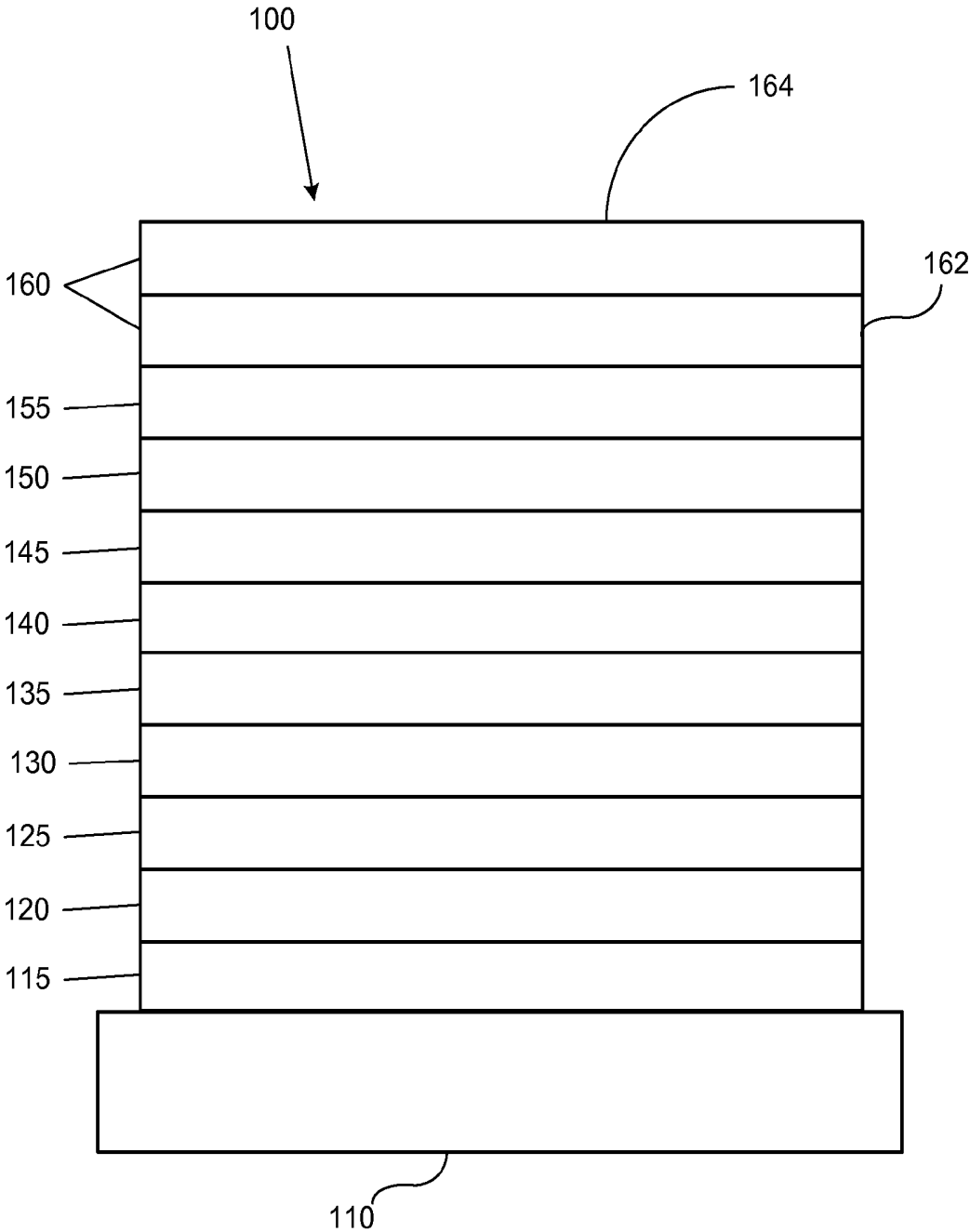


FIGURE 1

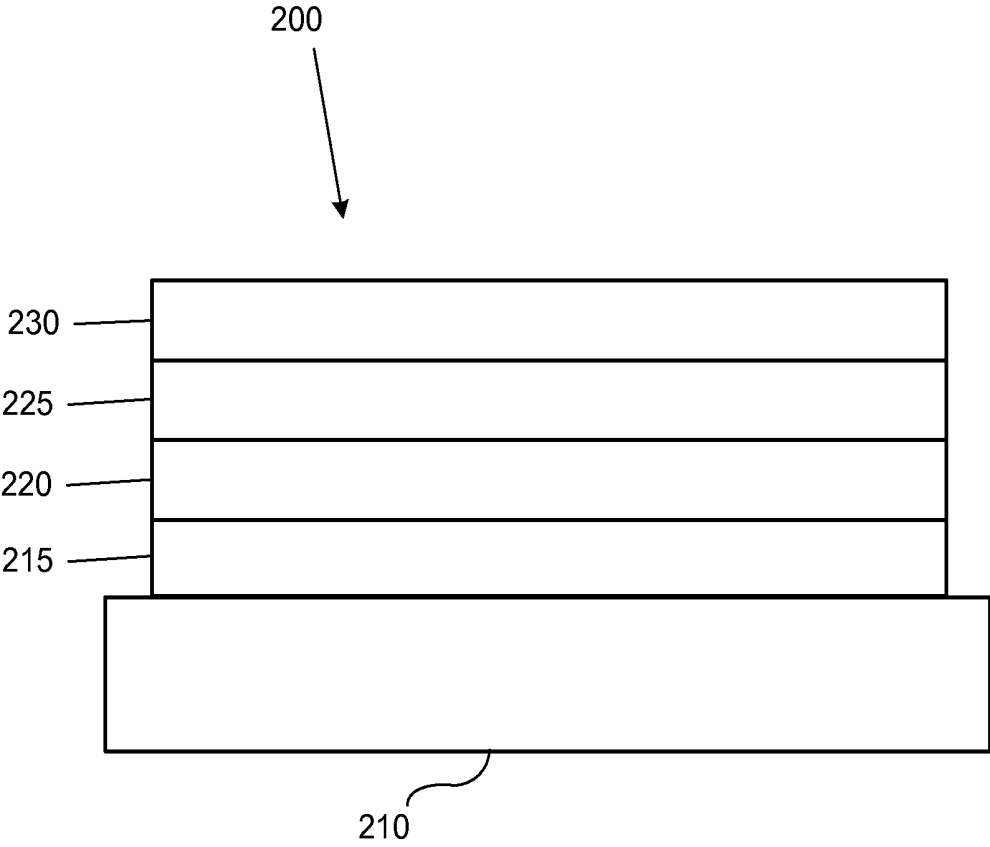


FIGURE 2

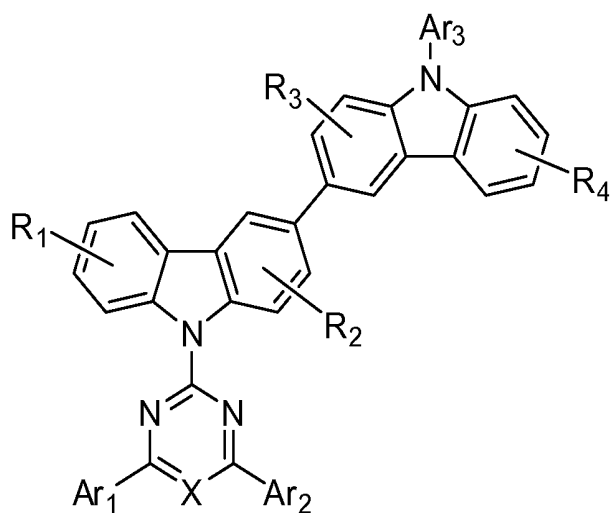


FIGURE 3

## BICARBAZOLE COMPOUNDS FOR OLEDS

[0001] The claimed invention was made by, on behalf of, and/or in connection with one or more of the following parties to a joint university corporation research agreement: Regents of the University of Michigan, Princeton University, The University of Southern California, and the Universal Display Corporation. The agreement was in effect on and before the date the claimed invention was made, and the claimed invention was made as a result of activities undertaken within the scope of the agreement.

### FIELD OF THE INVENTION

[0002] The present invention relates to organic light emitting devices (OLEDs). More specifically, the present invention pertains to phosphorescent organic materials comprising a bicarbazole having a nitrogen-containing heterocycle at the 9 position.

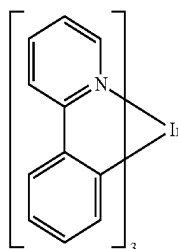
### BACKGROUND

[0003] Opto-electronic devices that make use of organic materials are becoming increasingly desirable for a number of reasons. Many of the materials used to make such devices are relatively inexpensive, so organic opto-electronic devices have the potential for cost advantages over inorganic devices. In addition, the inherent properties of organic materials, such as their flexibility, may make them well suited for particular applications such as fabrication on a flexible substrate. Examples of organic opto-electronic devices include organic light emitting devices (OLEDs), organic phototransistors, organic photovoltaic cells, and organic photodetectors. For OLEDs, the organic materials may have performance advantages over conventional materials. For example, the wavelength at which an organic emissive layer emits light may generally be readily tuned with appropriate dopants.

[0004] OLEDs make use of thin organic films that emit light when voltage is applied across the device. OLEDs are becoming an increasingly interesting technology for use in applications such as flat panel displays, illumination, and backlighting. Several OLED materials and configurations are described in U.S. Pat. Nos. 5,844,363, 6,303,238, and 5,707,745, which are incorporated herein by reference in their entirety.

[0005] One application for phosphorescent emissive molecules is a full color display. Industry standards for such a display call for pixels adapted to emit particular colors, referred to as “saturated” colors. In particular, these standards call for saturated red, green, and blue pixels. Color may be measured using CIE coordinates, which are well known to the art.

[0006] One example of a green emissive molecule is tris(2-phenylpyridine) iridium, denoted Ir(ppy)<sub>3</sub>, which has the structure:



[0007] In this, and later figures herein, we depict the dative bond from nitrogen to metal (here, Ir) as a straight line.

[0008] As used herein, the term “organic” includes polymeric materials as well as small molecule organic materials that may be used to fabricate organic opto-electronic devices. “Small molecule” refers to any organic material that is not a polymer, and “small molecules” may actually be quite large. Small molecules may include repeat units in some circumstances. For example, using a long chain alkyl group as a substituent does not remove a molecule from the “small molecule” class. Small molecules may also be incorporated into polymers, for example as a pendent group on a polymer backbone or as a part of the backbone. Small molecules may also serve as the core moiety of a dendrimer, which consists of a series of chemical shells built on the core moiety. The core moiety of a dendrimer may be a fluorescent or phosphorescent small molecule emitter. A dendrimer may be a “small molecule,” and it is believed that all dendrimers currently used in the field of OLEDs are small molecules.

[0009] As used herein, “top” means furthest away from the substrate, while “bottom” means closest to the substrate. Where a first layer is described as “disposed over” a second layer, the first layer is disposed further away from substrate. There may be other layers between the first and second layer, unless it is specified that the first layer is “in contact with” the second layer. For example, a cathode may be described as “disposed over” an anode, even though there are various organic layers in between.

[0010] As used herein, “solution processible” means capable of being dissolved, dispersed, or transported in and/or deposited from a liquid medium, either in solution or suspension form.

[0011] A ligand may be referred to as “photoactive” when it is believed that the ligand directly contributes to the photoactive properties of an emissive material. A ligand may be referred to as “ancillary” when it is believed that the ligand does not contribute to the photoactive properties of an emissive material, although an ancillary ligand may alter the properties of a photoactive ligand.

[0012] As used herein, and as would be generally understood by one skilled in the art, a first “Highest Occupied Molecular Orbital” (HOMO) or “Lowest Unoccupied Molecular Orbital” (LUMO) energy level is “greater than” or “higher than” a second HOMO or LUMO energy level if the first energy level is closer to the vacuum energy level. Since ionization potentials (IP) are measured as a negative energy relative to a vacuum level, a higher HOMO energy level corresponds to an IP having a smaller absolute value (an IP that is less negative). Similarly, a higher LUMO energy level corresponds to an electron affinity (EA) having a smaller absolute value (an EA that is less negative). On a conventional energy level diagram, with the vacuum level at the top, the LUMO energy level of a material is higher than the HOMO energy level of the same material. A “higher” HOMO or LUMO energy level appears closer to the top of such a diagram than a “lower” HOMO or LUMO energy level.

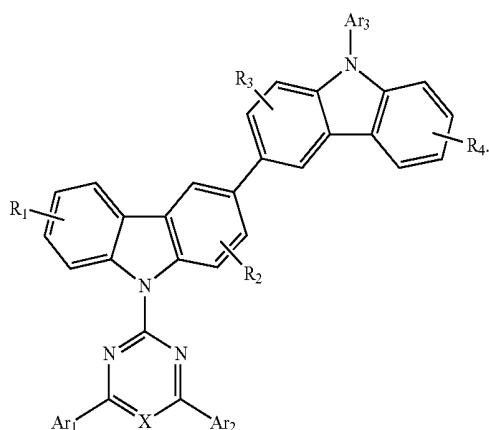
[0013] As used herein, and as would be generally understood by one skilled in the art, a first work function is “greater than” or “higher than” a second work function if the first work function has a higher absolute value. Because work functions

are generally measured as negative numbers relative to vacuum level, this means that a “higher” work function is more negative. On a conventional energy level diagram, with the vacuum level at the top, a “higher” work function is illustrated as further away from the vacuum level in the downward direction. Thus, the definitions of HOMO and LUMO energy levels follow a different convention than work functions.

[0014] More details on OLEDs, and the definitions described above, can be found in U.S. Pat. No. 7,279,704, which is incorporated herein by reference in its entirety.

#### SUMMARY OF THE INVENTION

[0015] Compounds comprising a bicarbazole are provided. The compounds have the formula:



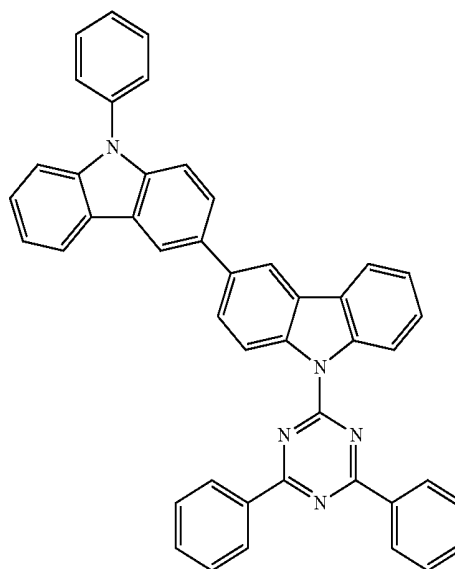
[0016]  $R_1$ ,  $R_2$ ,  $R_3$ , and  $R_4$  may represent mono, di, tri, or tetra substitutions.  $R_1$ ,  $R_2$ ,  $R_3$ , and  $R_4$  are independently selected from the group consisting of hydrogen, alkyl, alkoxy, amino, alkenyl, alkynyl, aryl and heteroaryl.  $Ar_1$ ,  $Ar_2$ , and  $Ar_3$  are independently selected from aryl or heteroaryl.  $Ar_1$ ,  $Ar_2$ , and  $Ar_3$  may be further substituted.  $X$  is C or N.

[0017] In one aspect,  $Ar_1$ ,  $Ar_2$ , and  $Ar_3$  are independently selected from the group consisting of phenyl, pyridine, naphthalene, biphenyl, terphenyl, fluorene, dibenzofuran, dibenzothiophene, phenanthrene, and triphenylene.  $Ar_1$ ,  $Ar_2$ , and  $Ar_3$  are independently further substituted with a substituent selected from the group consisting of hydrogen, alkyl, alkoxy, amino, alkenyl, alkynyl, aryl and heteroaryl, but the substituent is not an aryl or heteroaryl fused directly to  $Ar_1$ ,  $Ar_2$ , and  $Ar_3$ . Preferably,  $Ar_1$  and  $Ar_2$  are independently selected from the group consisting of phenyl, pyridine, and naphthalene. Preferably,  $Ar_3$  is selected from the group consisting of phenyl, biphenyl, dibenzofuran, and dibenzothiophene.

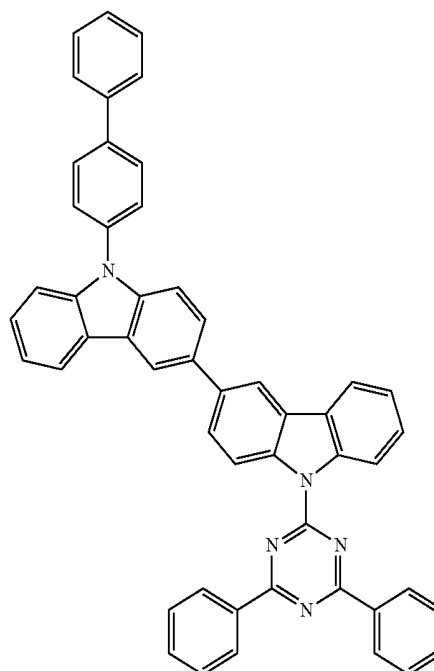
[0018] In another aspect,  $R_1$ ,  $R_2$ ,  $R_3$ , and  $R_4$  are hydrogen.

[0019] Specific examples of compounds comprising bicarbazole are also provided. In particular, the compound is selected from the group consisting of:

Compound 1



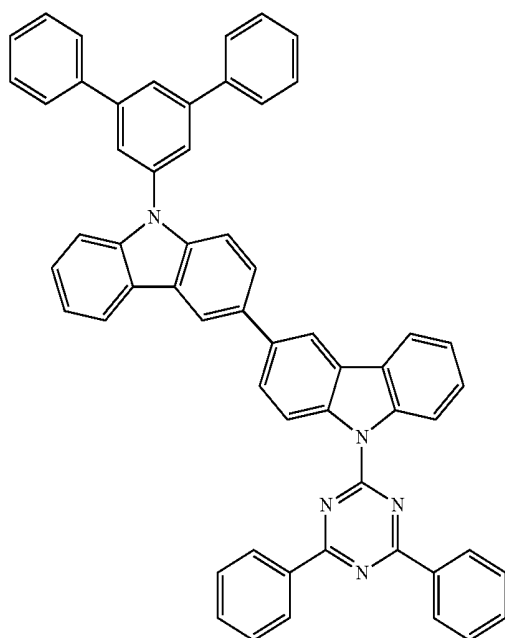
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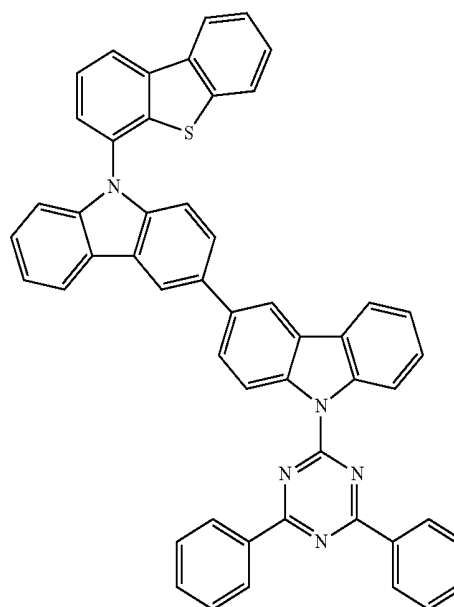
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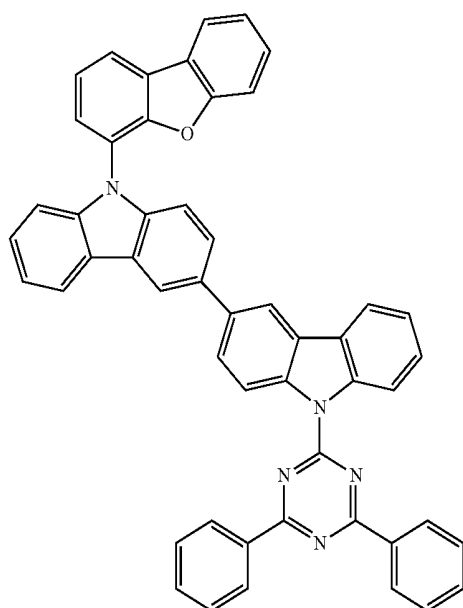
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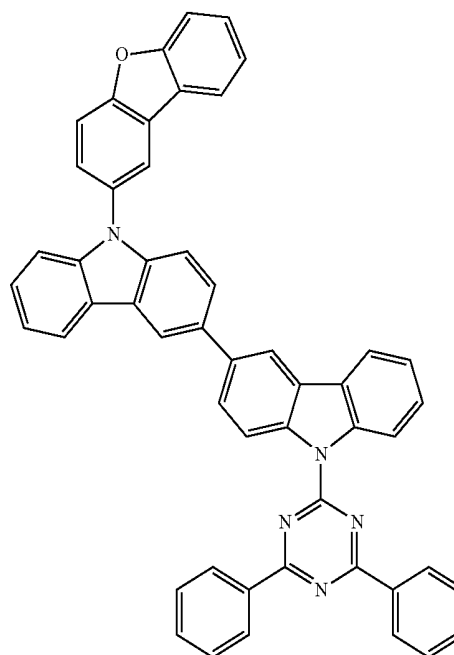
Compound 5



Compound 4



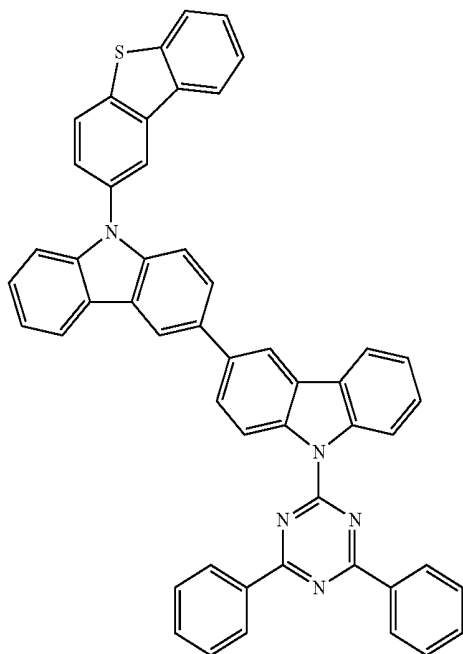
Compound 6



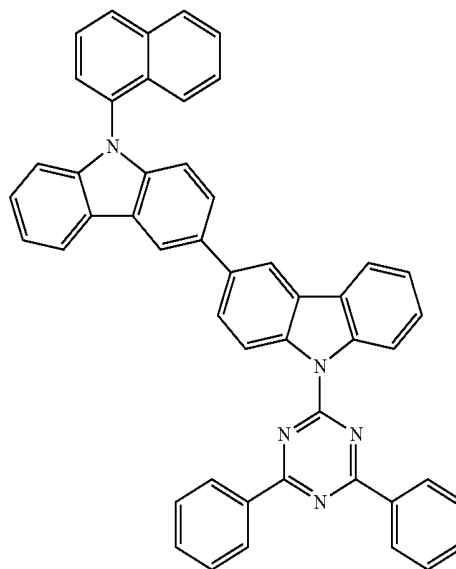
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Compound 7

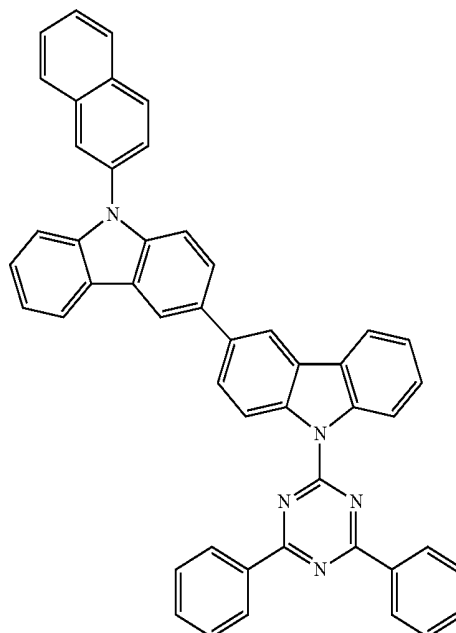
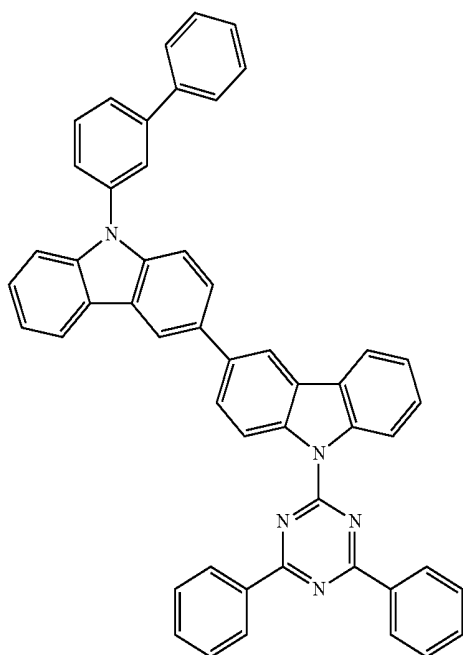


Compound 9



Compound 10

Compound 8

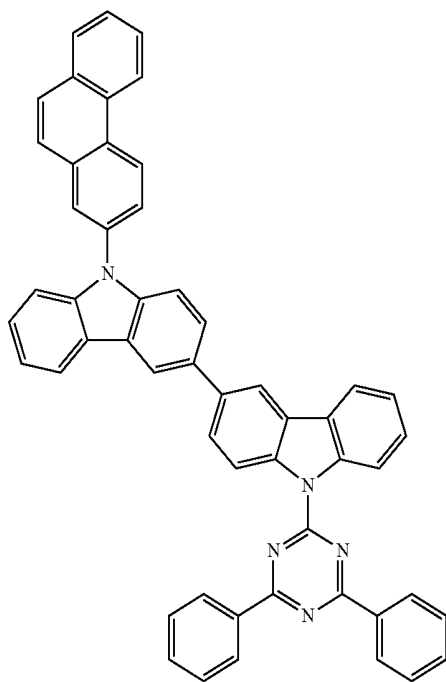




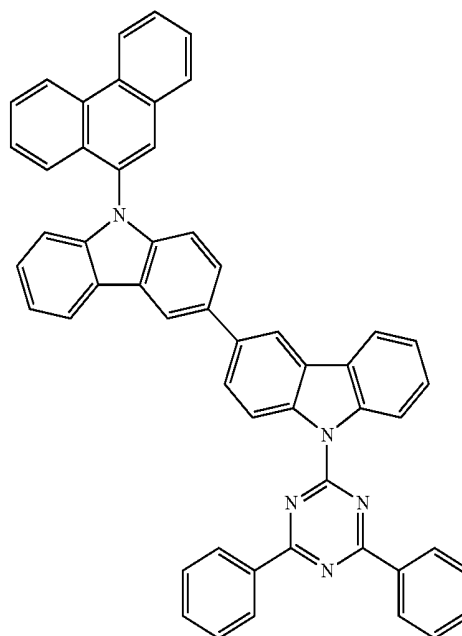
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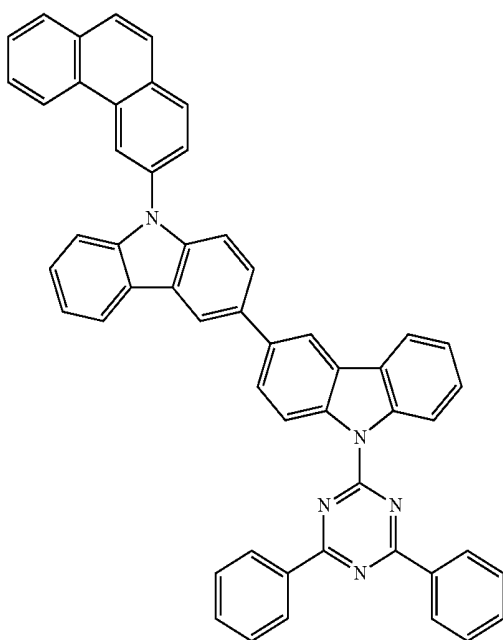
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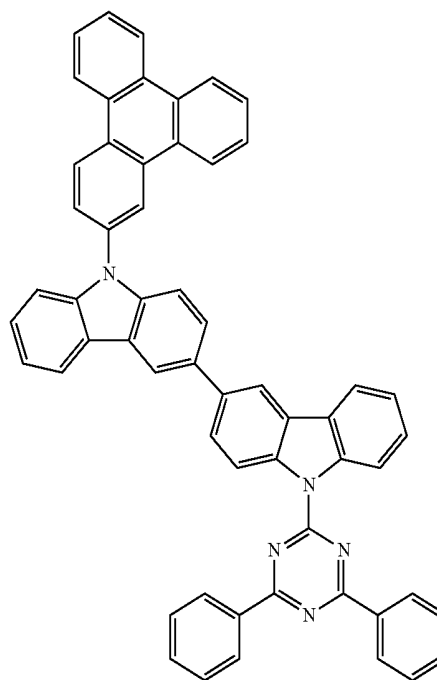
Compound 13



Compound 12



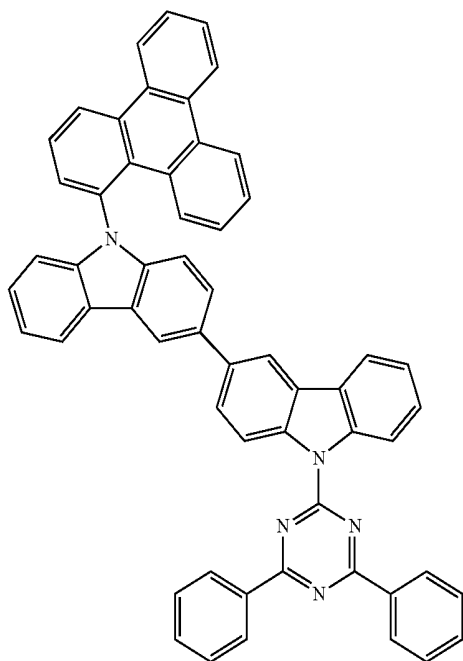
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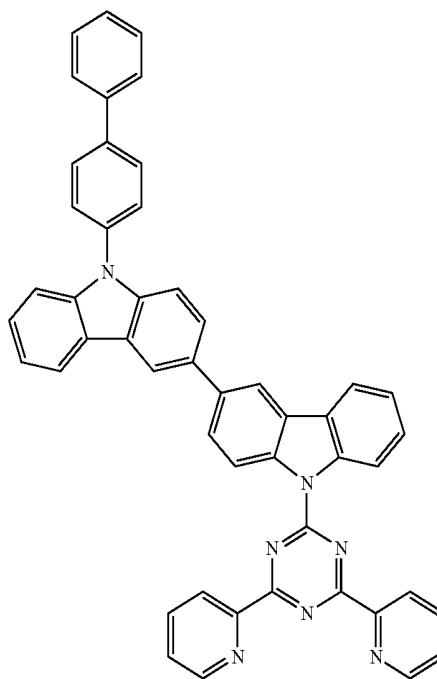
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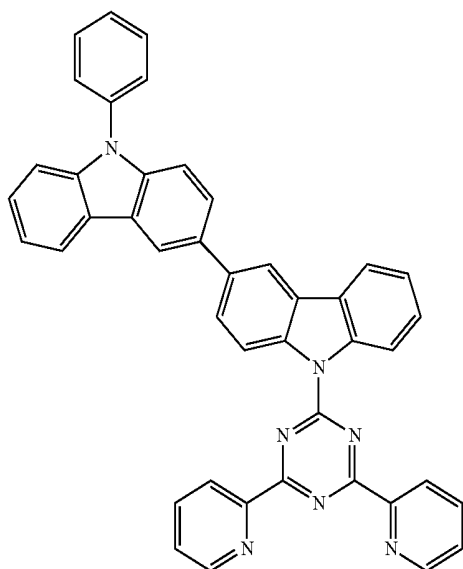
Compound 15



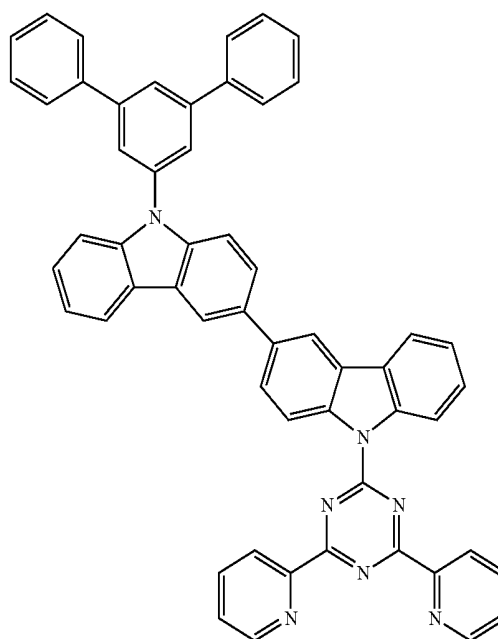
Compound 17



Compound 16



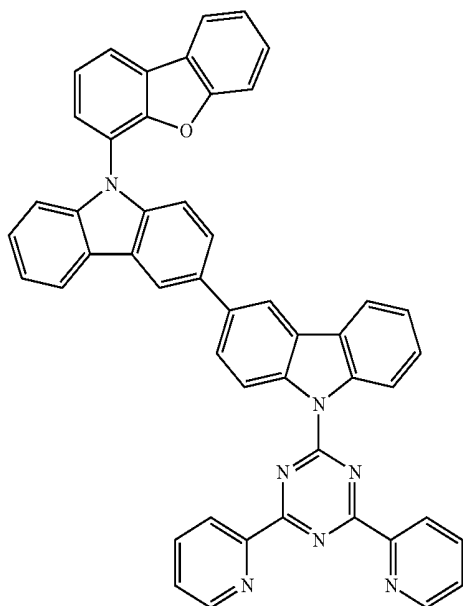
Compound 18



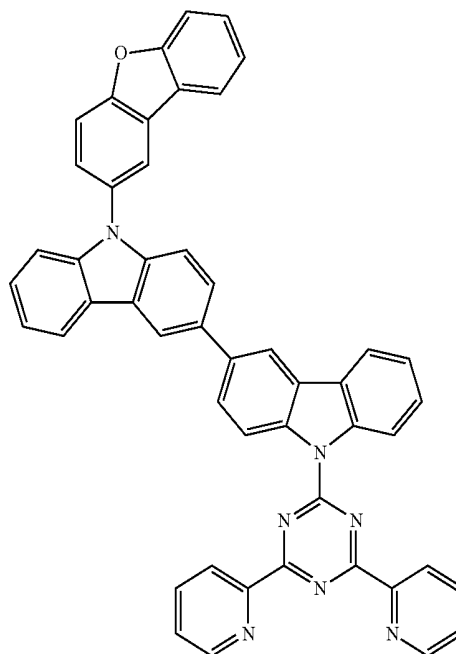
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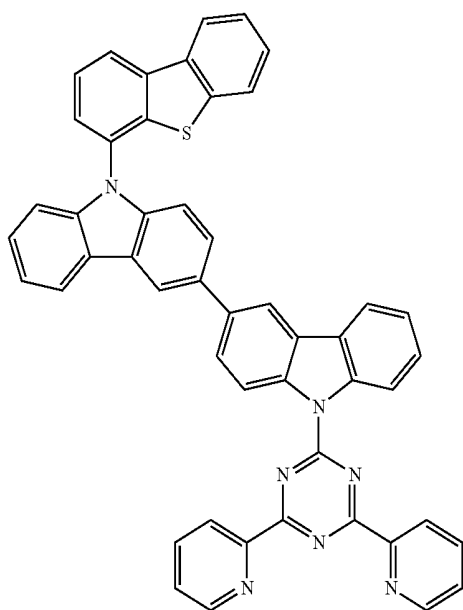
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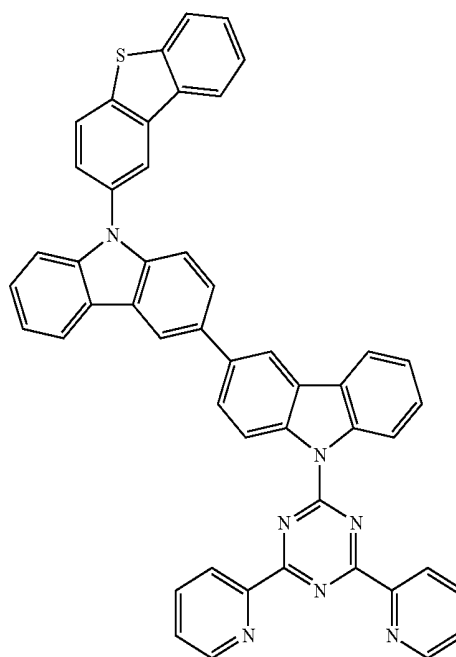
Compound 21



Compound 20



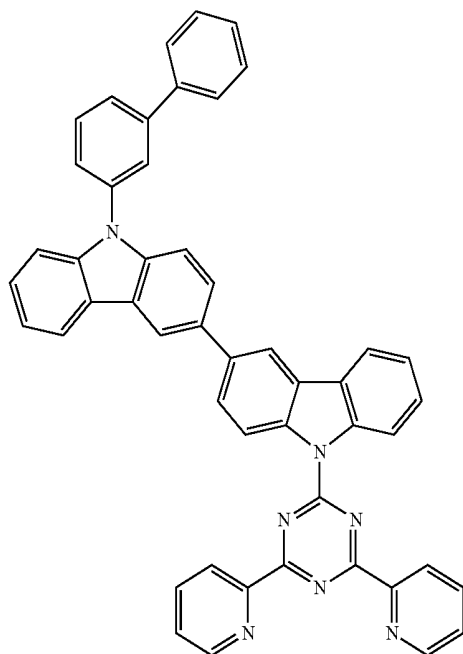
Compound 22



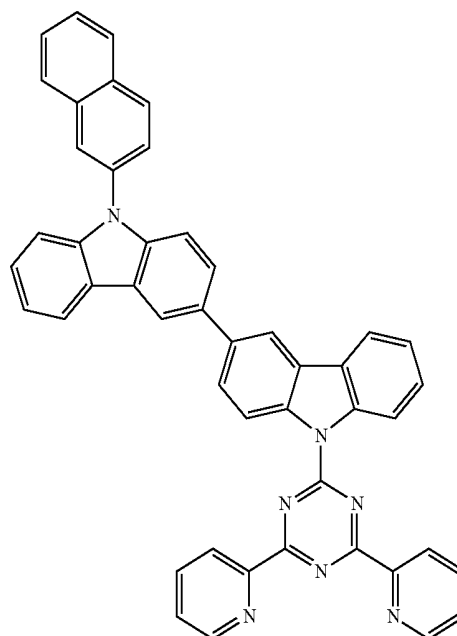
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Compound 23

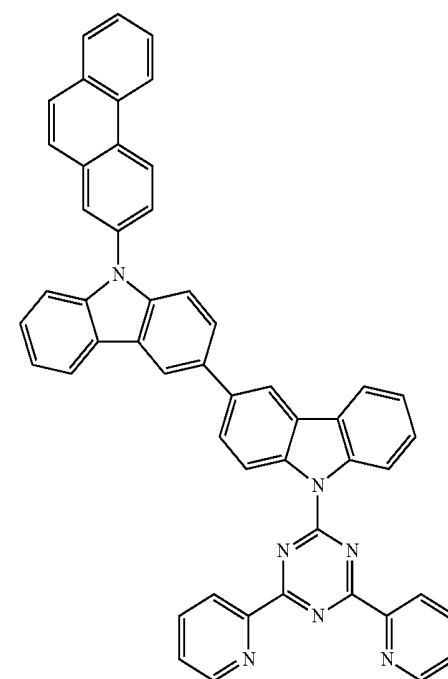
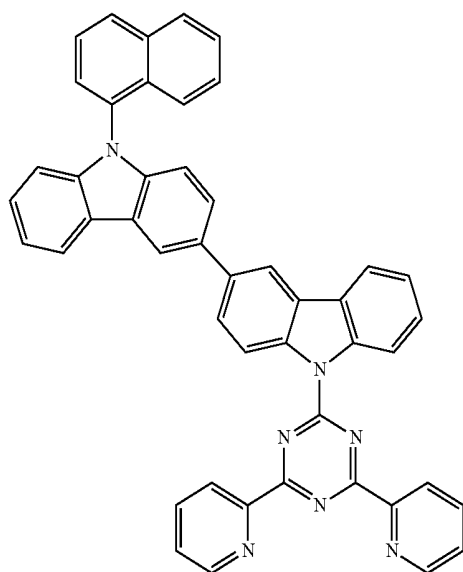


Compound 25



Compound 26

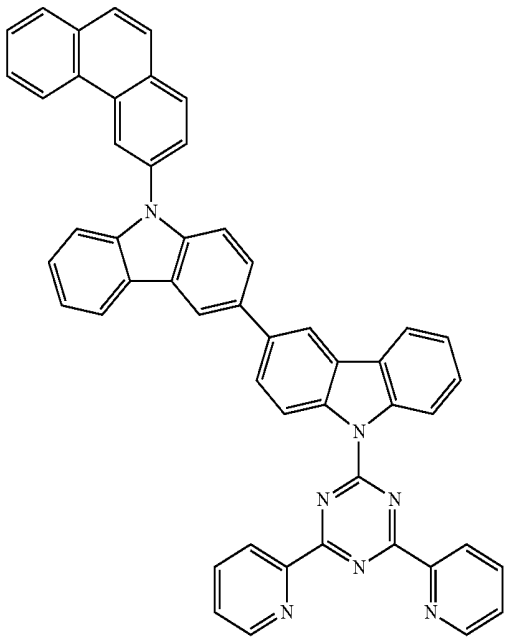
Compound 24



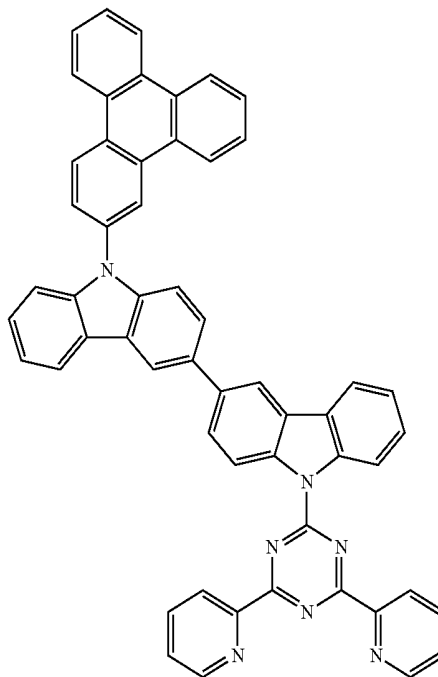
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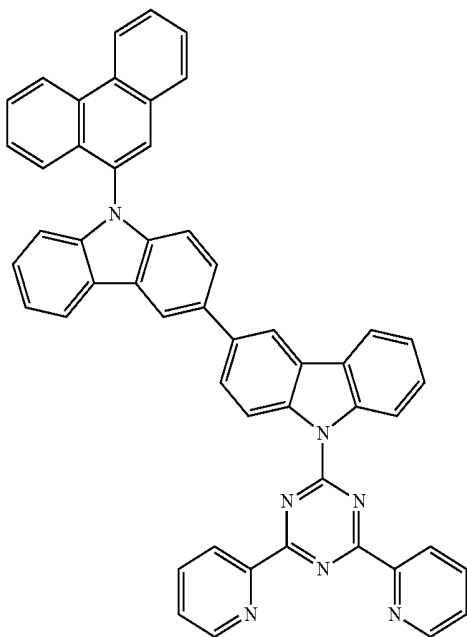
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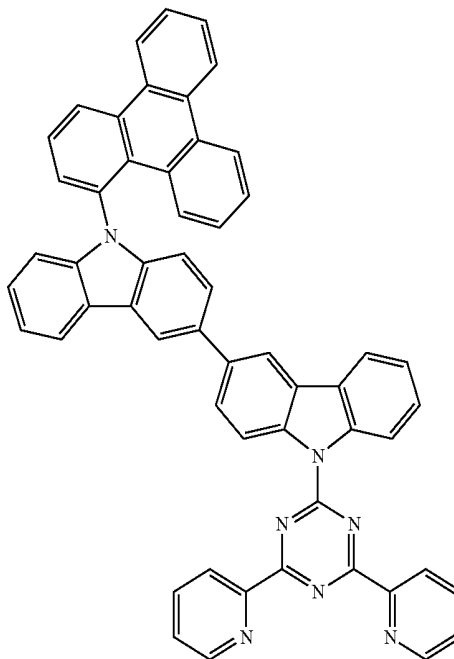
Compound 29



Compound 28



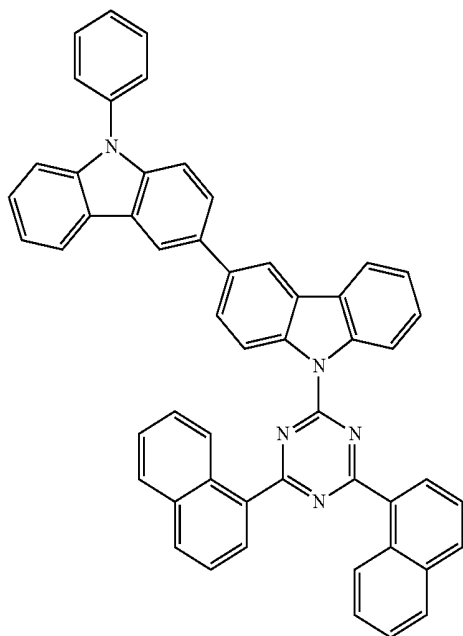
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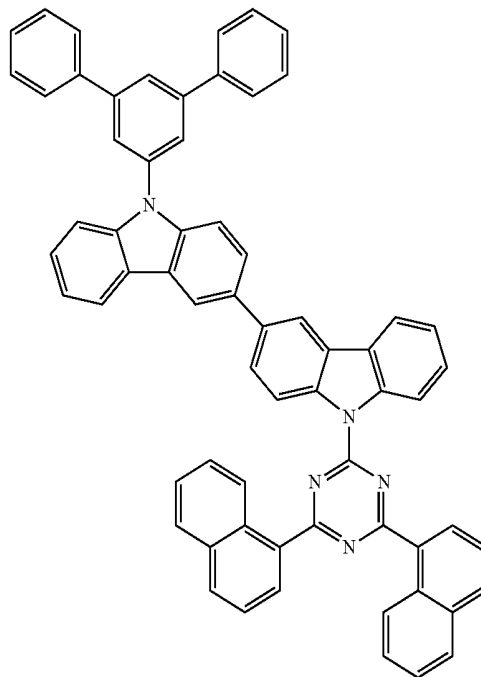
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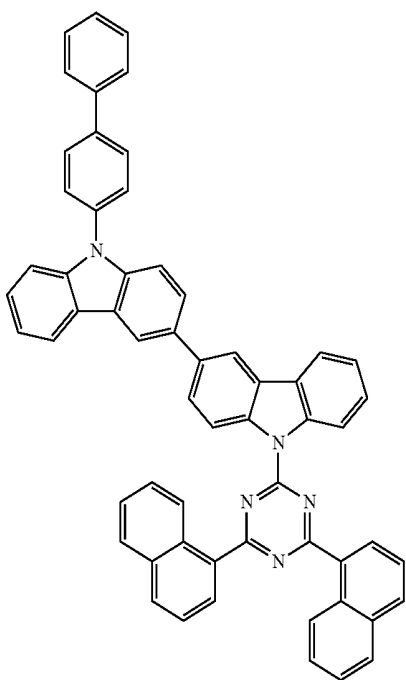
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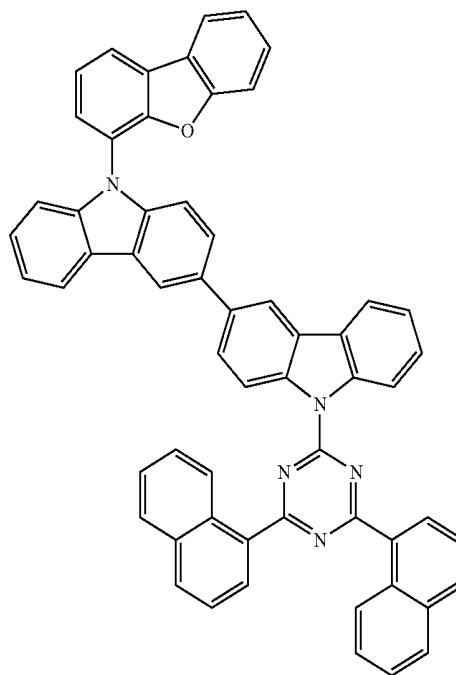
Compound 33



Compound 32



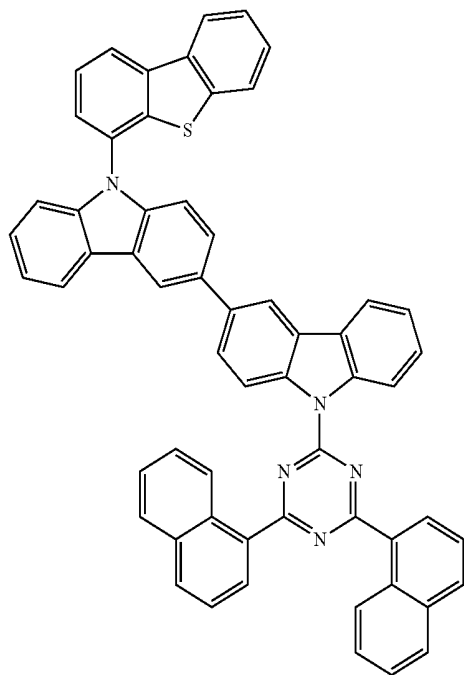
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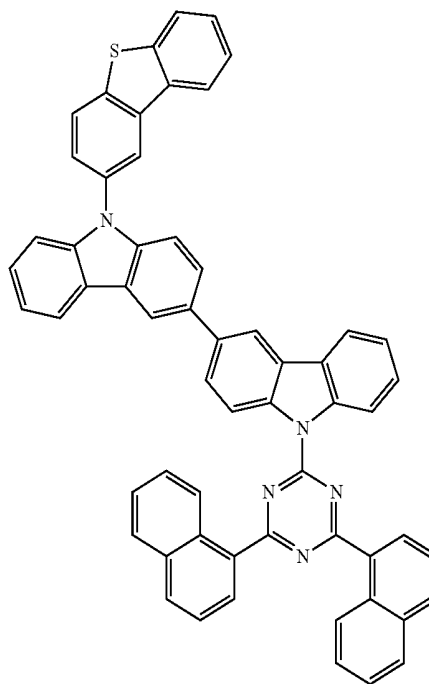
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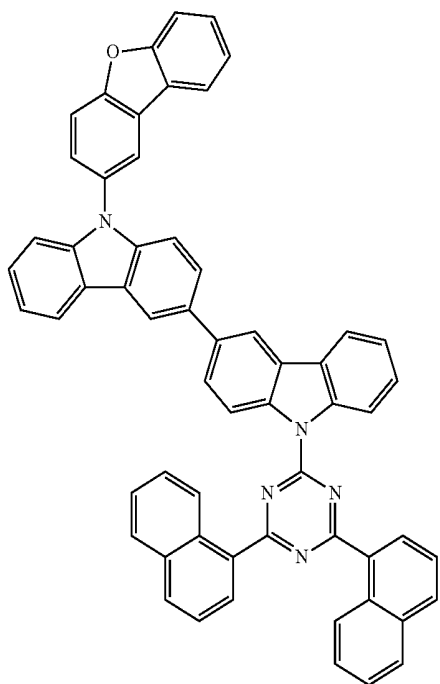
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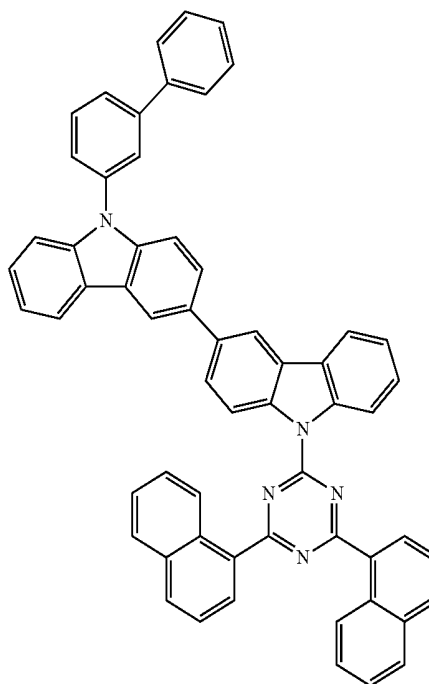
Compound 37



Compound 36



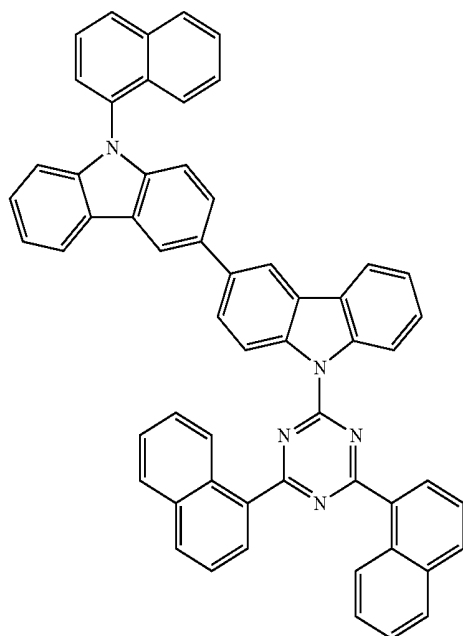
Compound 38



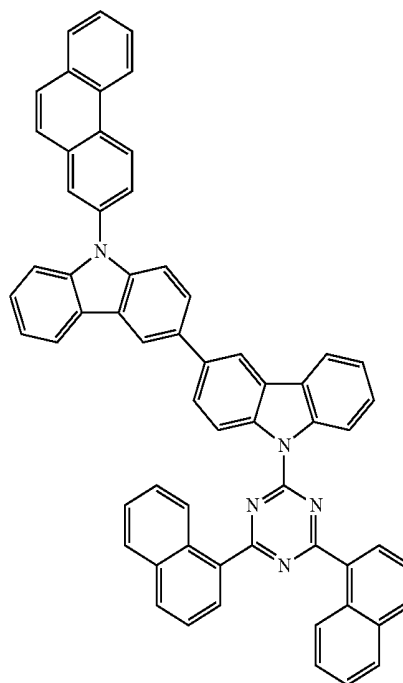
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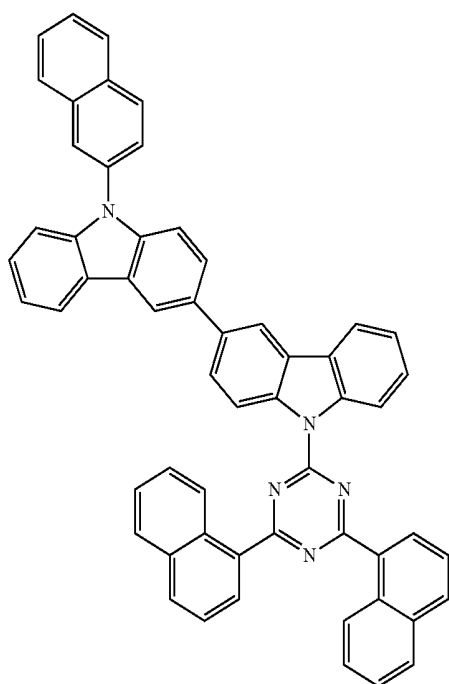
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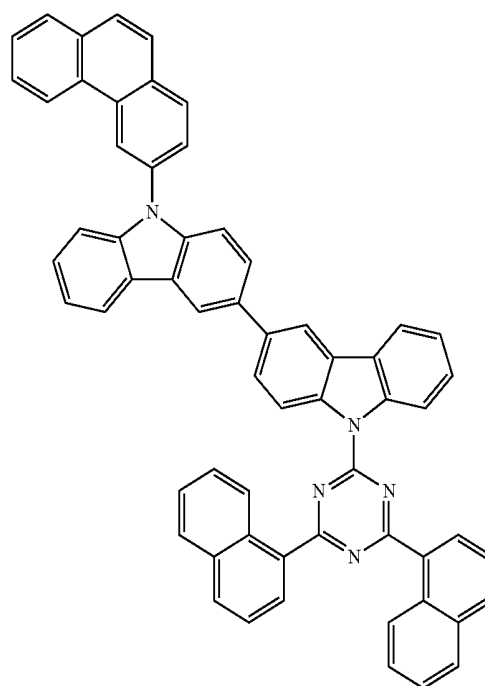
Compound 41



Compound 40



Compound 42

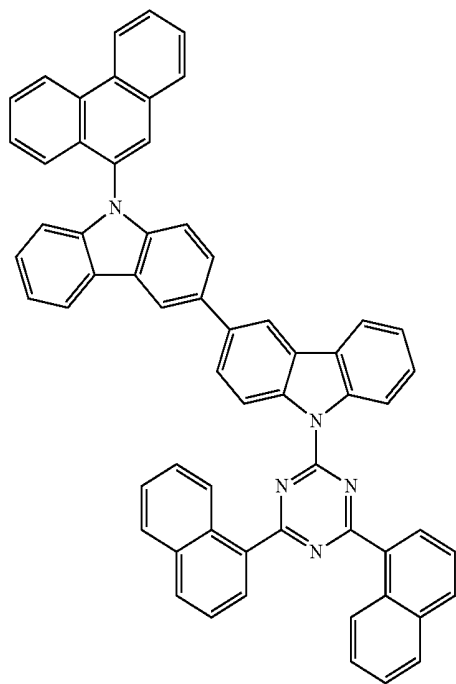




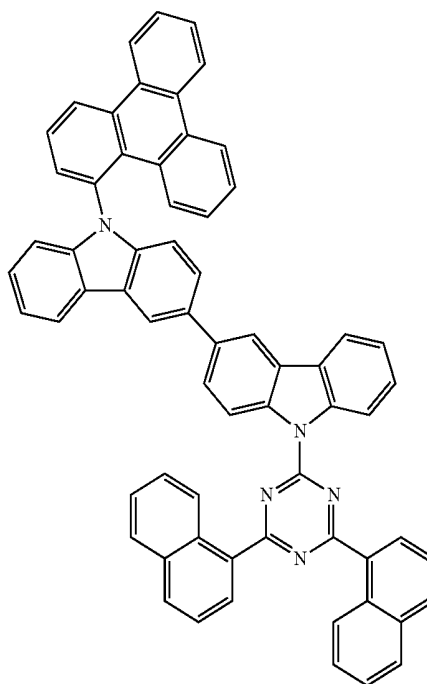
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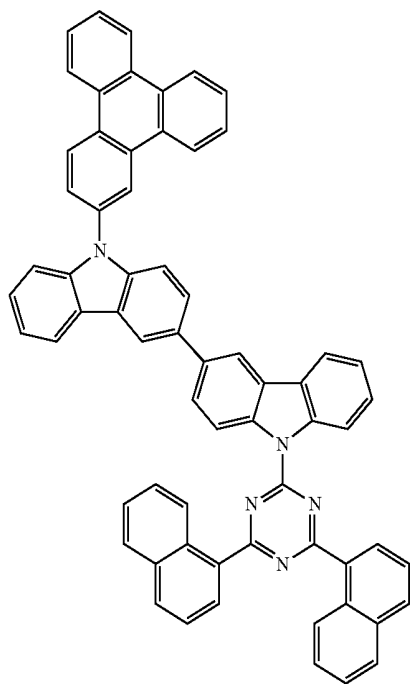
Compound 43



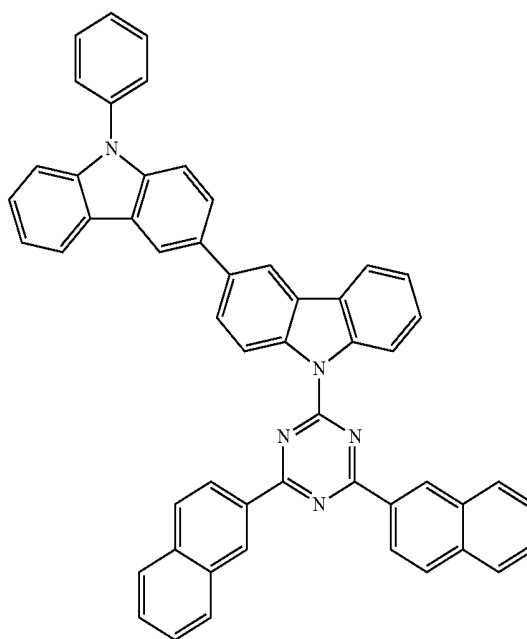
Compound 45



Compound 44



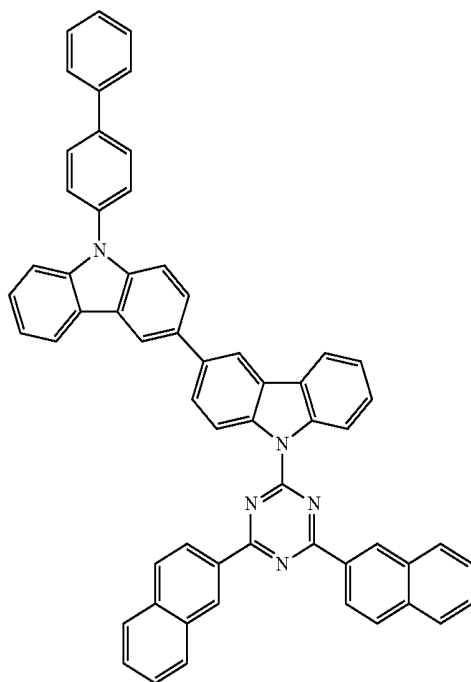
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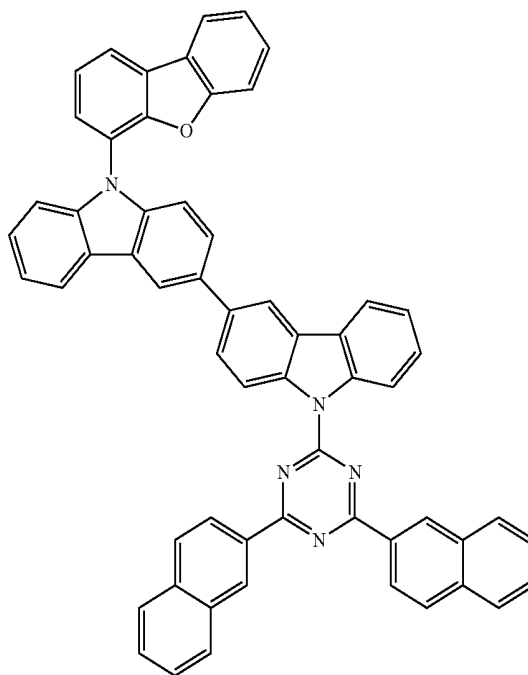
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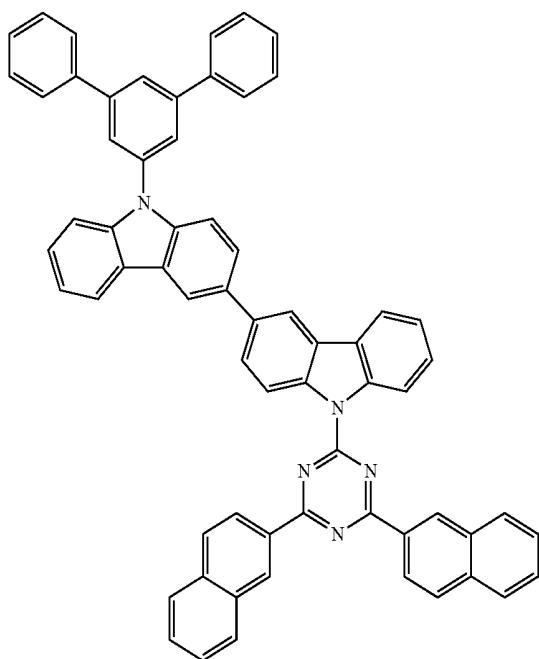
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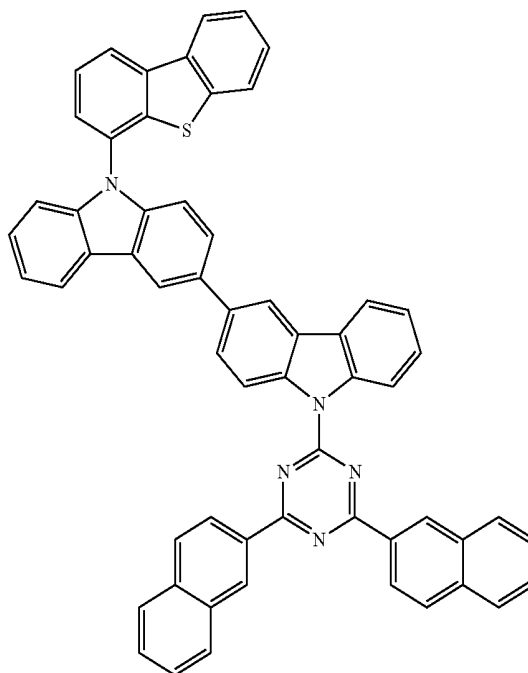
Compound 49



Compound 48



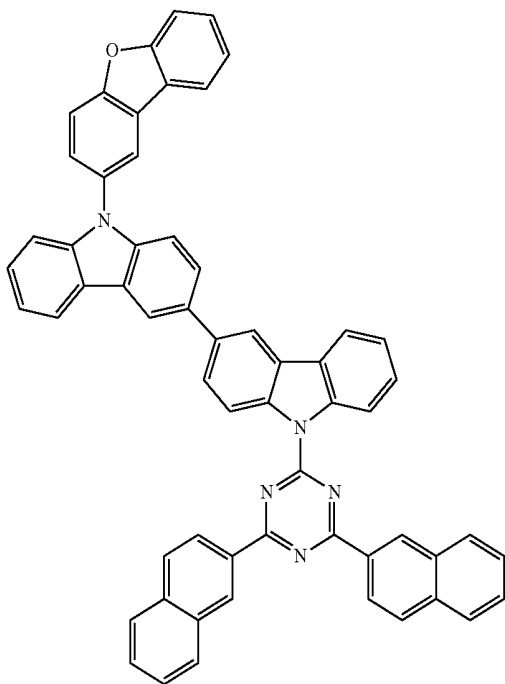
Compound 50



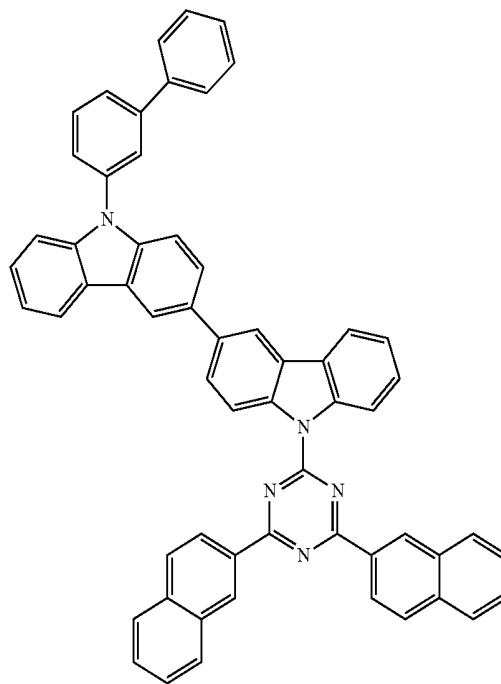
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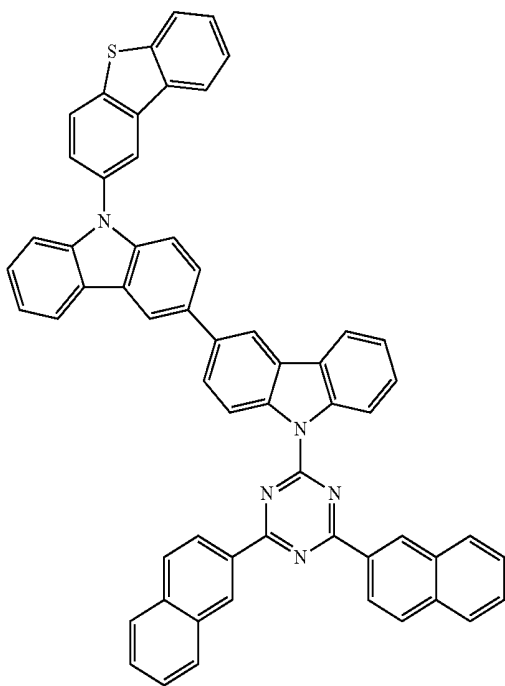
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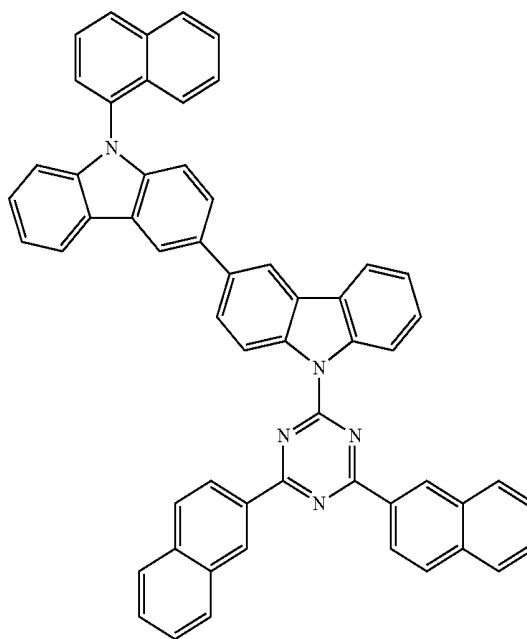
Compound 53



Compound 52



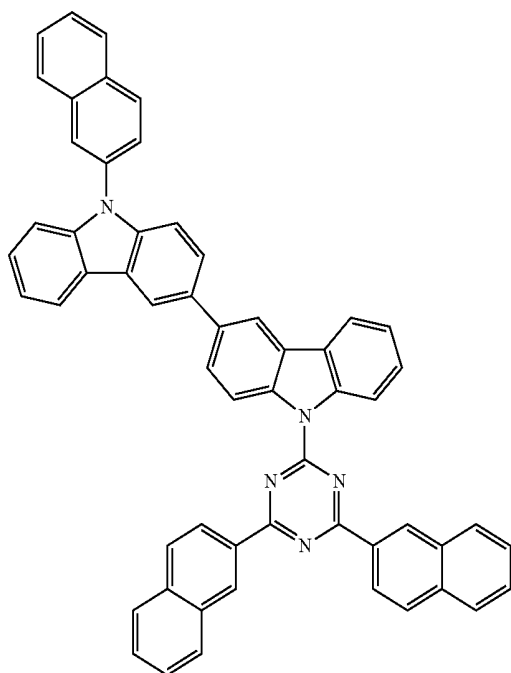
Compound 54



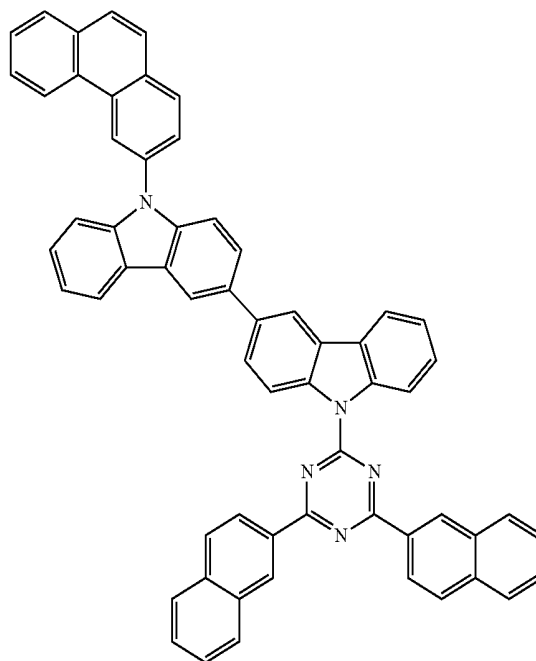
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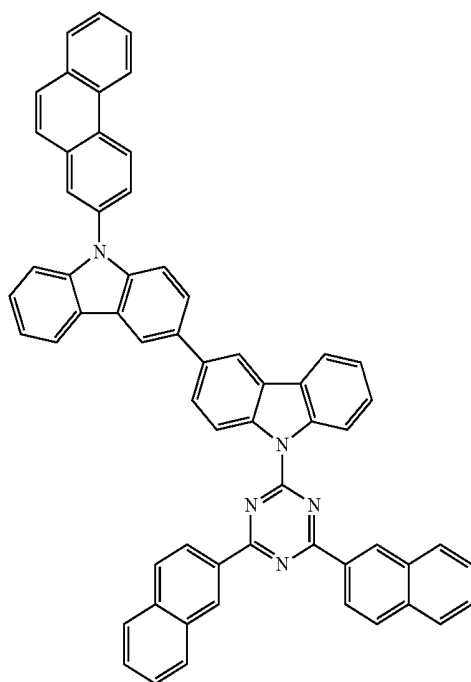
Compound 55



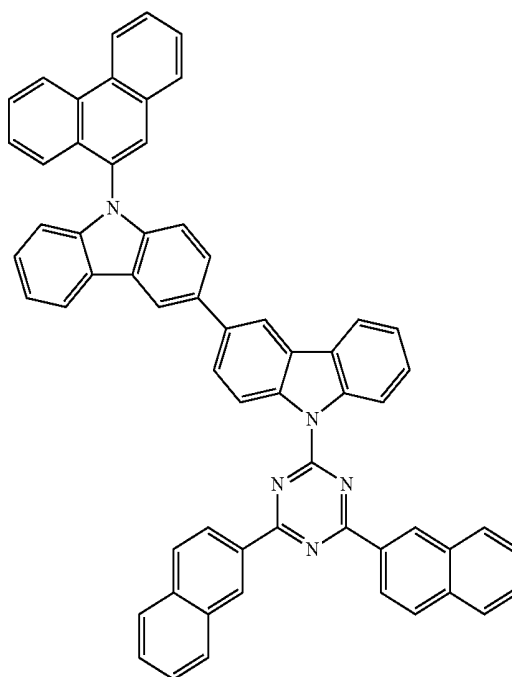
Compound 57



Compound 56



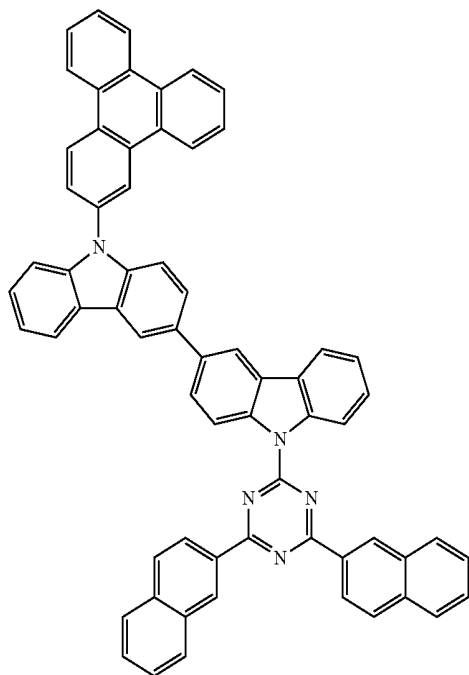
Compound 58



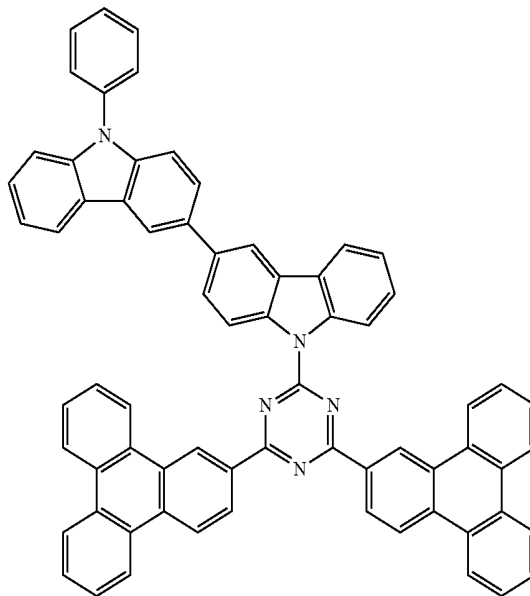
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Compound 59

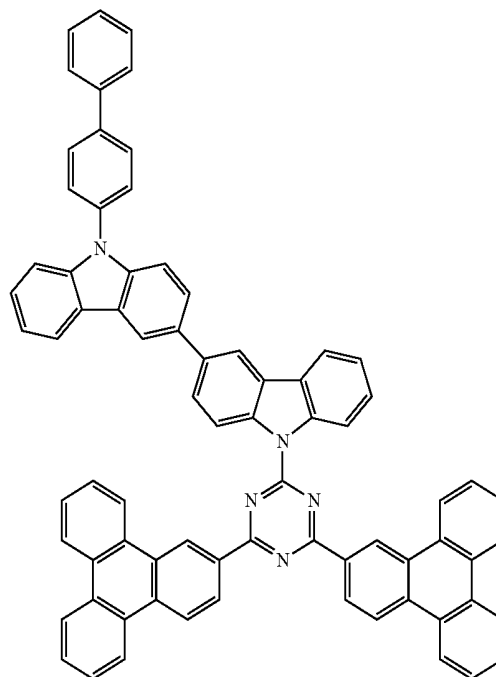
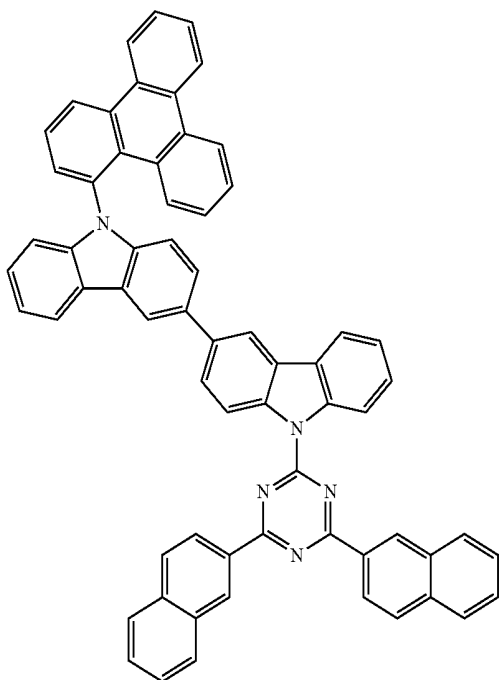


Compound 61



Compound 62

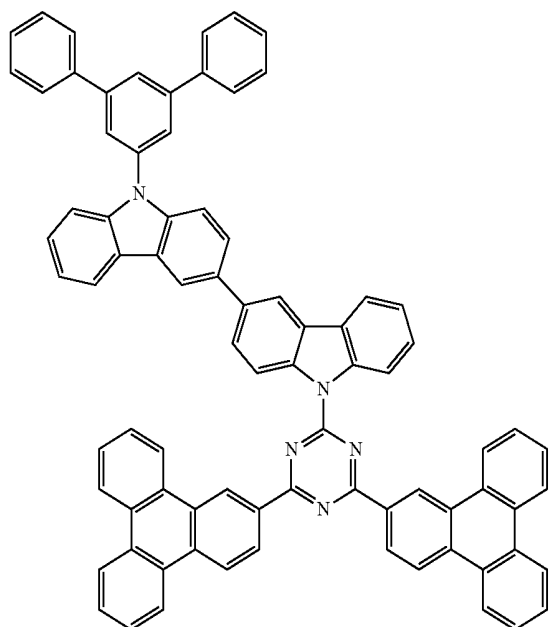
Compound 60



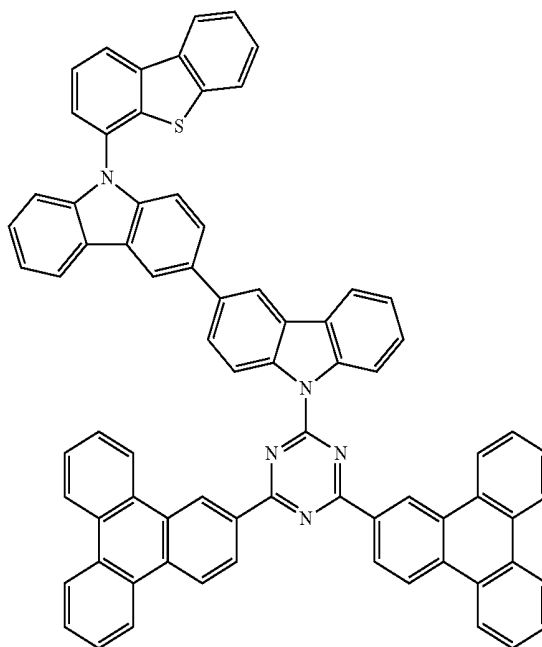
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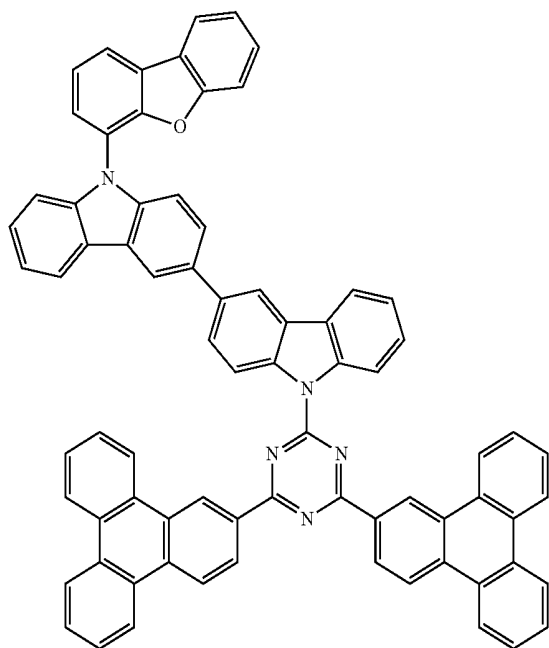
Compound 63



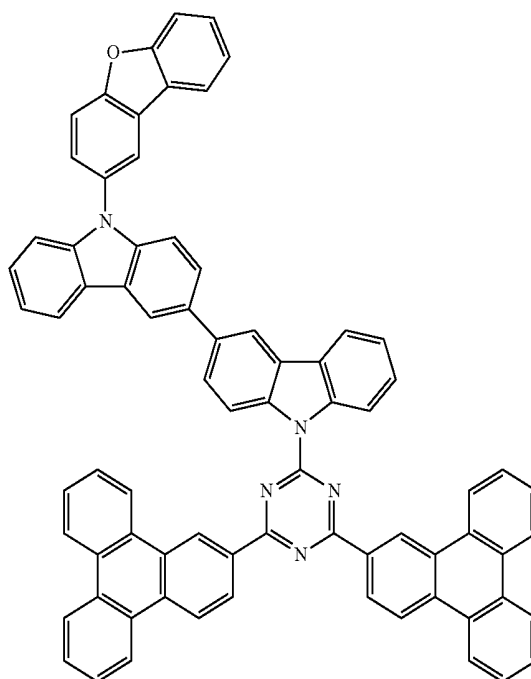
Compound 65



Compound 64



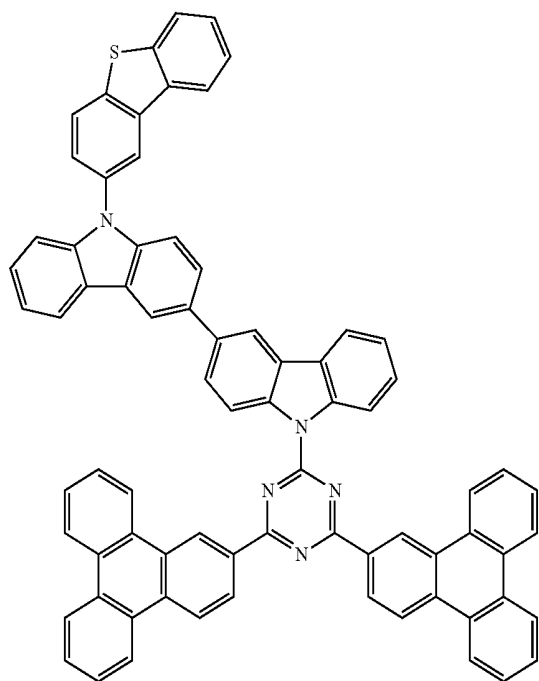
Compound 66



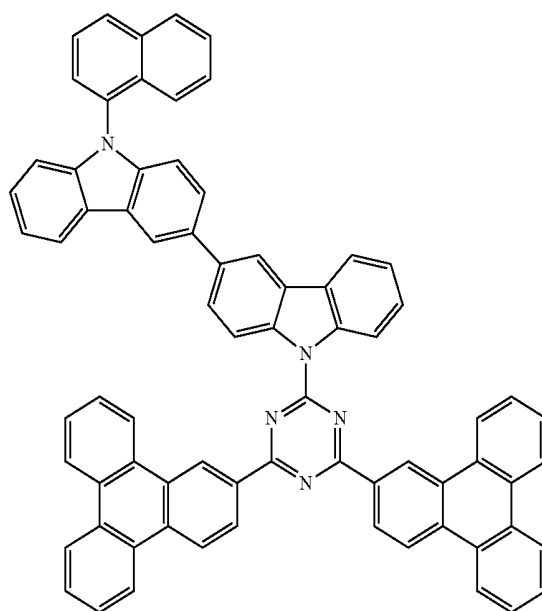
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Compound 67

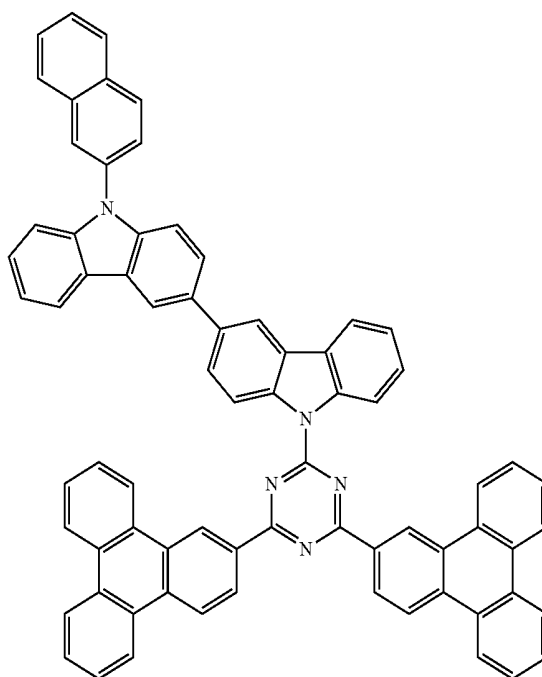
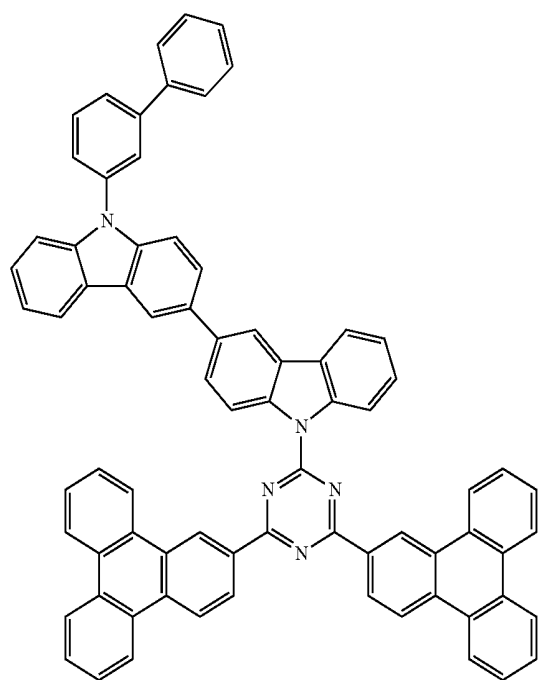


Compound 69



Compound 70

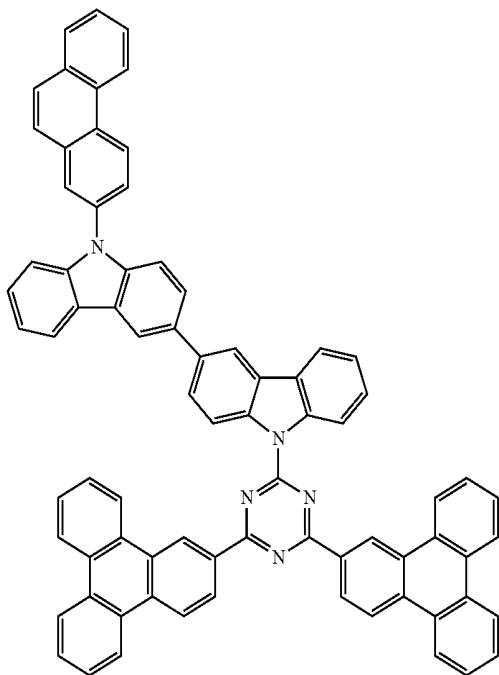
Compound 68



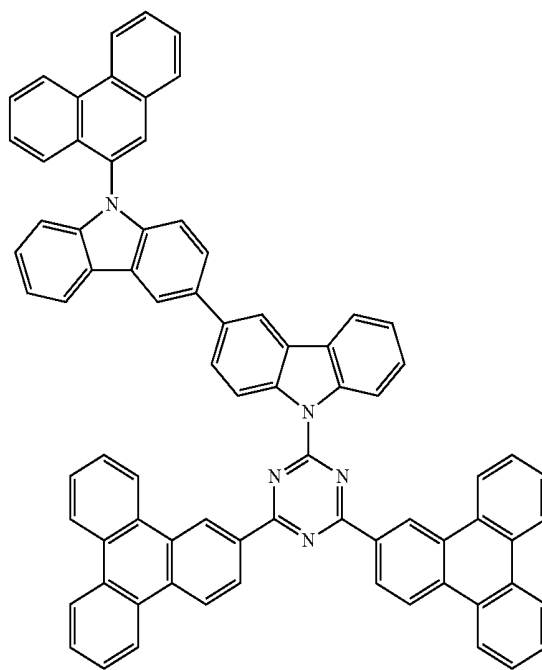
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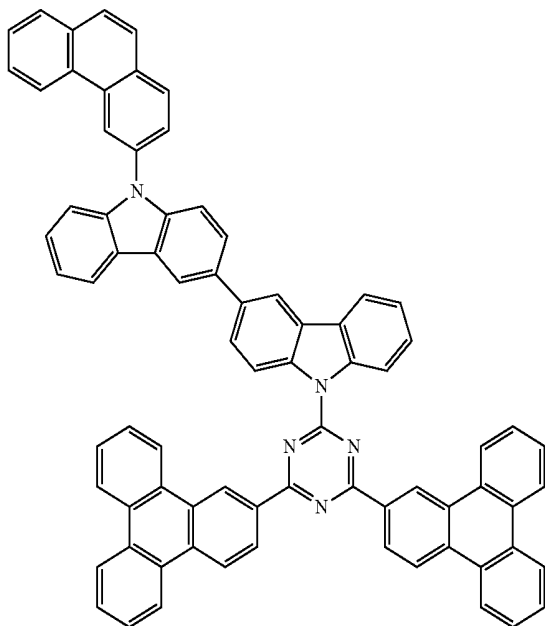
Compound 71



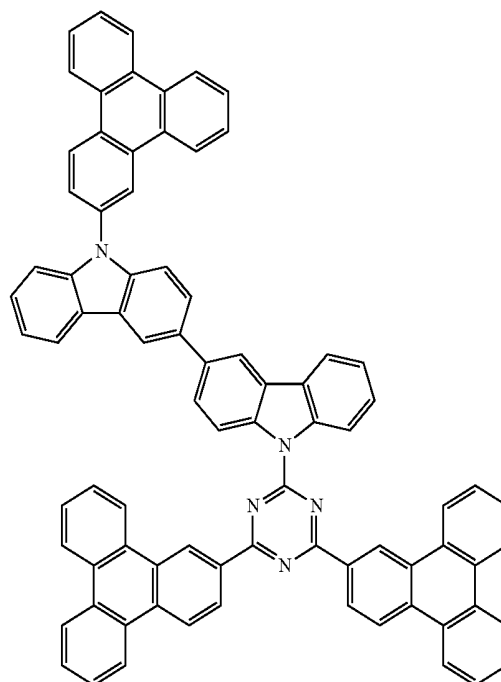
Compound 73



Compound 72



Compound 74

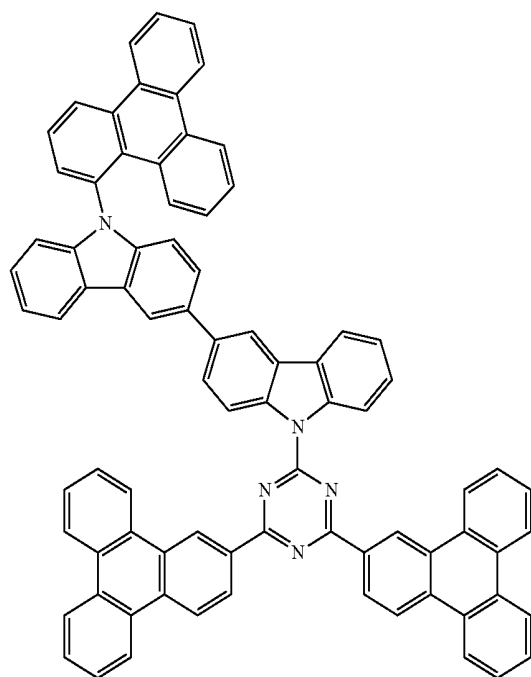




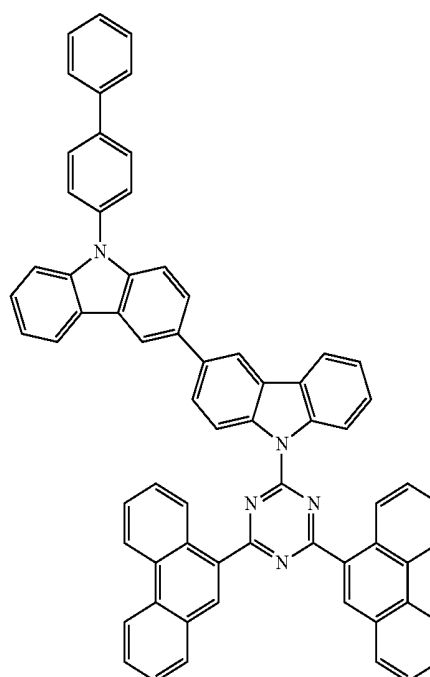
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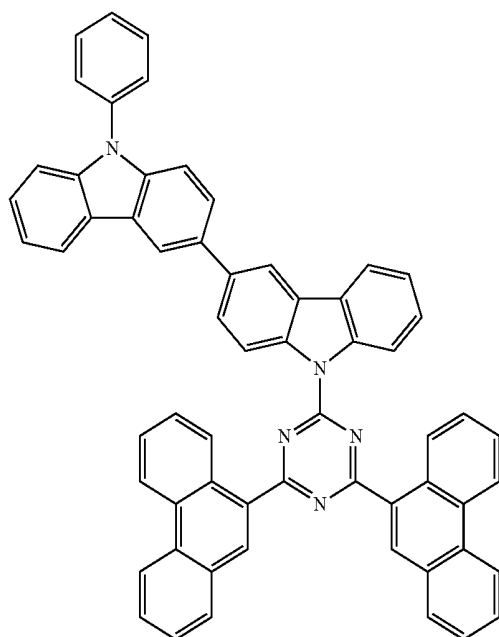
Compound 75



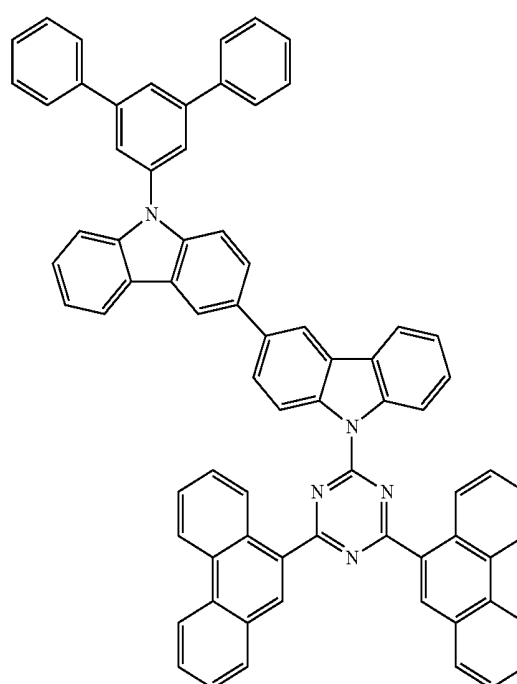
Compound 77



Compound 76

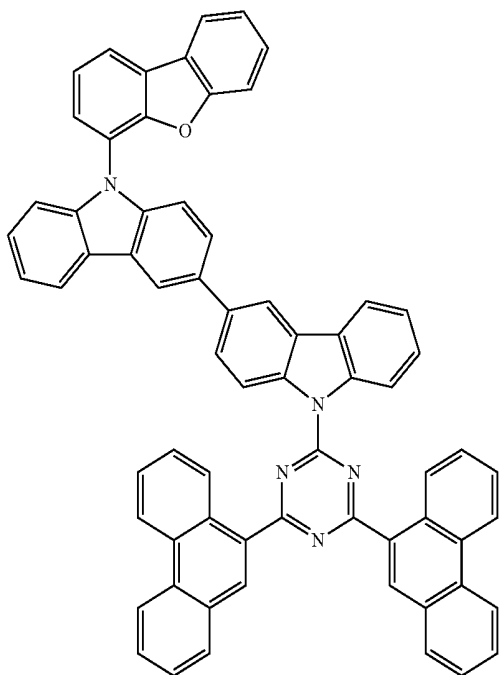


Compound 78



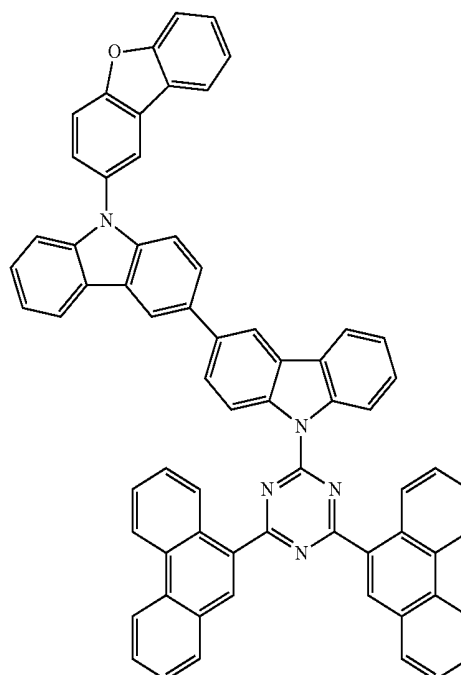
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Compound 79

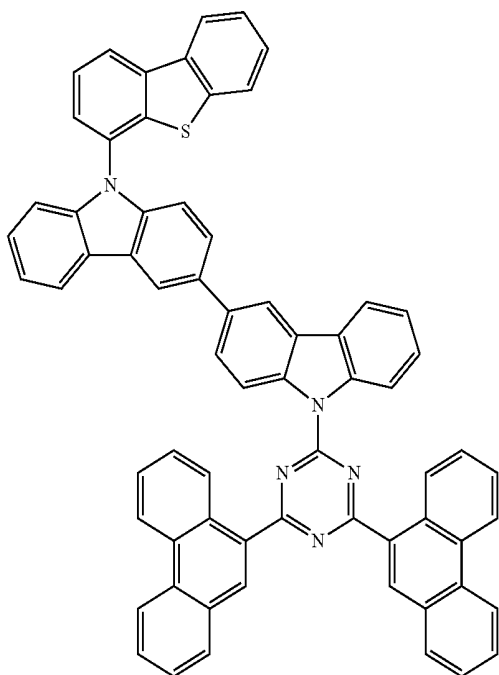


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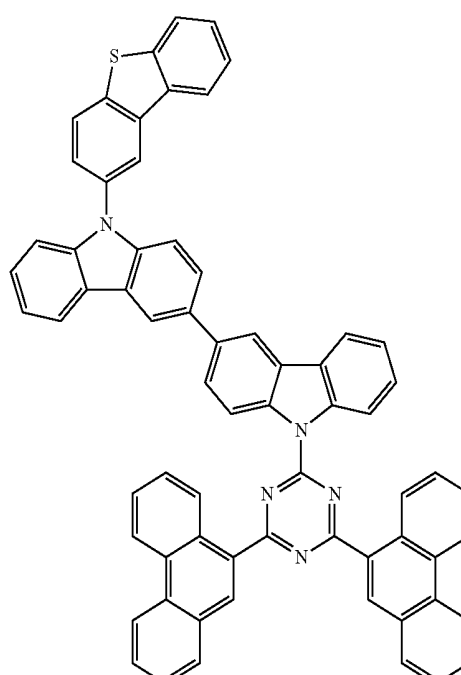
Compound 81



Compound 80



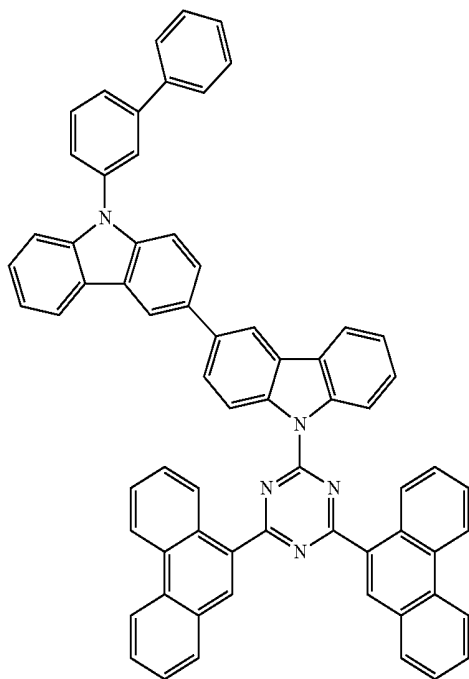
Compound 82



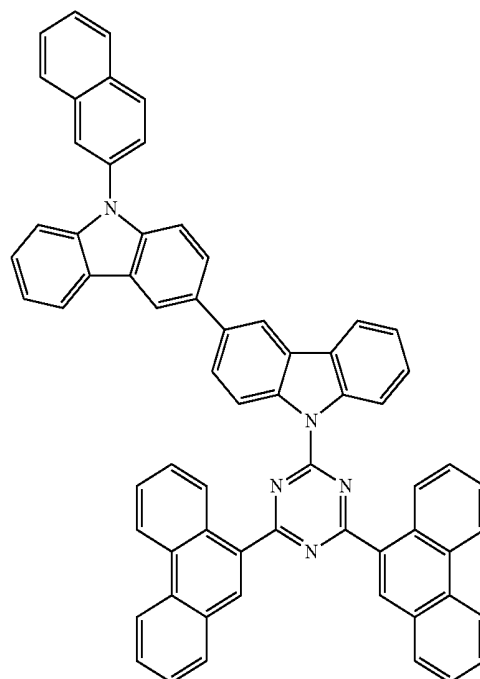
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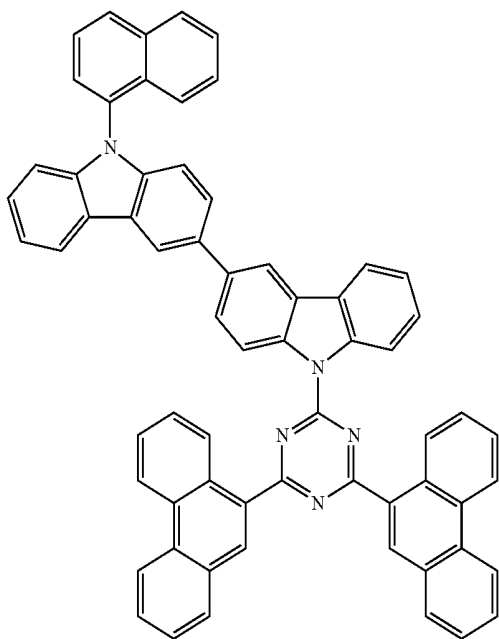
Compound 83



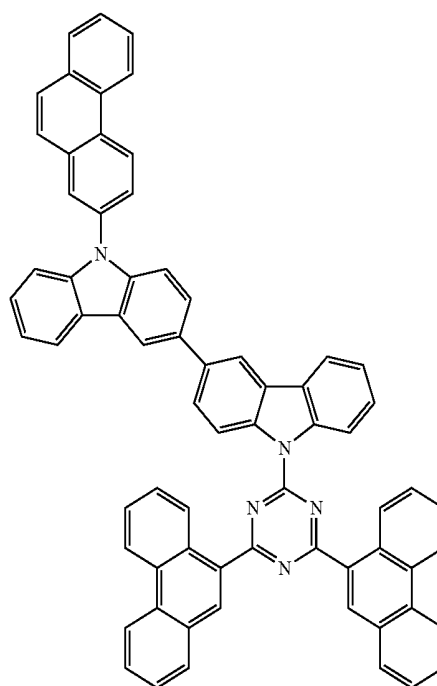
Compound 85



Compound 84



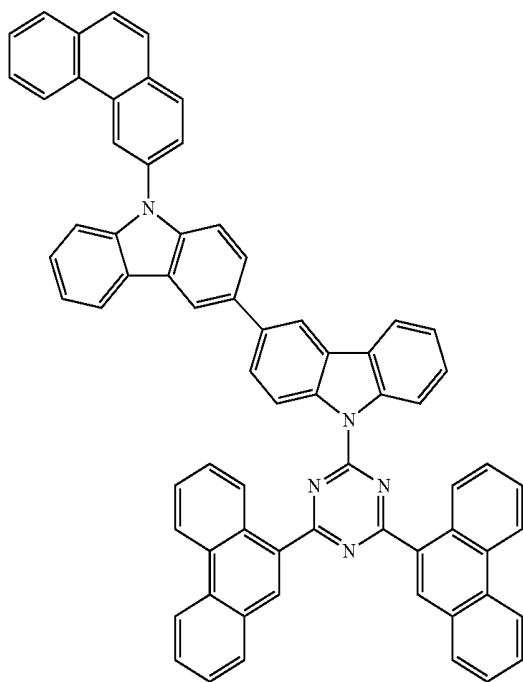
Compound 86



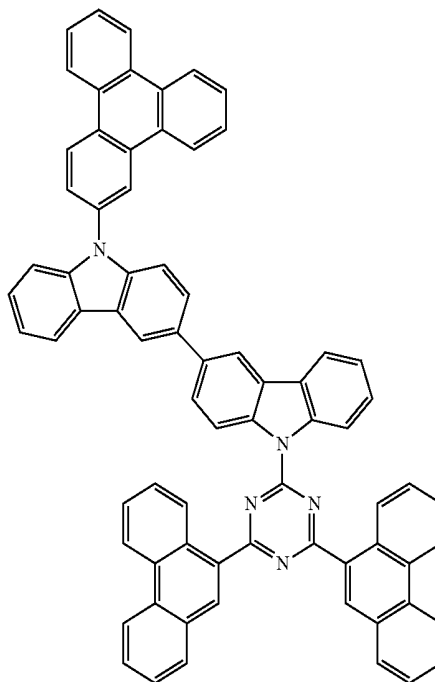
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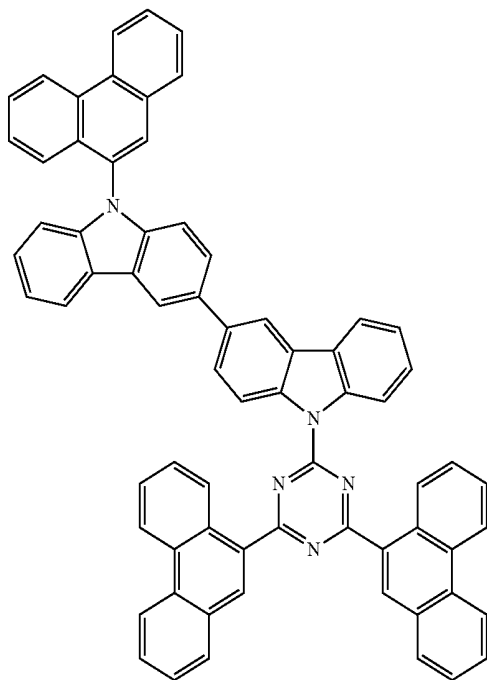
Compound 87



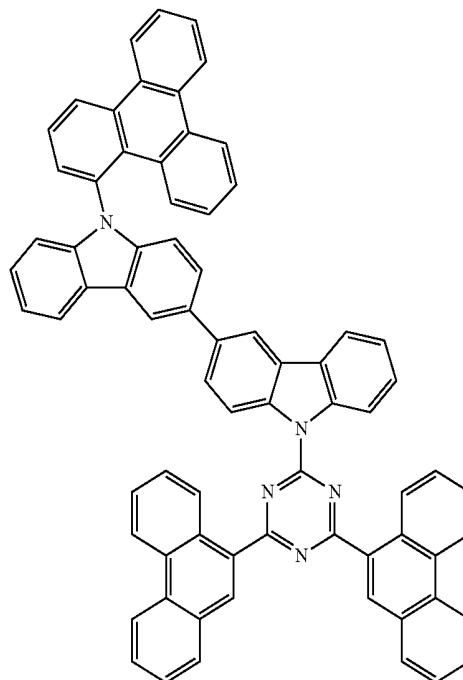
Compound 89



Compound 88



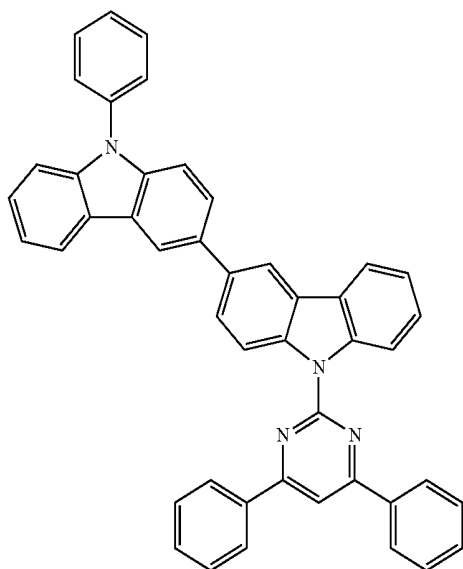
Compound 90



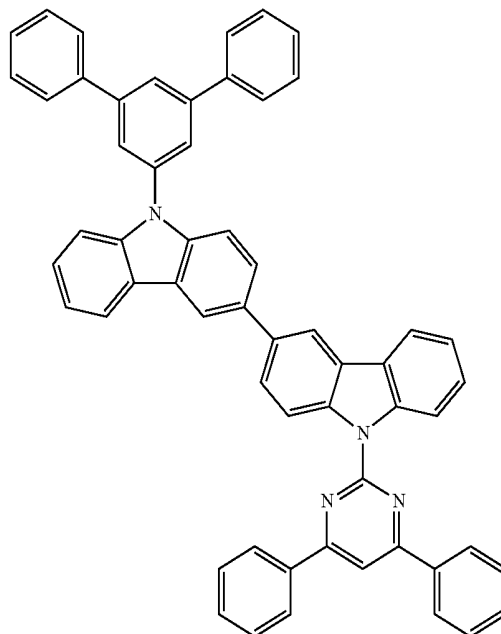
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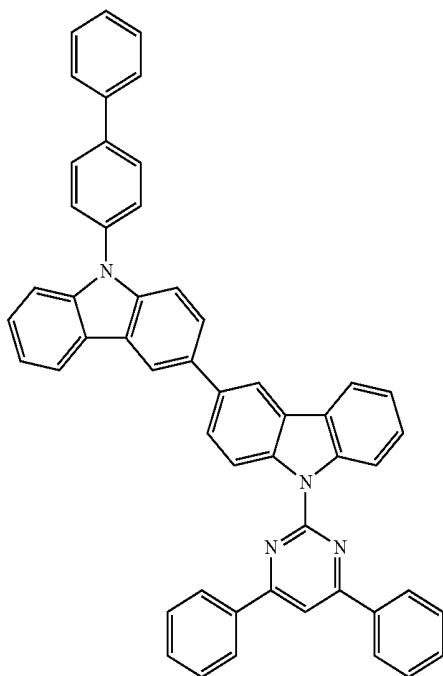
Compound 91



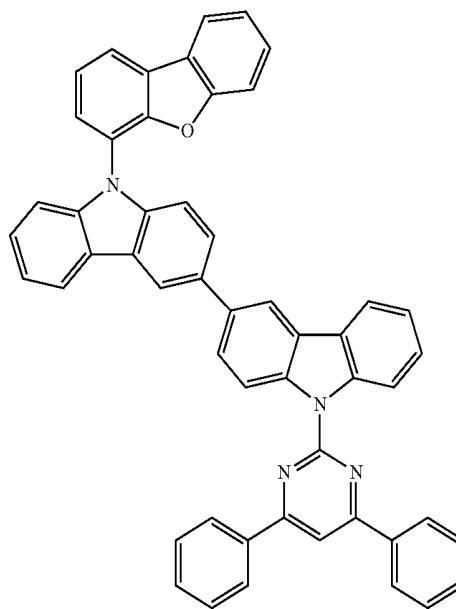
Compound 93



Compound 92



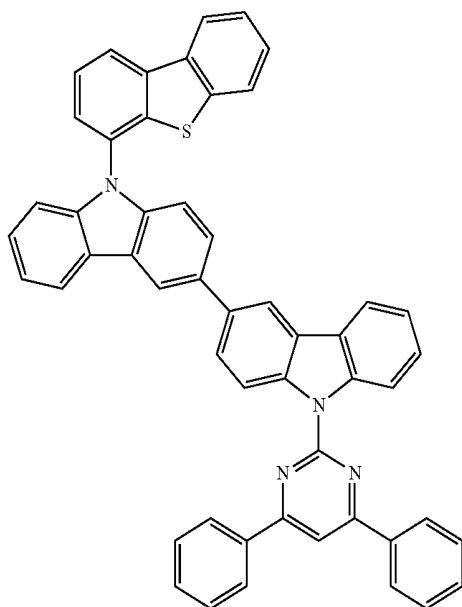
Compound 94



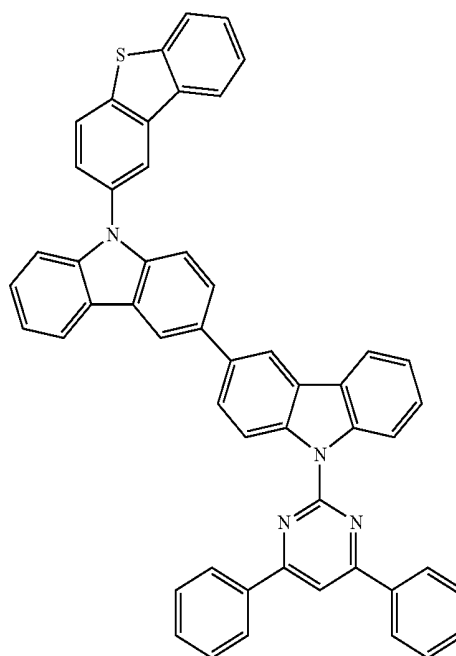
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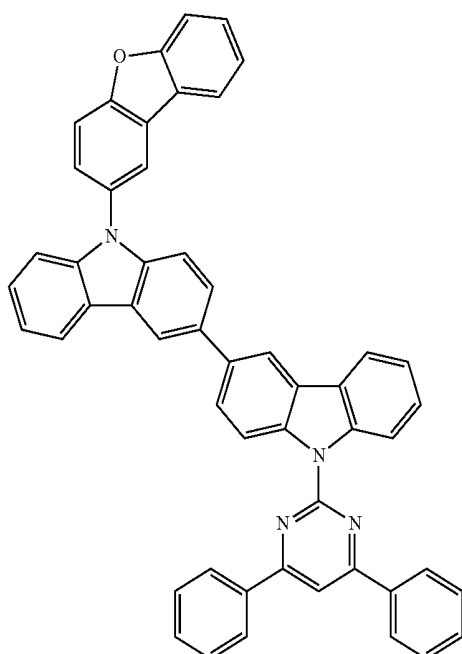
Compound 95



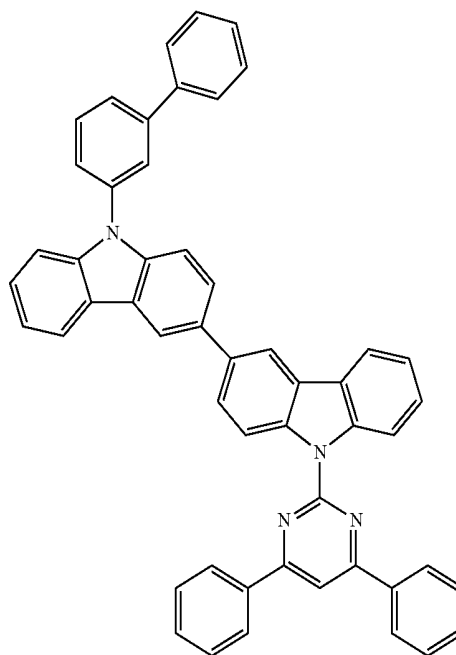
Compound 97



Compound 96



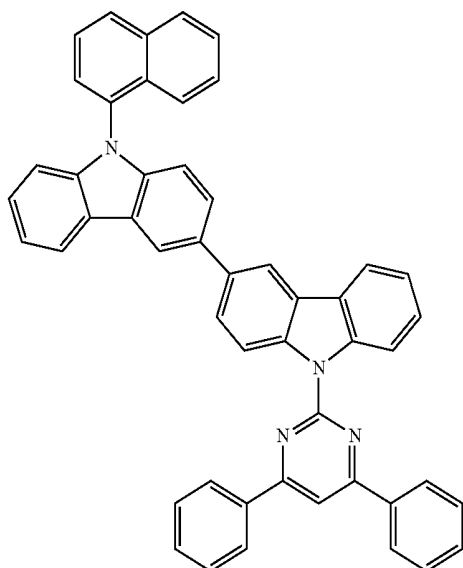
Compound 98



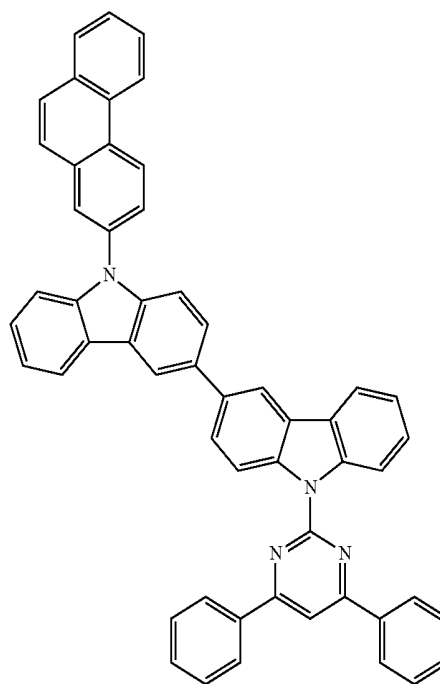
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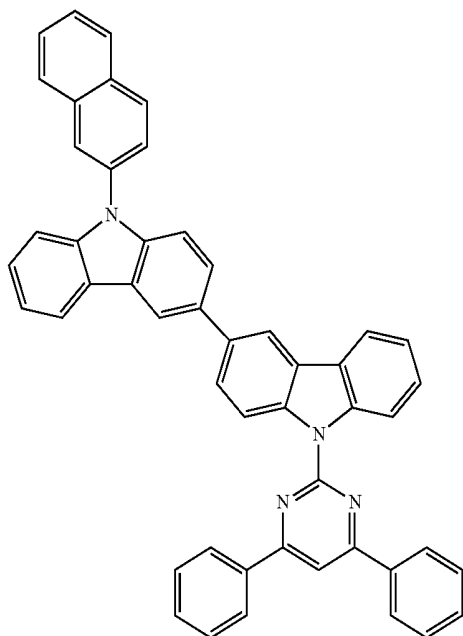
Compound 99



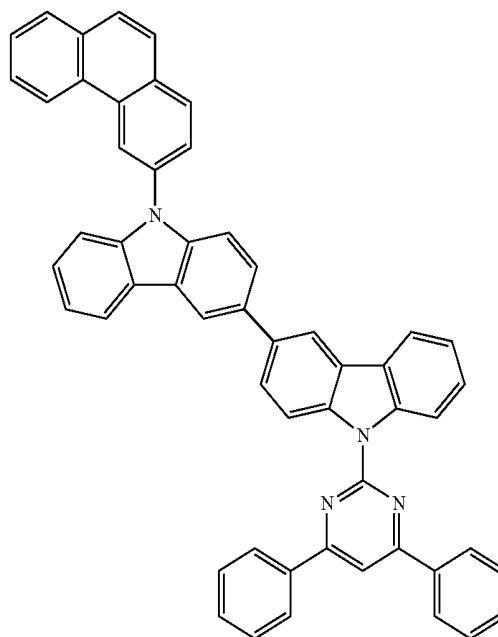
Compound 101



Compound 100



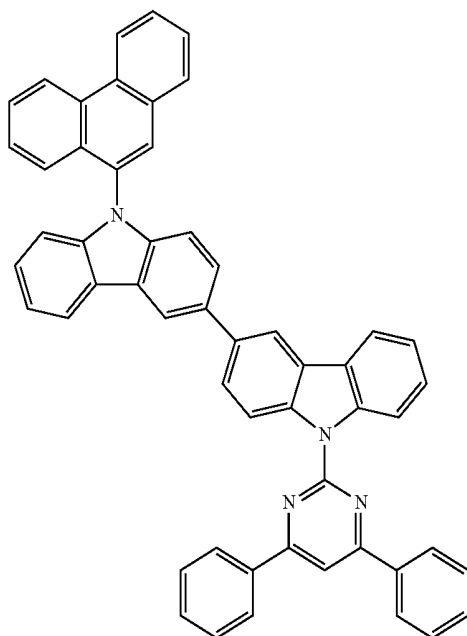
Compound 102



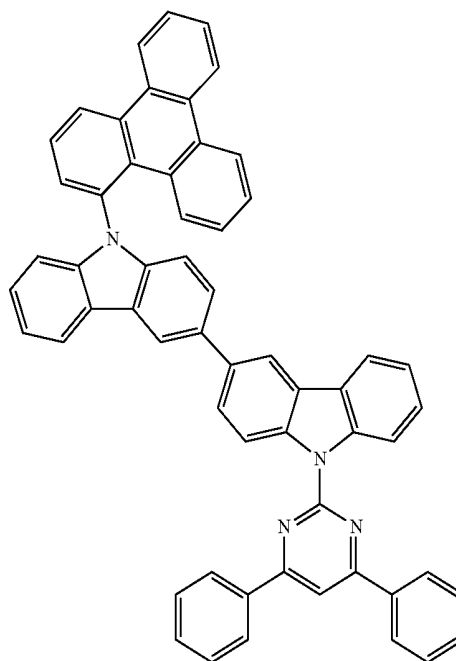
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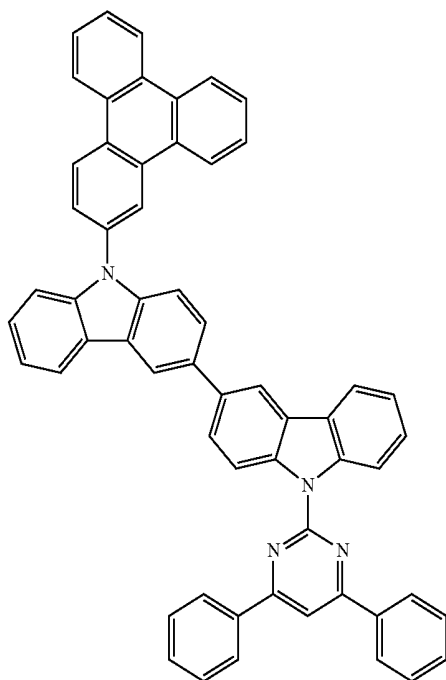
Compound 103



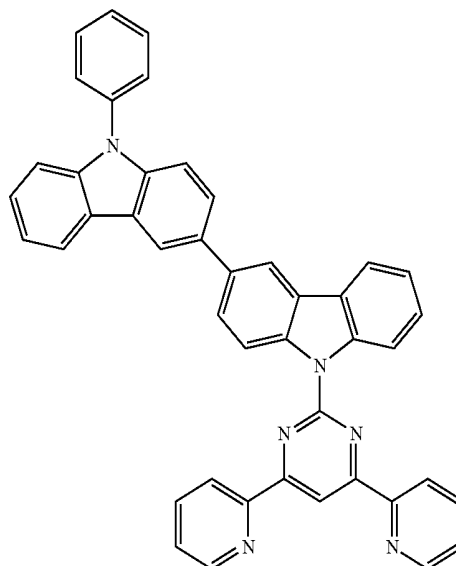
Compound 105



Compound 104



Compound 106

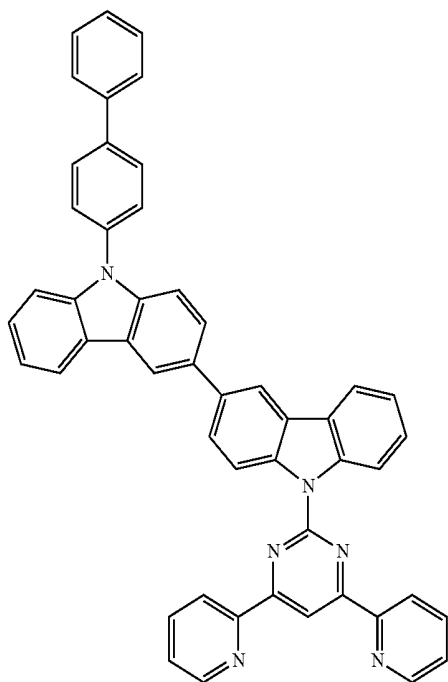




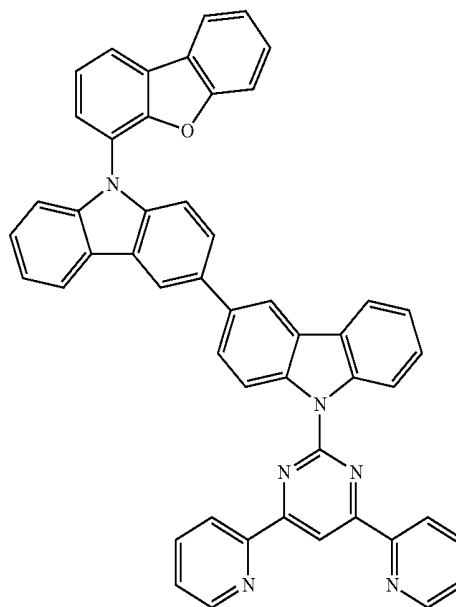
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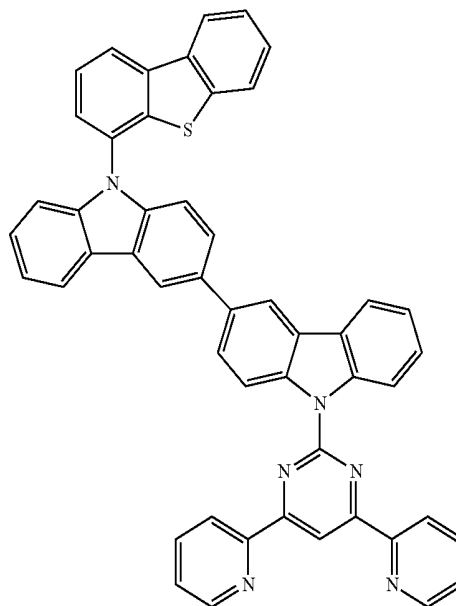
Compound 107



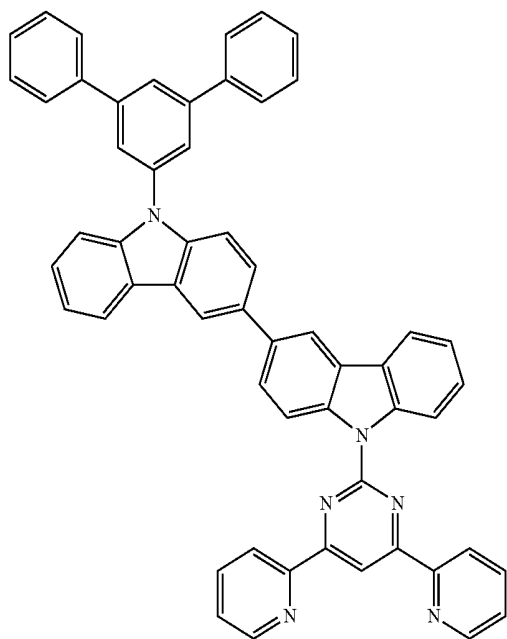
Compound 109



Compound 110



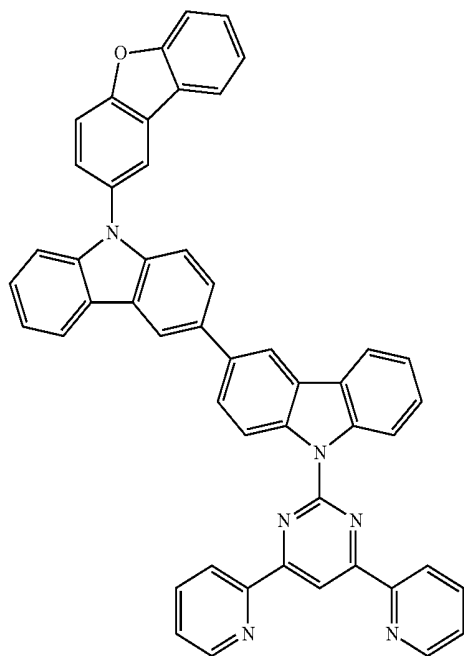
Compound 108



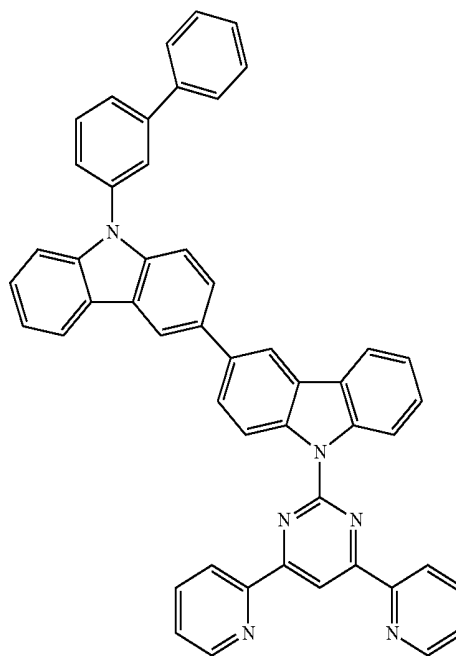
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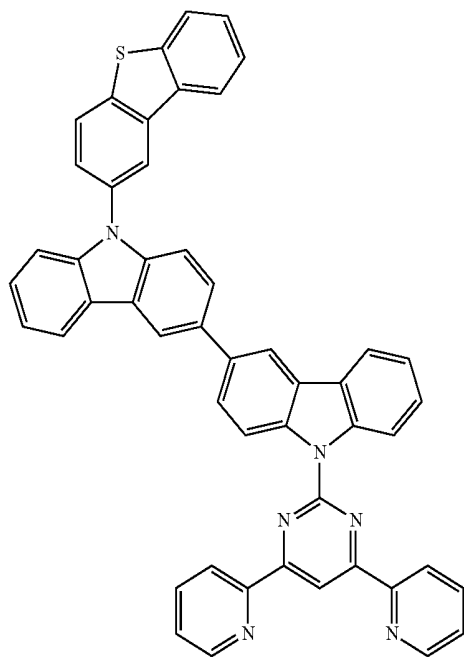
Compound 111



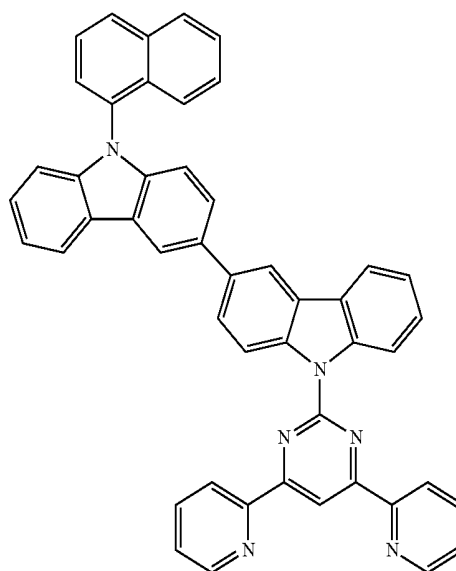
Compound 113



Compound 112



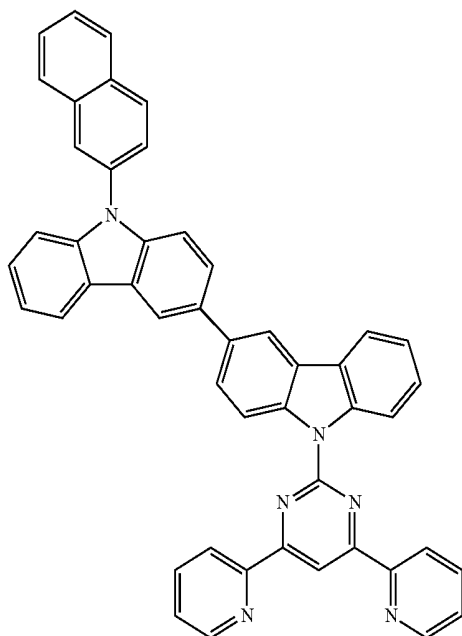
Compound 114



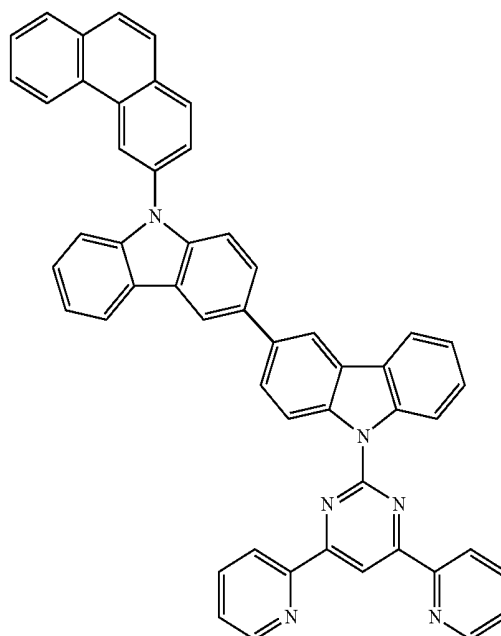
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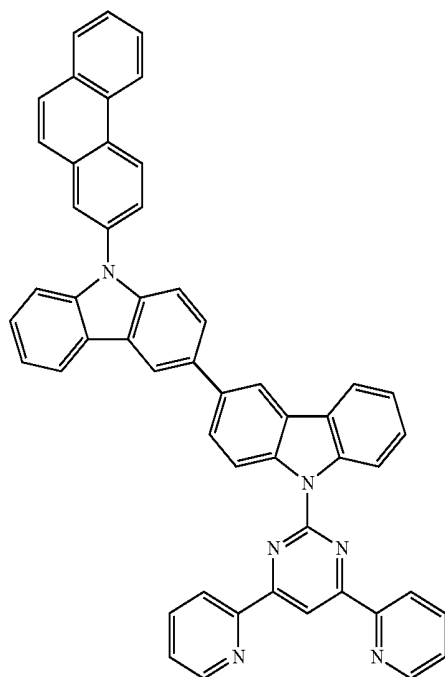
Compound 115



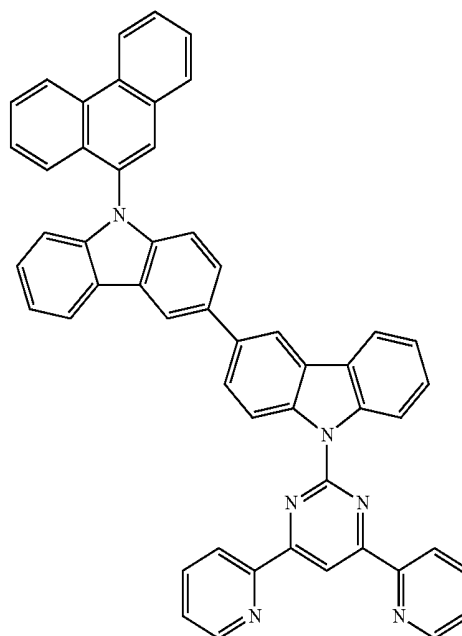
Compound 117



Compound 116



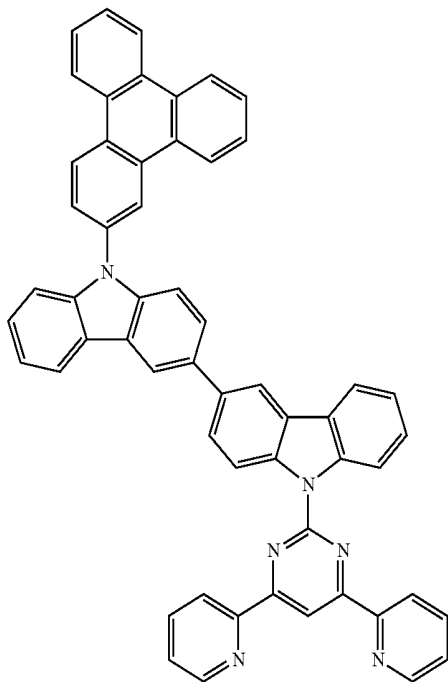
Compound 118



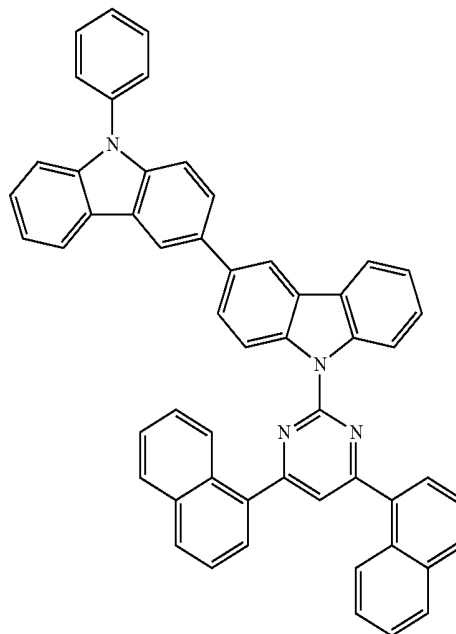
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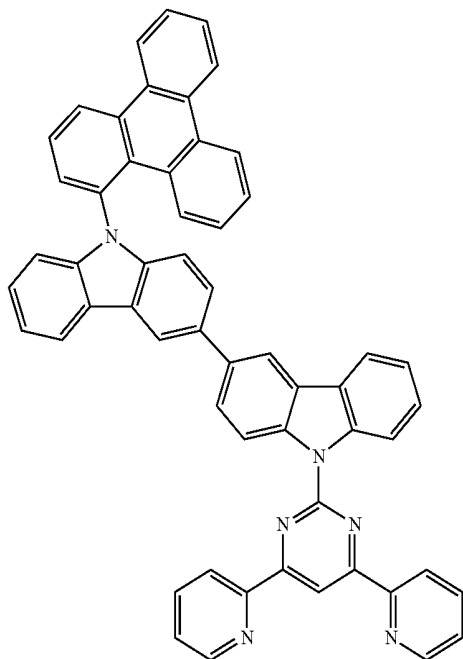
Compound 119



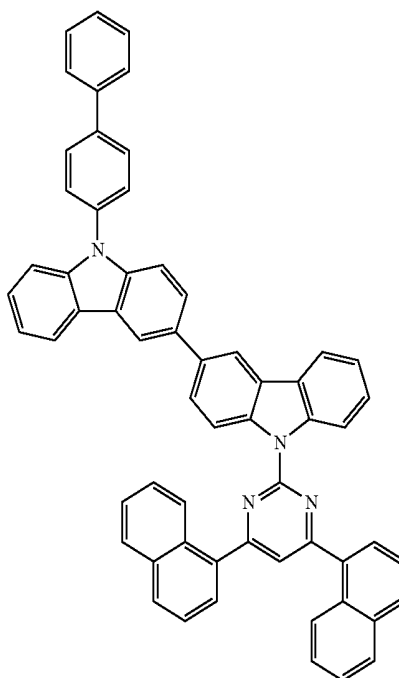
Compound 121



Compound 120

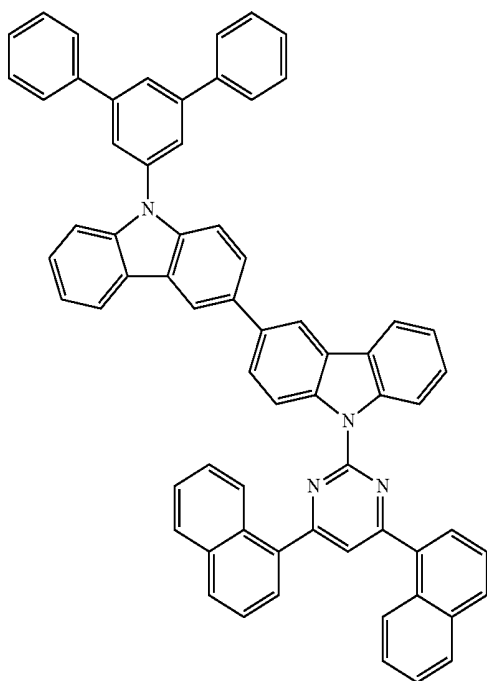


Compound 122



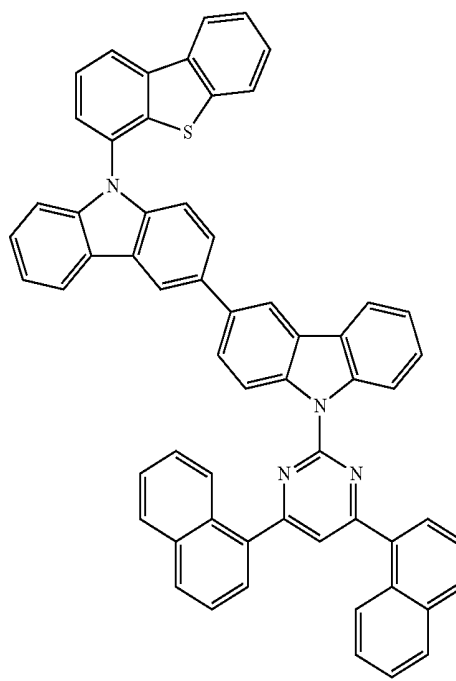
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Compound 123

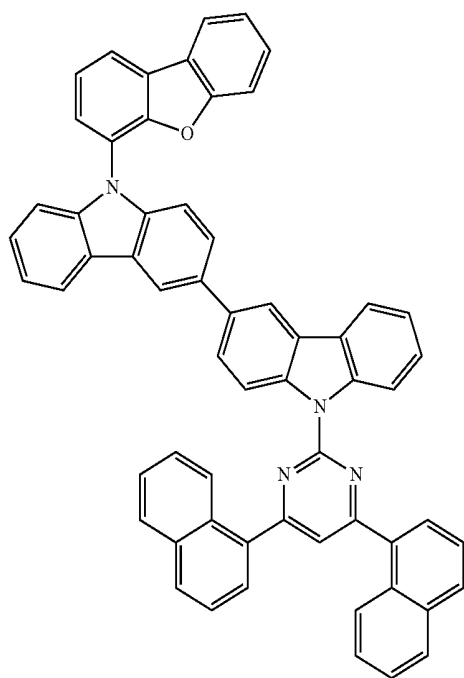


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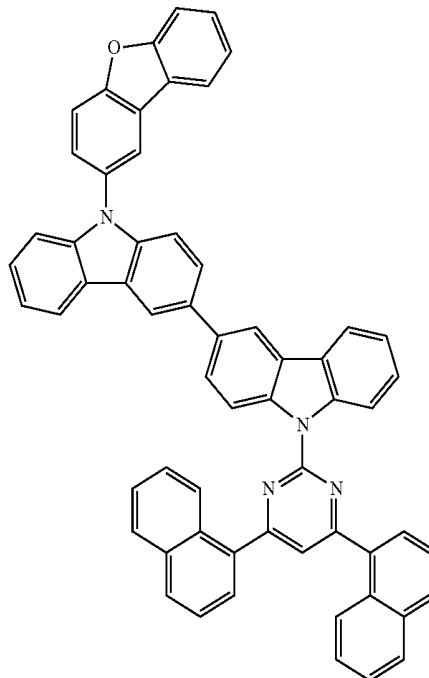
Compound 125



Compound 124

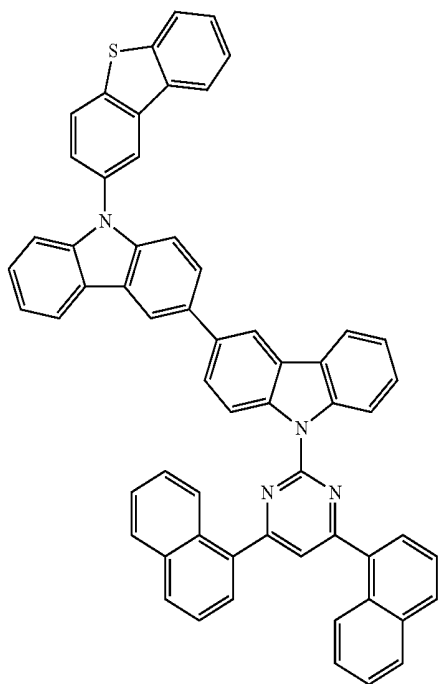


Compound 126



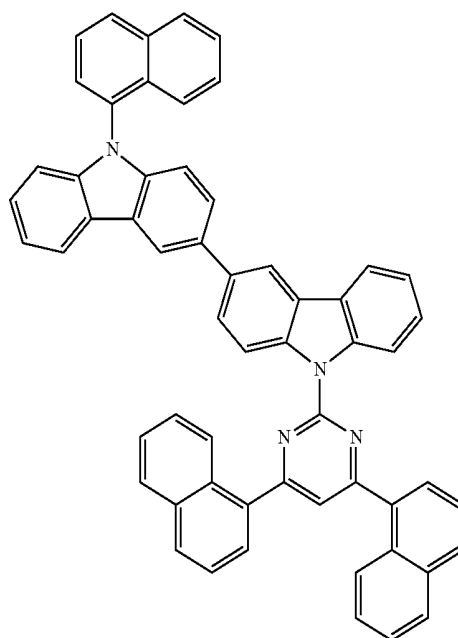
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Compound 127

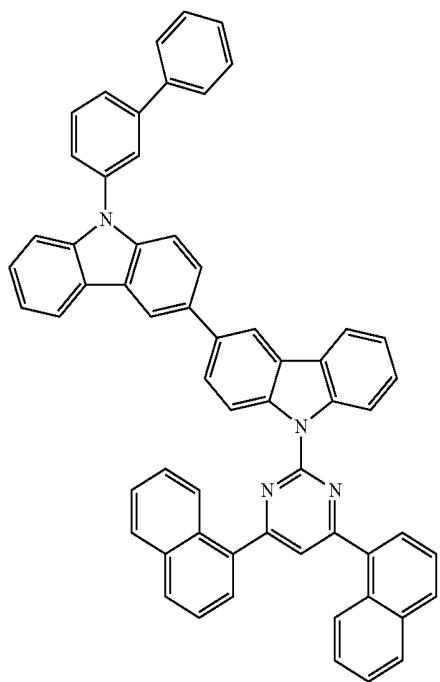


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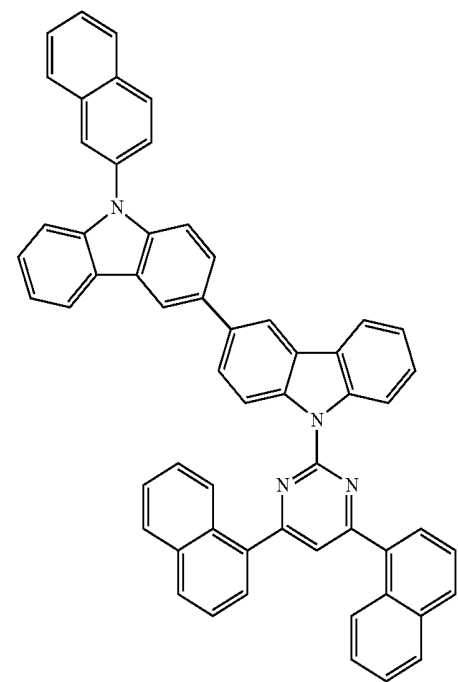
Compound 129



Compound 128



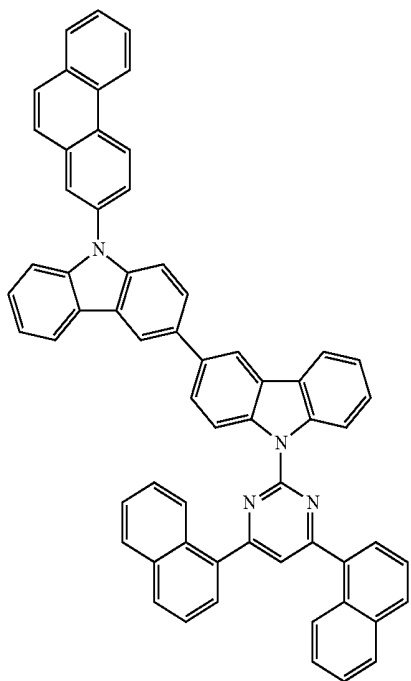
Compound 130



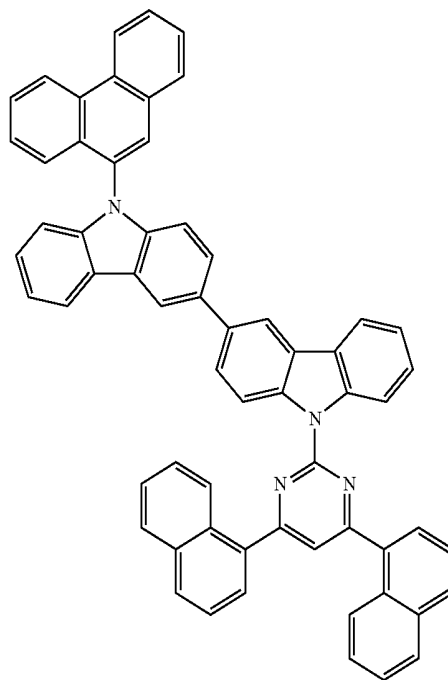
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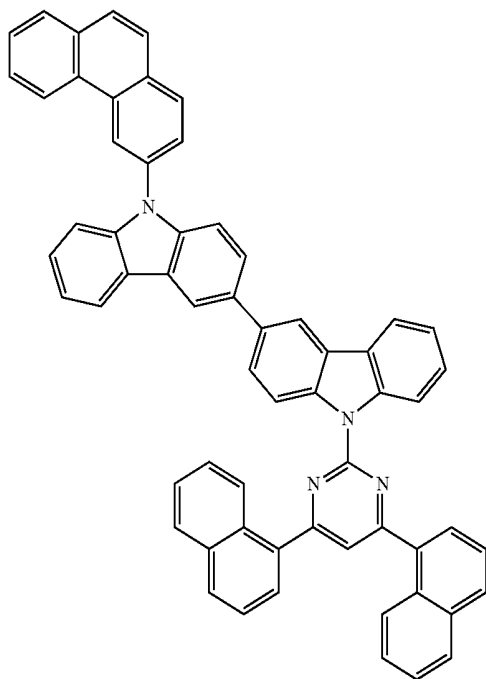
Compound 131



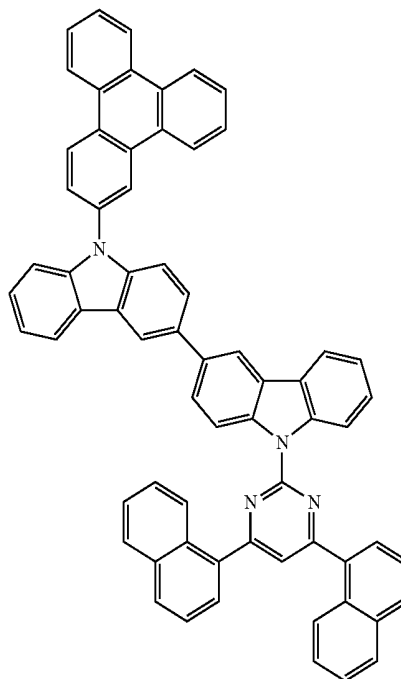
Compound 133



Compound 132



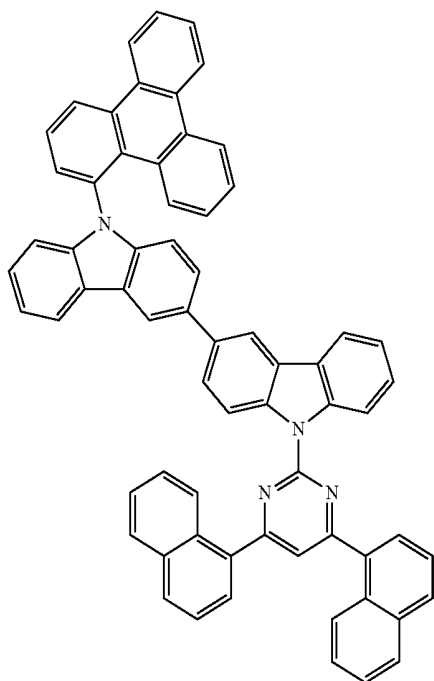
Compound 134



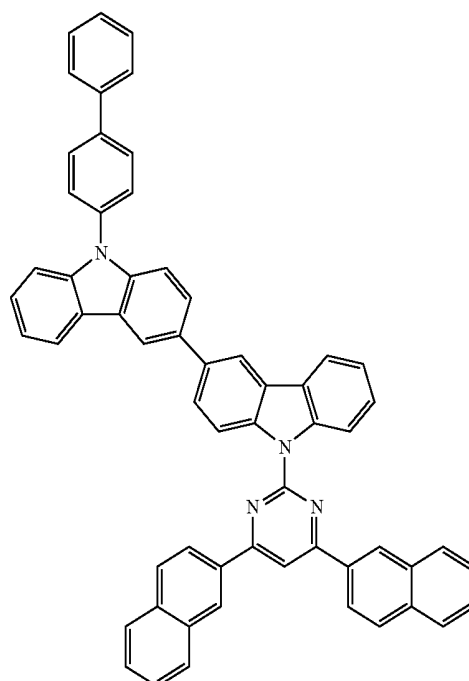
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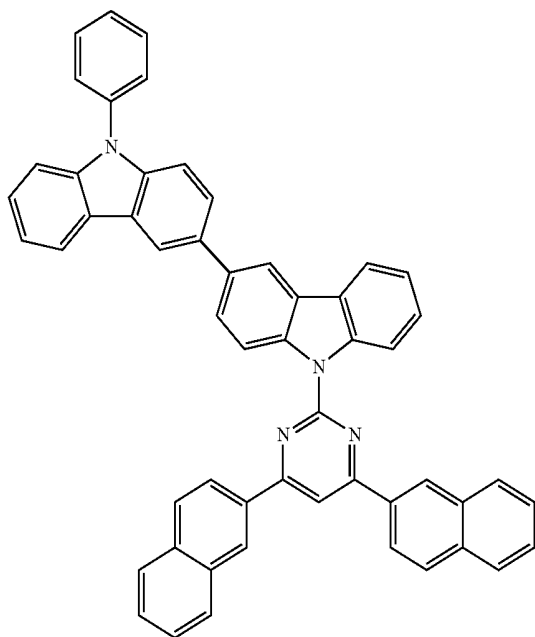
Compound 135



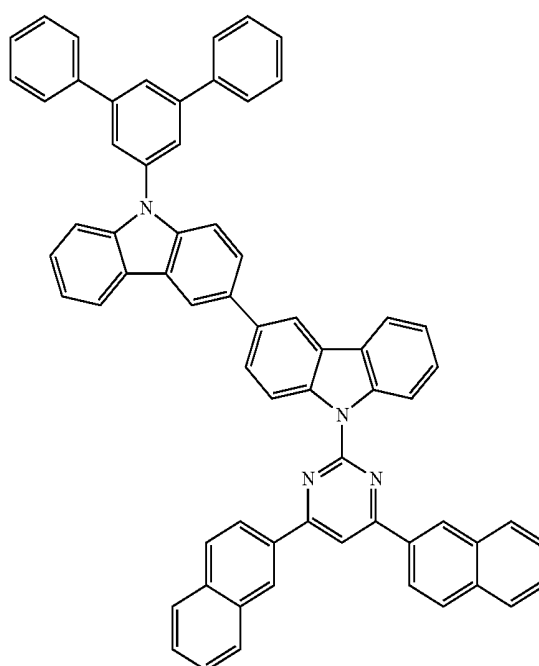
Compound 137



Compound 136



Compound 138

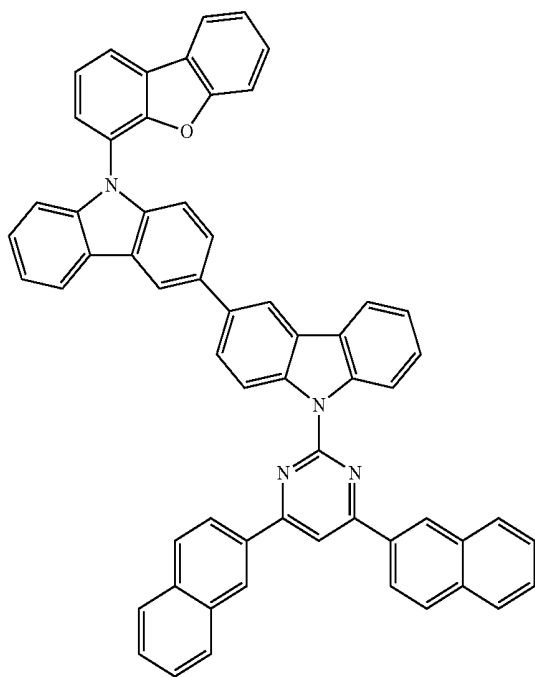




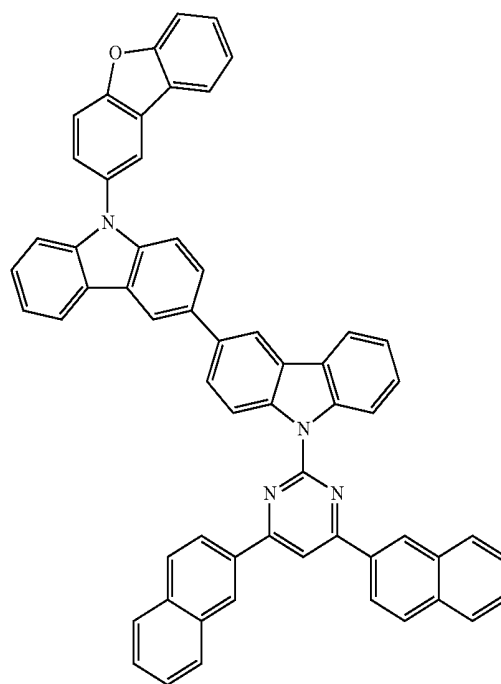
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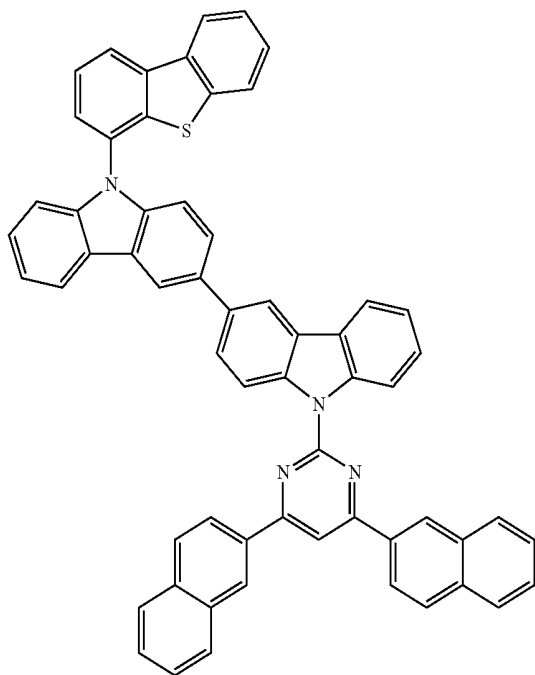
Compound 139



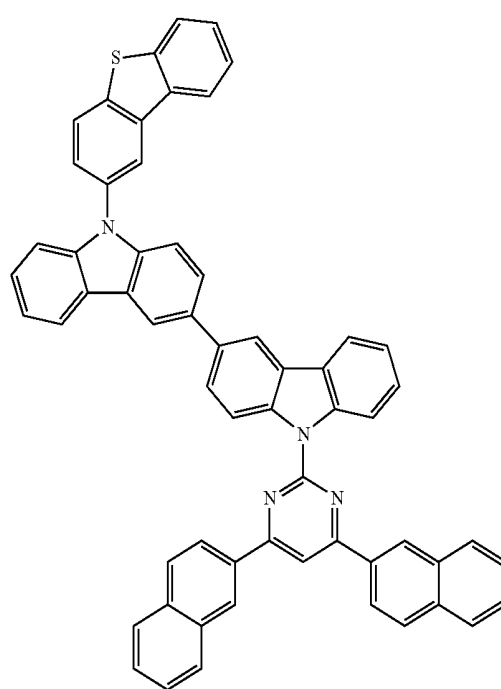
Compound 141



Compound 140



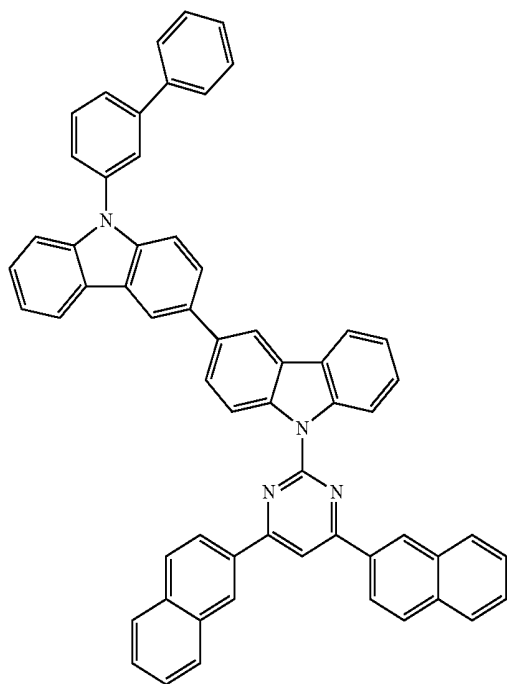
Compound 142



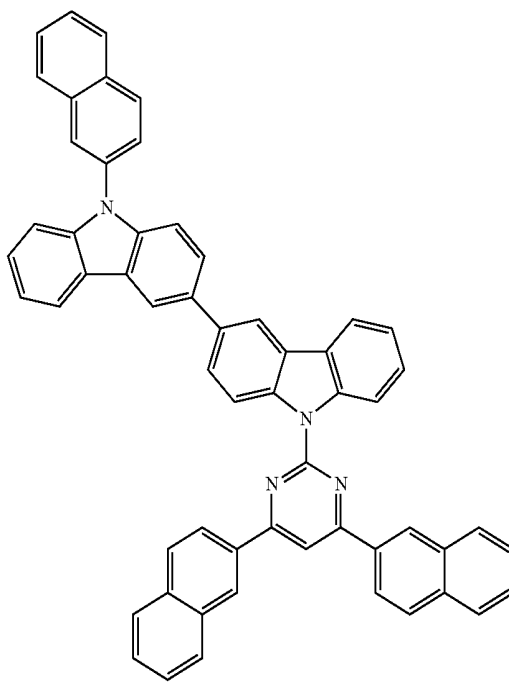
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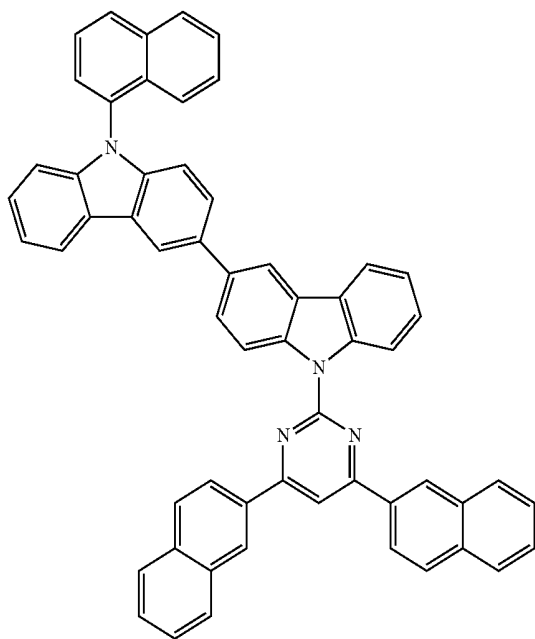
Compound 143



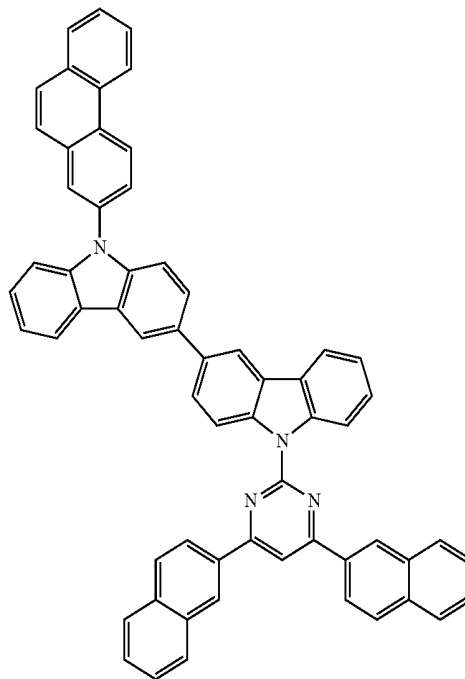
Compound 145



Compound 144



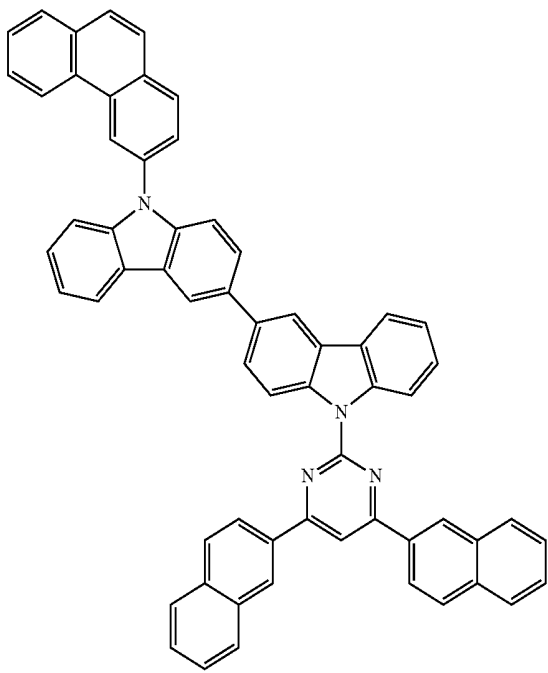
Compound 146



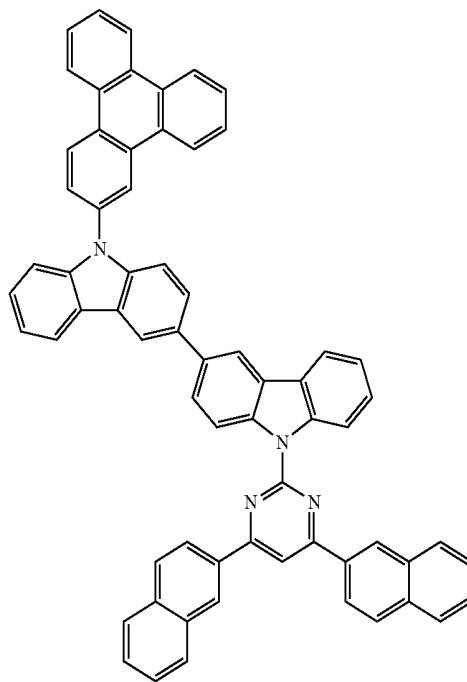
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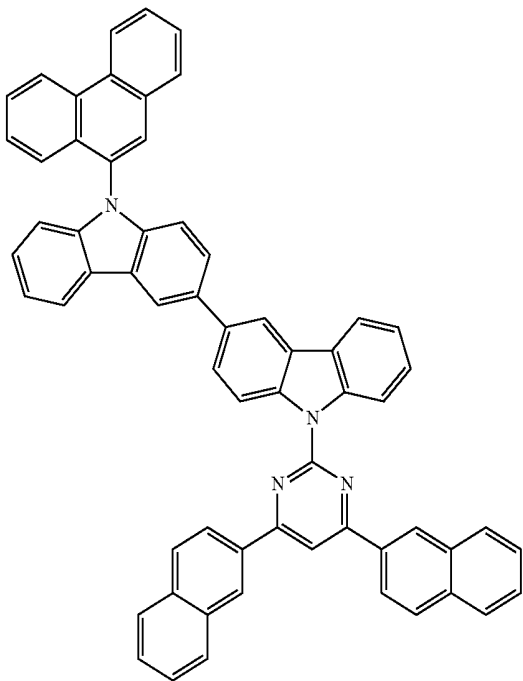
Compound 147



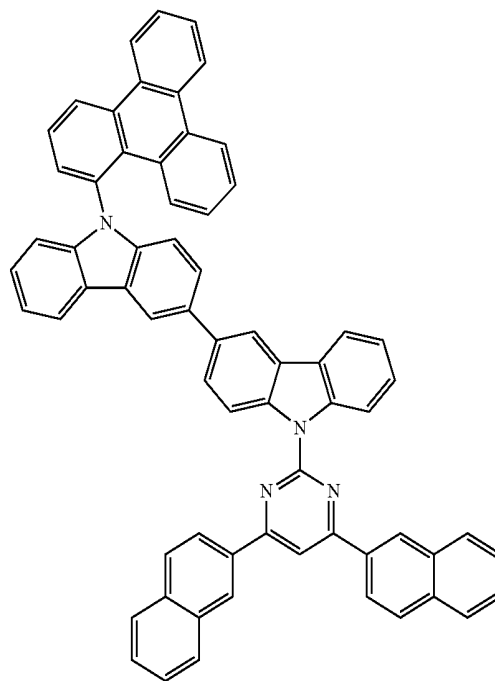
Compound 149



Compound 148



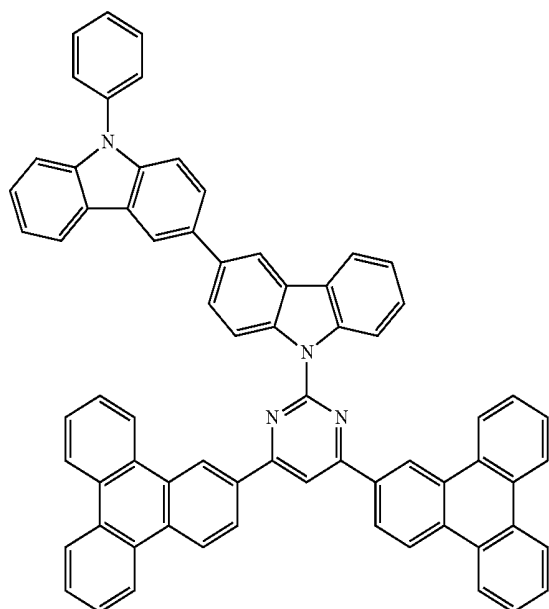
Compound 150



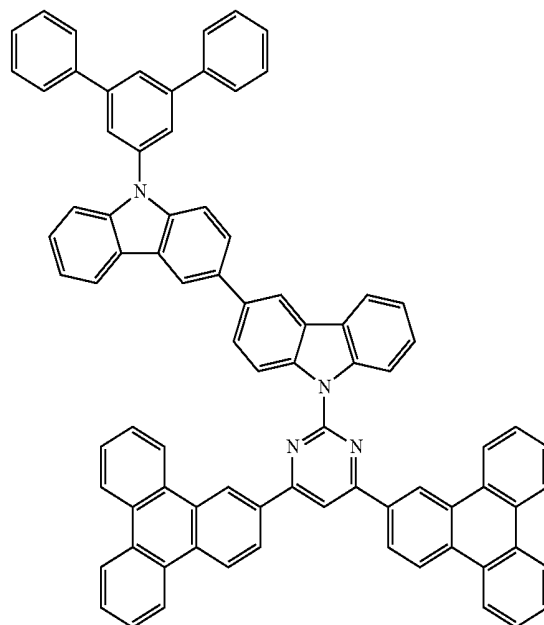
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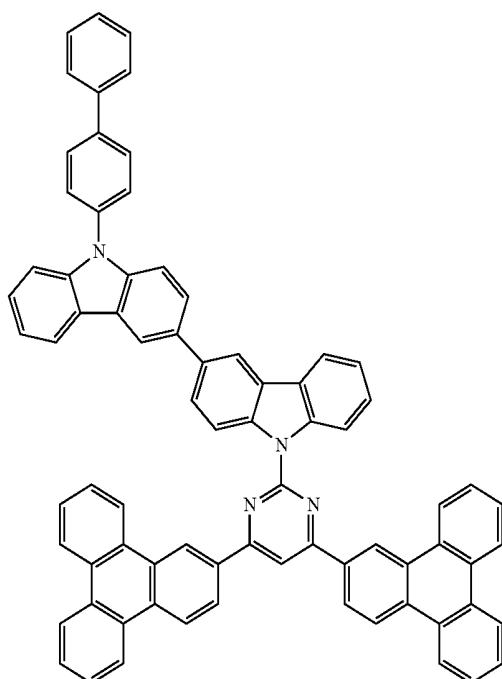
Compound 151



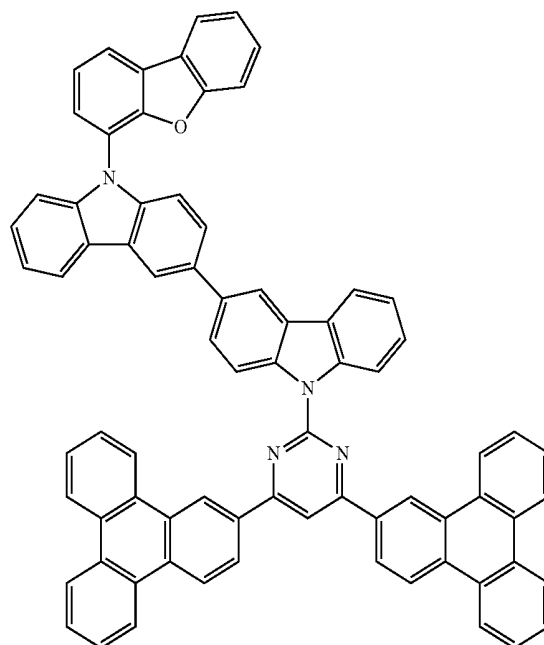
Compound 153



Compound 152



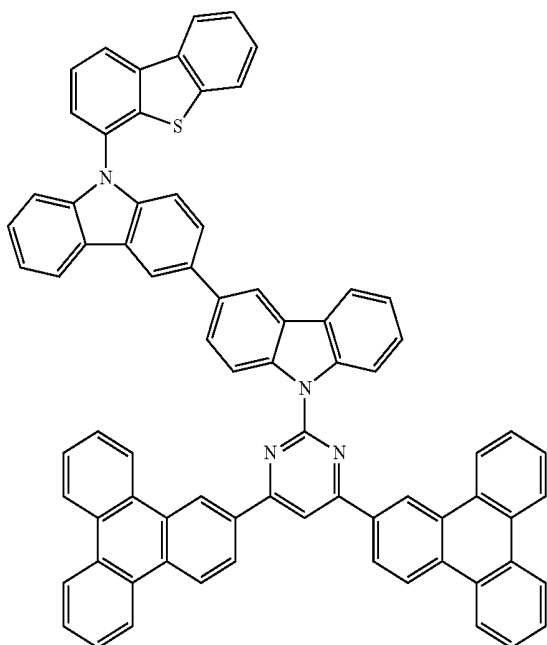
Compound 154



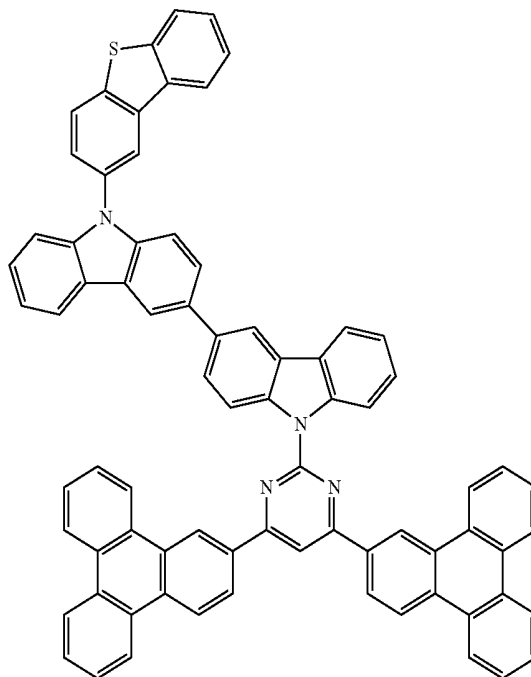
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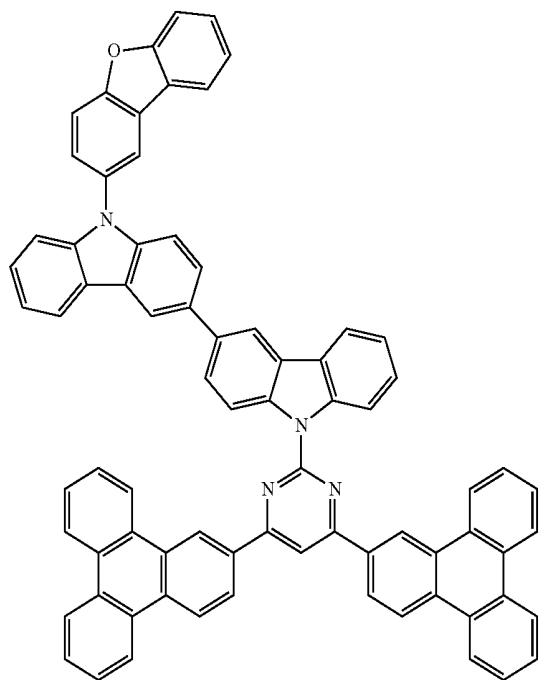
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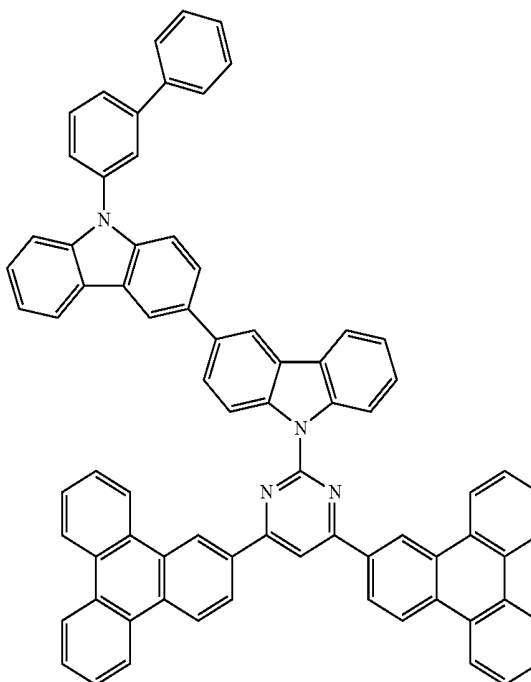
Compound 157



Compound 156



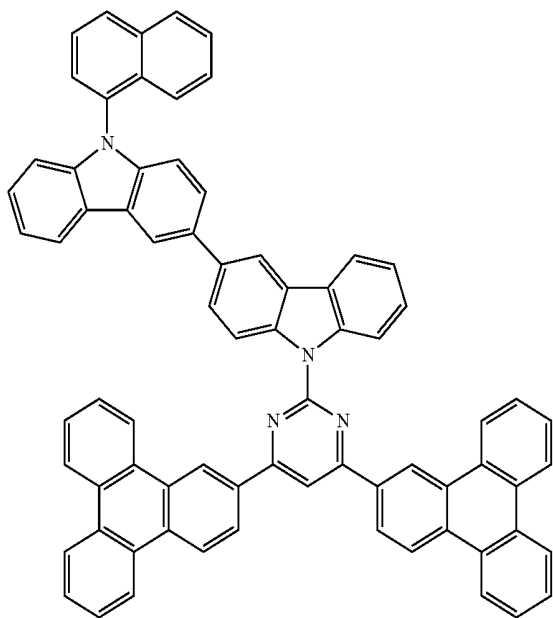
Compound 158



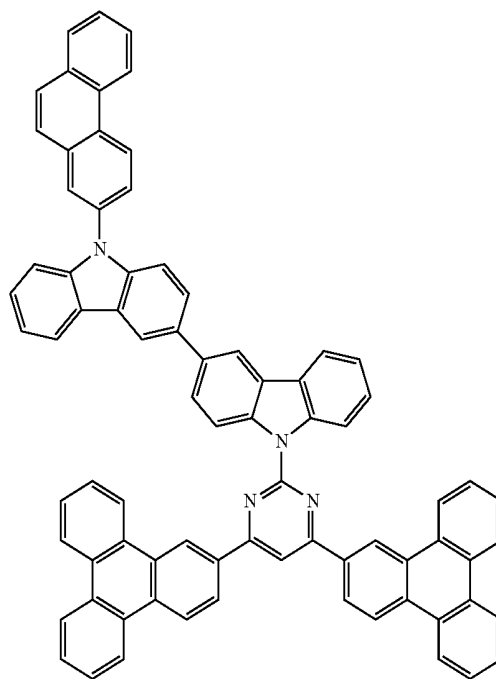
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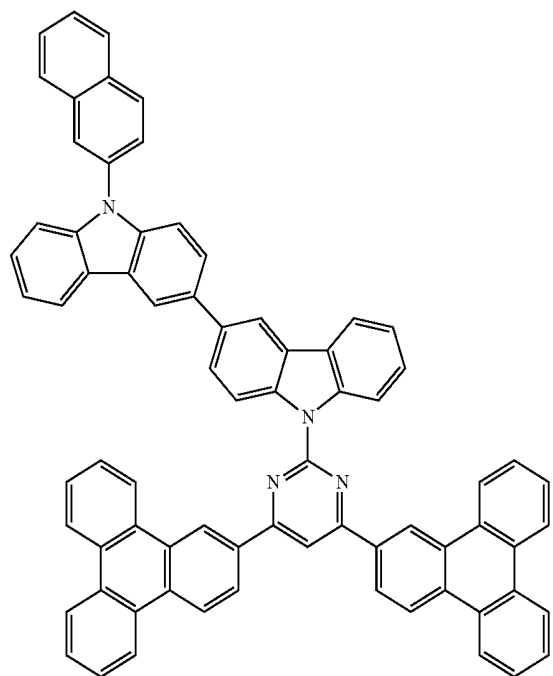
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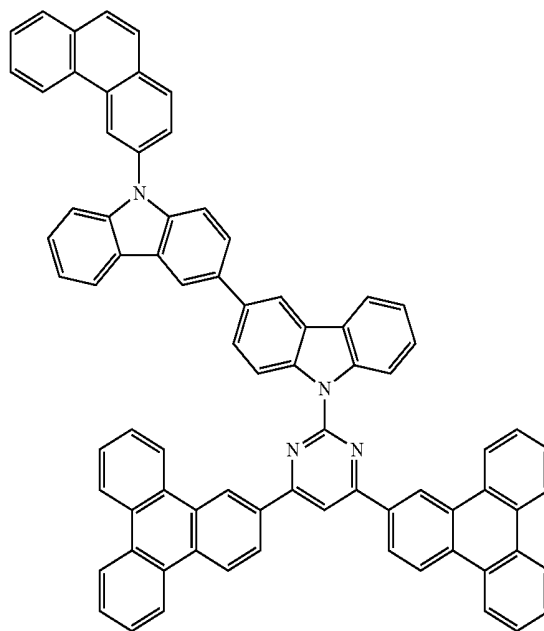
Compound 161



Compound 160



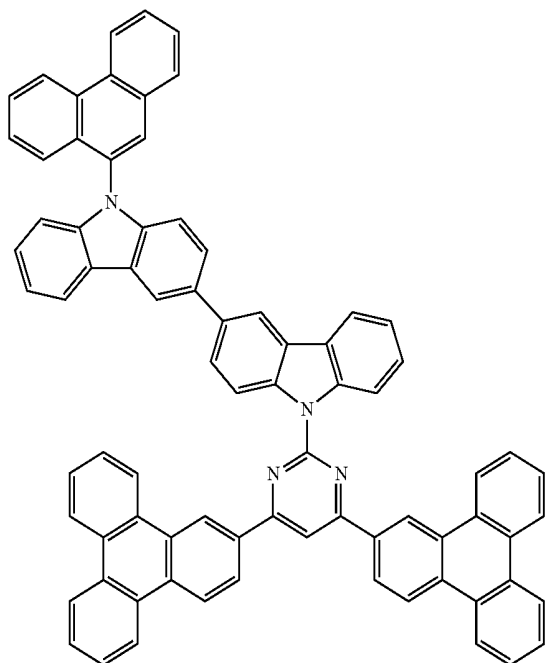
Compound 162



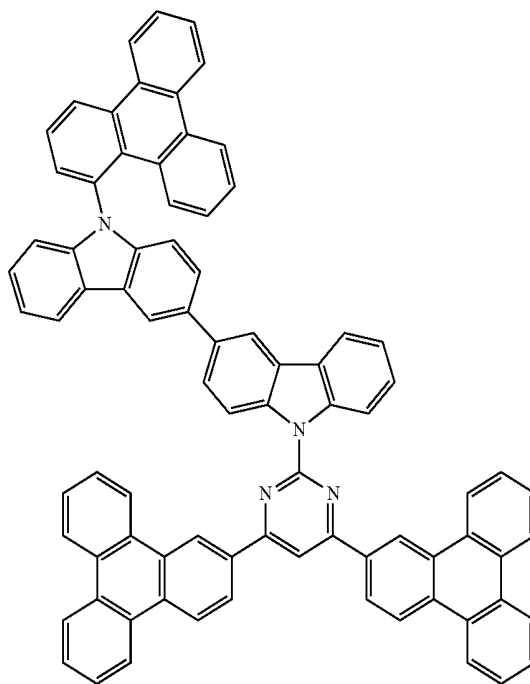
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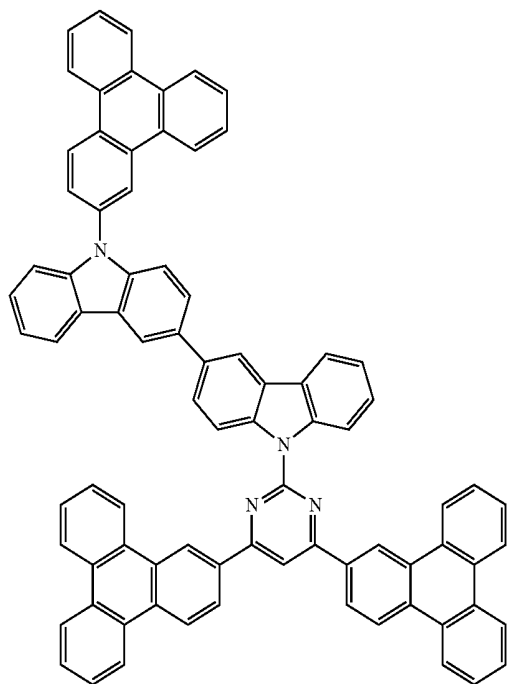
Compound 163



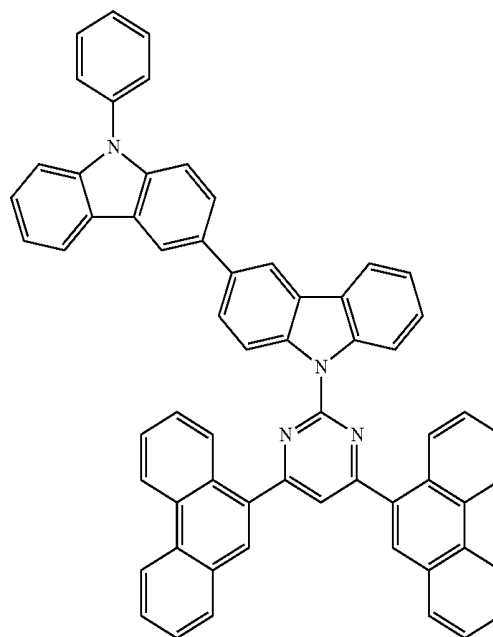
Compound 165



Compound 164



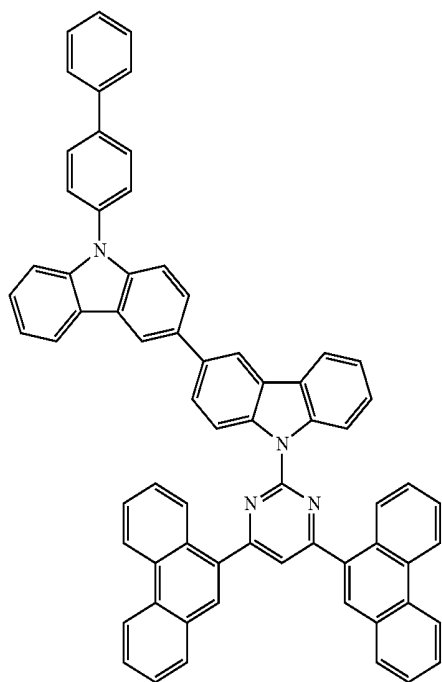
Compound 166



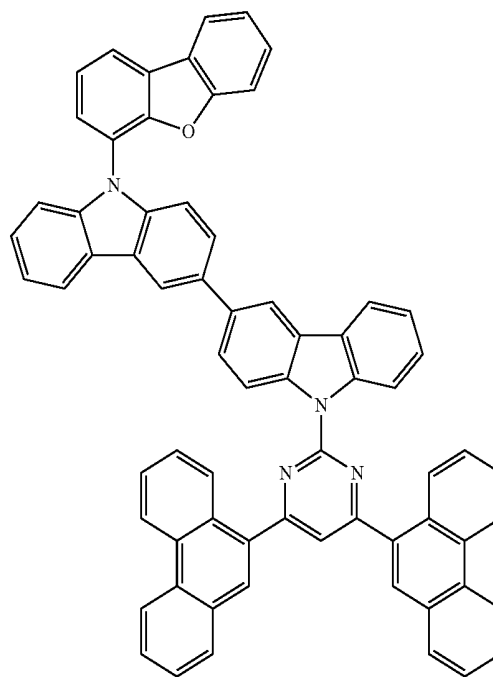
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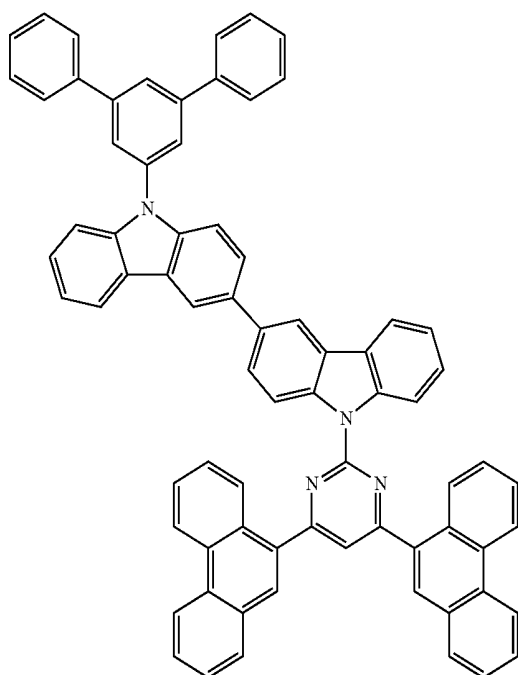
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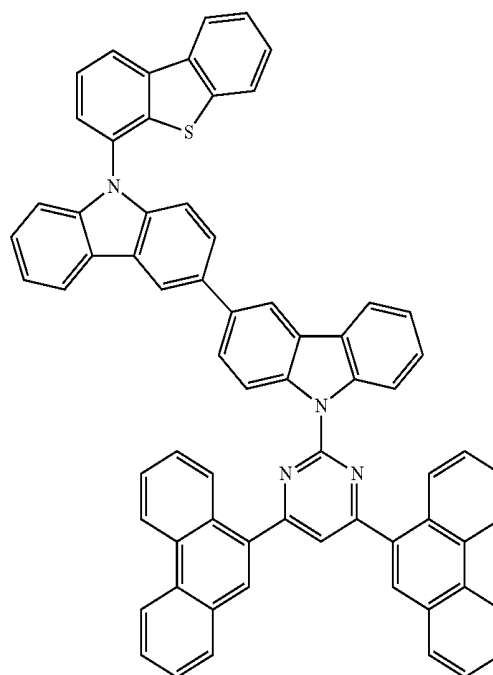
Compound 169



Compound 168



Compound 170

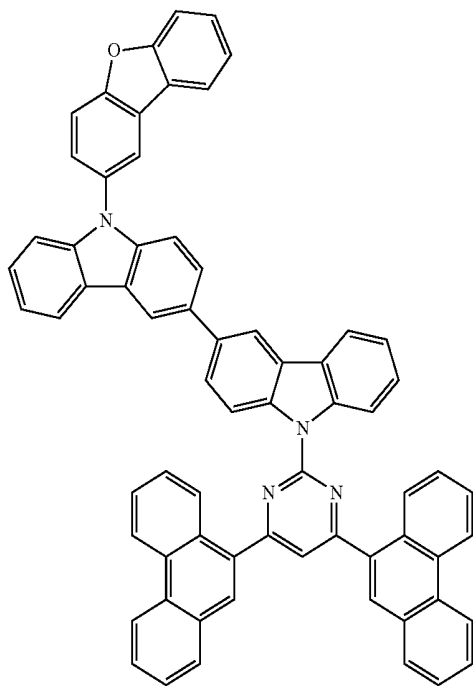




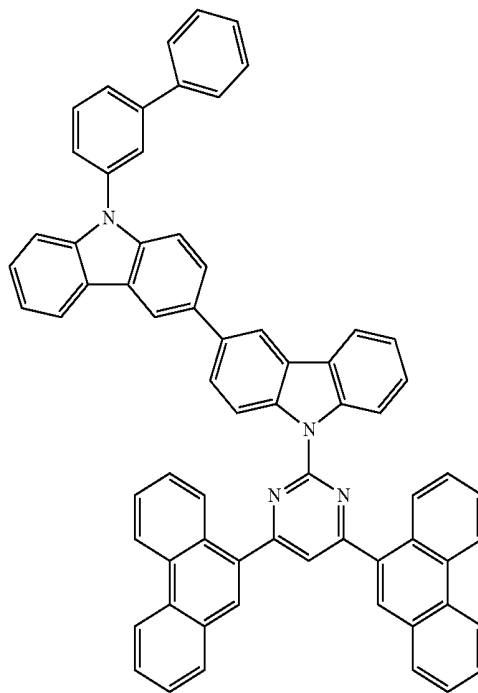
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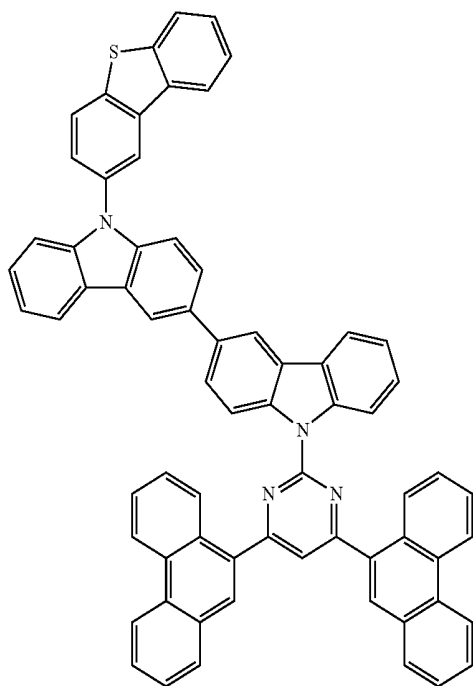
Compound 171



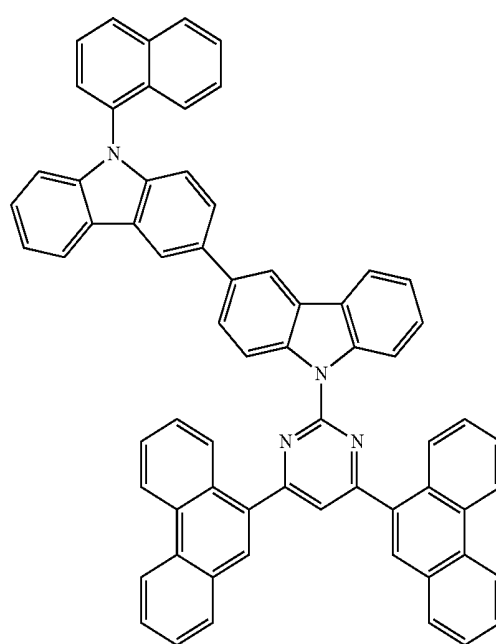
Compound 173



Compound 172



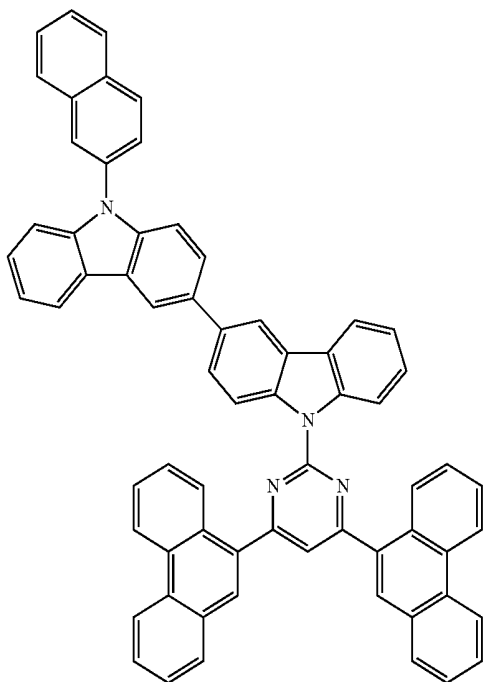
Compound 174



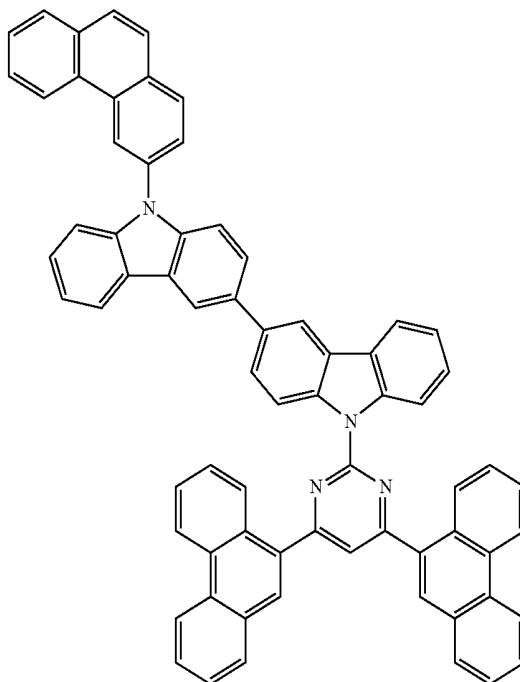
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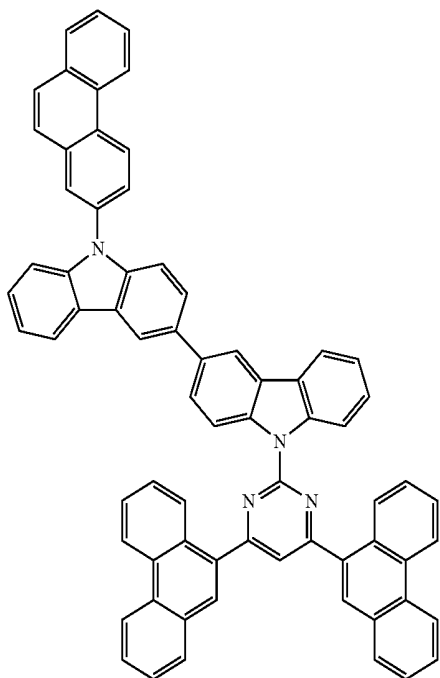
Compound 175



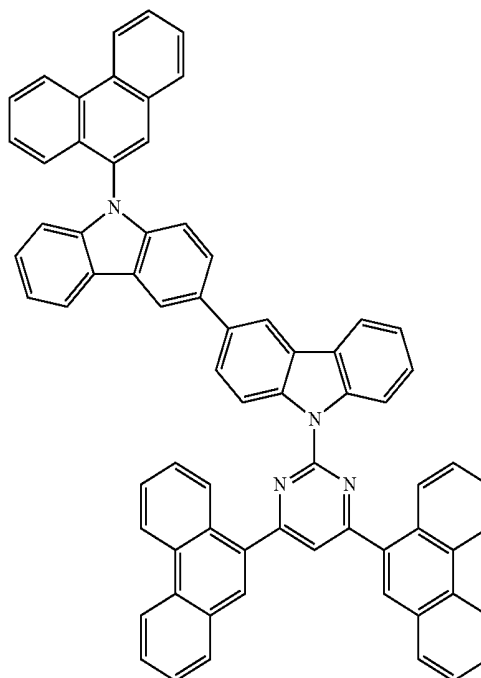
Compound 177



Compound 176

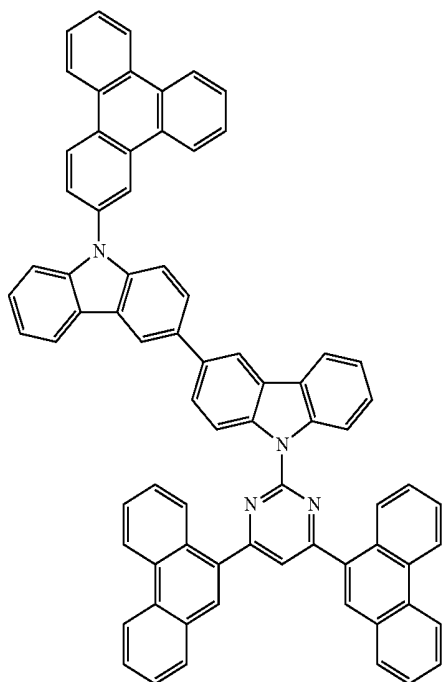


Compound 178

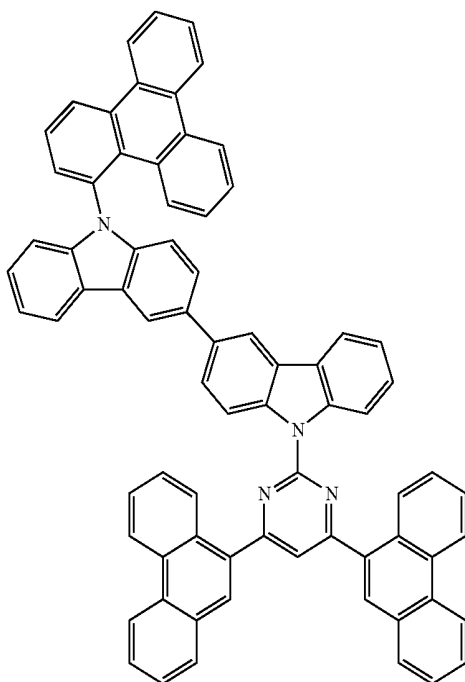


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Compound 179

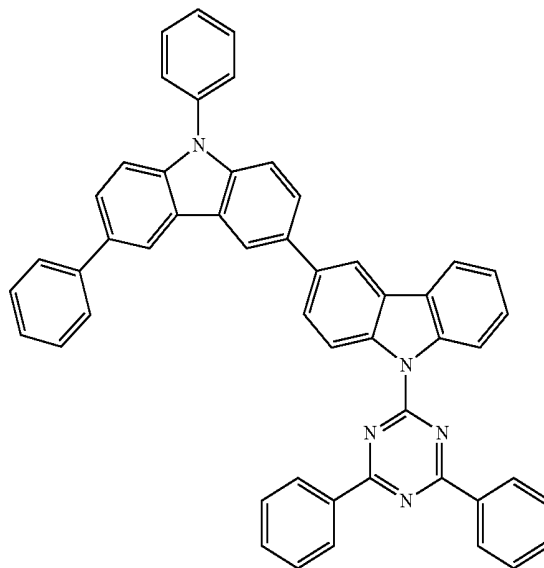


Compound 180

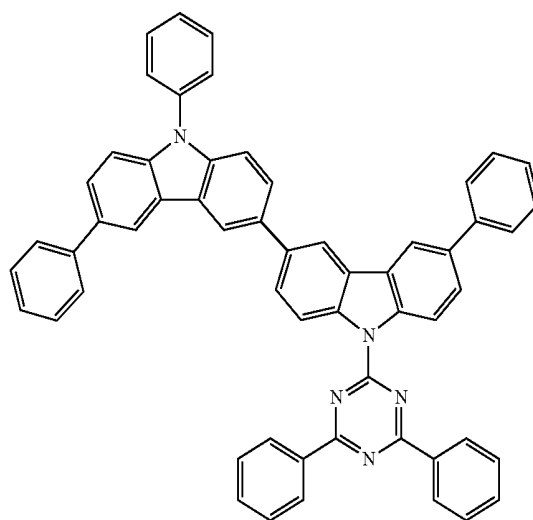


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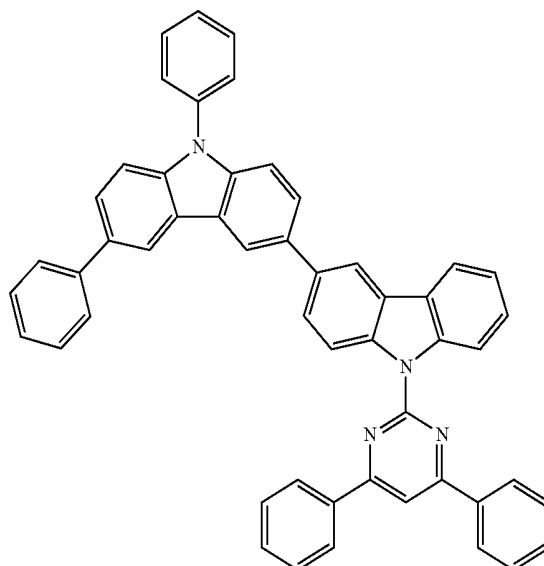
Compound 181



Compound 182

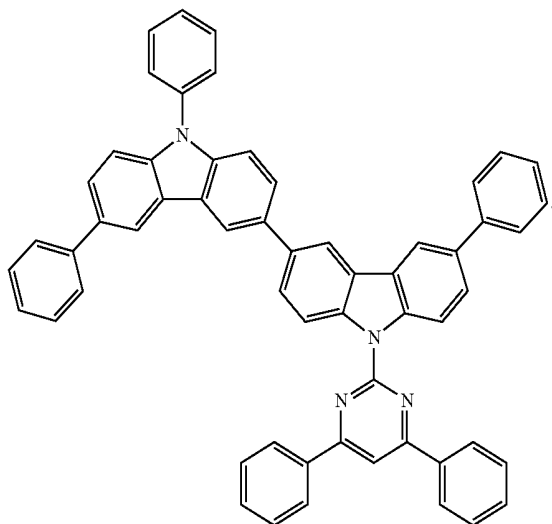


Compound 183



-continued

Compound 184



**[0020]** A first device comprising an organic light emitting device is also provided. The device further comprises an anode, a cathode, and an organic layer, disposed between the anode and the cathode. The organic layer comprises a compound having Formula I, as described above.

**[0021]**  $R_1$ ,  $R_2$ ,  $R_3$ , and  $R_4$  may represent mono, di, tri, or tetra substitutions.  $R_1$ ,  $R_2$ ,  $R_3$ , and  $R_4$  are independently selected from the group consisting of hydrogen, alkyl, alkoxy, amino, alkenyl, alkynyl, aryl and heteroaryl.  $Ar_1$ ,  $Ar_2$ , and  $Ar_3$  are independently selected from aryl or heteroaryl.  $Ar_1$ ,  $Ar_2$ , and  $Ar_3$  may be further substituted. X is C or N.

**[0022]** In one aspect,  $Ar_1$ ,  $Ar_2$ , and  $Ar_3$  are independently selected from the group consisting of phenyl, pyridine, naphthalene, biphenyl, terphenyl, fluorene, dibenzofuran, dibenzothiophene, phenanthrene, and triphenylene.  $Ar_1$ ,  $Ar_2$ , and  $Ar_3$  are independently further substituted with a substituent selected from the group consisting of hydrogen, alkyl, alkoxy, amino, alkenyl, alkynyl, aryl and heteroaryl, but the substituent is not an aryl or heteroaryl fused directly to  $Ar_1$ ,  $Ar_2$ , and  $Ar_3$ . Preferably,  $Ar_1$  and  $Ar_2$  are independently selected from the group consisting of phenyl, pyridine, and naphthalene. Preferably,  $Ar_3$  is selected from the group consisting of phenyl, biphenyl, dibenzofuran, and dibenzothiophene.

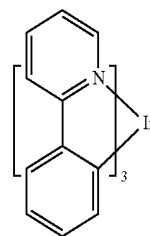
**[0023]** In another aspect,  $R_1$ ,  $R_2$ ,  $R_3$ , and  $R_4$  are hydrogen.

**[0024]** Specific examples of devices containing compounds comprising bicarbazole are also provided. In particular, the compound is selected from the group consisting of Compound 1-Compound 184.

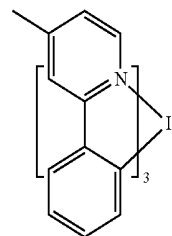
**[0025]** In one aspect, the organic layer is deposited using solution processing.

**[0026]** In one aspect, the organic layer is an emissive layer and the compound having Formula I is a host.

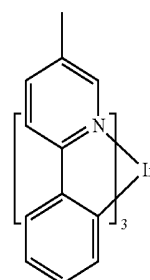
**[0027]** In another aspect, the organic layer further comprises an emissive dopant having the formula:



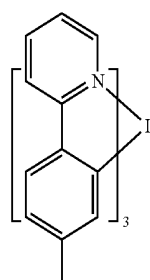
D1



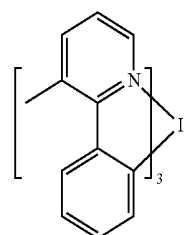
D2



D3

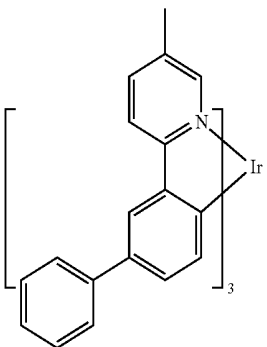
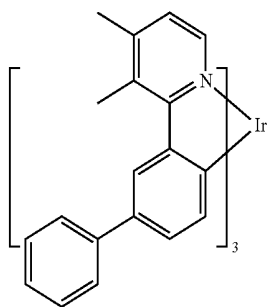
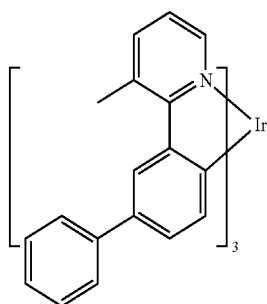
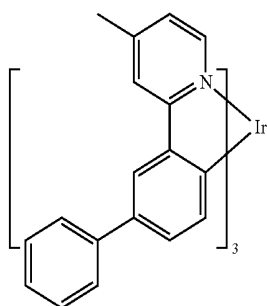
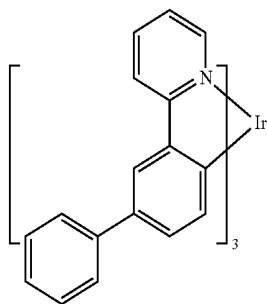


D4



D5

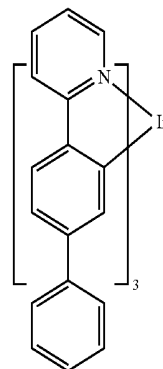
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D6

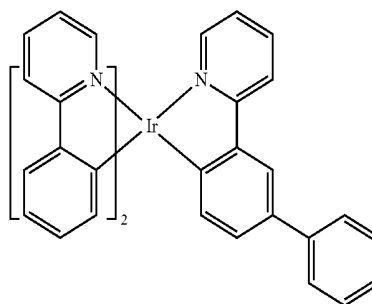
D11



D7

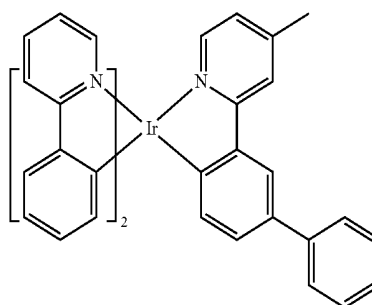
D8

D12



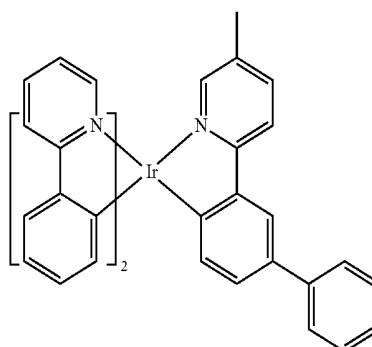
D9

D13

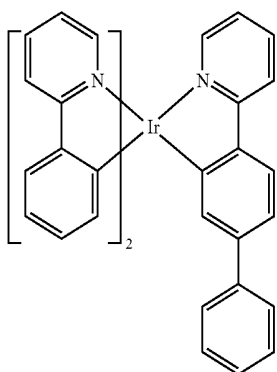


D10

D14

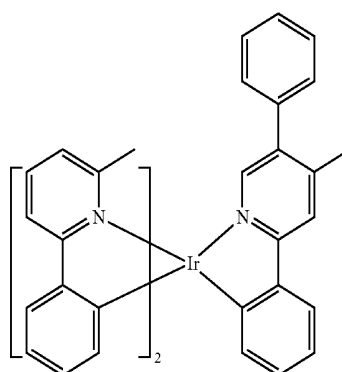


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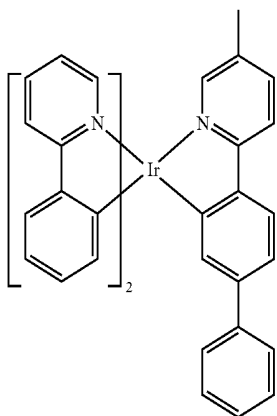


D15

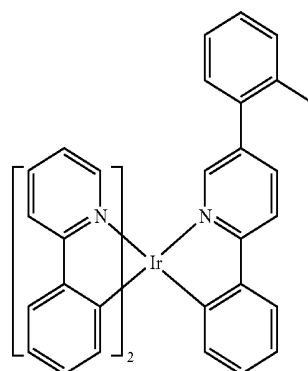
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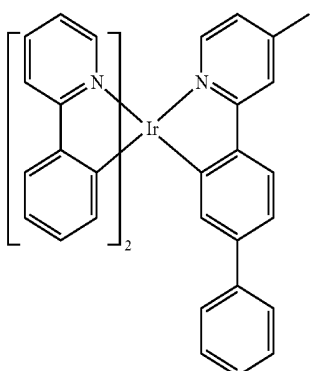
D19



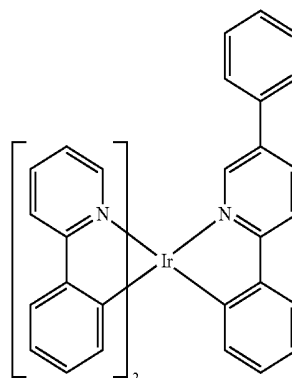
D16



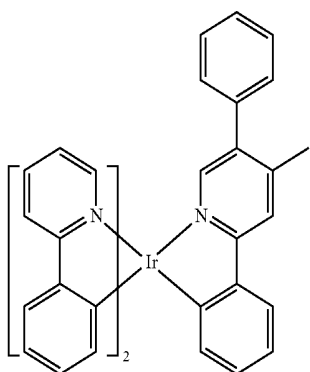
D20



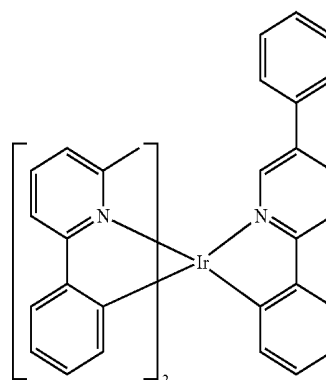
D17



D21

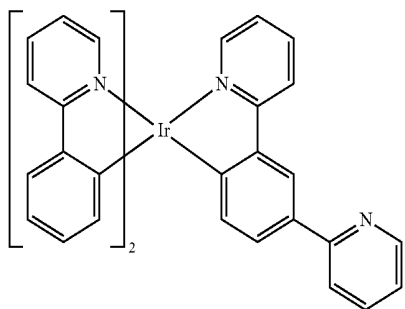


D18

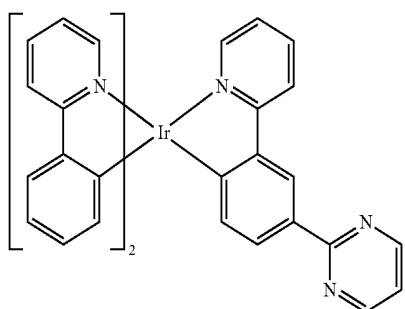


D22

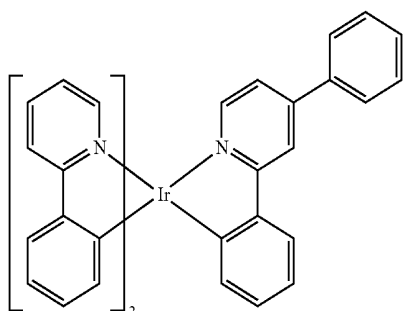
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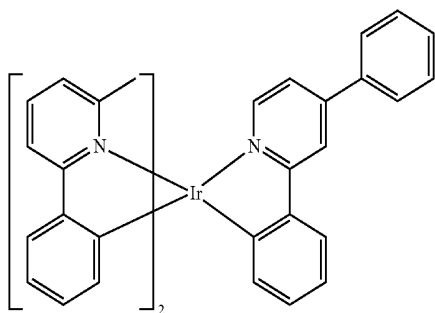
D23



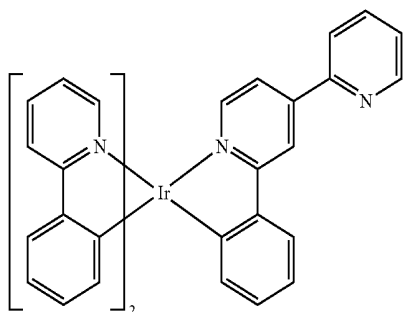
D24



D25

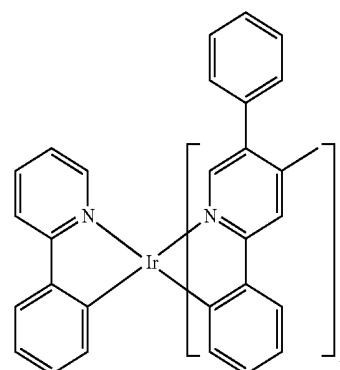


D26

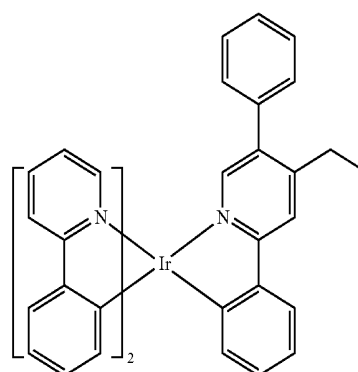


D27

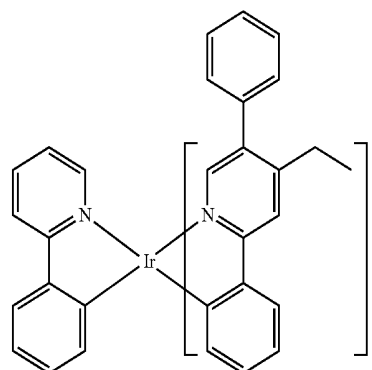
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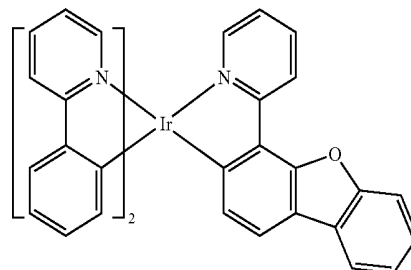
D28



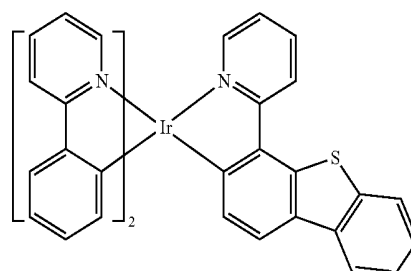
D29



D30

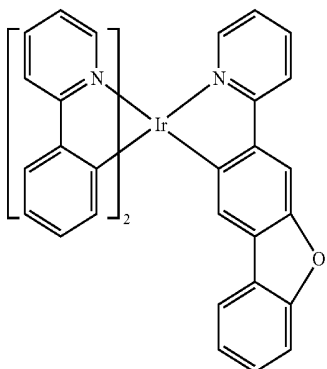


D31



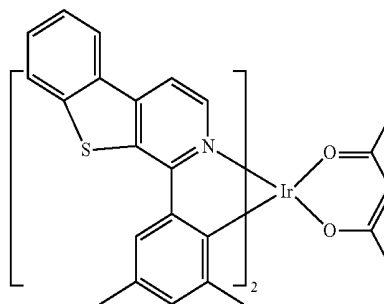
D32

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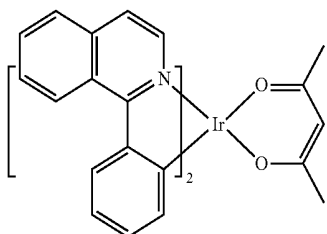


D33

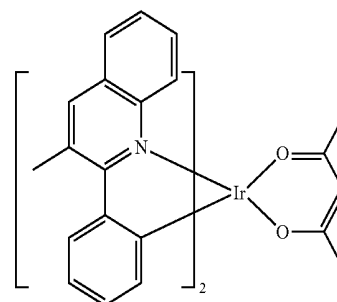
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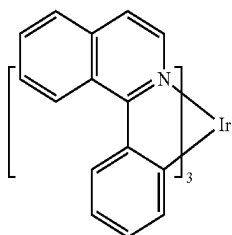
D38



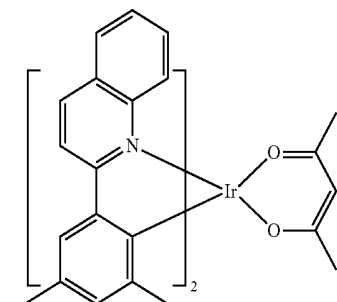
D34



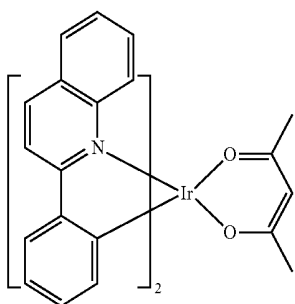
D39



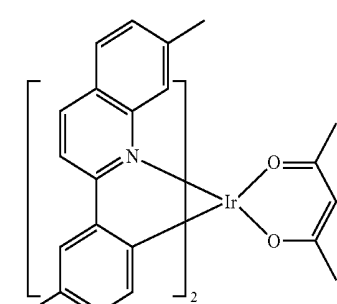
D35



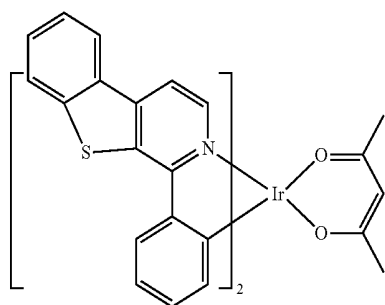
D40



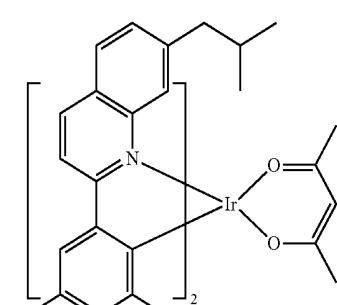
D36



D41



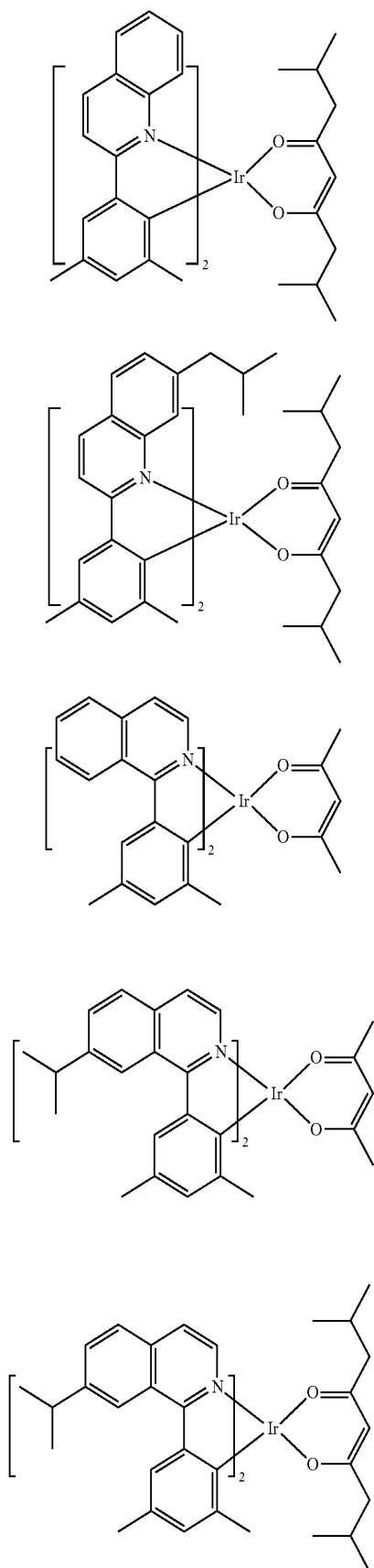
D37



D42



-continued



**[0028]** In one aspect, the first device is a consumer product. In another aspect, the first device is an organic light emitting device.

#### BRIEF DESCRIPTION OF THE DRAWINGS

**[0029]** FIG. 1 shows an organic light emitting device.

**[0030]** FIG. 2 shows an inverted organic light emitting device that does not have a separate electron transport layer.

**[0031]** FIG. 3 shows a bicarbazole compound with a nitrogen-containing heterocycle substitution at the 9-position.

#### DETAILED DESCRIPTION

**[0032]** Generally, an OLED comprises at least one organic layer disposed between and electrically connected to an anode and a cathode. When a current is applied, the anode injects holes and the cathode injects electrons into the organic layer(s). The injected holes and electrons each migrate toward the oppositely charged electrode. When an electron and hole localize on the same molecule, an “exciton,” which is a localized electron-hole pair having an excited energy state, is formed. Light is emitted when the exciton relaxes via a photoemissive mechanism. In some cases, the exciton may be localized on an excimer or an exciplex. Non-radiative mechanisms, such as thermal relaxation, may also occur, but are generally considered undesirable.

**[0033]** The initial OLEDs used emissive molecules that emitted light from their singlet states (“fluorescence”) as disclosed, for example, in U.S. Pat. No. 4,769,292, which is incorporated by reference in its entirety. Fluorescent emission generally occurs in a time frame of less than 10 nanoseconds.

**[0034]** More recently, OLEDs having emissive materials that emit light from triplet states (“phosphorescence”) have been demonstrated. Baldo et al., “Highly Efficient Phosphorescent Emission from Organic Electroluminescent Devices,” *Nature*, vol. 395, 151-154, 1998; (“Baldo-I”) and Baldo et al., “Very high-efficiency green organic light-emitting devices based on electrophosphorescence,” *Appl. Phys. Lett.*, vol. 75, No. 3, 4-6 (1999) (“Baldo-II”), which are incorporated by reference in their entireties. Phosphorescence is described in more detail in U.S. Pat. No. 7,279,704 at cols. 5-6, which are incorporated by reference.

**[0035]** FIG. 1 shows an organic light emitting device 100. The figures are not necessarily drawn to scale. Device 100 may include a substrate 110, an anode 115, a hole injection layer 120, a hole transport layer 125, an electron blocking layer 130, an emissive layer 135, a hole blocking layer 140, an electron transport layer 145, an electron injection layer 150, a protective layer 155, and a cathode 160. Cathode 160 is a compound cathode having a first conductive layer 162 and a second conductive layer 164. Device 100 may be fabricated by depositing the layers described, in order. The properties and functions of these various layers, as well as example materials, are described in more detail in U.S. Pat. No. 7,279,704 at cols. 6-10, which are incorporated by reference.

**[0036]** More examples for each of these layers are available. For example, a flexible and transparent substrate-anode combination is disclosed in U.S. Pat. No. 5,844,363, which is incorporated by reference in its entirety. An example of a p-doped hole transport layer is m-MTDATA doped with F.sub.4-TCNQ at a molar ratio of 50:1, as disclosed in U.S. Patent Application Publication No. 2003/0230980, which is incorporated by reference in its entirety. Examples of emissive and host materials are disclosed in U.S. Pat. No. 6,303,

238 to Thompson et al., which is incorporated by reference in its entirety. An example of an n-doped electron transport layer is BPhen doped with Li at a molar ratio of 1:1, as disclosed in U.S. Patent Application Publication No. 2003/0230980, which is incorporated by reference in its entirety. U.S. Pat. Nos. 5,703,436 and 5,707,745, which are incorporated by reference in their entireties, disclose examples of cathodes including compound cathodes having a thin layer of metal such as Mg:Ag with an overlying transparent, electrically-conductive, sputter-deposited ITO layer. The theory and use of blocking layers is described in more detail in U.S. Pat. No. 6,097,147 and U.S. Patent Application Publication No. 2003/0230980, which are incorporated by reference in their entireties. Examples of injection layers are provided in U.S. Patent Application Publication No. 2004/0174116, which is incorporated by reference in its entirety. A description of protective layers may be found in U.S. Patent Application Publication No. 2004/0174116, which is incorporated by reference in its entirety.

**[0037]** FIG. 2 shows an inverted OLED 200. The device includes a substrate 210, a cathode 215, an emissive layer 220, a hole transport layer 225, and an anode 230. Device 200 may be fabricated by depositing the layers described, in order. Because the most common OLED configuration has a cathode disposed over the anode, and device 200 has cathode 215 disposed under anode 230, device 200 may be referred to as an "inverted" OLED. Materials similar to those described with respect to device 100 may be used in the corresponding layers of device 200. FIG. 2 provides one example of how some layers may be omitted from the structure of device 100.

**[0038]** The simple layered structure illustrated in FIGS. 1 and 2 is provided by way of non-limiting example, and it is understood that embodiments of the invention may be used in connection with a wide variety of other structures. The specific materials and structures described are exemplary in nature, and other materials and structures may be used. Functional OLEDs may be achieved by combining the various layers described in different ways, or layers may be omitted entirely, based on design, performance, and cost factors. Other layers not specifically described may also be included. Materials other than those specifically described may be used. Although many of the examples provided herein describe various layers as comprising a single material, it is understood that combinations of materials, such as a mixture of host and dopant, or more generally a mixture, may be used. Also, the layers may have various sublayers. The names given to the various layers herein are not intended to be strictly limiting. For example, in device 200, hole transport layer 225 transports holes and injects holes into emissive layer 220, and may be described as a hole transport layer or a hole injection layer. In one embodiment, an OLED may be described as having an "organic layer" disposed between a cathode and an anode. This organic layer may comprise a single layer, or may further comprise multiple layers of different organic materials as described, for example, with respect to FIGS. 1 and 2.

**[0039]** Structures and materials not specifically described may also be used, such as OLEDs comprised of polymeric materials (PLEDs) such as disclosed in U.S. Pat. No. 5,247,190 to Friend et al., which is incorporated by reference in its entirety. By way of further example, OLEDs having a single organic layer may be used. OLEDs may be stacked, for example as described in U.S. Pat. No. 5,707,745 to Forrest et al, which is incorporated by reference in its entirety. The OLED structure may deviate from the simple layered struc-

ture illustrated in FIGS. 1 and 2. For example, the substrate may include an angled reflective surface to improve out-coupling, such as a mesa structure as described in U.S. Pat. No. 6,091,195 to Forrest et al., and/or a pit structure as described in U.S. Pat. No. 5,834,893 to Bulovic et al., which are incorporated by reference in their entireties.

**[0040]** Unless otherwise specified, any of the layers of the various embodiments may be deposited by any suitable method. For the organic layers, preferred methods include thermal evaporation, ink-jet, such as described in U.S. Pat. Nos. 6,013,982 and 6,087,196, which are incorporated by reference in their entireties, organic vapor phase deposition (OVPD), such as described in U.S. Pat. No. 6,337,102 to Forrest et al., which is incorporated by reference in its entirety, and deposition by organic vapor jet printing (OVJP), such as described in U.S. patent application Ser. No. 10/233,470, which is incorporated by reference in its entirety. Other suitable deposition methods include spin coating and other solution based processes. Solution based processes are preferably carried out in nitrogen or an inert atmosphere. For the other layers, preferred methods include thermal evaporation. Preferred patterning methods include deposition through a mask, cold welding such as described in U.S. Pat. Nos. 6,294,398 and 6,468,819, which are incorporated by reference in their entireties, and patterning associated with some of the deposition methods such as ink jet and OVJD. Other methods may also be used. The materials to be deposited may be modified to make them compatible with a particular deposition method. For example, substituents such as alkyl and aryl groups, branched or unbranched, and preferably containing at least 3 carbons, may be used in small molecules to enhance their ability to undergo solution processing. Substituents having 20 carbons or more may be used, and 3-20 carbons is a preferred range. Materials with asymmetric structures may have better solution processibility than those having symmetric structures, because asymmetric materials may have a lower tendency to recrystallize. Dendrimer substituents may be used to enhance the ability of small molecules to undergo solution processing.

**[0041]** Devices fabricated in accordance with embodiments of the invention may be incorporated into a wide variety of consumer products, including flat panel displays, computer monitors, televisions, billboards, lights for interior or exterior illumination and/or signaling, heads up displays, fully transparent displays, flexible displays, laser printers, telephones, cell phones, personal digital assistants (PDAs), laptop computers, digital cameras, camcorders, viewfinders, micro-displays, vehicles, a large area wall, theater or stadium screen, or a sign. Various control mechanisms may be used to control devices fabricated in accordance with the present invention, including passive matrix and active matrix. Many of the devices are intended for use in a temperature range comfortable to humans, such as 18 degrees C. to 30 degrees C., and more preferably at room temperature (20-25 degrees C.).

**[0042]** The materials and structures described herein may have applications in devices other than OLEDs. For example, other optoelectronic devices such as organic solar cells and organic photodetectors may employ the materials and structures. More generally, organic devices, such as organic transistors, may employ the materials and structures.

**[0043]** The terms halo, halogen, alkyl, cycloalkyl, alkenyl, alkynyl, arylkyl, heterocyclic group, aryl, aromatic group,

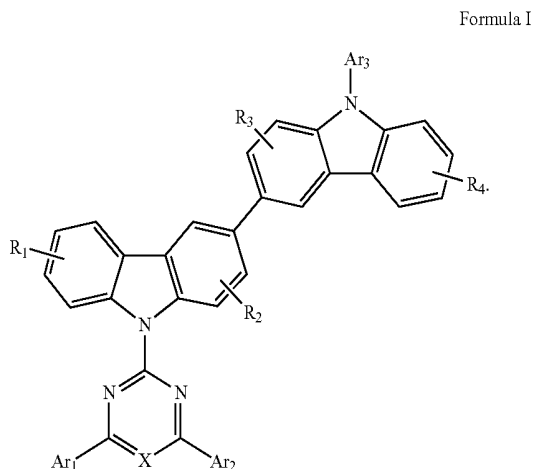
and heteroaryl are known to the art, and are defined in U.S. Pat. No. 7,279,704 at cols. 31-32, which are incorporated herein by reference.

**[0044]** Novel bicarbazole containing compounds are provided (illustrated in FIG. 3). More specifically, these compounds contain a 3,3'-bicarbazole core and triazine or pyrimidine substitution at the 9-position. These compounds may be used as hosts for phosphorescent OLEDs.

**[0045]** Carbazole containing compounds for use as OLED materials have been previously described. In particular, 3,3'-bicarbazole compounds have good hole transporting properties, but have poor stability toward electrons. Alkyl and aryl substituted 3,3'-bicarbazole compounds have been used as hole transporting materials and hosts in OLEDs; however, these compounds also have imbalanced charge transporting properties and poor electron stability and may provide devices with low efficiency and limited lifetime. For example, a diaryl substituted 3,3'-bicarbazole, i.e. H1, has a HOMO around 5.6 eV, very good for hole transporting but poor for electron transporting and stability. Therefore, the 3,3'-bicarbazole compounds reported in the literature may have limited use.

**[0046]** In the present invention, nitrogen containing electron deficient heterocycles were introduced to 3,3'-bicarbazole compounds. In particular, the compounds contain a 3,3'-bicarbazole core and triazine or pyrimidine substitution at the 9 position. The nitrogen containing heterocycle tunes the HOMO/LUMO levels as well as increases the compound's stability toward electrons. In addition, these compounds contain a donor part, i.e. bicarbazole, and an acceptor part, i.e. electron deficient nitrogen heterocycle. Without being bound by theory, it is believed that these donor-acceptor type molecules can shrink singlet and triplet gap and improve stability to both hole and electrons. Therefore, these 3,3'-bicarbazole compounds containing a nitrogen heterocycle may provide devices having better stability and lower operating voltage.

**[0047]** Compounds comprising a bicarbazole are provided. The compounds have the formula:



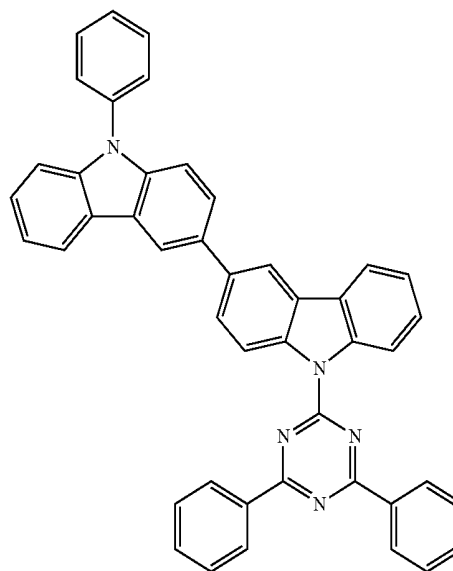
**[0048]**  $R_1$ ,  $R_2$ ,  $R_3$ , and  $R_4$  may represent mono, di, tri, or tetra substitutions.  $R_1$ ,  $R_2$ ,  $R_3$ , and  $R_4$  are independently selected from the group consisting of hydrogen, alkyl, alkoxy, amino, alkenyl, alkynyl, aryl and heteroaryl.  $Ar_1$ ,  $Ar_2$ , and  $Ar_3$  are independently selected from aryl or heteroaryl.  $Ar_1$ ,  $Ar_2$ , and  $Ar_3$  may be further substituted. X is C or N.

**[0049]** In one aspect,  $Ar_1$ ,  $Ar_2$ , and  $Ar_3$  are independently selected from the group consisting of phenyl, pyridine, naphthalene, biphenyl, terphenyl, fluorene, dibenzofuran, dibenzothiophene, phenanthrene, and triphenylene, and  $Ar_1$ ,  $Ar_2$ , and  $Ar_3$  are independently further substituted with a substituent selected from the group consisting of hydrogen, alkyl, alkoxy, amino, alkenyl, alkynyl, aryl and heteroaryl, but the substituent is not an aryl or heteroaryl fused directly to  $Ar_1$ ,  $Ar_2$ , and  $Ar_3$ . Preferably,  $Ar_1$  and  $Ar_2$  are independently selected from the group consisting of phenyl, pyridine, and naphthalene. Preferably,  $Ar_3$  is selected from the group consisting of phenyl, biphenyl, dibenzofuran, and dibenzothiophene.

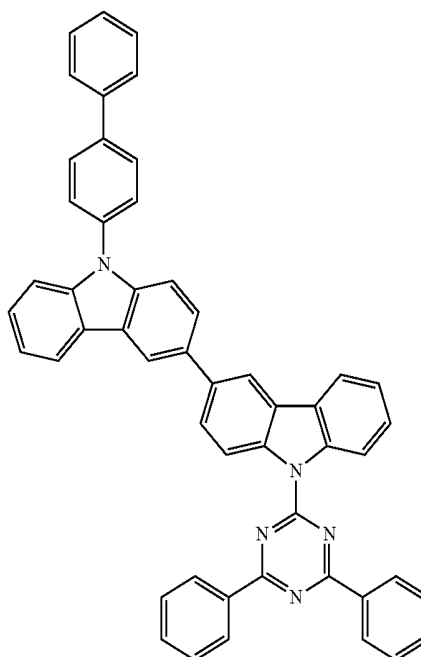
**[0050]** In another aspect,  $R_1$ ,  $R_2$ ,  $R_3$ , and  $R_4$  are hydrogen.

**[0051]** Specific examples of compounds comprising bicarbazole are also provided. In particular, the compound is selected from the group consisting of:

Compound 1



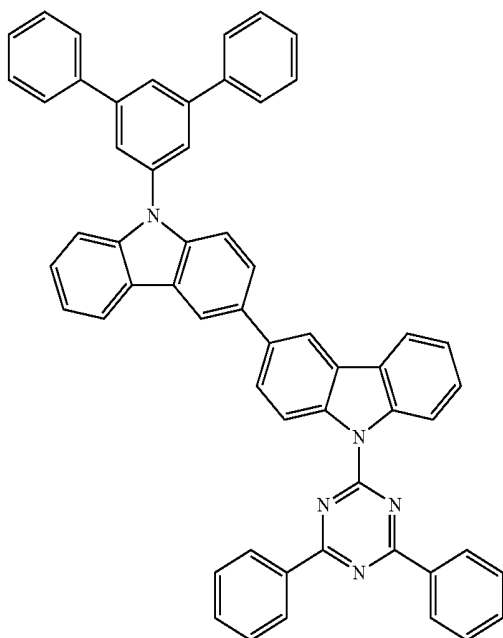
Compound 2



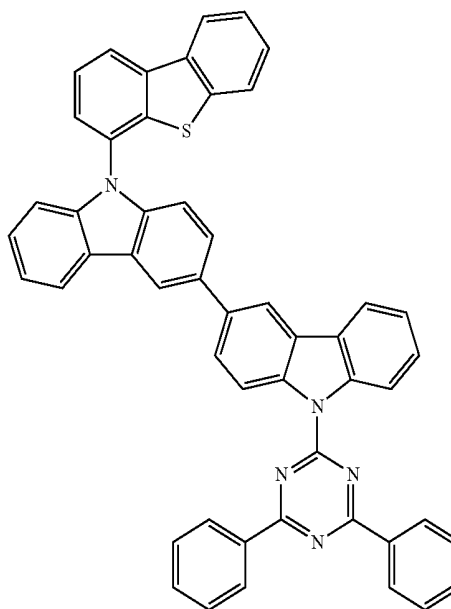
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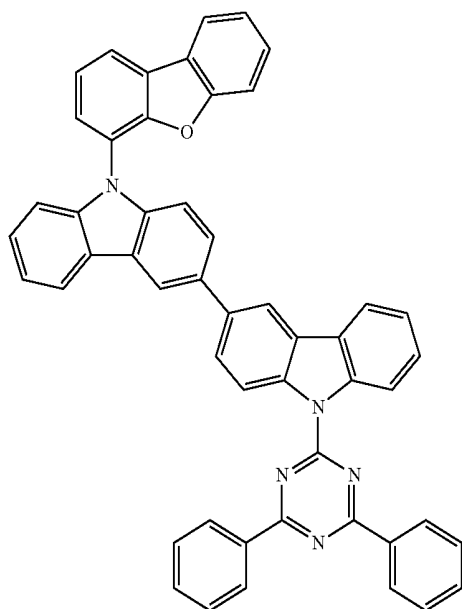
Compound 3



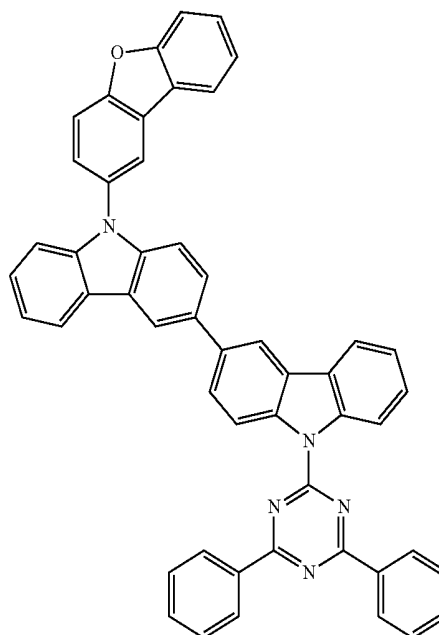
Compound 5



Compound 4



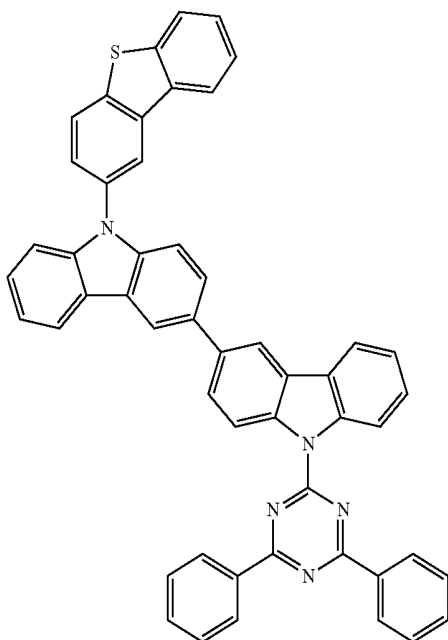
Compound 6



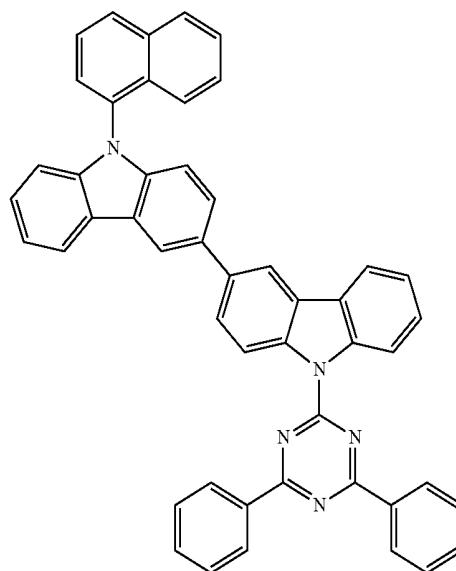
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Compound 7

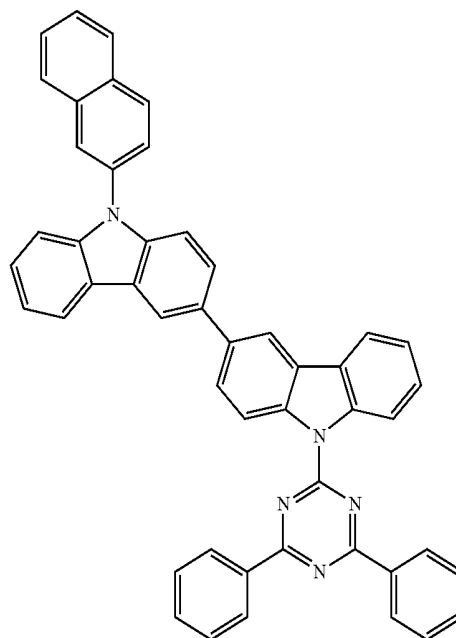
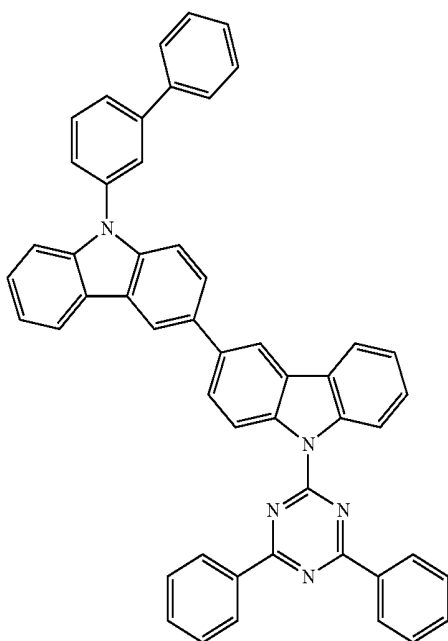


Compound 9



Compound 10

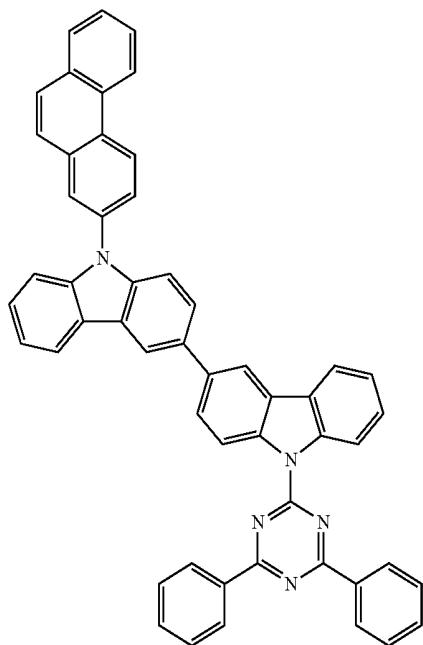
Compound 8



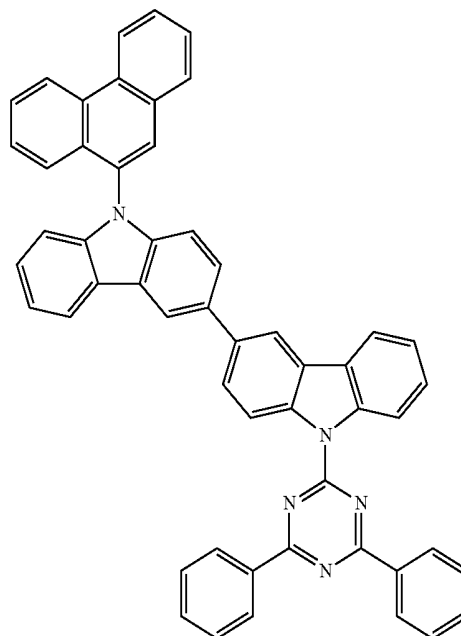
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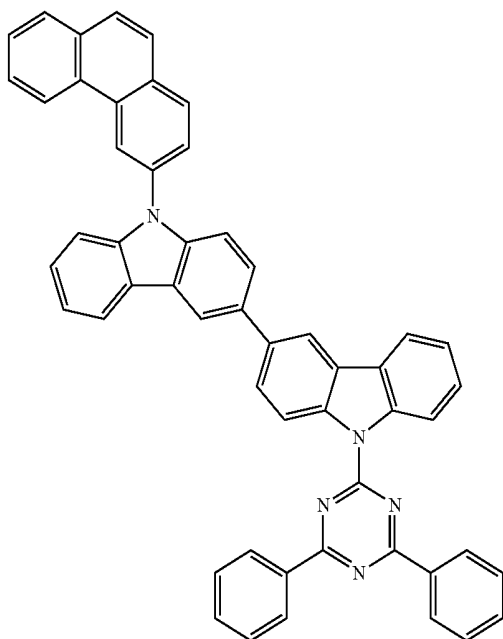
Compound 11



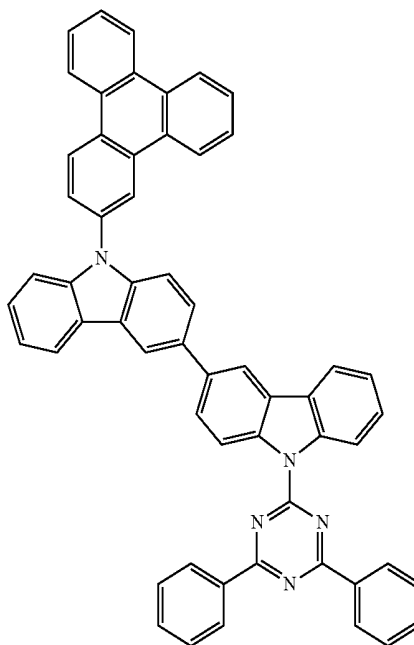
Compound 13



Compound 12



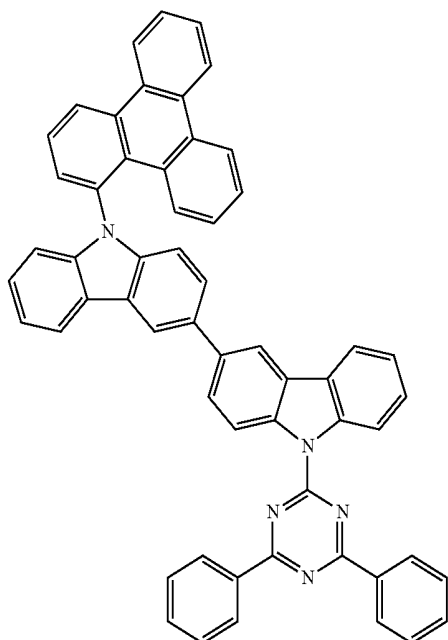
Compound 14



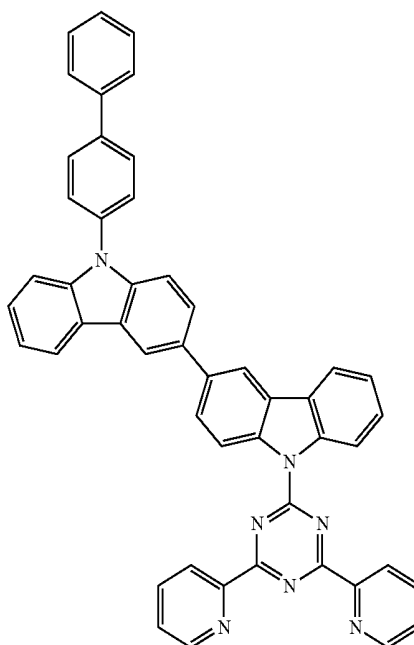
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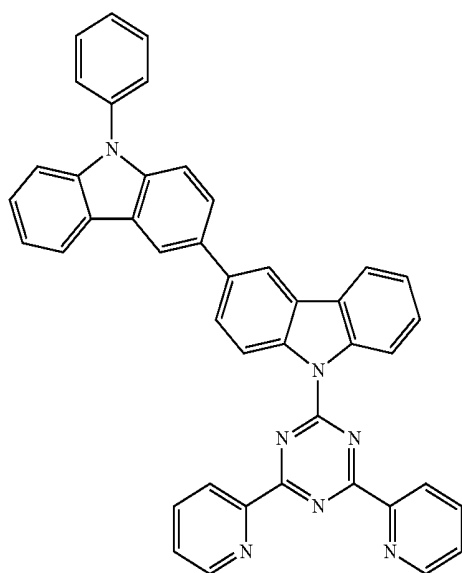
Compound 15



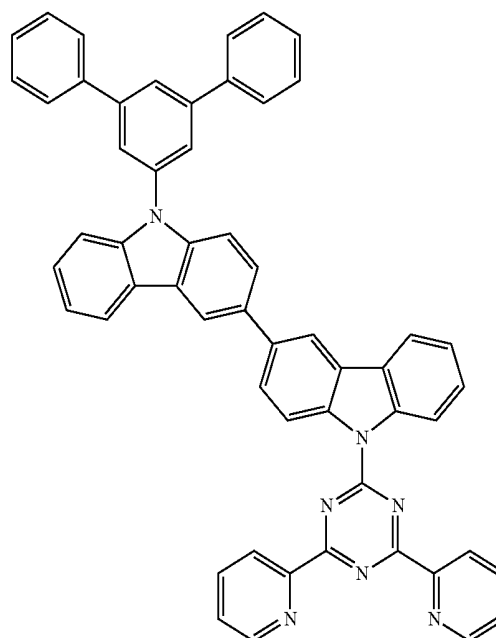
Compound 17



Compound 16



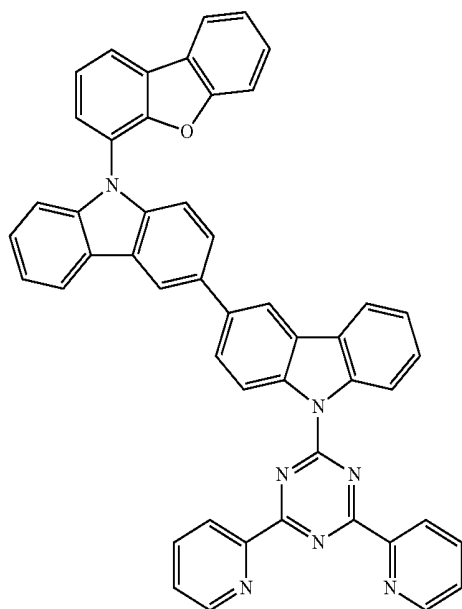
Compound 18



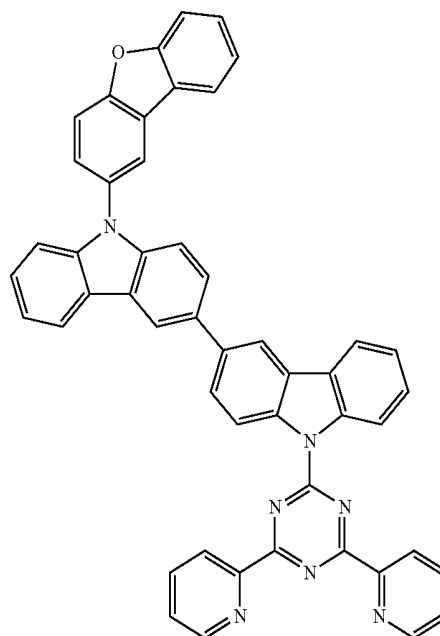
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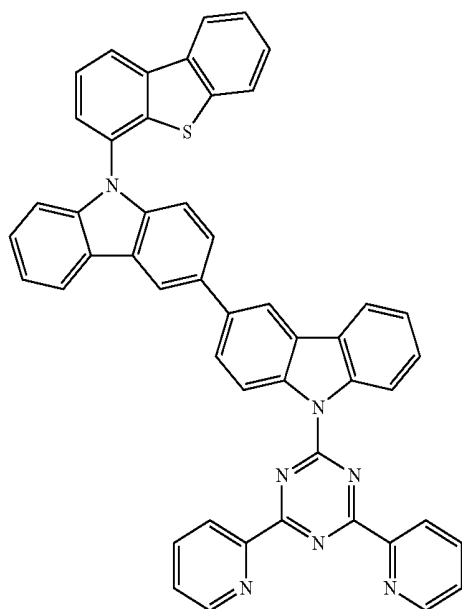
Compound 19



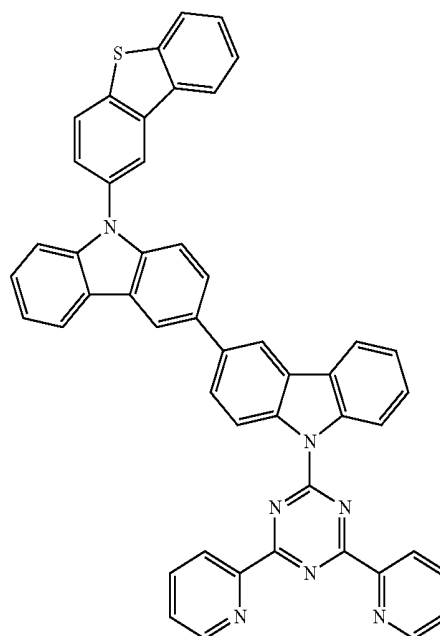
Compound 21



Compound 20



Compound 22

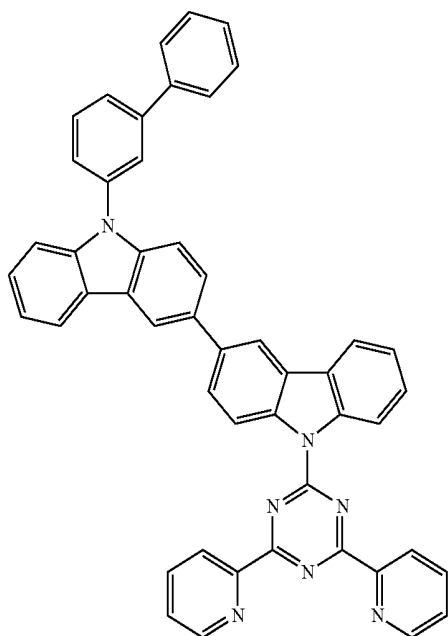




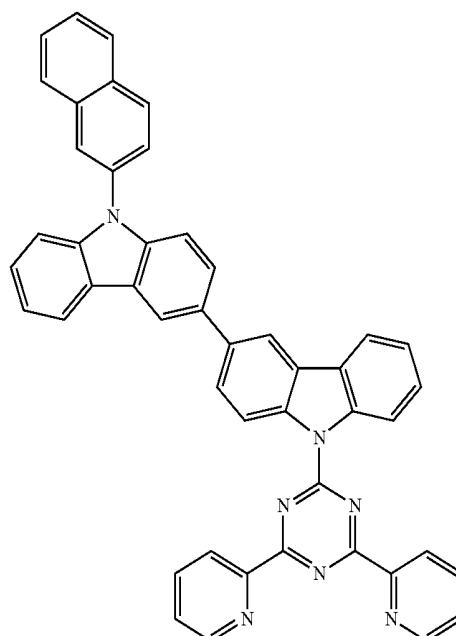
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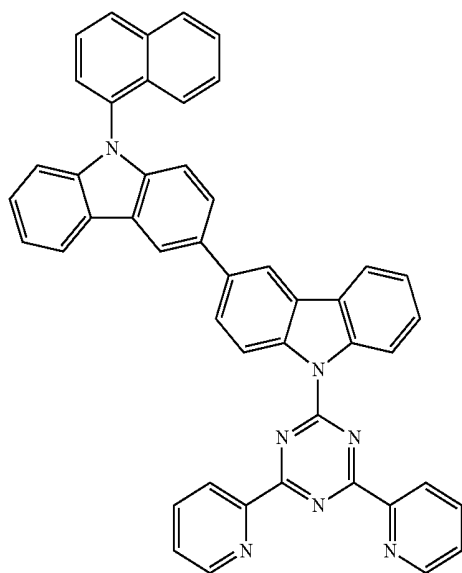
Compound 23



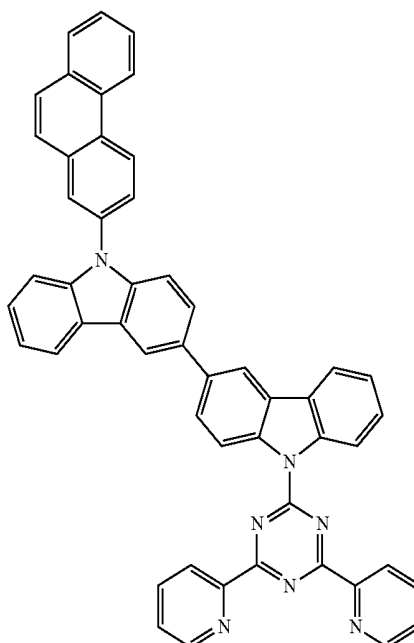
Compound 25



Compound 24



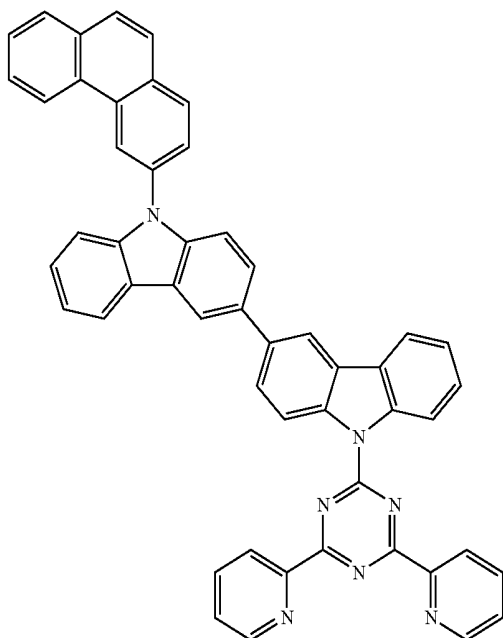
Compound 26



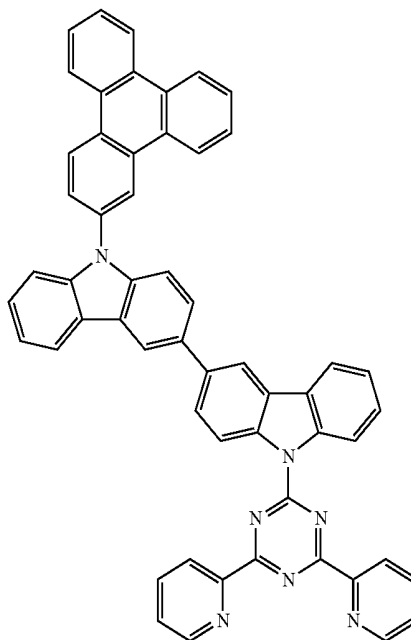
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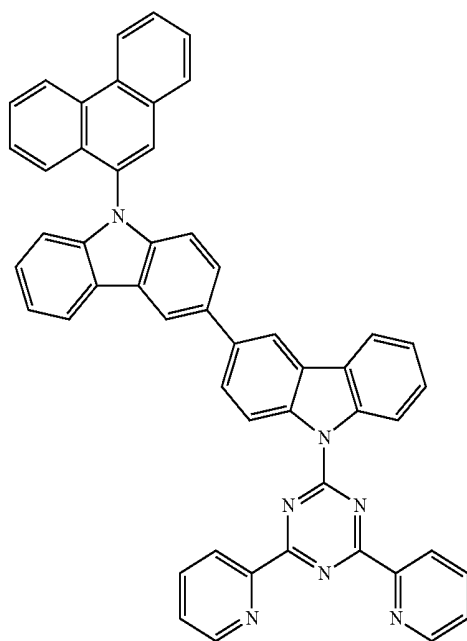
Compound 27



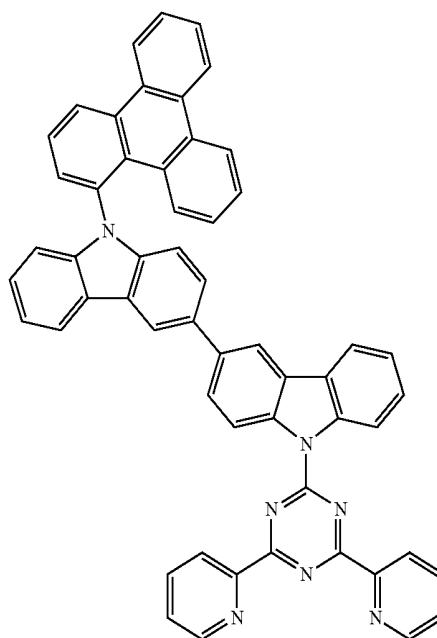
Compound 29



Compound 28



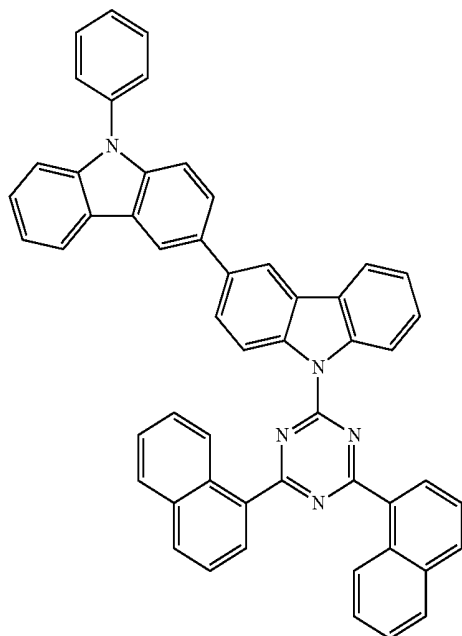
Compound 30



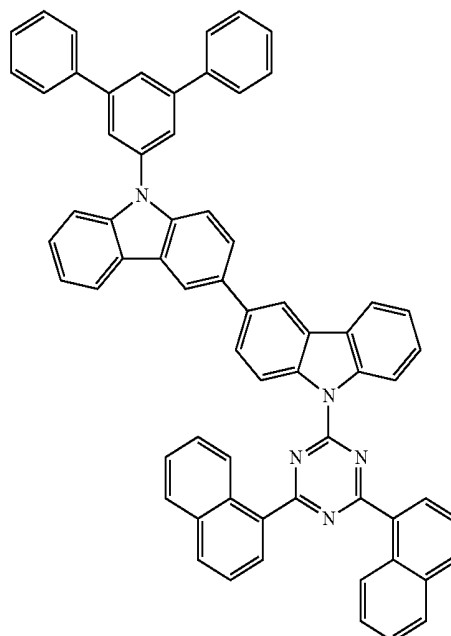
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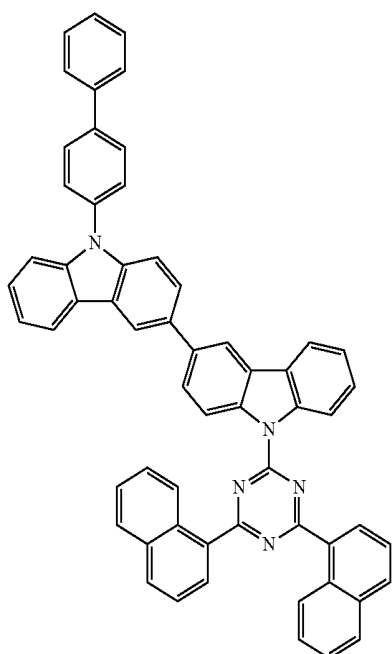
Compound 31



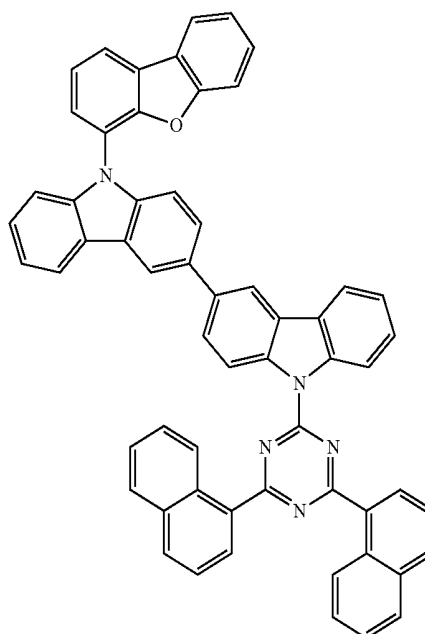
Compound 33



Compound 32



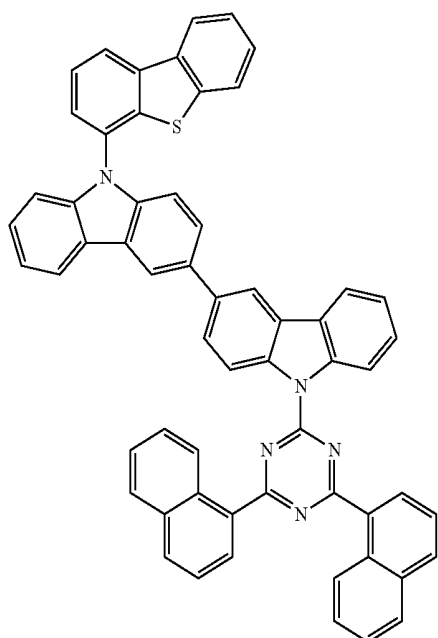
Compound 34



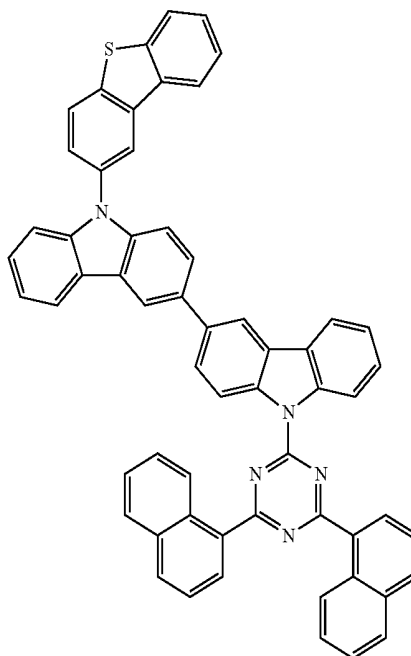
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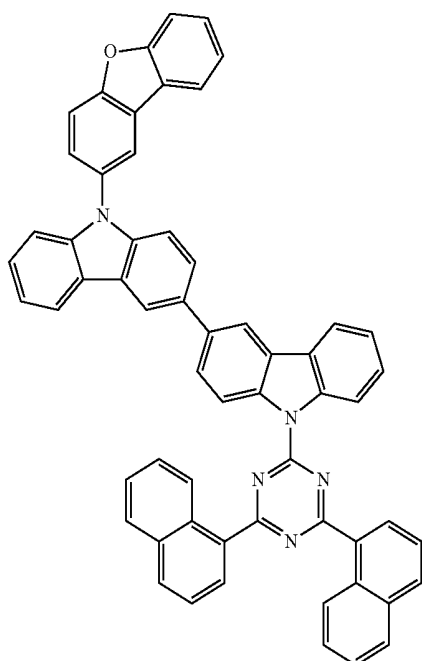
Compound 35



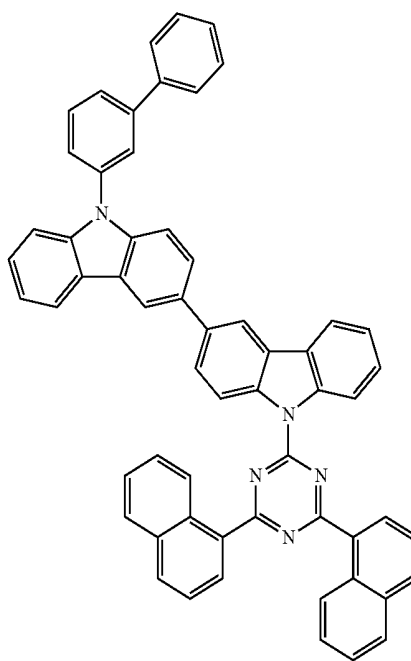
Compound 37



Compound 36



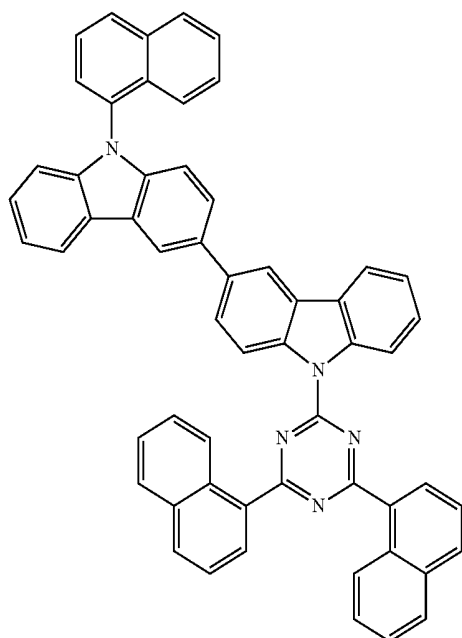
Compound 38



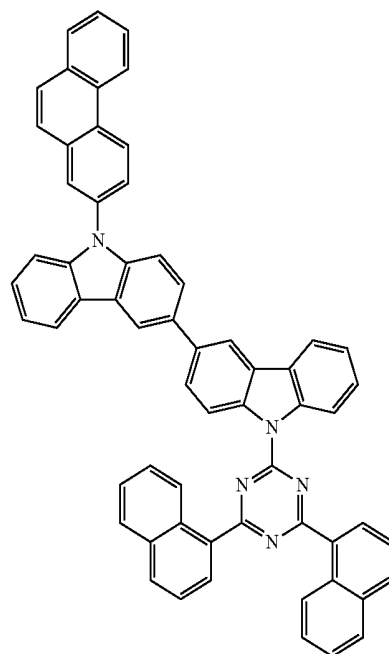
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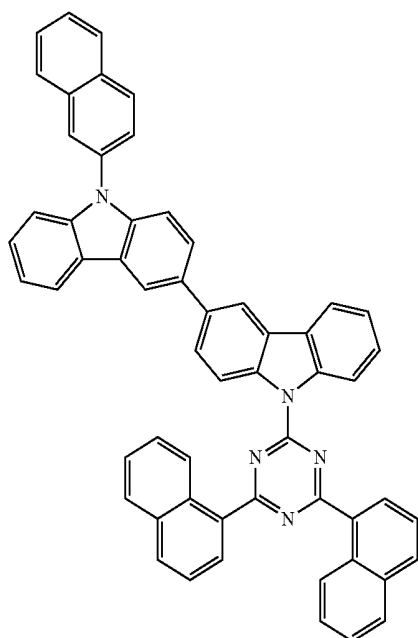
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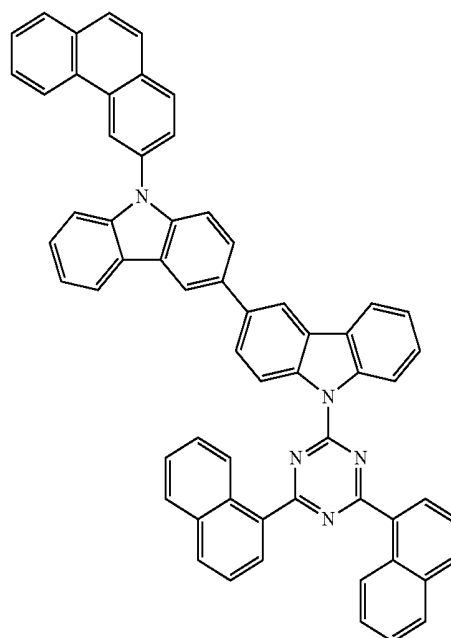
Compound 41



Compound 40



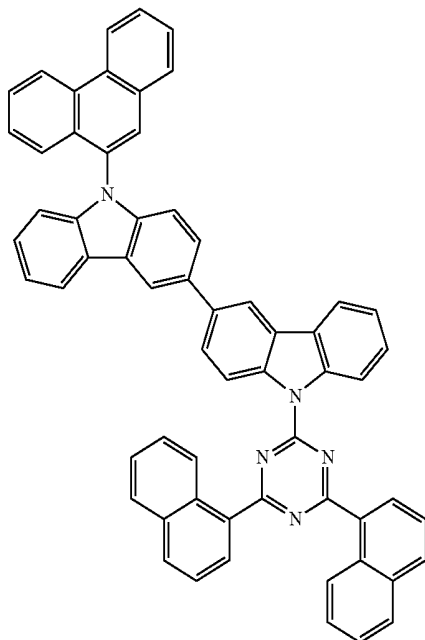
Compound 42



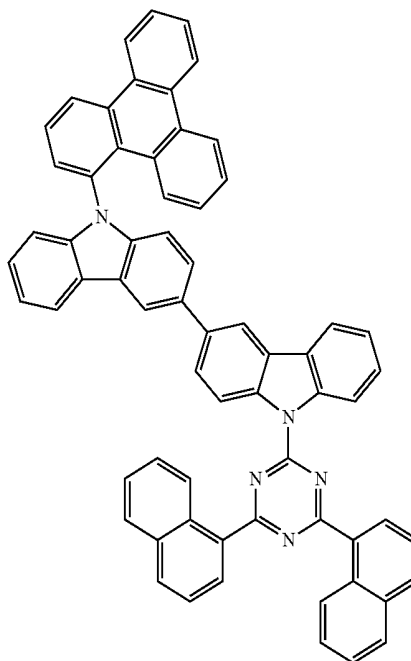
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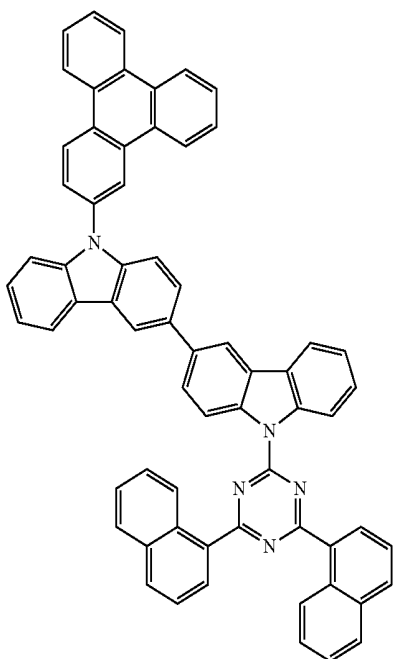
Compound 43



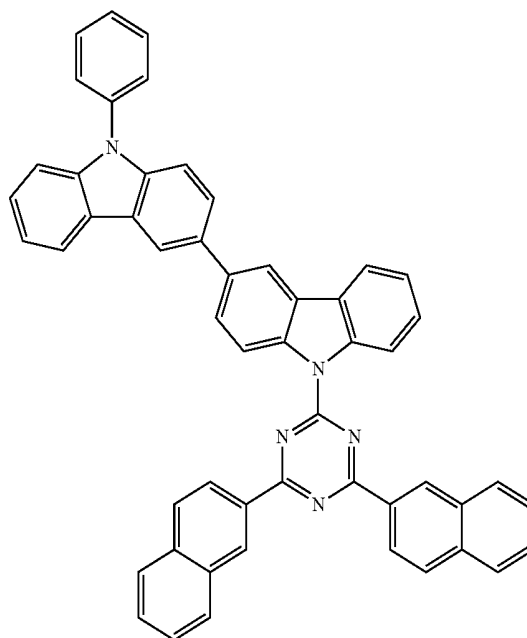
Compound 45



Compound 44



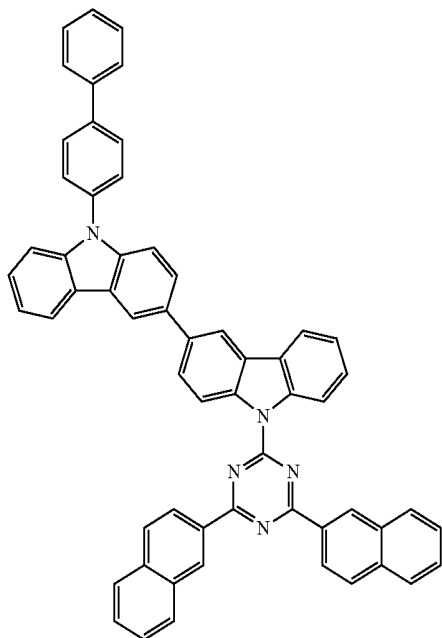
Compound 46



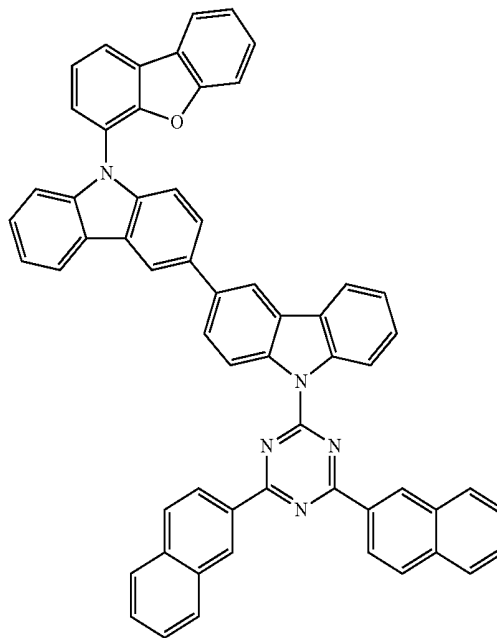
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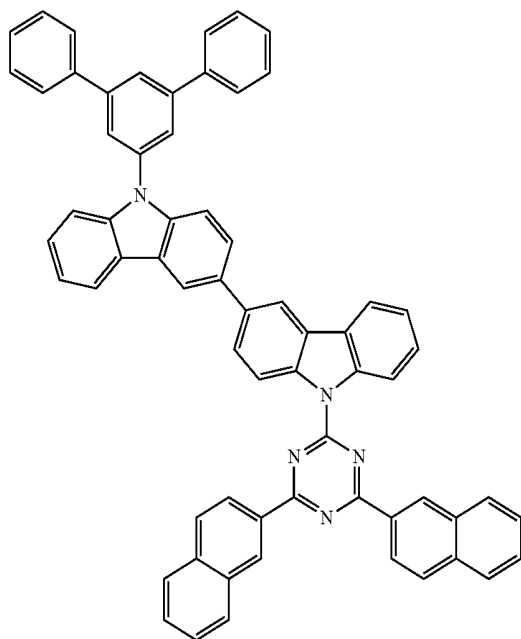
Compound 47



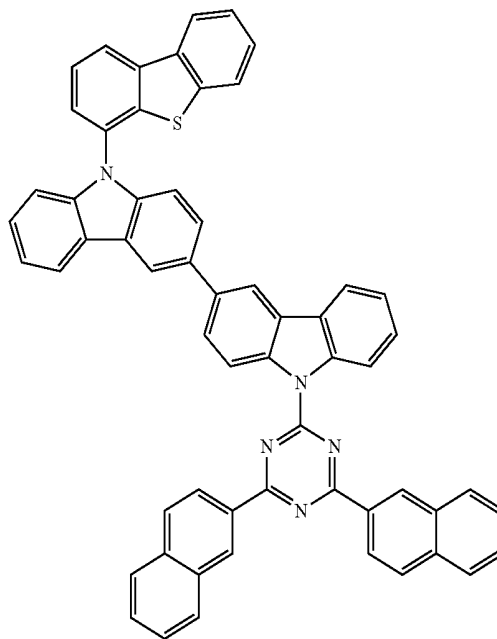
Compound 49



Compound 48



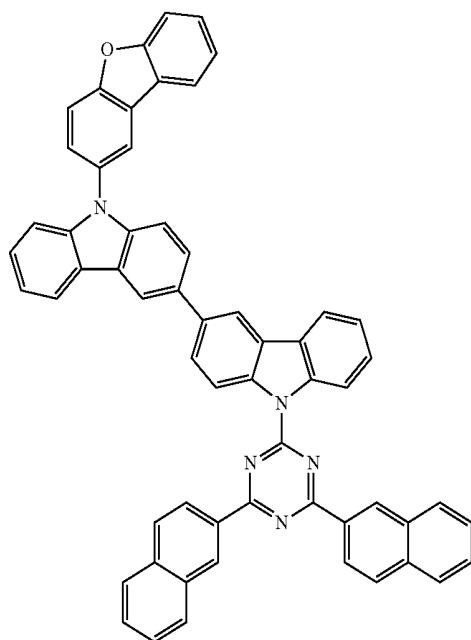
Compound 50



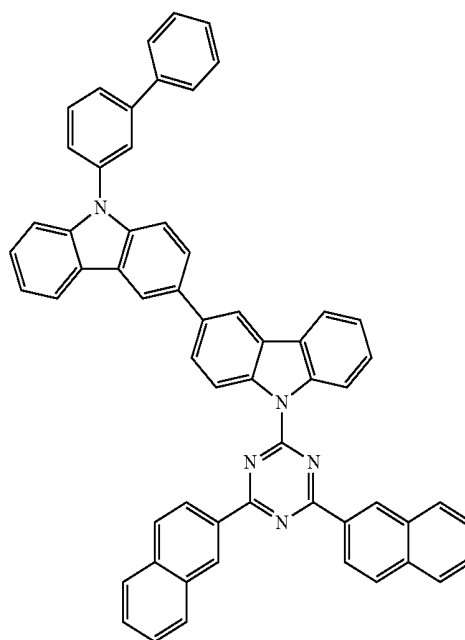
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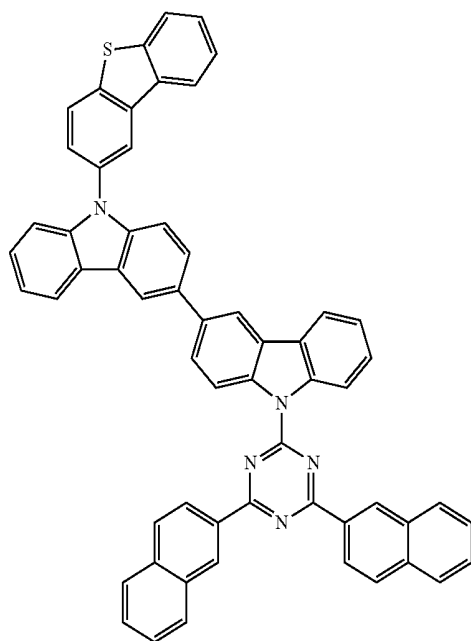
Compound 51



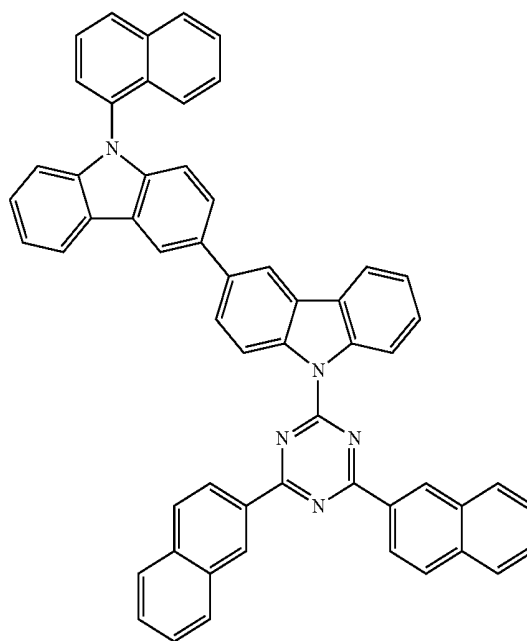
Compound 53



Compound 52



Compound 54

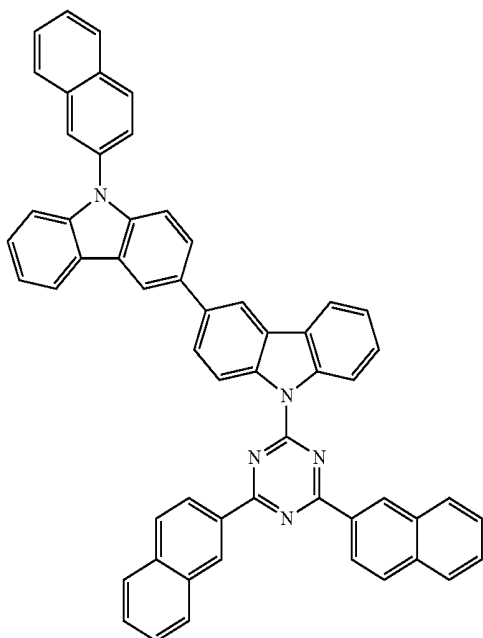




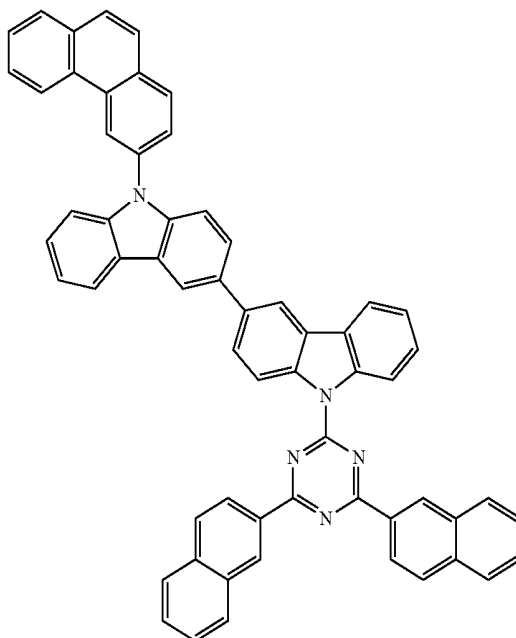
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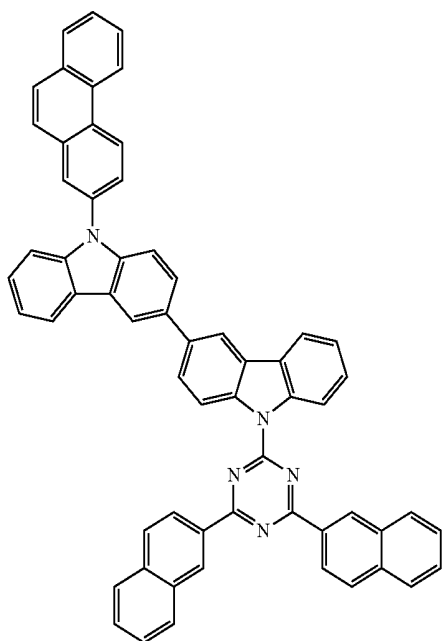
Compound 55



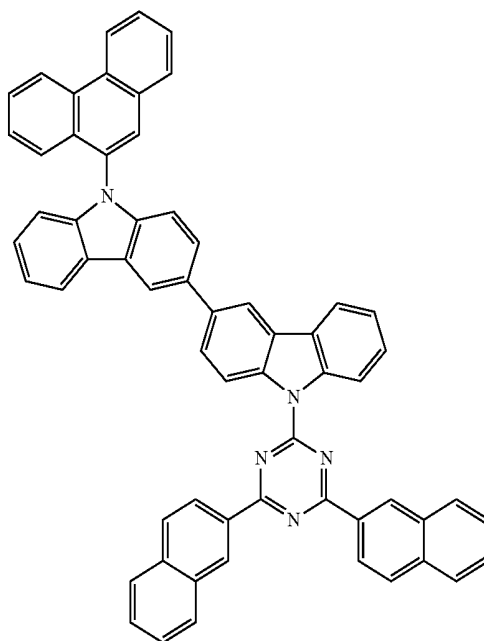
Compound 57



Compound 56



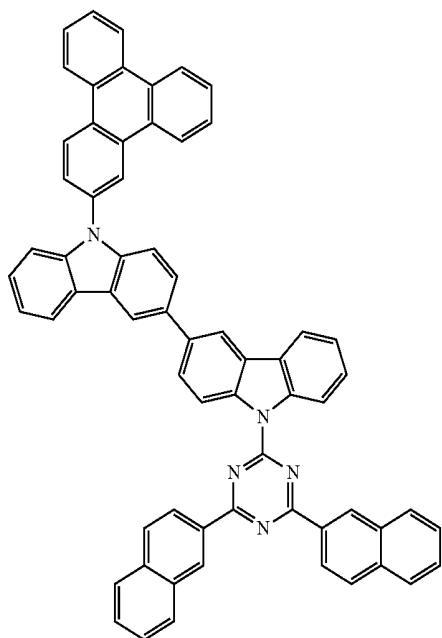
Compound 58



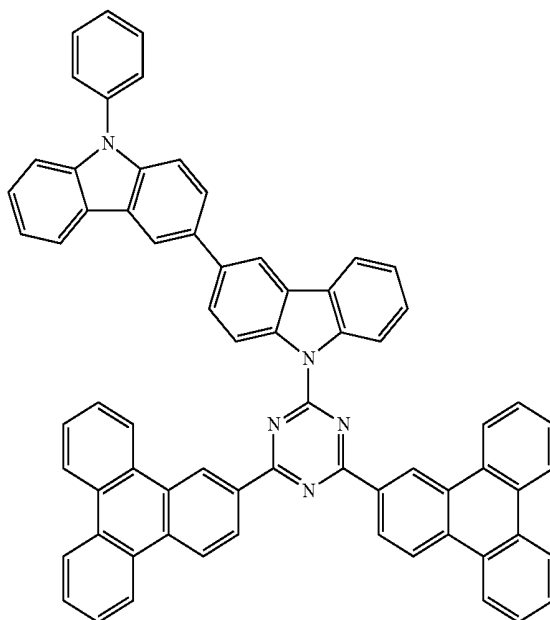
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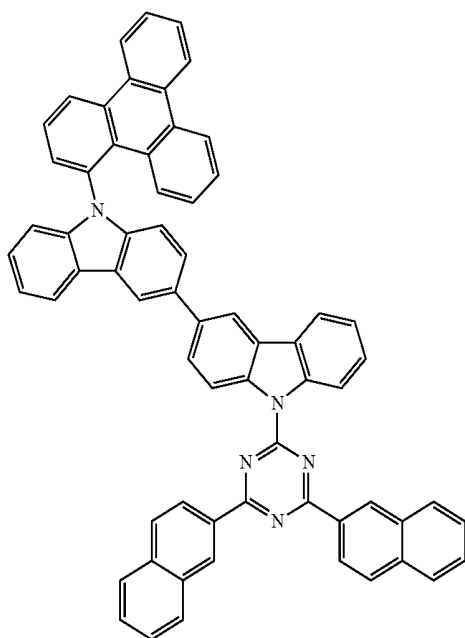
Compound 59



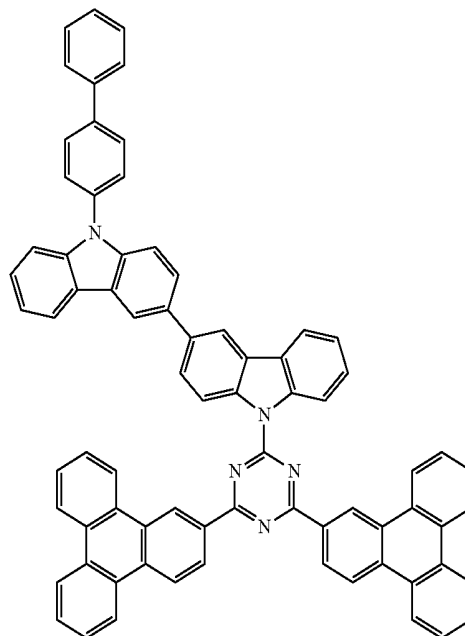
Compound 61



Compound 60



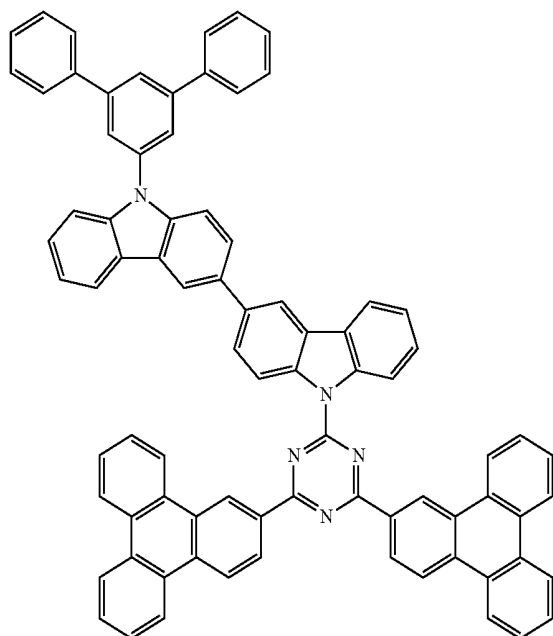
Compound 62



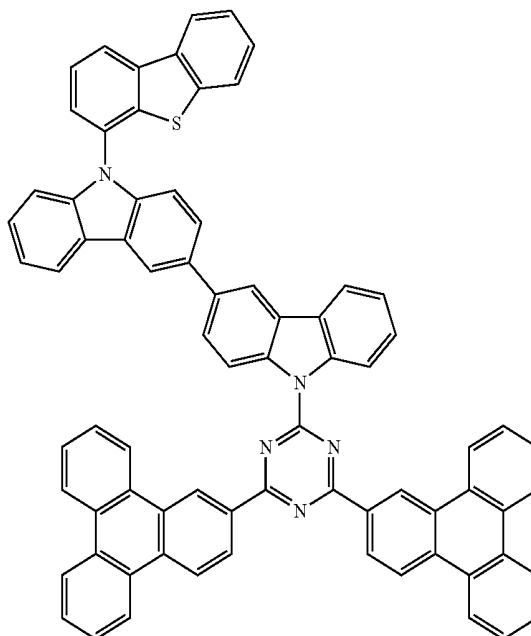
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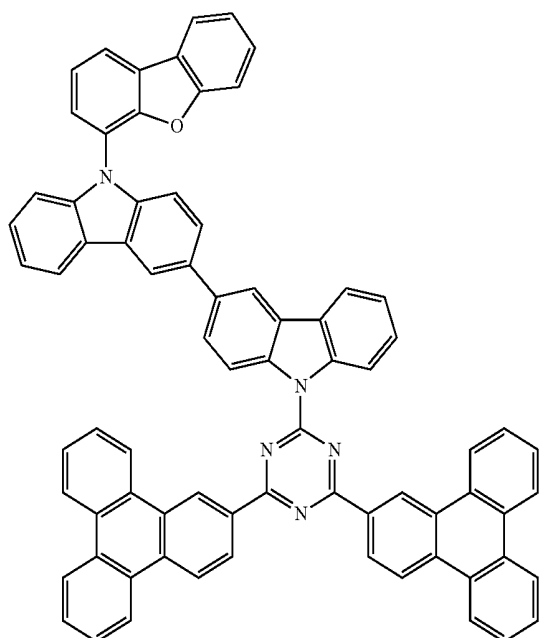
Compound 63



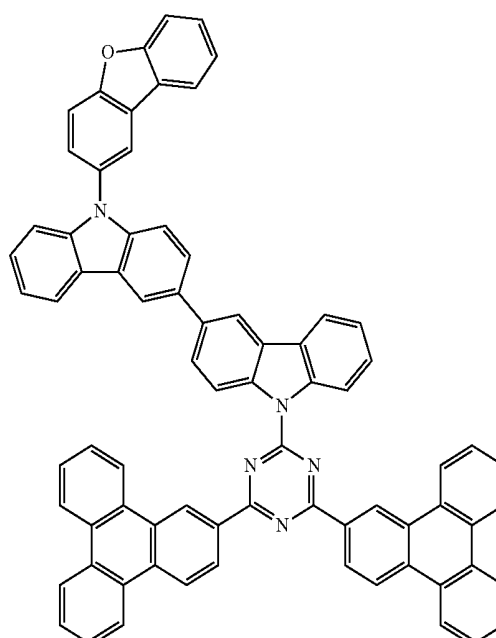
Compound 65



Compound 64



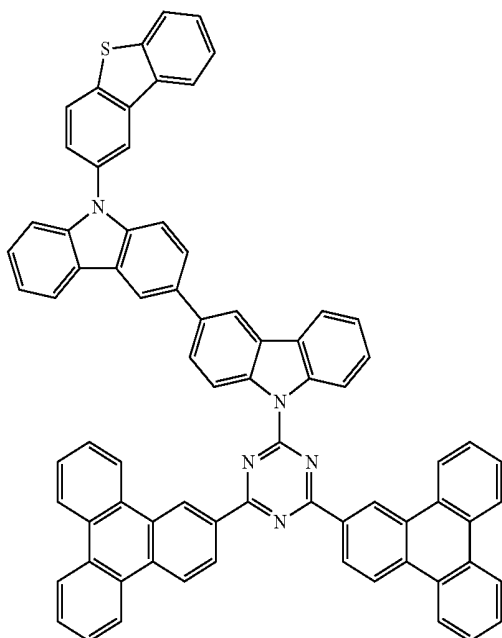
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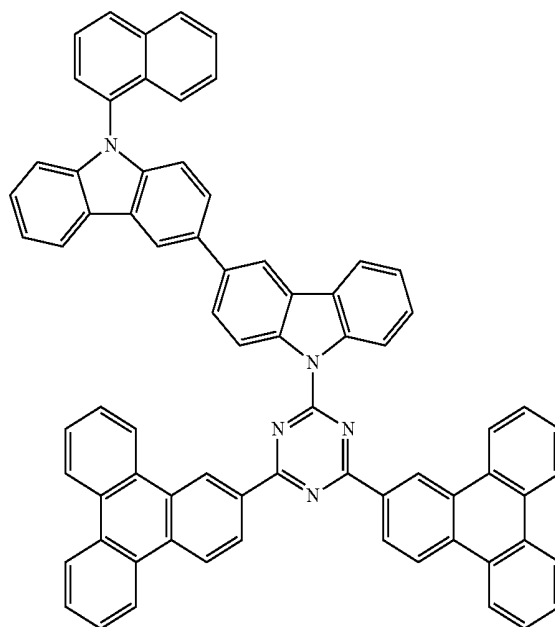
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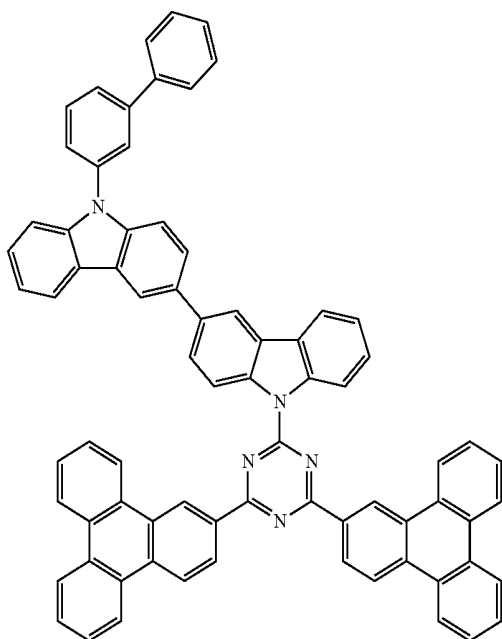
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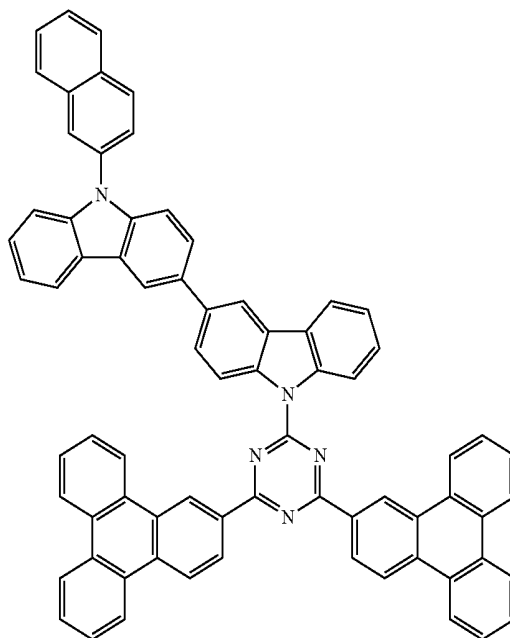
Compound 69



Compound 68



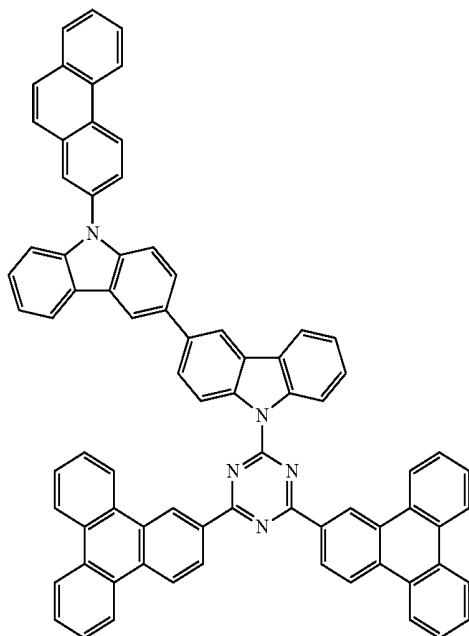
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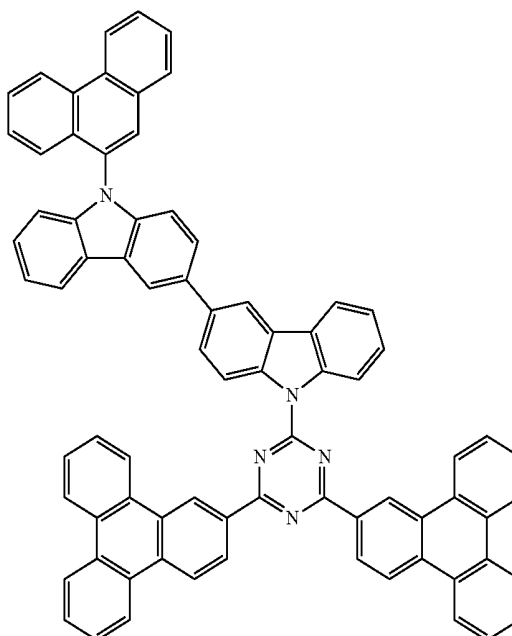
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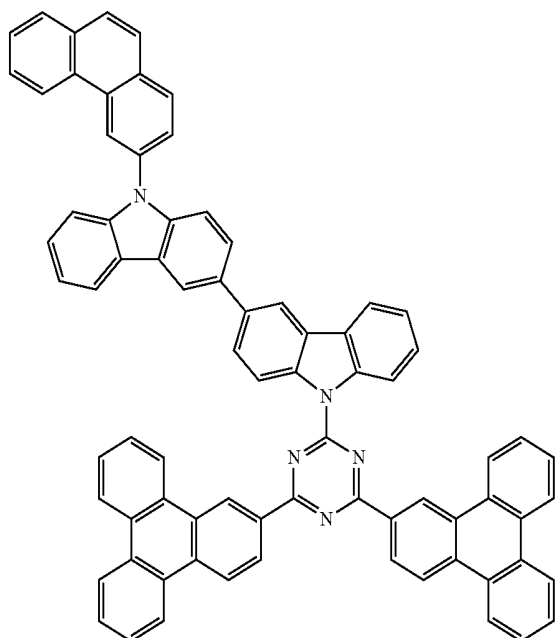
Compound 71



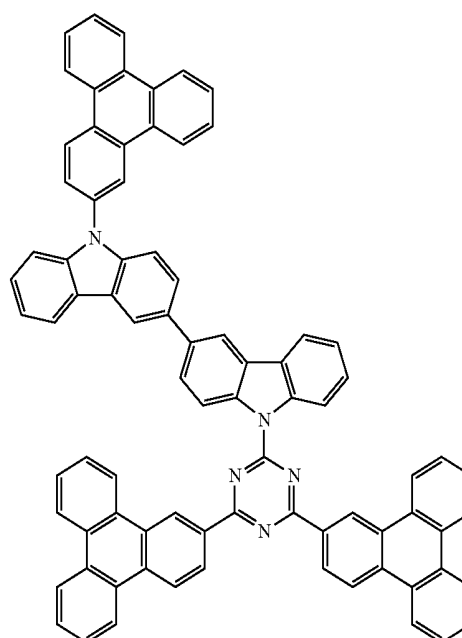
Compound 73



Compound 72



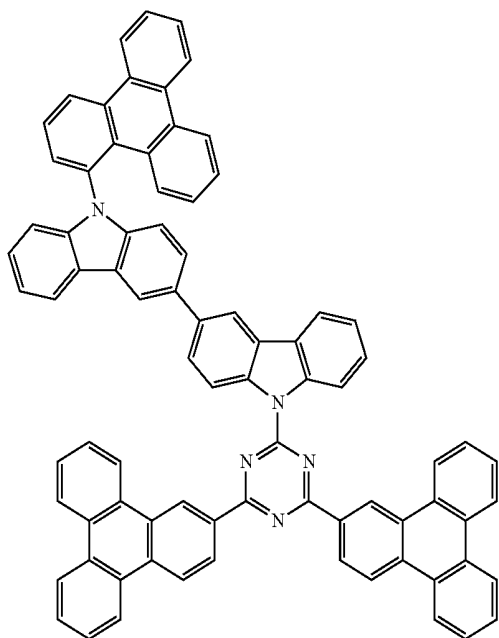
Compound 74



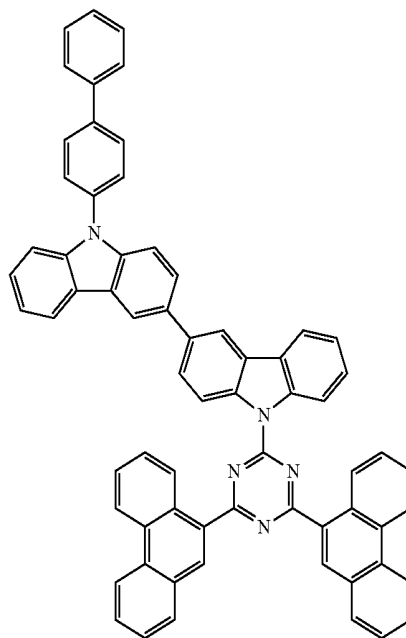
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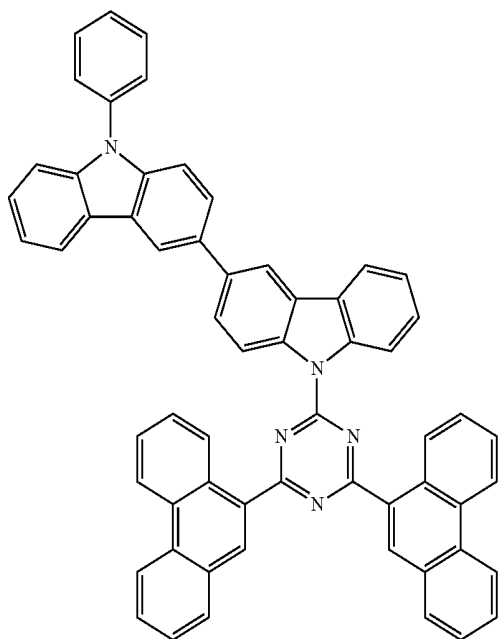
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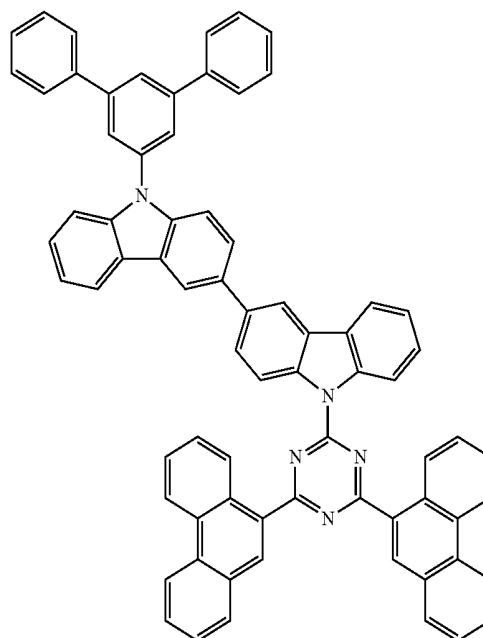
Compound 77



Compound 76



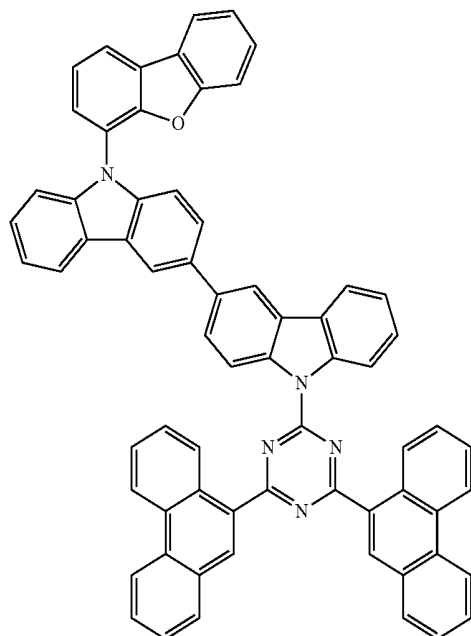
Compound 78



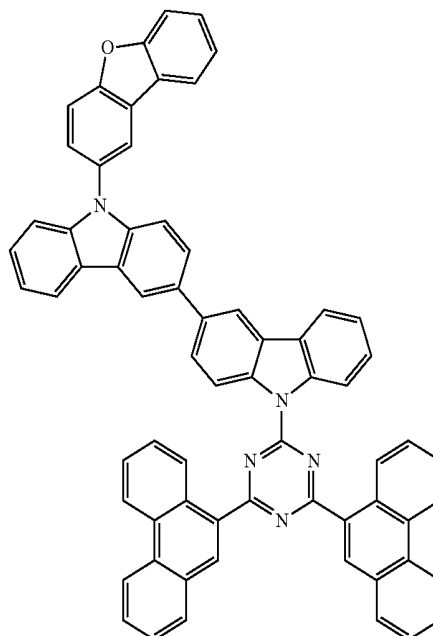
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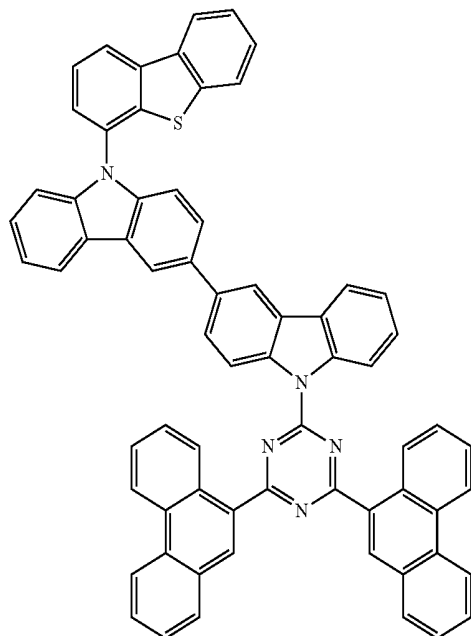
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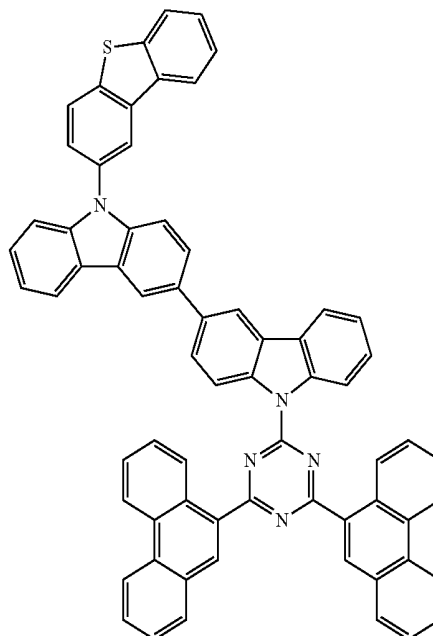
Compound 81



Compound 80



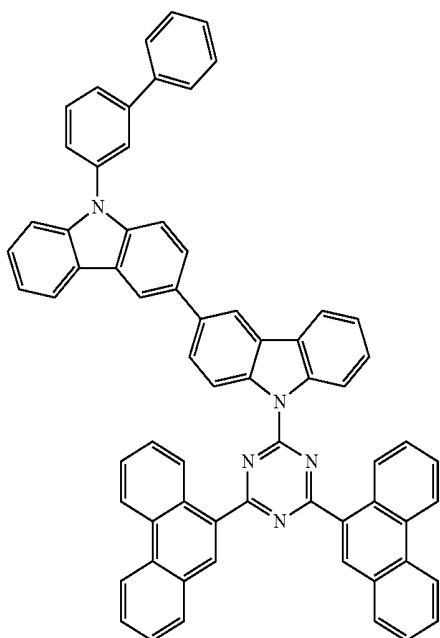
Compound 82



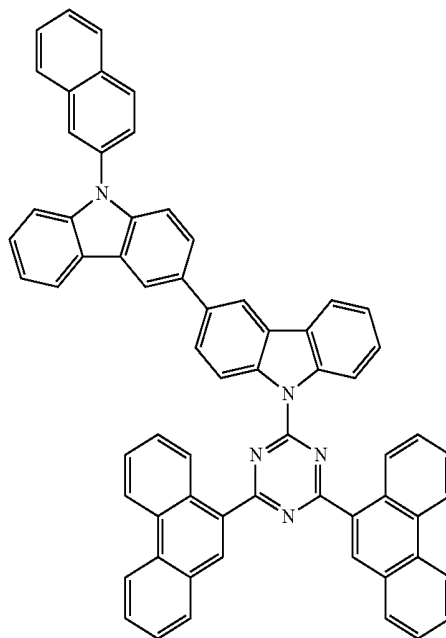
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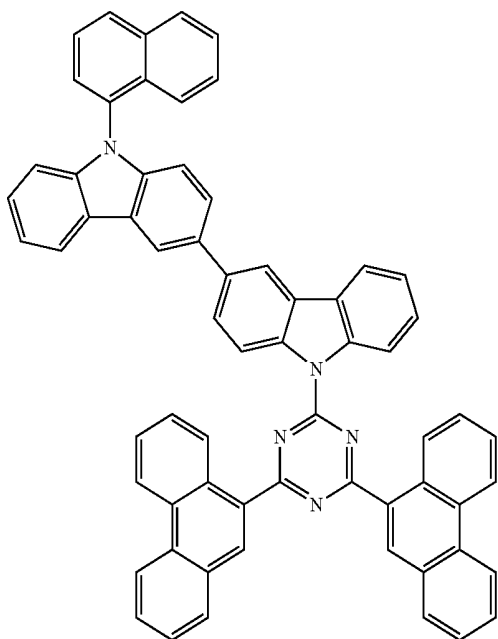
Compound 83



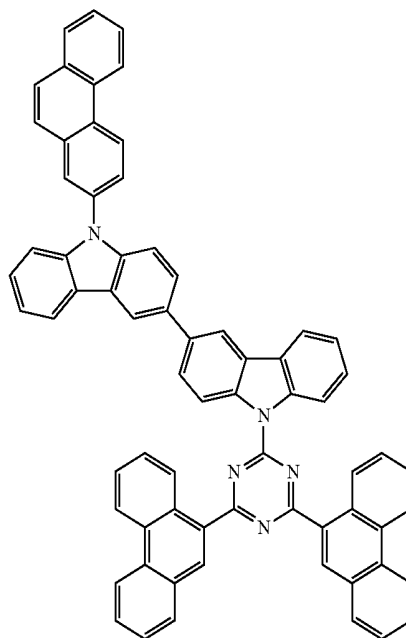
Compound 85



Compound 84



Compound 86

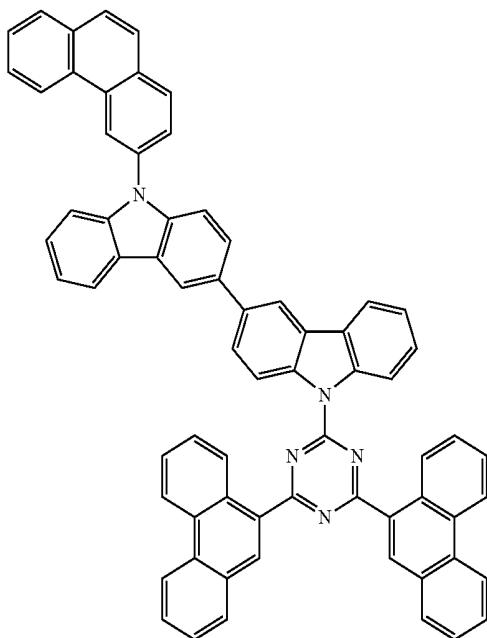




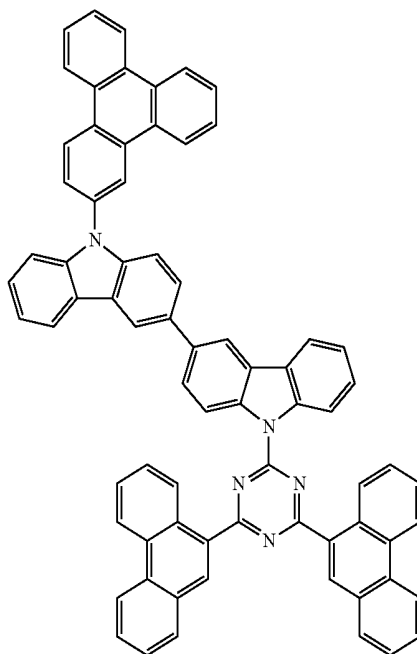
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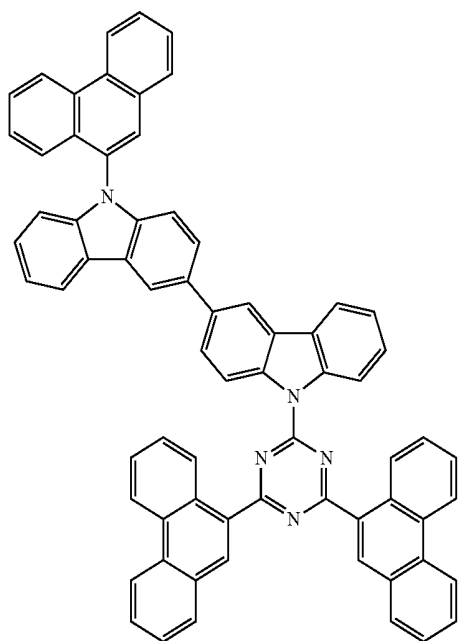
Compound 87



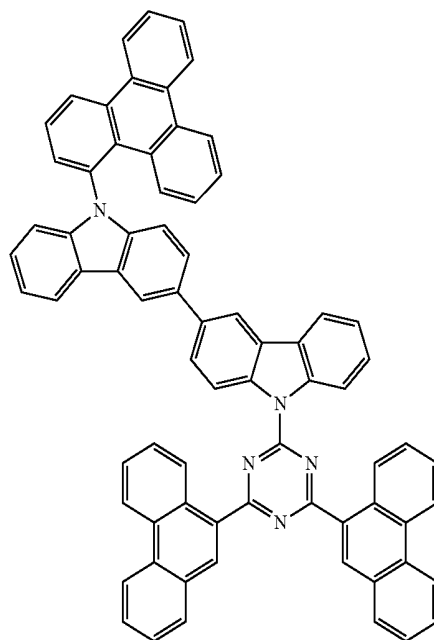
Compound 89



Compound 88



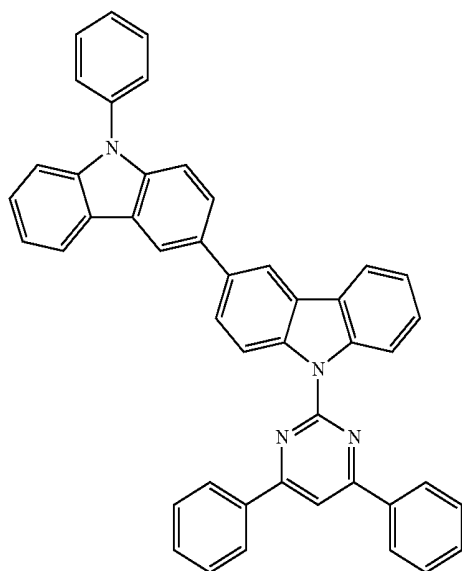
Compound 90



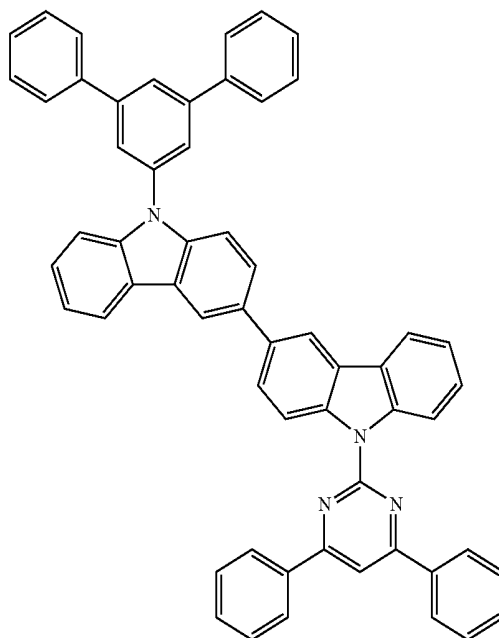
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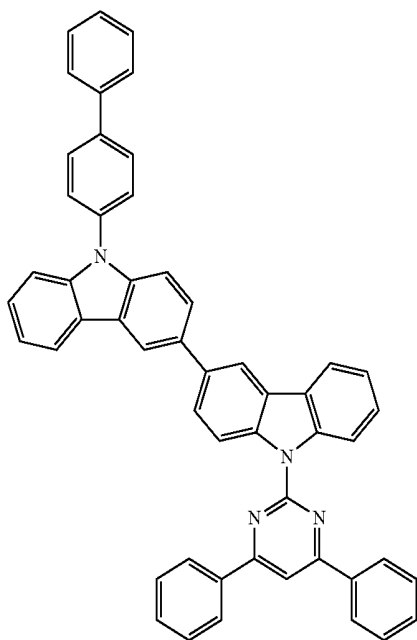
Compound 91



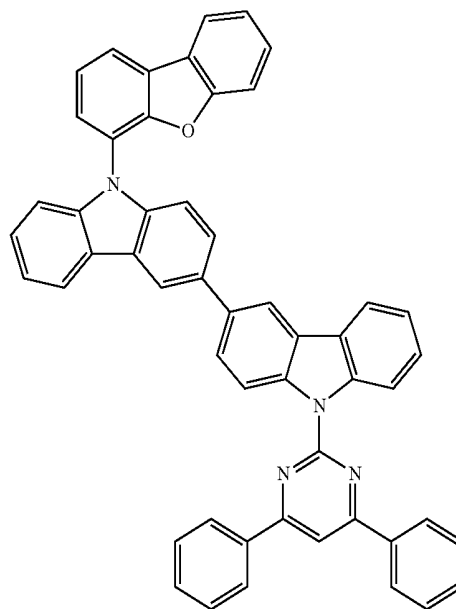
Compound 93



Compound 92



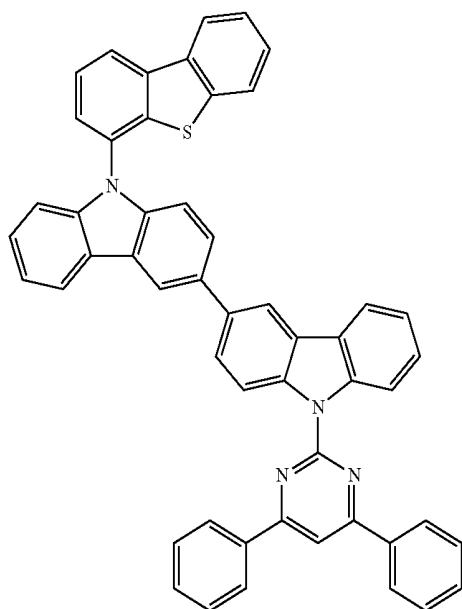
Compound 94



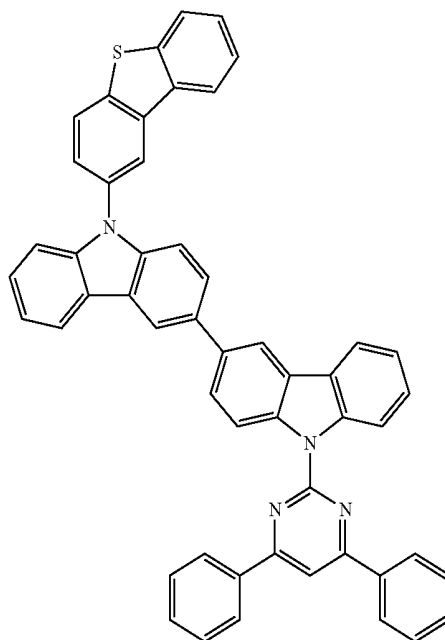
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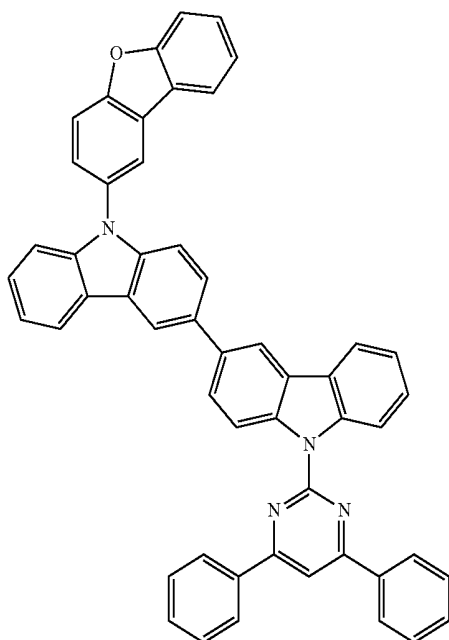
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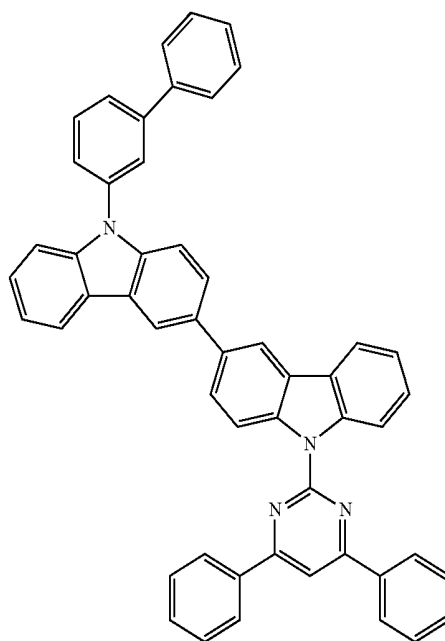
Compound 97



Compound 96



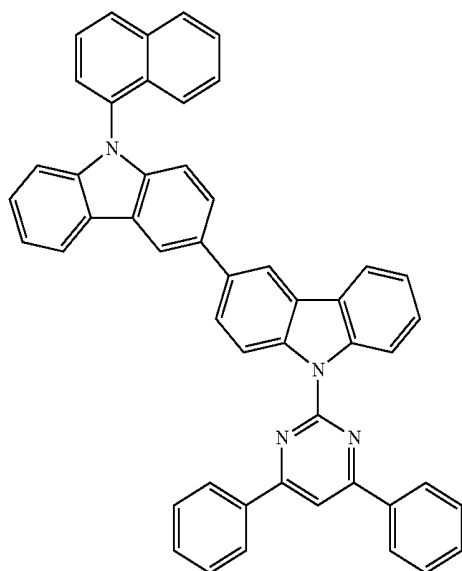
Compound 98



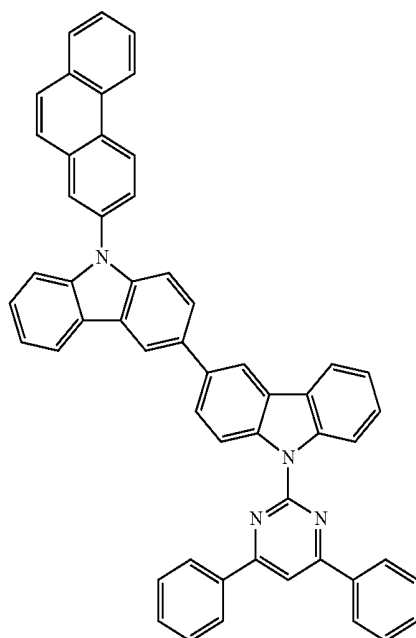
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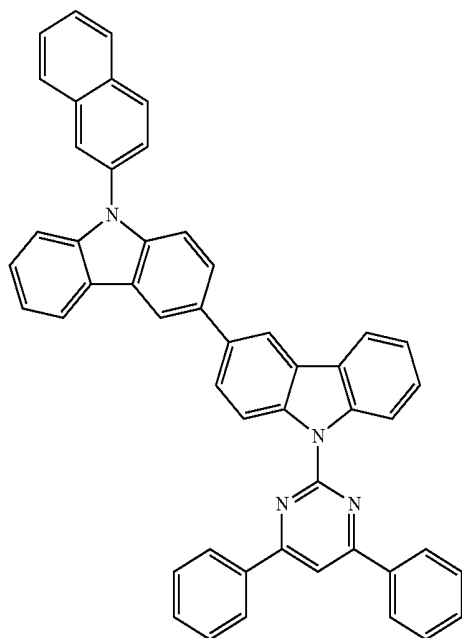
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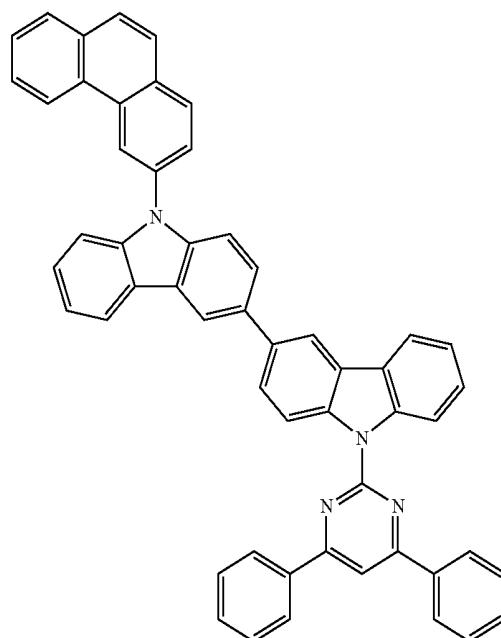
Compound 101



Compound 100



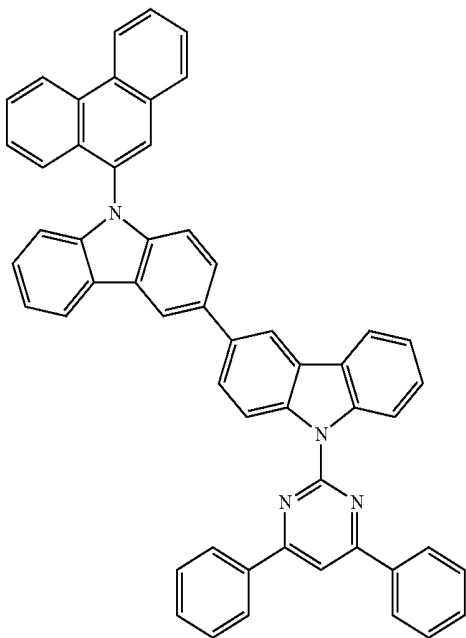
Compound 102



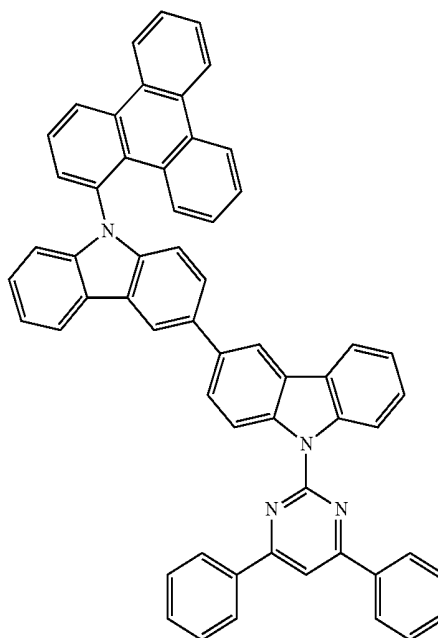
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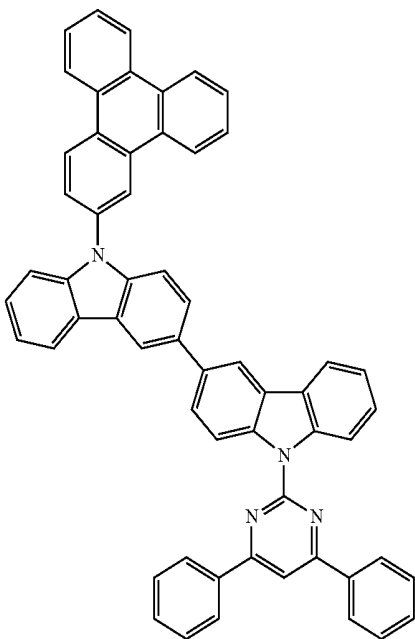
Compound 103



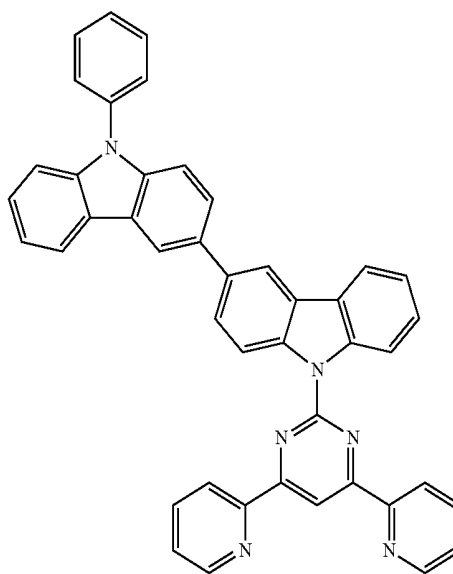
Compound 105



Compound 104

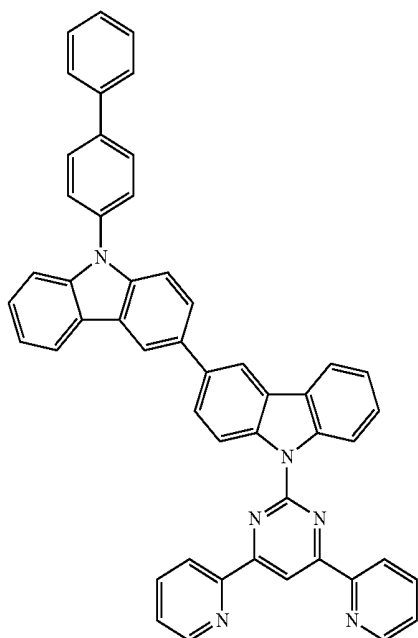


Compound 106



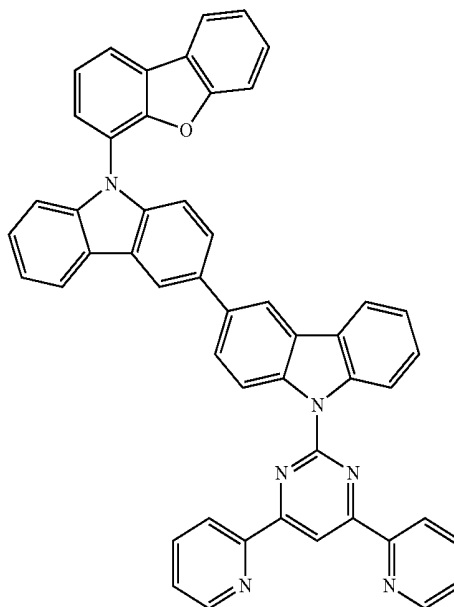
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Compound 107

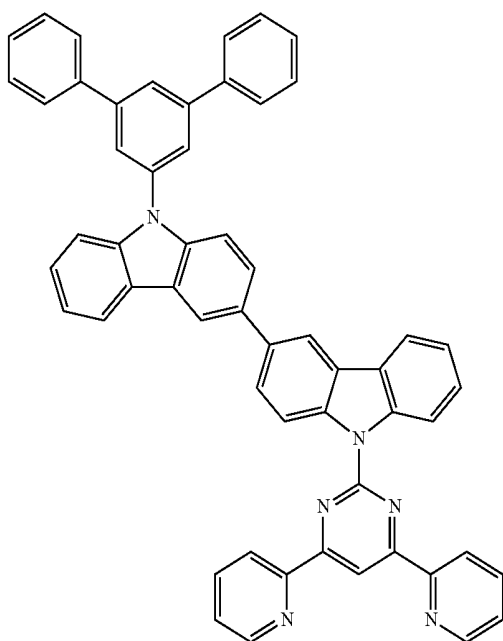


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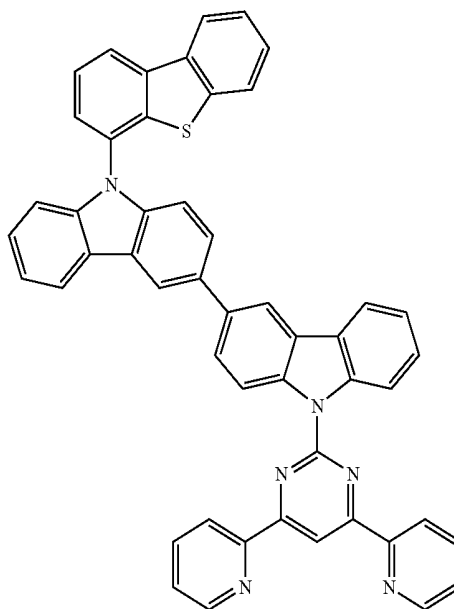
Compound 109



Compound 108



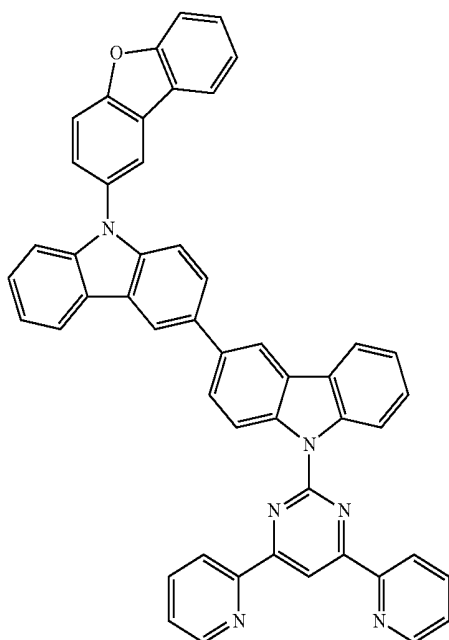
Compound 110



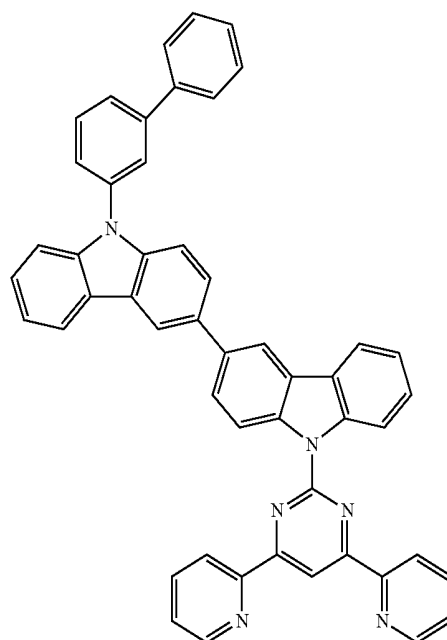
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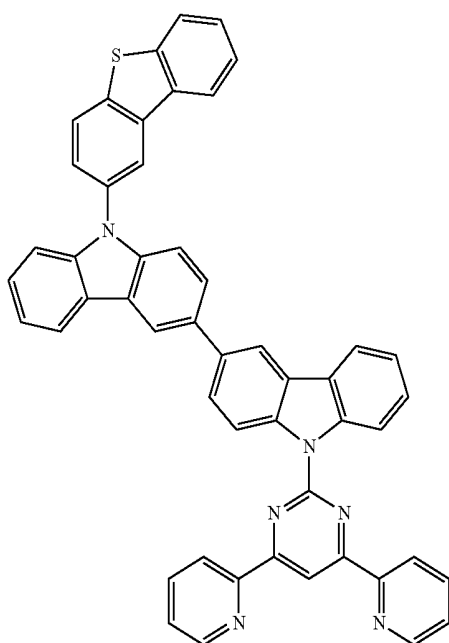
Compound 111



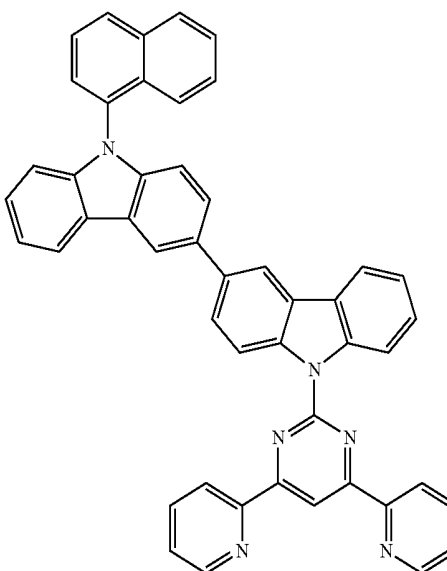
Compound 113



Compound 112



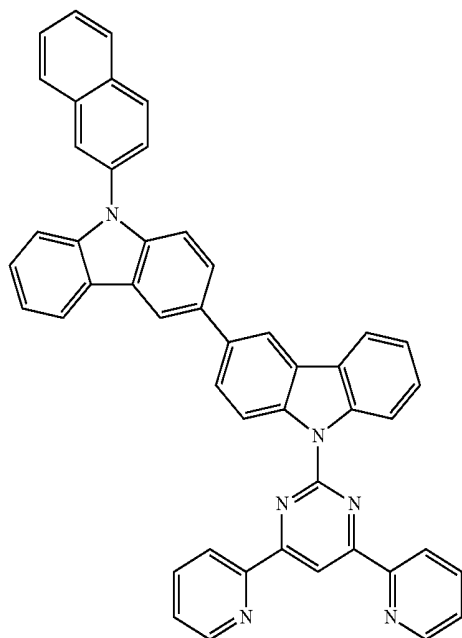
Compound 114



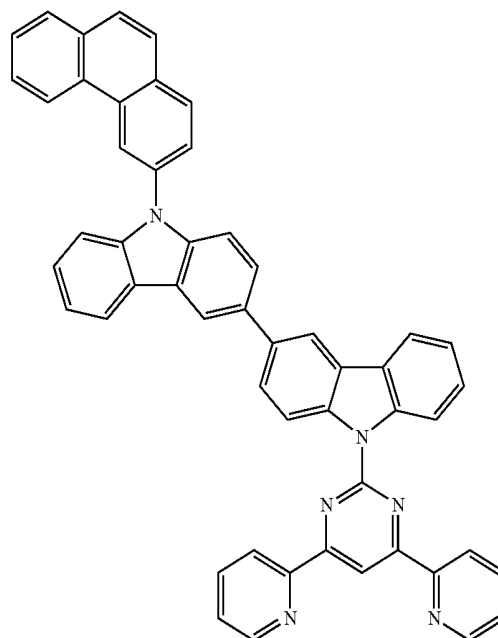
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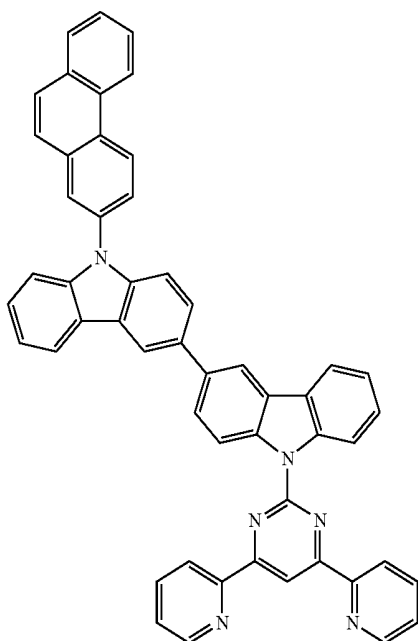
Compound 115



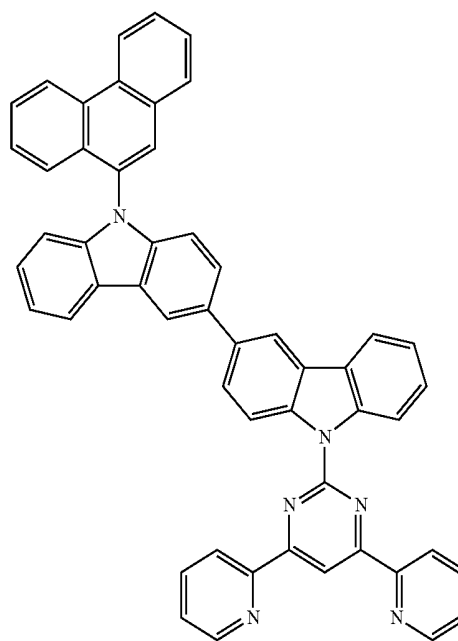
Compound 117



Compound 116



Compound 118

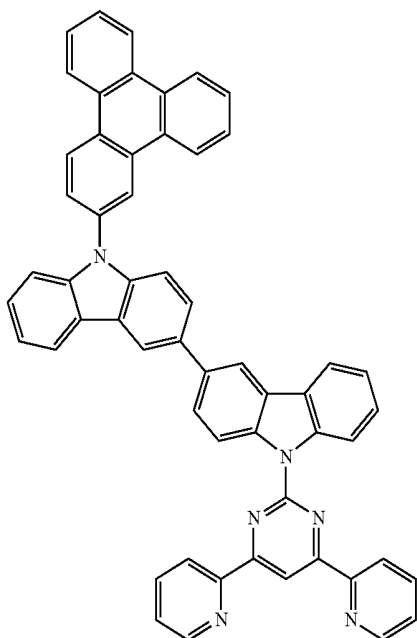




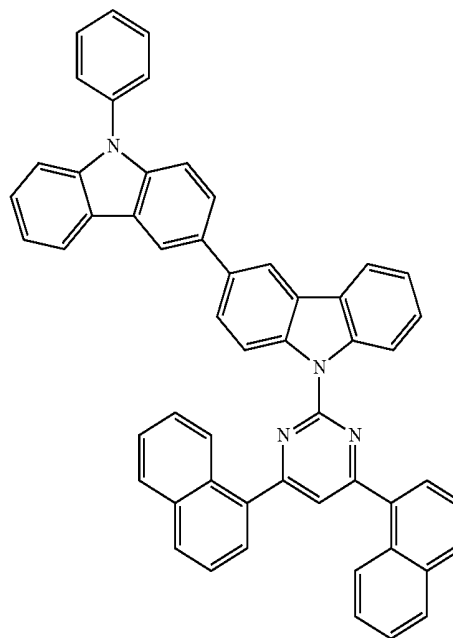
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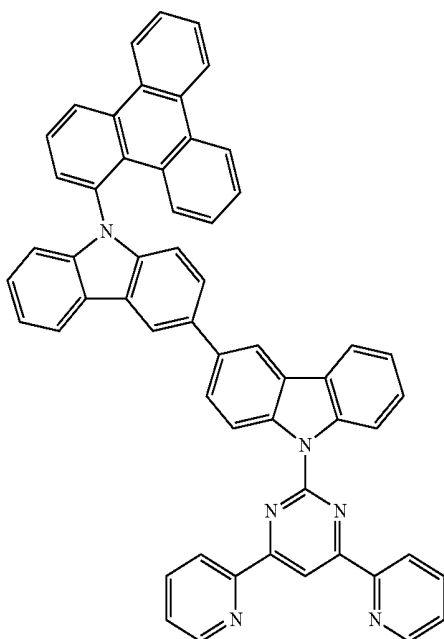
Compound 119



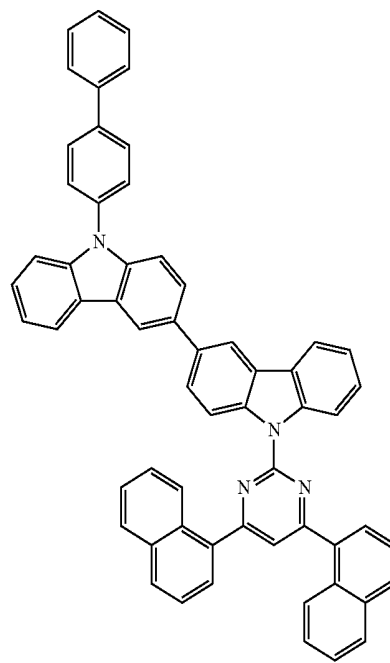
Compound 121



Compound 120



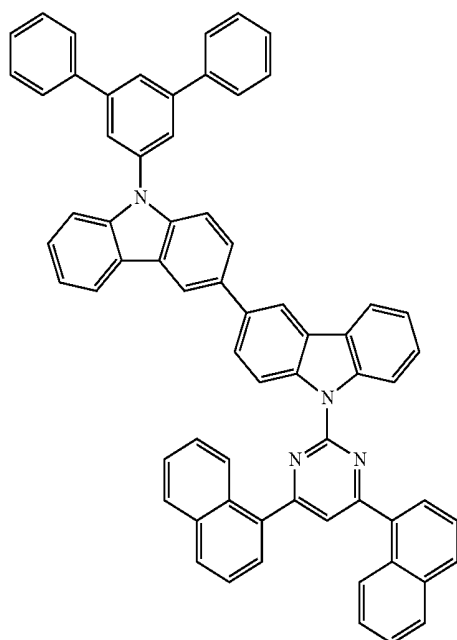
Compound 122



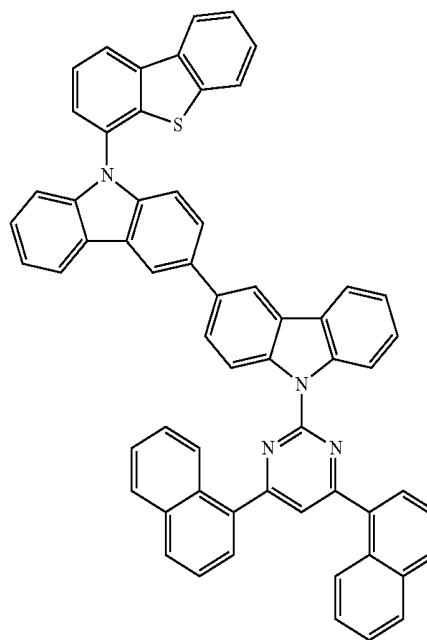
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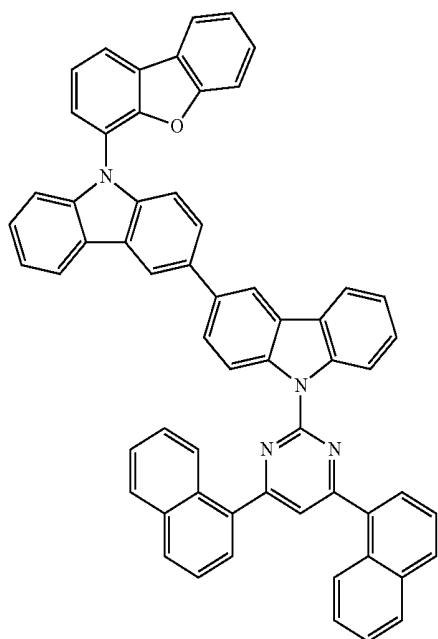
Compound 123



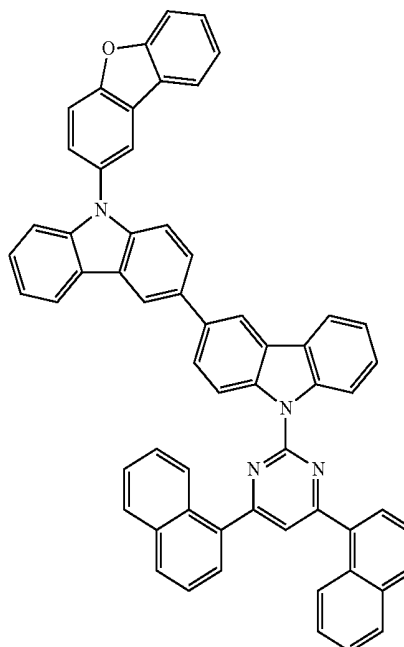
Compound 125



Compound 124



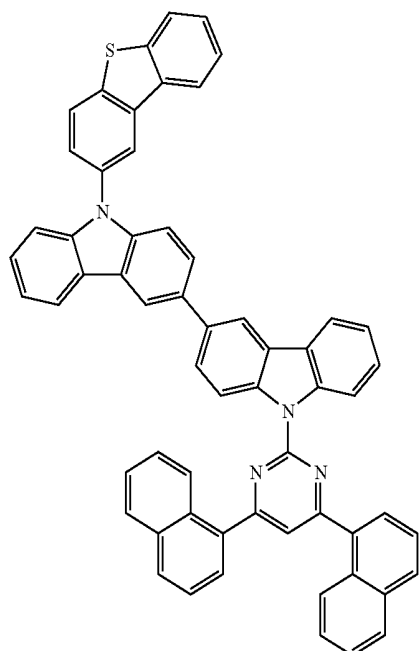
Compound 126



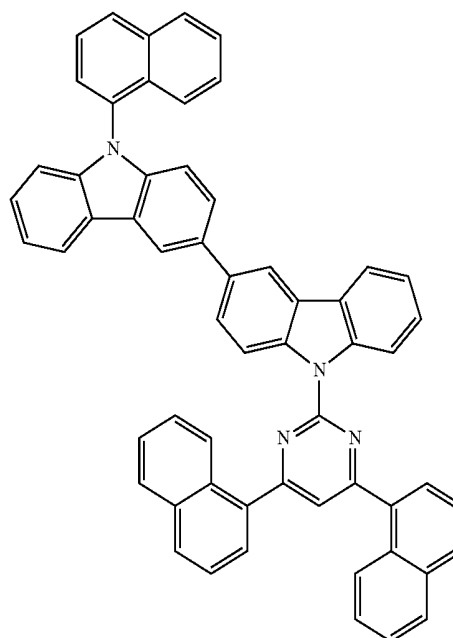
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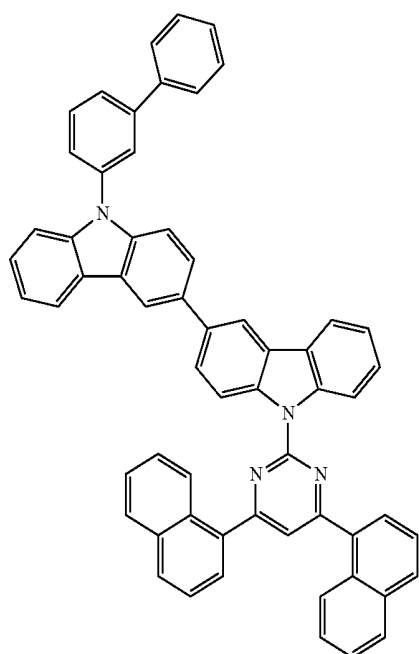
Compound 127



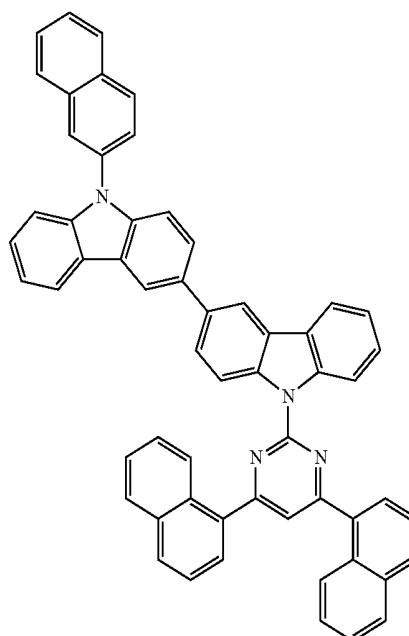
Compound 129



Compound 128



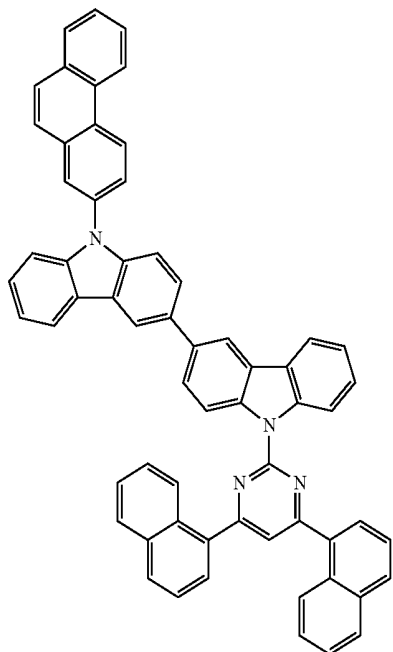
Compound 130



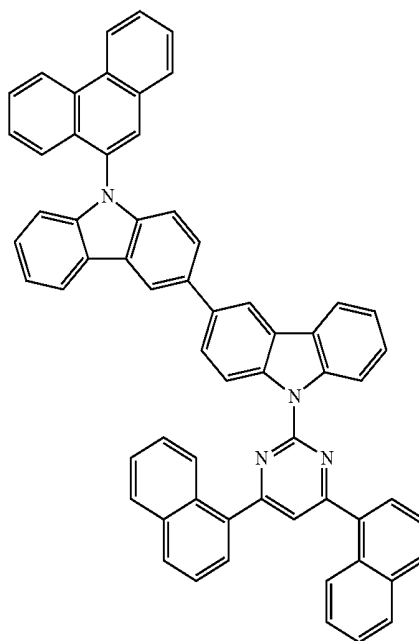
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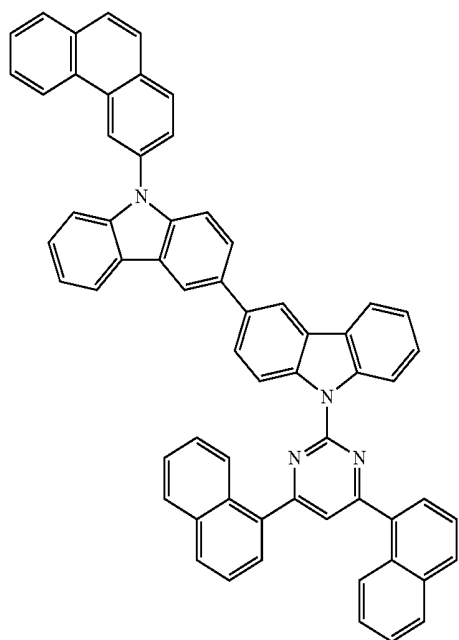
Compound 131



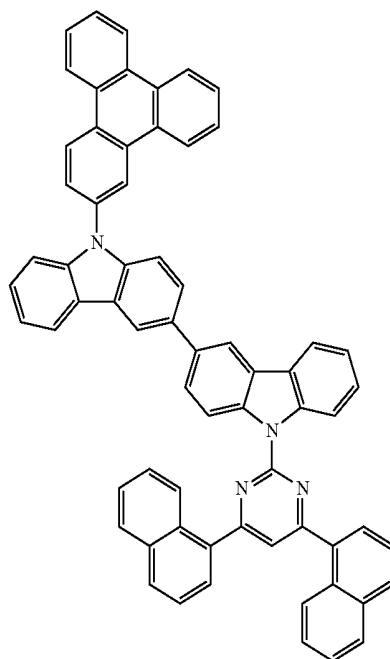
Compound 133



Compound 132



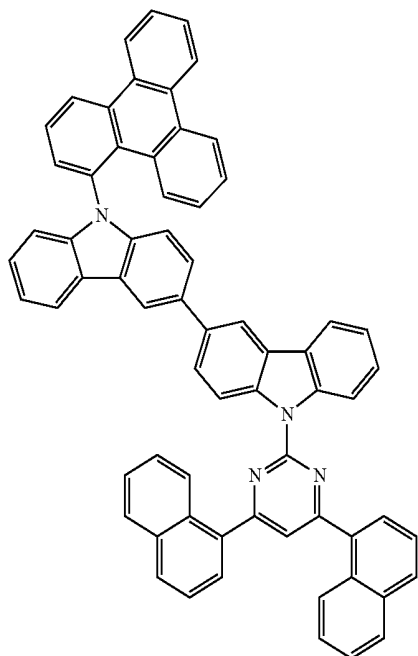
Compound 134



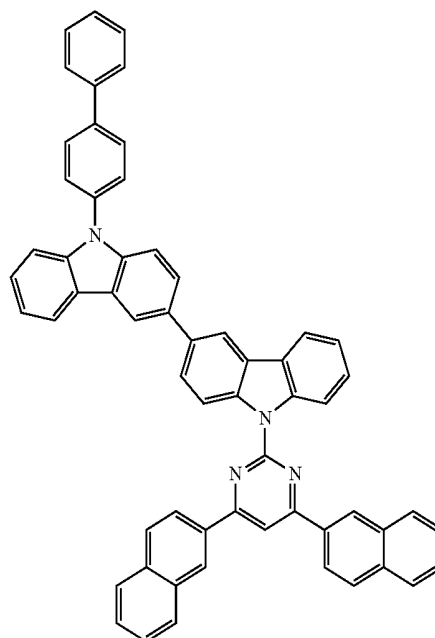
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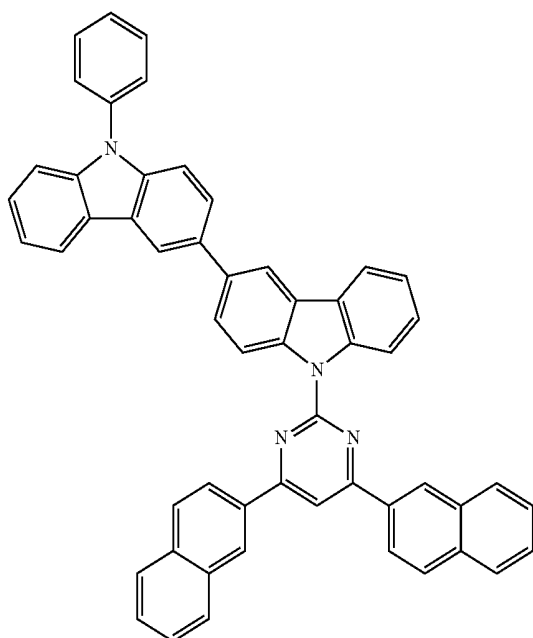
Compound 135



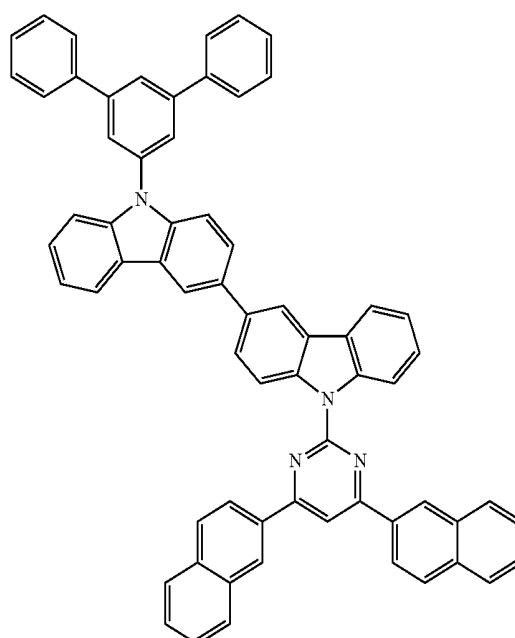
Compound 137



Compound 136



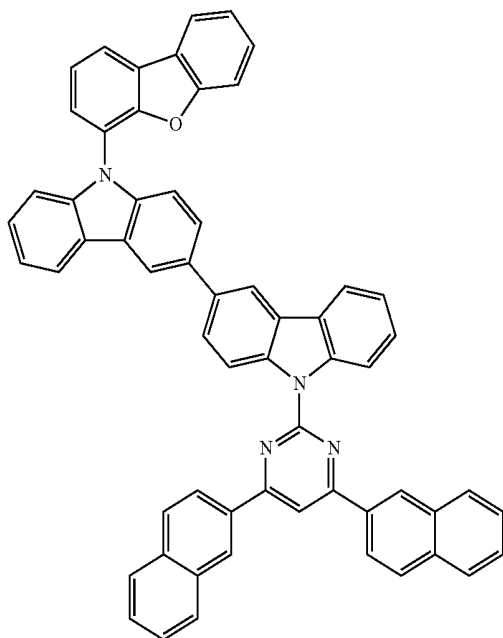
Compound 138



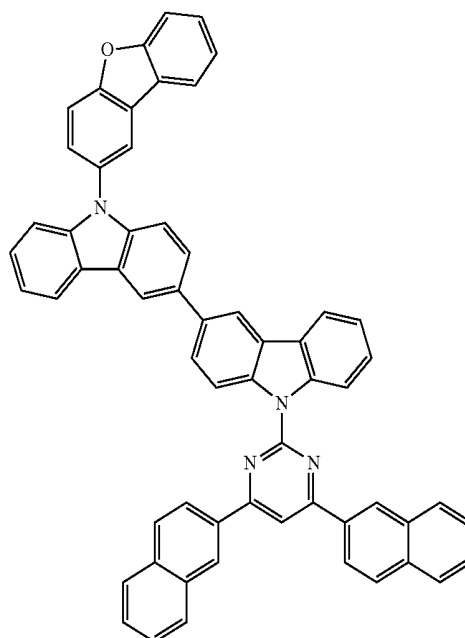
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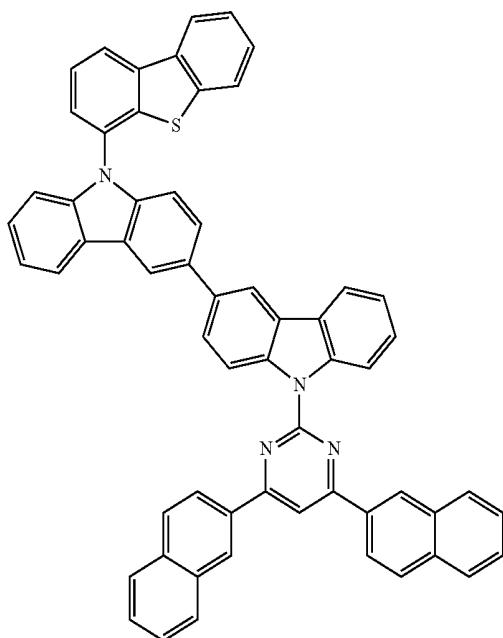
Compound 139



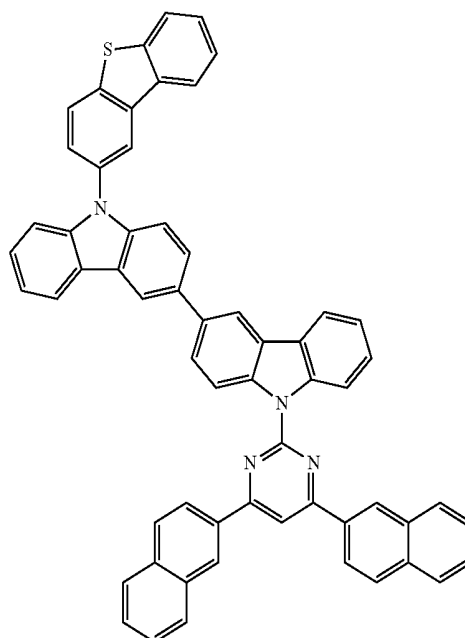
Compound 141



Compound 140



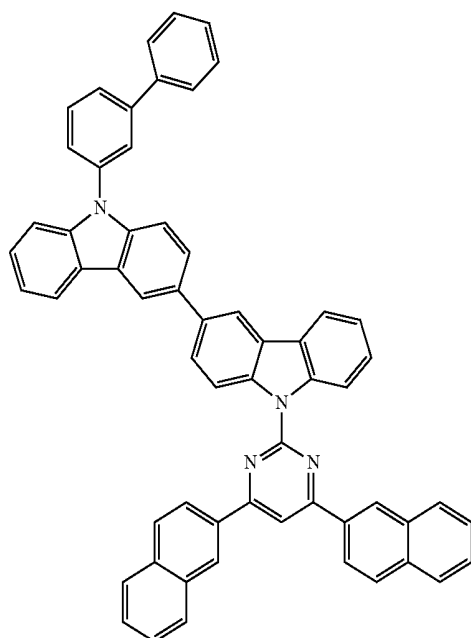
Compound 142



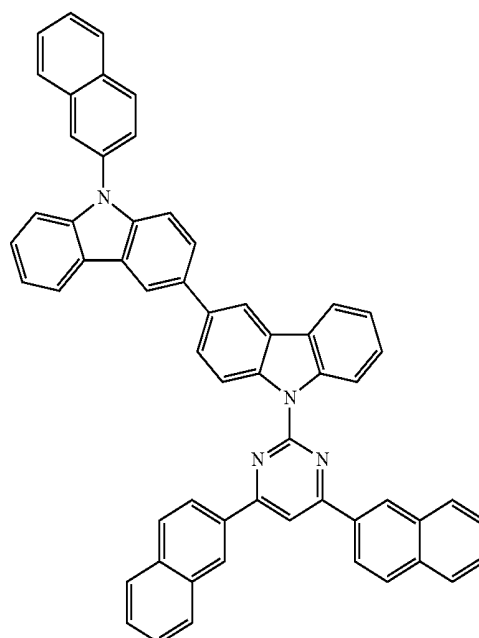
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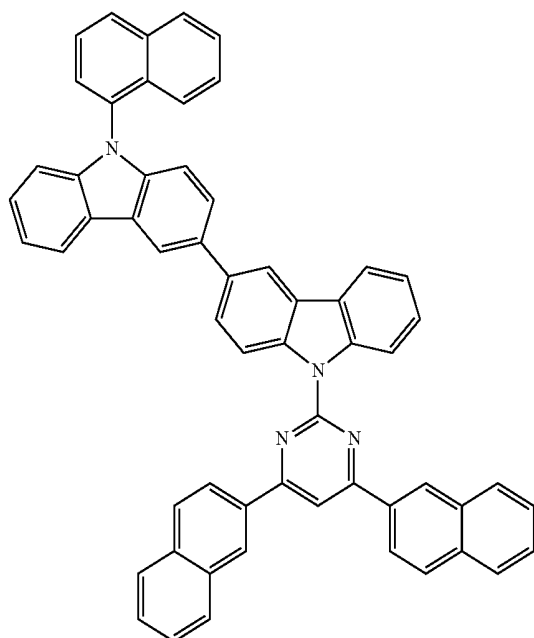
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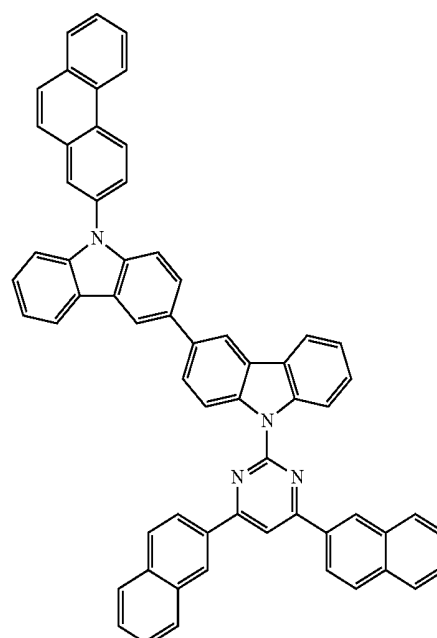
Compound 145



Compound 144



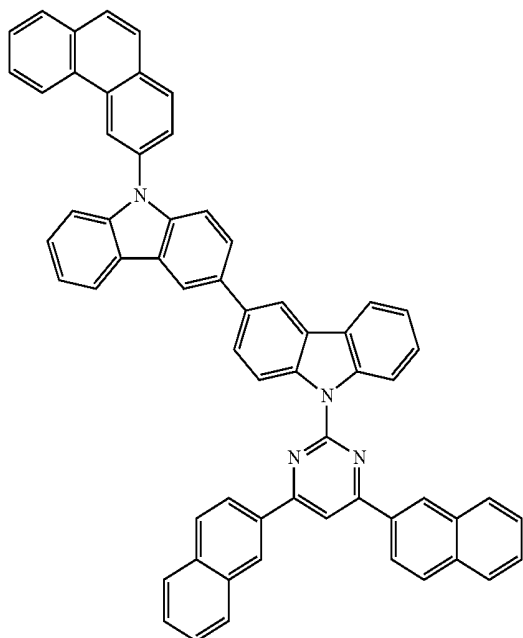
Compound 146



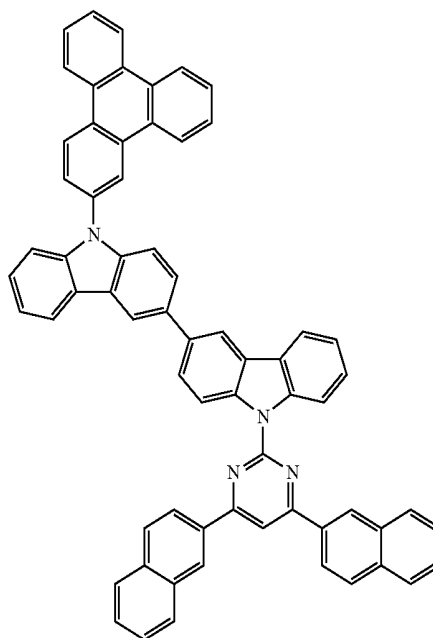
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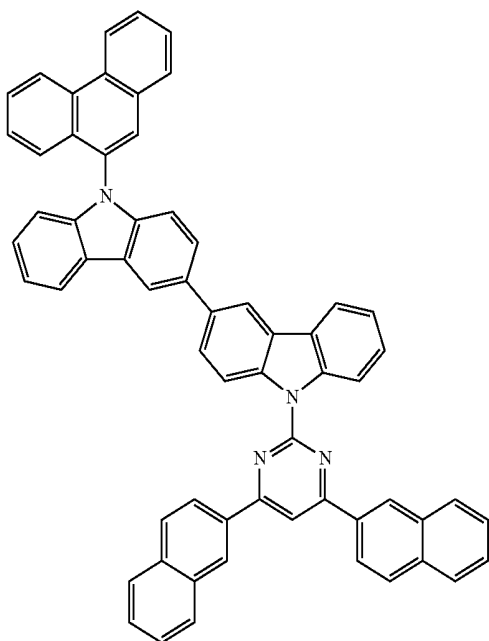
Compound 147



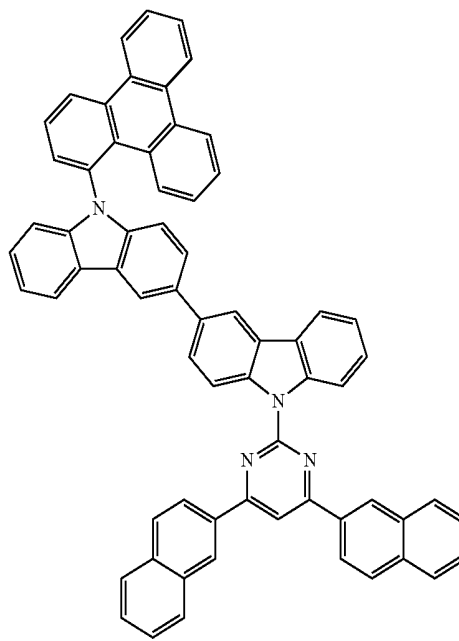
Compound 149



Compound 148



Compound 150

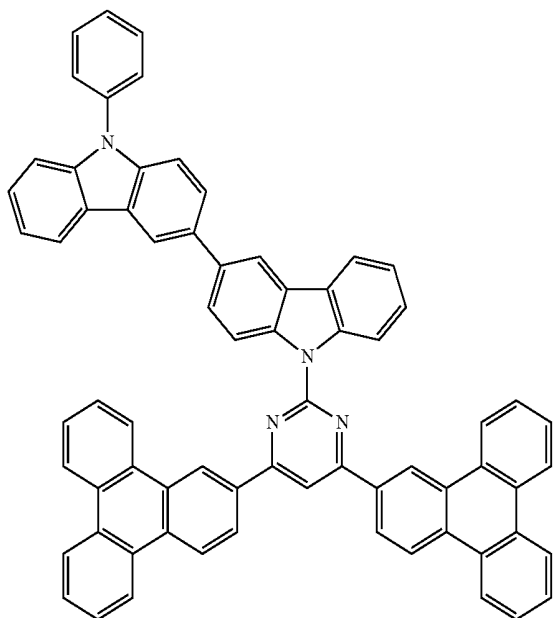




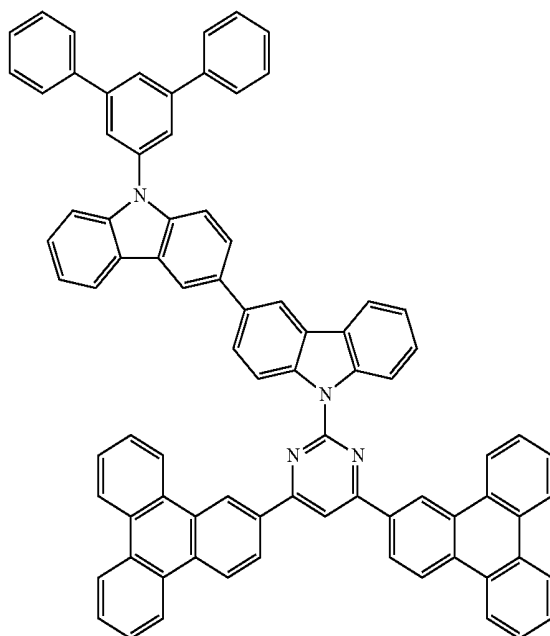
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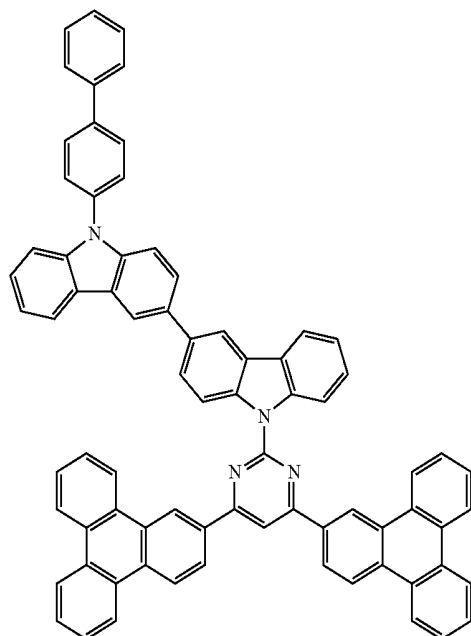
Compound 151



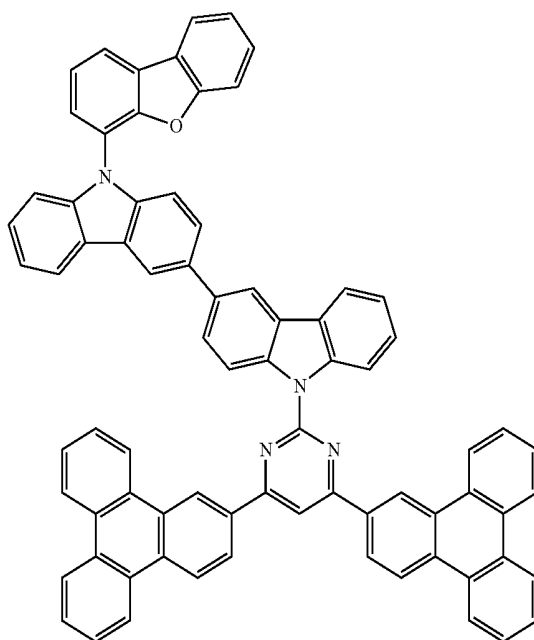
Compound 153



Compound 152



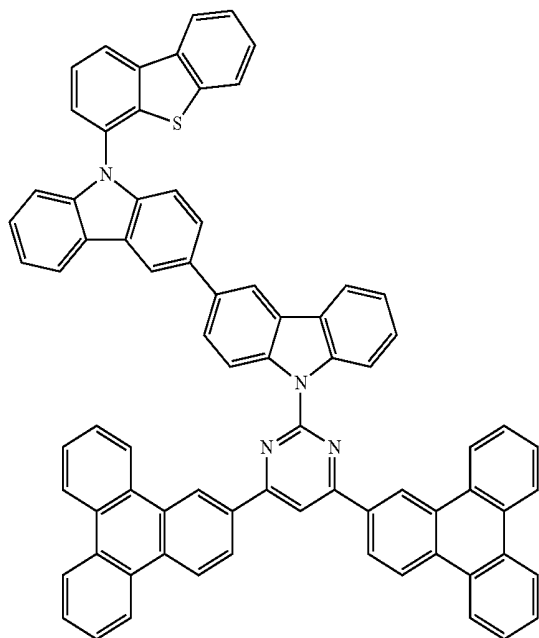
Compound 154



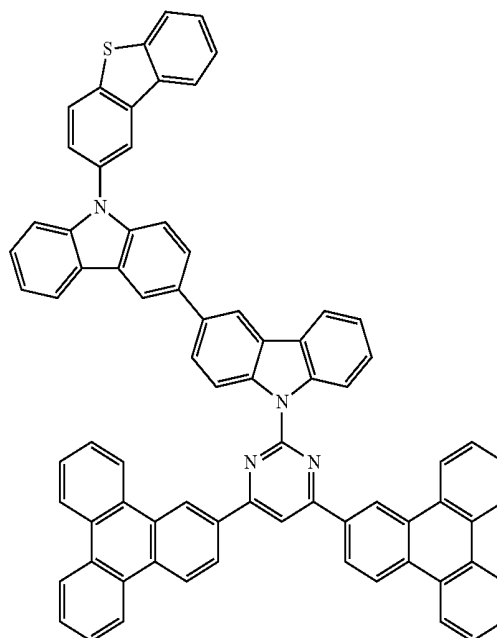
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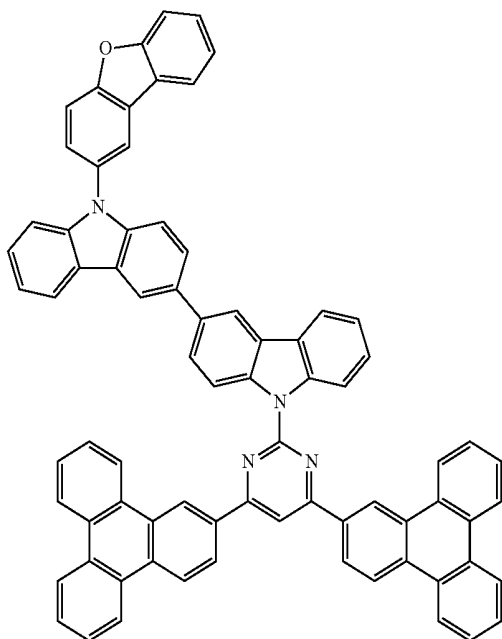
Compound 155



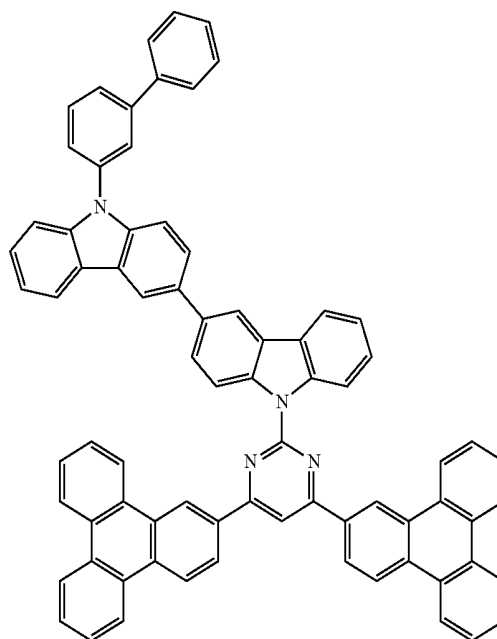
Compound 157



Compound 156



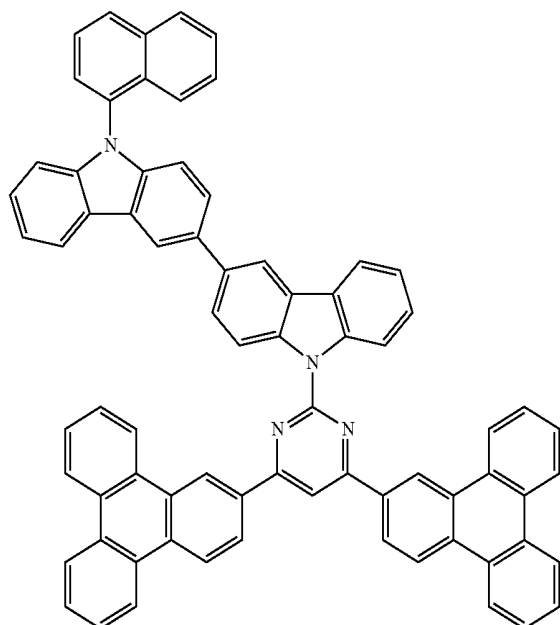
Compound 158



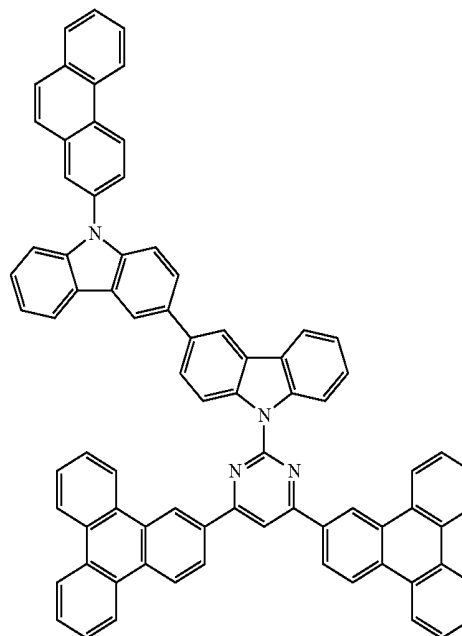
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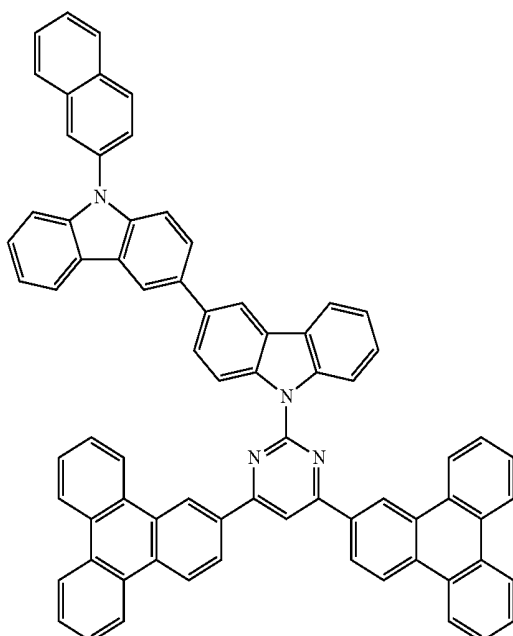
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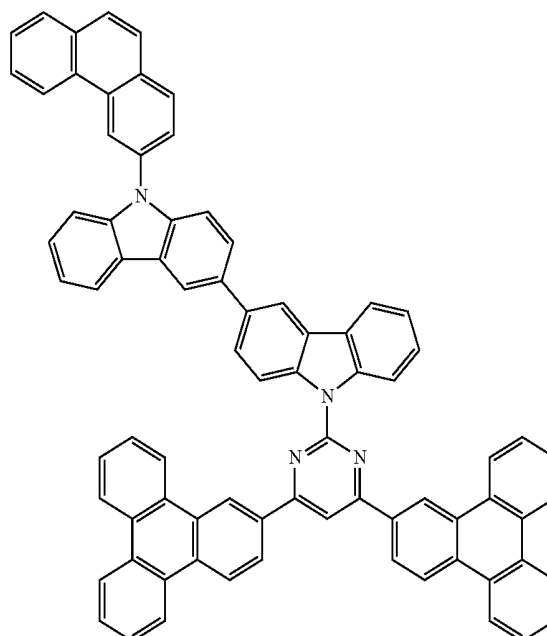
Compound 161



Compound 160



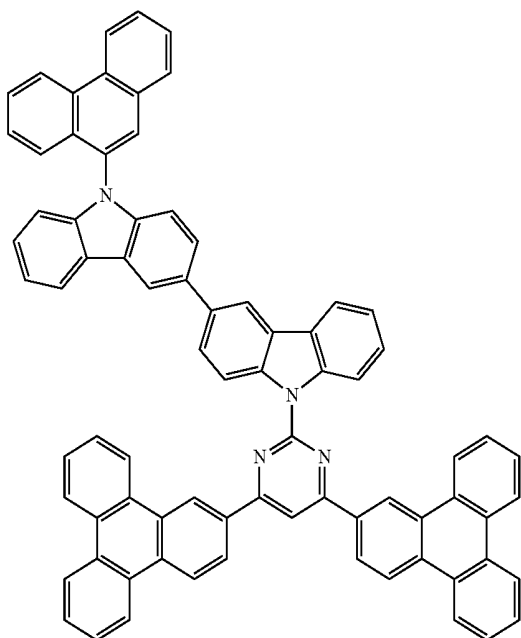
Compound 162



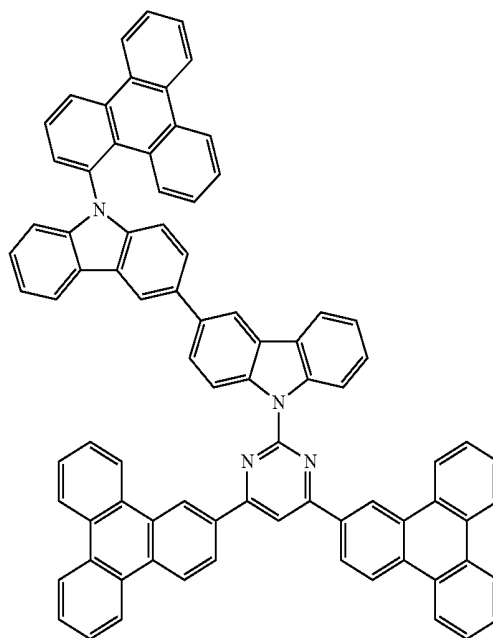
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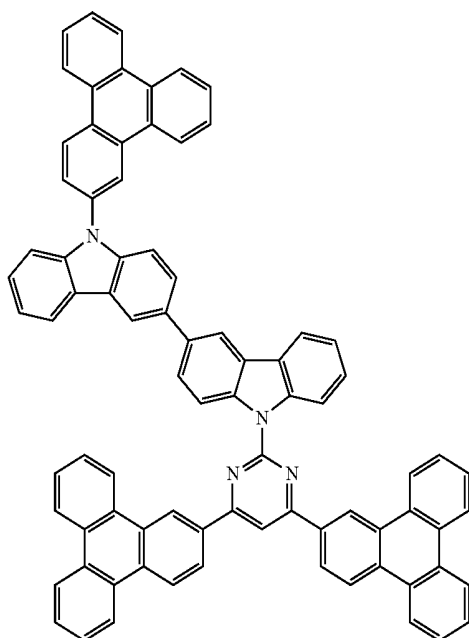
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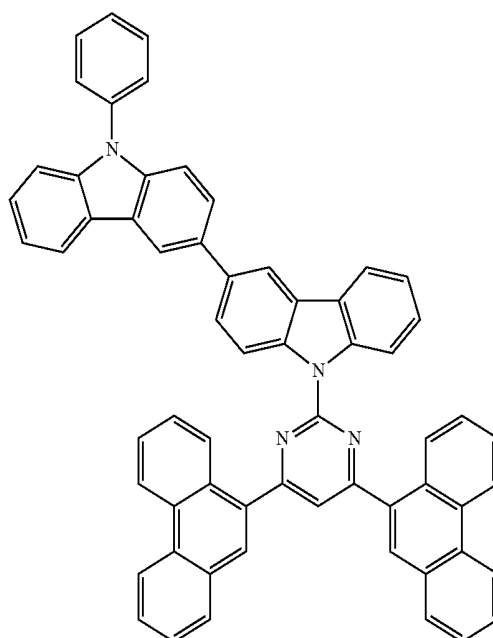
Compound 165



Compound 164



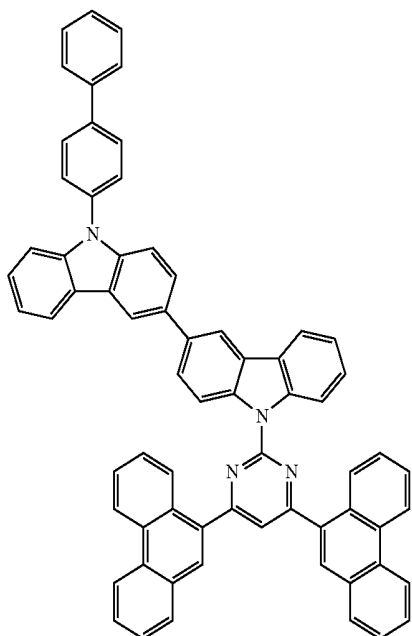
Compound 166



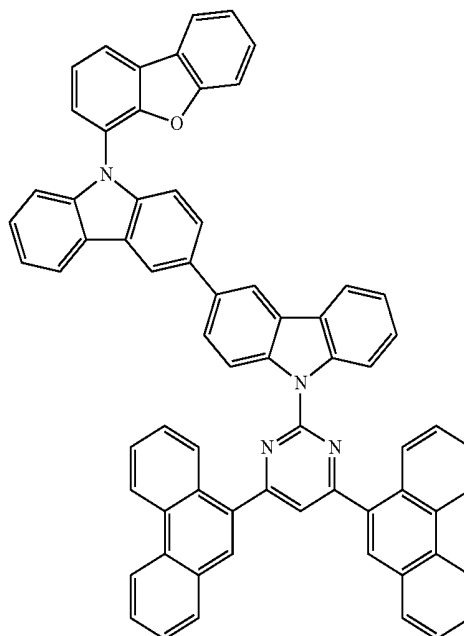
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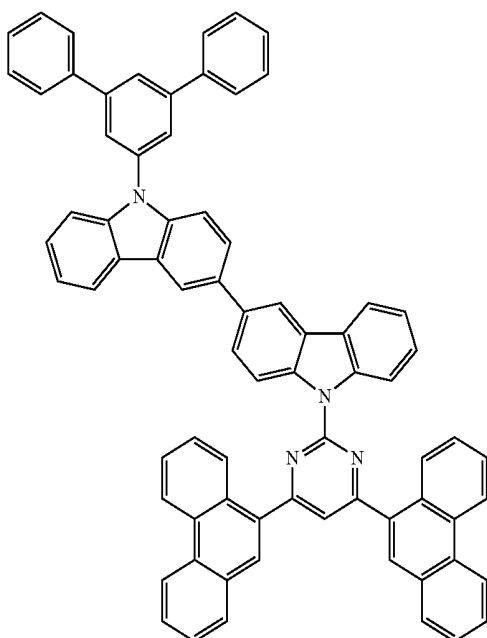
Compound 167



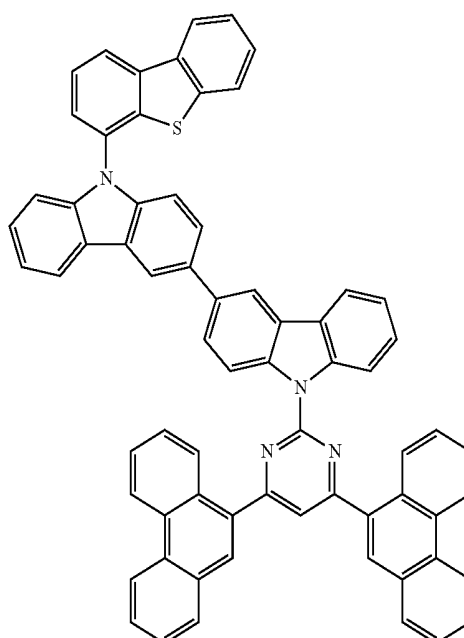
Compound 169



Compound 168



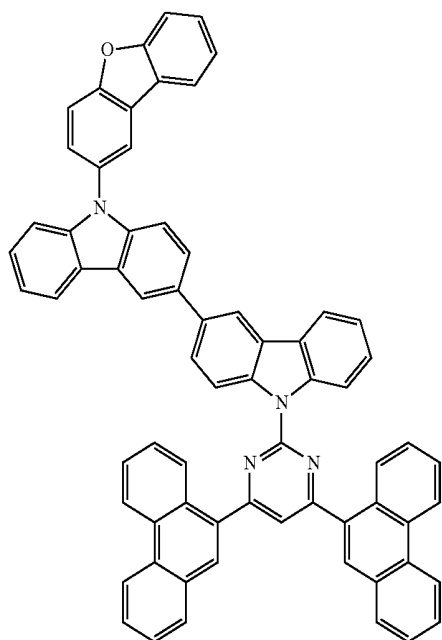
Compound 170



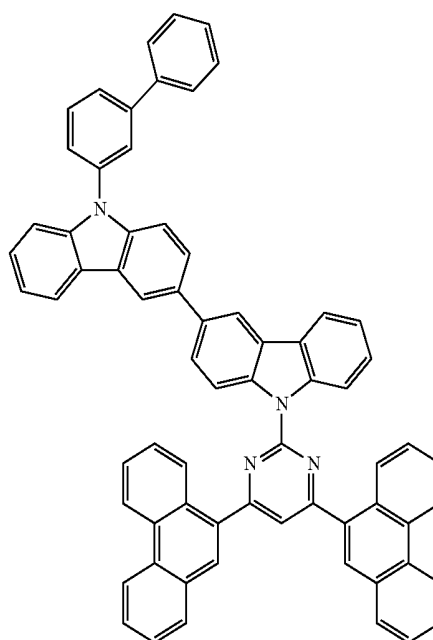
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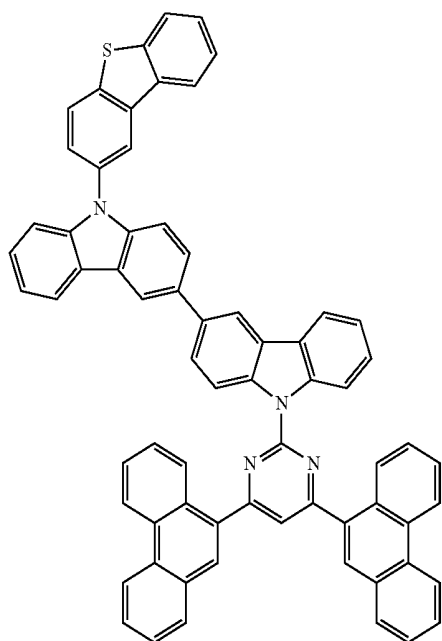
Compound 171



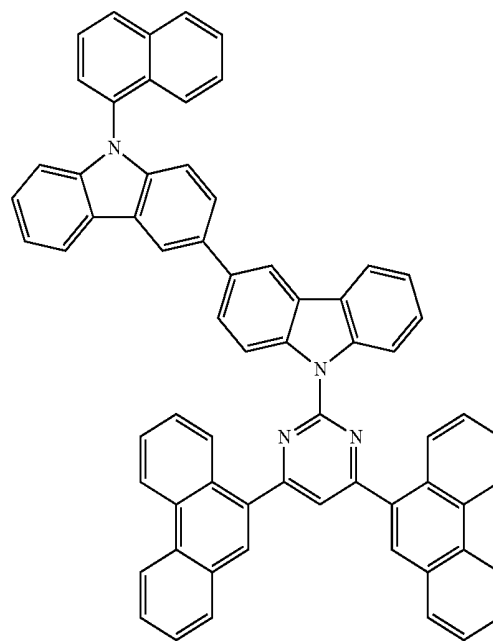
Compound 173



Compound 172



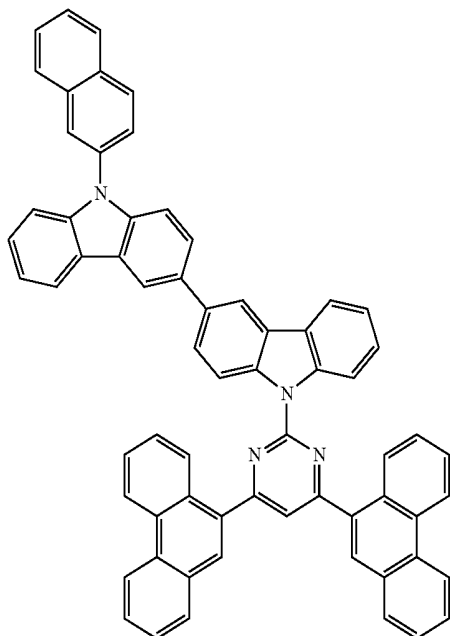
Compound 174



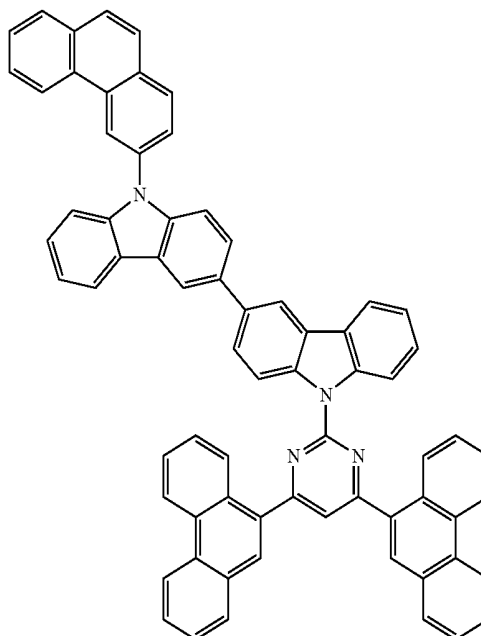
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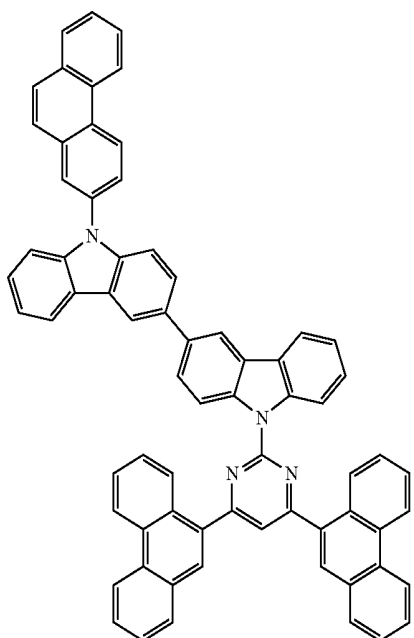
Compound 175



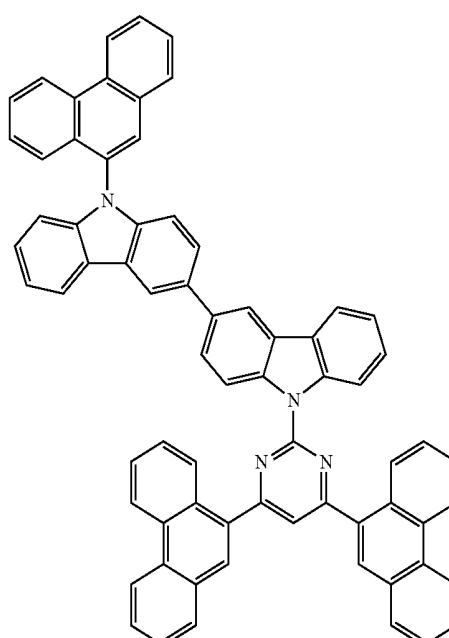
Compound 177



Compound 176



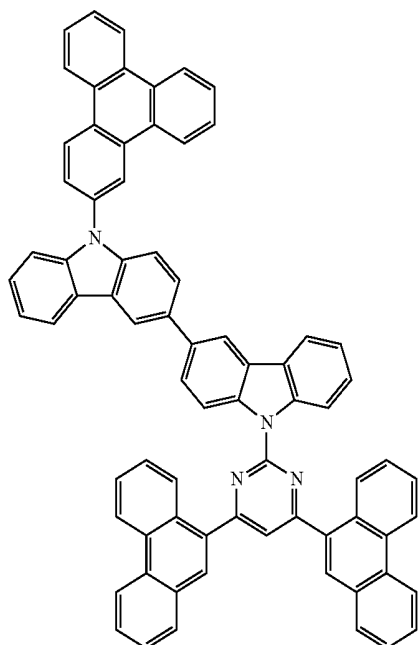
Compound 178



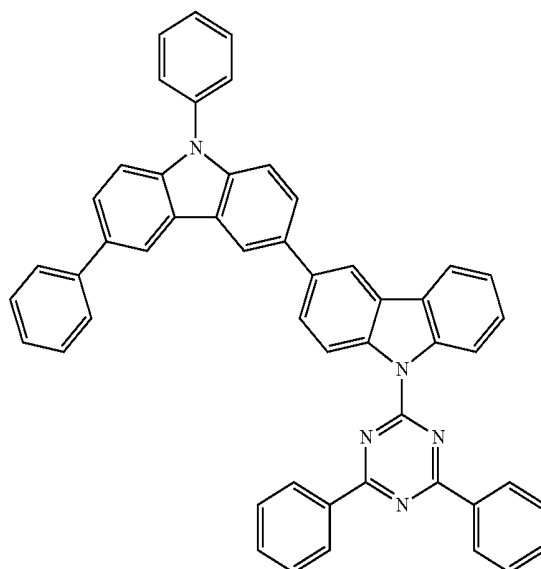
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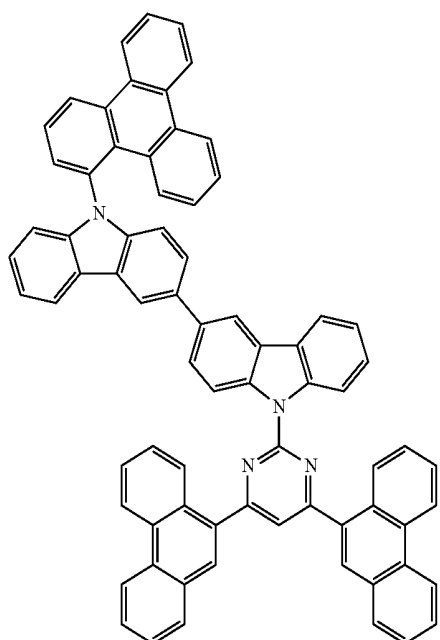
Compound 179



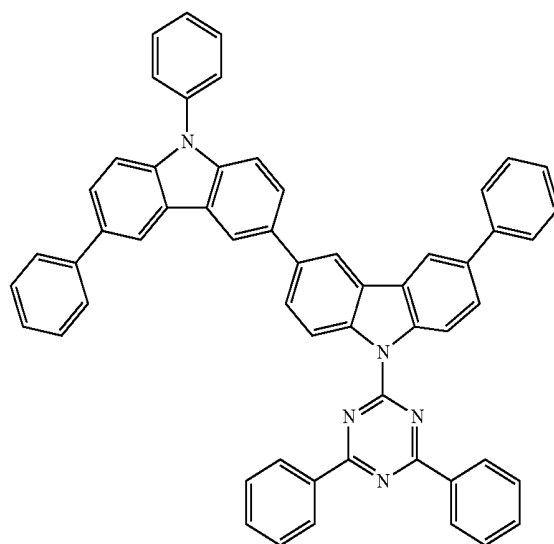
Compound 181



Compound 180



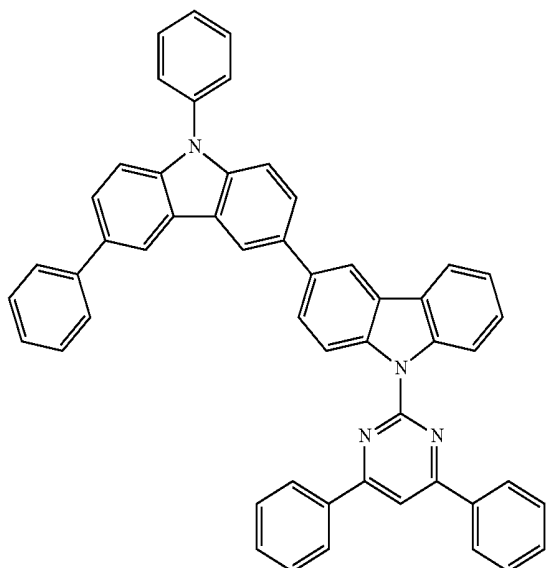
Compound 182



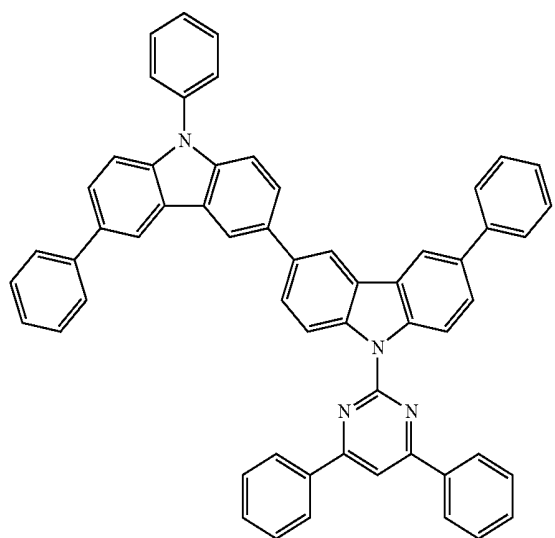


-continued

Compound 183



Compound 184



[0052] A first device comprising an organic light emitting device is also provided. The device further comprises an anode, a cathode, and an organic layer, disposed between the anode and the cathode. The organic layer comprises a compound having Formula I, as described above.

[0053]  $R_1$ ,  $R_2$ ,  $R_3$ , and  $R_4$  may represent mono, di, tri, or tetra substitutions.  $R_1$ ,  $R_2$ ,  $R_3$ , and  $R_4$  are independently selected from the group consisting of hydrogen, alkyl, alkoxy, amino, alkenyl, alkynyl, aryl and heteroaryl.  $Ar_1$ ,  $Ar_2$ , and  $Ar_3$  are independently selected from aryl or heteroaryl.  $Ar_1$ ,  $Ar_2$ , and  $Ar_3$  may be further substituted. X is C or N.

[0054] In one aspect,  $Ar_1$ ,  $Ar_2$ , and  $Ar_3$  are independently selected from the group consisting of phenyl, pyridine, naphthalene, biphenyl, terphenyl, fluorene, dibenzofuran, dibenzothiophene, phenanthrene, and triphenylene.  $Ar_1$ ,  $Ar_2$ , and  $Ar_3$  are independently further substituted with a substituent selected from the group consisting of hydrogen, alkyl, alkoxy, amino, alkenyl, alkynyl, aryl and heteroaryl, but the substituent is not an aryl or heteroaryl fused directly to  $Ar_1$ ,  $Ar_2$ , and

$Ar_3$ . Preferably,  $Ar_1$  and  $Ar_2$  are independently selected from the group consisting of phenyl, pyridine, and naphthalene. Preferably,  $Ar_3$  is selected from the group consisting of phenyl, biphenyl, dibenzofuran, and dibenzothiophene.

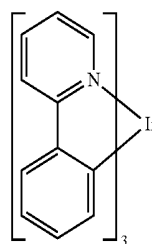
[0055] In another aspect,  $R_1$ ,  $R_2$ ,  $R_3$ , and  $R_4$  are hydrogen.

[0056] Specific examples of devices containing compounds comprising bicarbazole are also provided. In particular, the compound is selected from the group consisting of Compound 1-Compound 184.

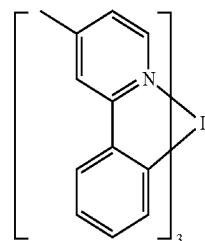
[0057] In one aspect, the organic layer is deposited using solution processing.

[0058] In one aspect, the organic layer is an emissive layer and the compound having Formula I is a host.

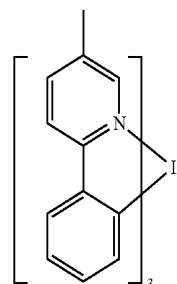
[0059] In another aspect, the organic layer further comprises an emissive dopant having the formula:



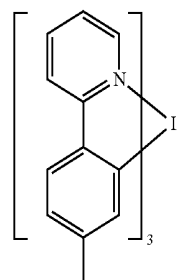
D1



D2

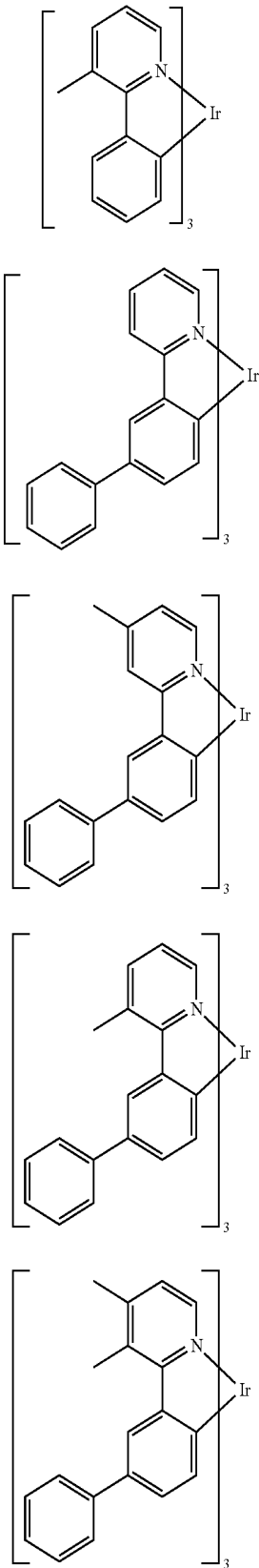


D3

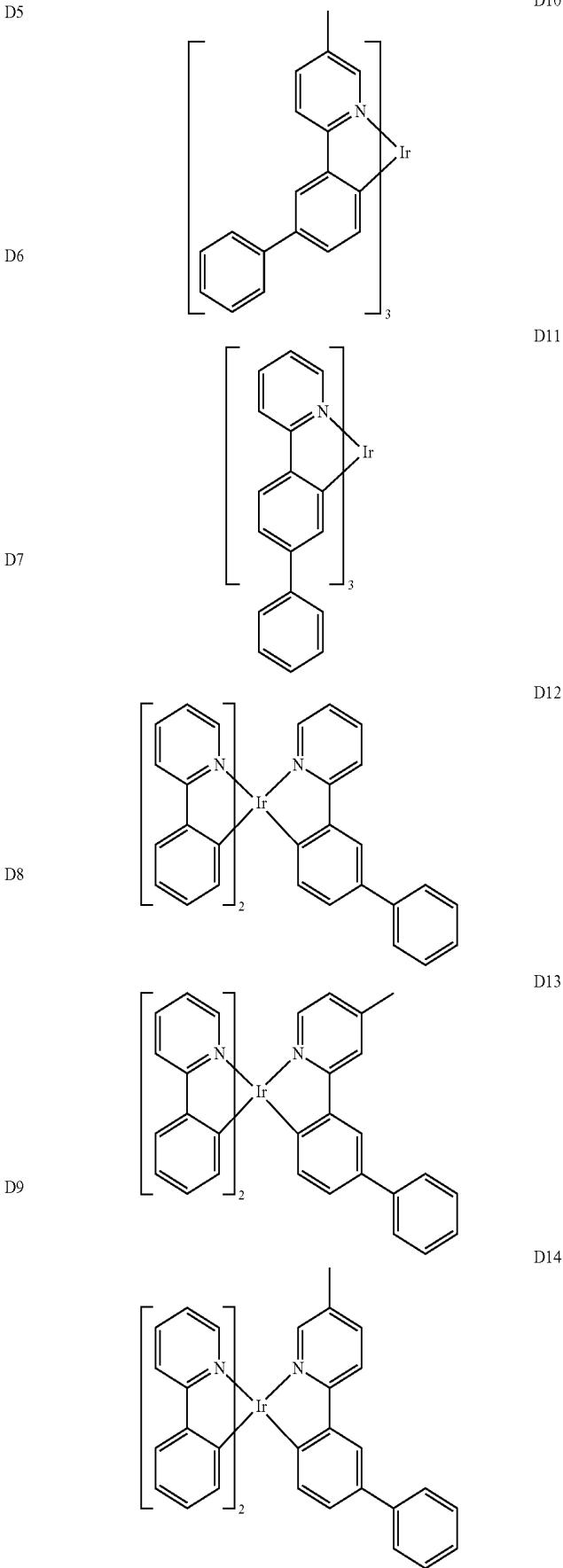


D4

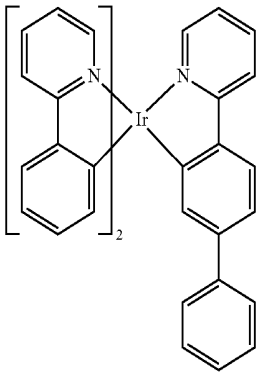
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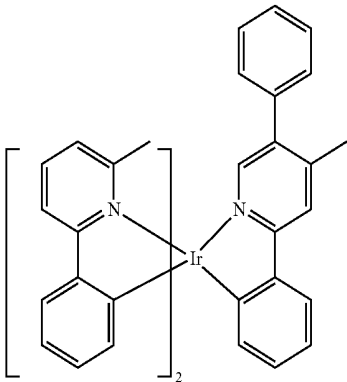


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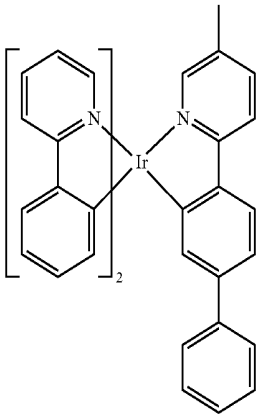


D15

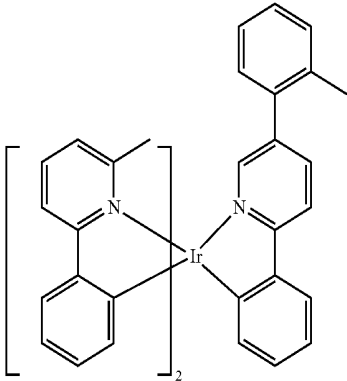
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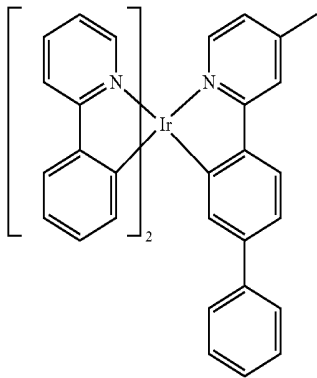
D19



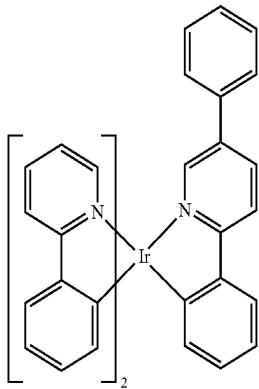
D16



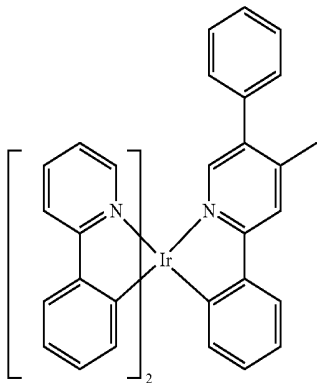
D20



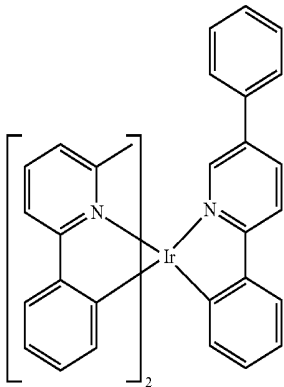
D17



D21

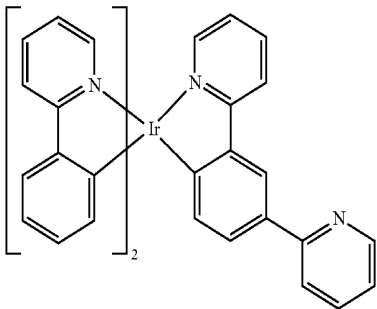


D18

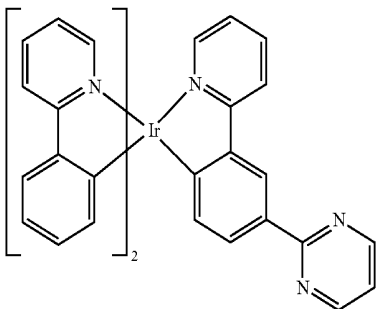


D22

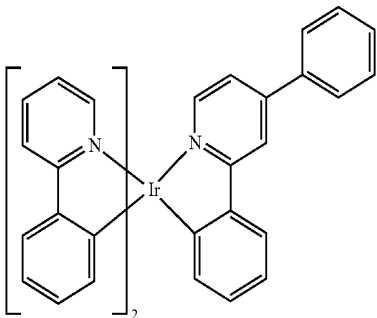
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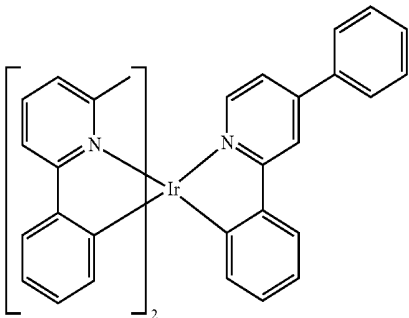
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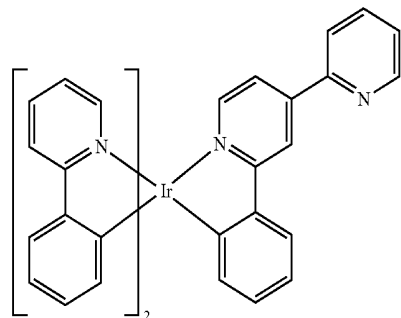
D24



D25

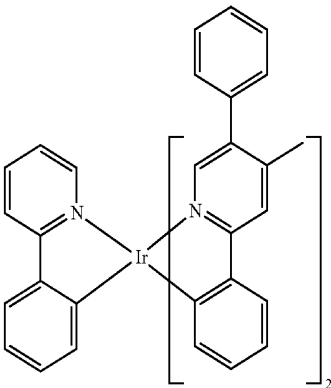


D26

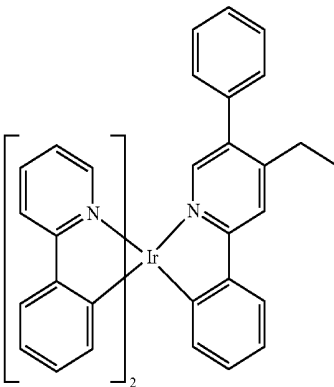


D27

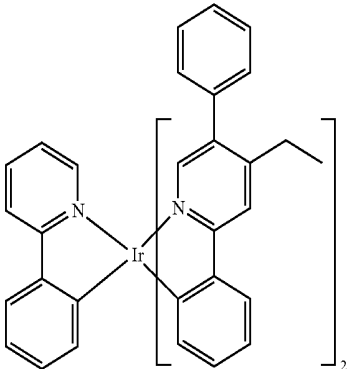
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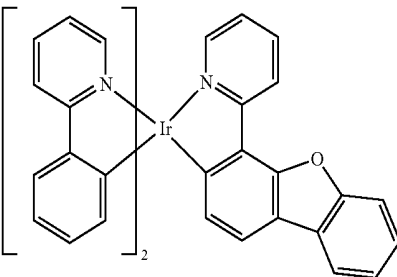
D28



D29

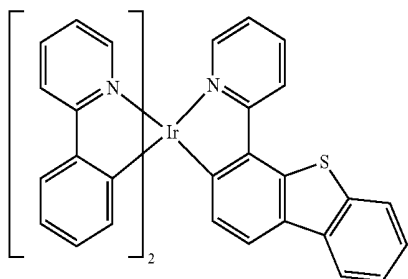


D30



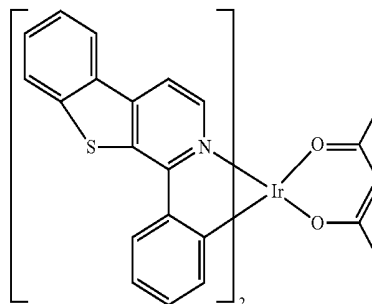
D31

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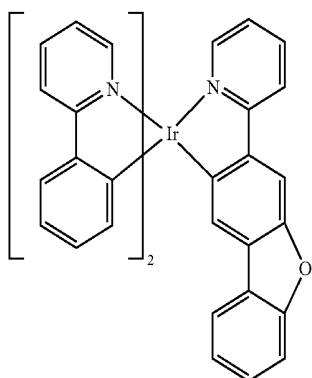


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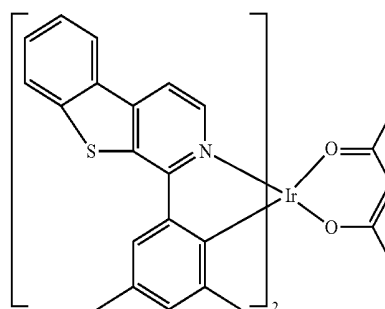
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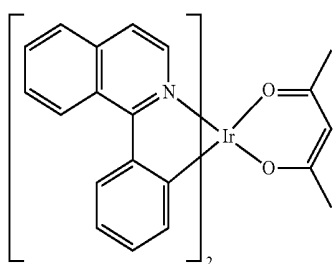
D37



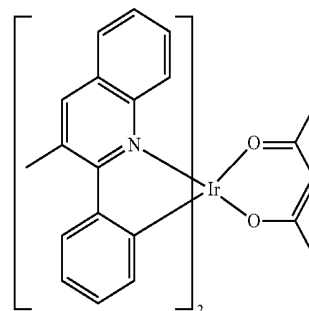
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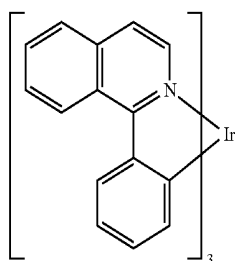
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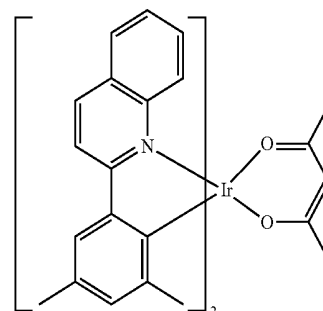
D34



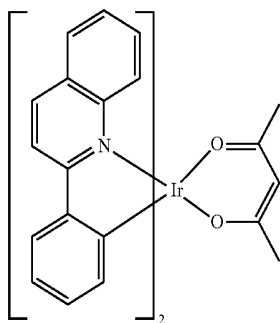
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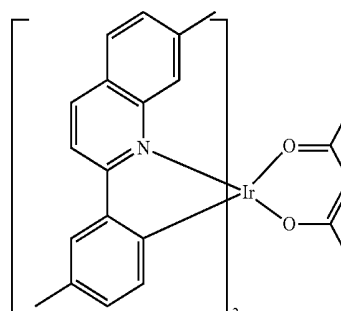
D35



D40

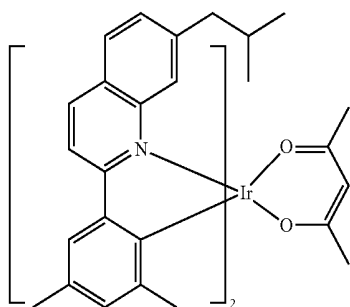


D36

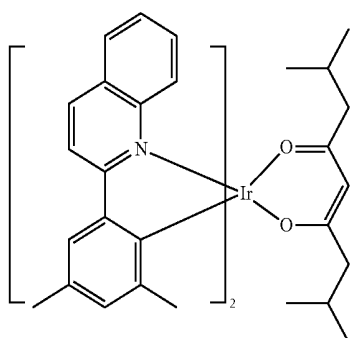


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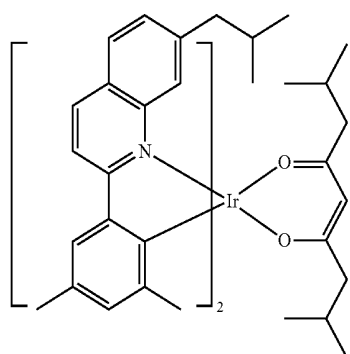
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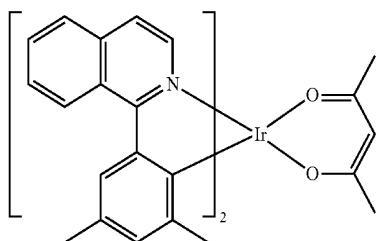
D42



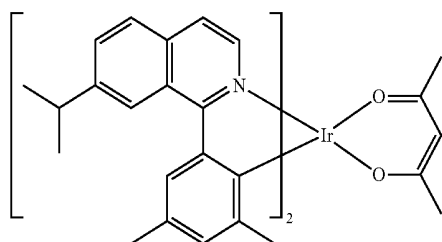
D43



D44

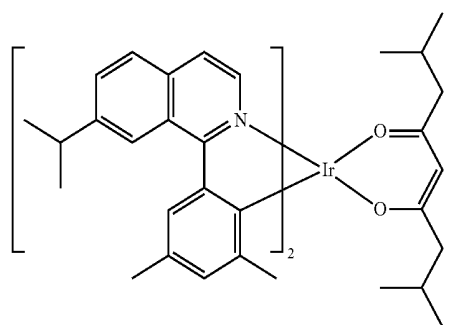


D45



D46

-continued



D47

**[0060]** In one aspect, the first device is a consumer product. In another aspect, the first device is an organic light emitting device.

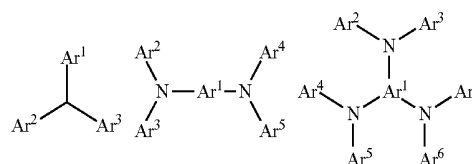
#### Combination with Other Materials

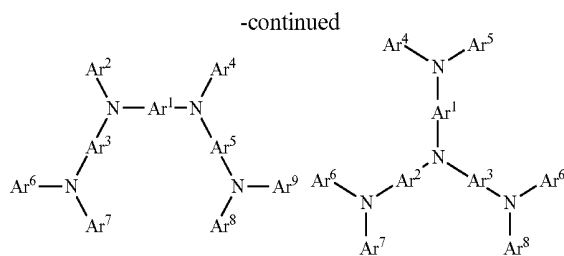
**[0061]** The materials described herein as useful for a particular layer in an organic light emitting device may be used in combination with a wide variety of other materials present in the device. For example, emissive dopants disclosed herein may be used in conjunction with a wide variety of hosts, transport layers, blocking layers, injection layers, electrodes and other layers that may be present. The materials described or referred to below are non-limiting examples of materials that may be useful in combination with the compounds disclosed herein, and one of skill in the art can readily consult the literature to identify other materials that may be useful in combination.

#### HIL/HTL:

**[0062]** A hole injecting/transporting material to be used in the present invention is not particularly limited, and any compound may be used as long as the compound is typically used as a hole injecting/transporting material. Examples of the material include, but not limit to: a phthalocyanine or porphyrin derivative; an aromatic amine derivative; an indolo-carbazole derivative; a polymer containing fluorohydrocarbon; a polymer with conductivity dopants; a conducting polymer, such as PEDOT/PSS; a self-assembly monomer derived from compounds such as phosphonic acid and silane derivatives; a metal oxide derivative, such as  $\text{MoO}_x$ ; a p-type semiconducting organic compound, such as 1,4,5,8,9,12-Hexaazatriphenylenehexacarbonitrile; a metal complex, and a cross-linkable compounds.

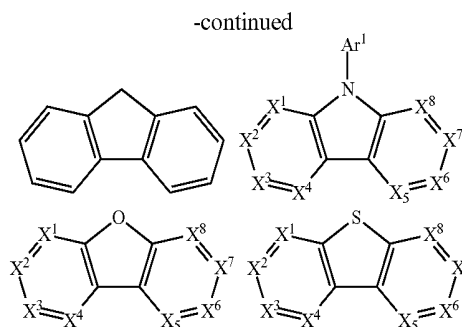
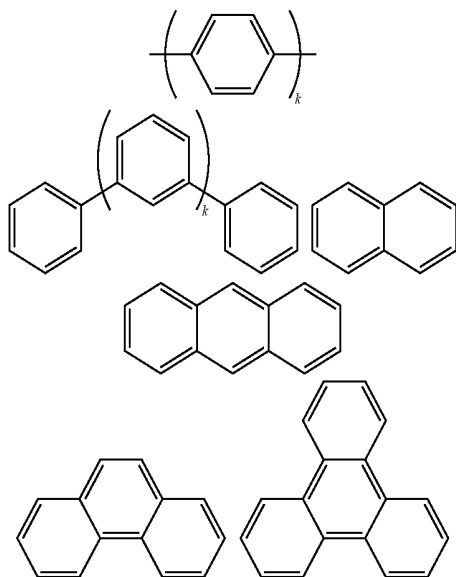
**[0063]** Examples of aromatic amine derivatives used in HIL or HTL include, but not limit to the following general structures:





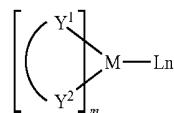
**[0064]** Each of Ar<sup>1</sup> to Ar<sup>9</sup> is selected from the group consisting aromatic hydrocarbon cyclic compounds such as benzene, biphenyl, triphenyl, triphenylene, naphthalene, anthracene, phenalene, phenanthrene, fluorene, pyrene, chrysene, perylene, azulene; group consisting aromatic heterocyclic compounds such as dibenzothiophene, dibenzofuran, dibenzoselenophene, furan, thiophene, benzofuran, benzothiophene, benzoselenophene, carbazole, indolocarbazole, pyridylindole, pyrrolo-dipyrindine, pyrazole, imidazole, triazole, oxazole, thiazole, oxadiazole, oxatriazole, dioxazole, thiadiazole, pyridine, pyridazine, pyrimidine, pyrazine, triazine, oxazine, oxathiazine, oxadiazine, indole, benzimidazole, indazole, indoxazine, benzoxazole, benzisoxazole, benzothiazole, quinoline, isoquinoline, cinnoline, quinazoline, quinoxaline, naphthyridine, phthalazine, pteridine, xanthene, acridine, phenazine, phenothiazine, phenoxazine, benzofuro-pyridine, furodipyrindine, benzothienopyridine, thienodipyrindine, benzoselenophenopyridine, and selenophenodipyrindine; and group consisting 2 to 10 cyclic structural units which are groups of the same type or different types selected from the aromatic hydrocarbon cyclic group and the aromatic heterocyclic group and are bonded to each other directly or via at least one of oxygen atom, nitrogen atom, sulfur atom, silicon atom, phosphorus atom, boron atom, chain structural unit and the aliphatic cyclic group. Wherein each Ar is further substituted by a substituent selected from the group consisting of hydrogen, alkyl, alkoxy, amino, alkenyl, alkynyl, arylalkyl, heteroalkyl, aryl and heteroaryl.

**[0065]** In one aspect, Ar<sup>1</sup> to Ar<sup>9</sup> is independently selected from the group consisting of:



**[0066]** k is an integer from 1 to 20; X<sup>1</sup> to X<sup>8</sup> is CH or N; Ar<sup>1</sup> has the same group defined above.

**[0067]** Examples of metal complexes used in HIL or HTL include, but not limit to the following general formula:



**[0068]** M is a metal, having an atomic weight greater than 40; (Y<sup>1</sup>-Y<sup>2</sup>) is a bidentate ligand, Y<sup>1</sup> and Y<sup>2</sup> are independently selected from C, N, O, P, and S; L is an ancillary ligand; m is an integer value from 1 to the maximum number of ligands that may be attached to the metal; and m+n is the maximum number of ligands that may be attached to the metal.

**[0069]** In one aspect, (Y<sup>1</sup>-Y<sup>2</sup>) is a 2-phenylpyridine derivative.

**[0070]** In another aspect, (Y<sup>1</sup>-Y<sup>2</sup>) is a carbene ligand.

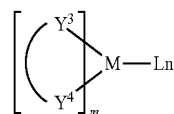
**[0071]** In another aspect, M is selected from Ir, Pt, Os, and Zn.

**[0072]** In a further aspect, the metal complex has a smallest oxidation potential in solution vs. Fe<sup>+</sup>/Fe couple less than about 0.6 V.

Host:

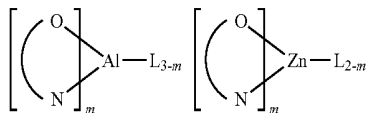
**[0073]** The light emitting layer of the organic EL device of the present invention preferably contains at least a metal complex as light emitting material, and may contain a host material using the metal complex as a dopant material. Examples of the host material are not particularly limited, and any metal complexes or organic compounds may be used as long as the triplet energy of the host is larger than that of the dopant.

**[0074]** Examples of metal complexes used as host are preferred to have the following general formula:



**[0075]** M is a metal; (Y<sup>3</sup>-Y<sup>4</sup>) is a bidentate ligand, Y<sup>3</sup> and Y<sup>4</sup> are independently selected from C, N, O, P, and S; L is an ancillary ligand; m is an integer value from 1 to the maximum number of ligands that may be attached to the metal; and m+n is the maximum number of ligands that may be attached to the metal.

[0076] In one aspect, the metal complexes are:



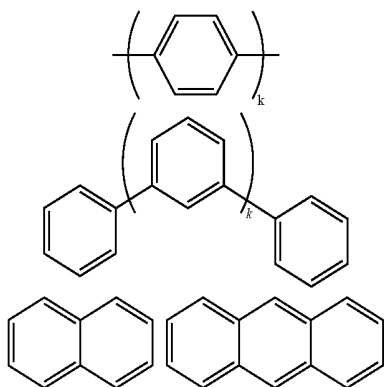
[0077] (O—N) is a bidentate ligand, having metal coordinated to atoms O and N.

[0078] In another aspect, M is selected from Ir and Pt.

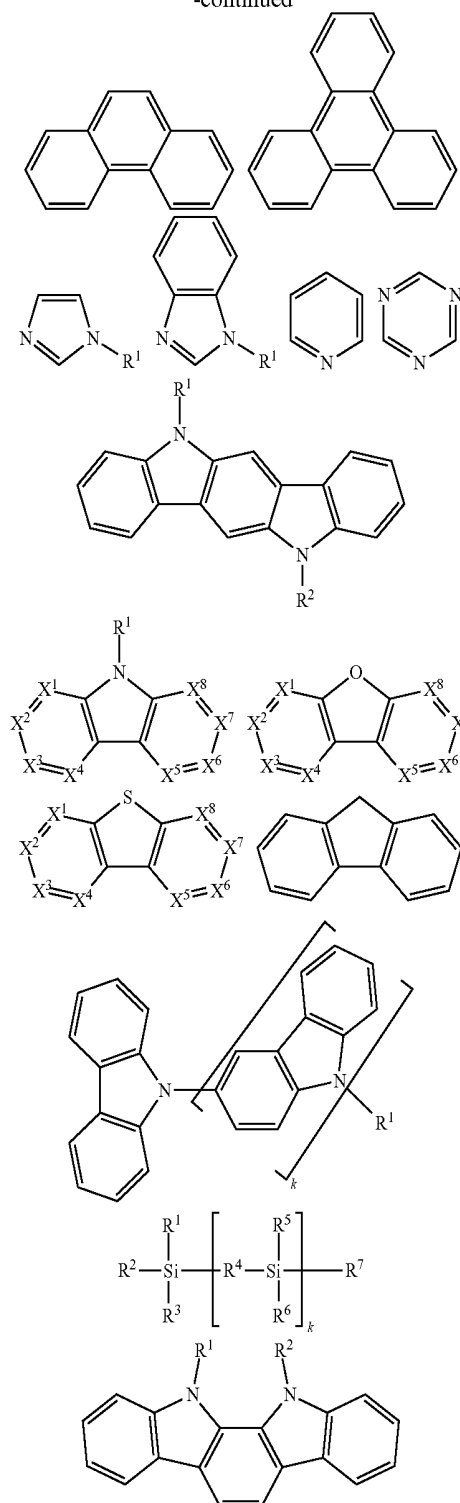
[0079] In a further aspect, (Y<sup>3</sup>-Y<sup>4</sup>) is a carbene ligand.

[0080] Examples of organic compounds used as host are selected from the group consisting aromatic hydrocarbon cyclic compounds such as benzene, biphenyl, triphenyl, triphenylene, naphthalene, anthracene, phenalene, phenanthrene, fluorene, pyrene, chrysene, perylene, azulene; group consisting aromatic heterocyclic compounds such as dibenzothiophene, dibenzofuran, dibenzoselenophene, furan, thiophene, benzofuran, benzothiophene, benzoselenophene, carbazole, indolocarbazole, pyridylindole, pyrrolodipyridine, pyrazole, imidazole, triazole, oxazole, thiazole, oxadiazole, oxatriazole, dioxazole, thiadiazole, pyridine, pyridazine, pyrimidine, pyrazine, triazine, oxazine, oxathiazine, oxadiazine, indole, benzimidazole, indazole, indoxazine, benzoxazole, benzisoxazole, benzothiazole, quinoline, isoquinoline, cinnoline, quinoxaline, naphthyridine, phthalazine, pteridine, xanthene, acridine, phenazine, phenothiazine, phenoxazine, benzofuropyridine, furodipyridine, benzothienopyridine, thienodipyridine, benzoselenophenopyridine, and selenophenodipyridine; and group consisting 2 to 10 cyclic structural units which are groups of the same type or different types selected from the aromatic hydrocarbon cyclic group and the aromatic heterocyclic group and are bonded to each other directly or via at least one of oxygen atom, nitrogen atom, sulfur atom, silicon atom, phosphorus atom, boron atom, chain structural unit and the aliphatic cyclic group. Wherein each group is further substituted by a substituent selected from the group consisting of hydrogen, alkyl, alkoxy, amino, alkenyl, alkynyl, arylalkyl, heteroalkyl, aryl and heteroaryl.

[0081] In one aspect, host compound contains at least one of the following groups in the molecule:



-continued



[0082] R<sup>1</sup> to R<sup>7</sup> is independently selected from the group consisting of hydrogen, alkyl, alkoxy, amino, alkenyl, alkynyl, arylalkyl, heteroalkyl, aryl and heteroaryl, when it is aryl or heteroaryl, it has the similar definition as Ar's mentioned above.

[0083] k is an integer from 0 to 20.

[0084] X<sup>1</sup> to X<sup>8</sup> is selected from CH or N.





**[0098]** The materials described herein as useful for a particular layer in an organic light emitting device may be used in combination with a wide variety of other materials present in the device. For example, emissive dopants disclosed herein may be used in conjunction with a wide variety of hosts, transport layers, blocking layers, injection layers, electrodes and other layers that may be present. The materials described or referred to below are non-limiting examples of materials that may be useful in combination with the compounds disclosed herein, and one of skill in the art can readily consult the literature to identify other materials that may be useful in combination.

**[0099]** In addition to and/or in combination with the materials disclosed herein, many hole injection materials, hole transporting materials, host materials, dopant materials, exciton/hole blocking layer materials, electron transporting and electron injecting materials may be used in an OLED. Non-limiting examples of the materials that may be used in an OLED in combination with materials disclosed herein are listed in Table 1 below. Table 1 lists non-limiting classes of materials, non-limiting examples of compounds for each class, and references that disclose the materials.

TABLE 1

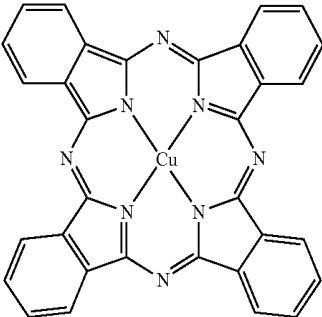
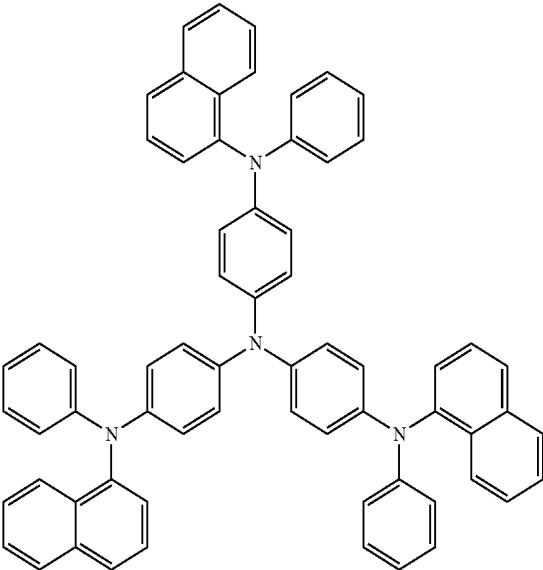
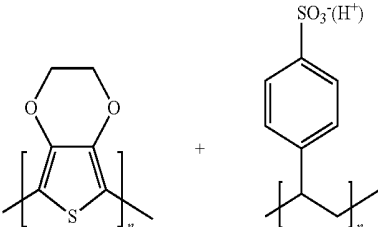
| MATERIAL  | EXAMPLES OF MATERIAL   | PUBLICATIONS                               |
|---|--|--|
| Hole injection materials  |  |  |
| Phthalocyanine and porphyrin compounds                            |    | Appl. Phys. Lett. 69, 2160 (1996)          |
| Starburst triarylamines   |  | J. Lumin. 72-74, 985 (1997)                |
| CF <sub>x</sub> Fluorohydrocarbon polymer                         | $\text{---}[\text{CH}_x\text{F}_y]_n\text{---}$                                      | Appl. Phys. Lett. 78, 673 (2001)           |
| Conducting polymers (e.g., PEDOT:PSS, polyaniline, polythiophene) |   | Synth. Met. 87, 171 (1997)<br>WO2007002683 |

TABLE 1-continued

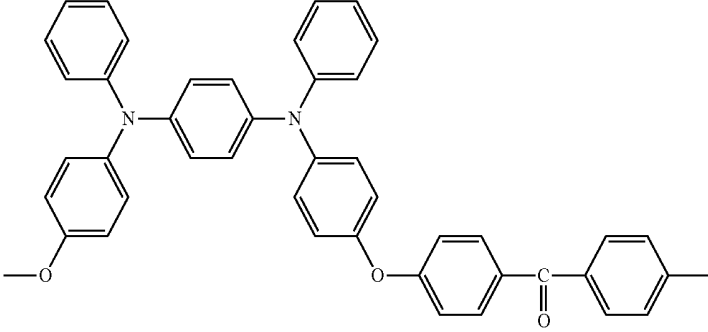
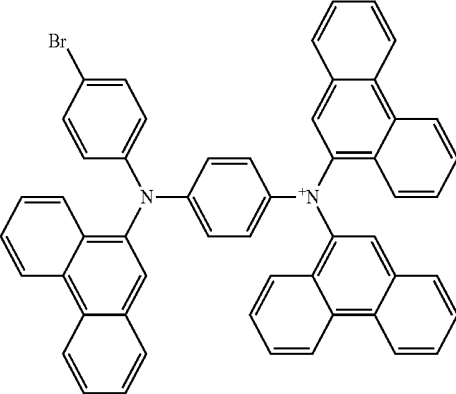
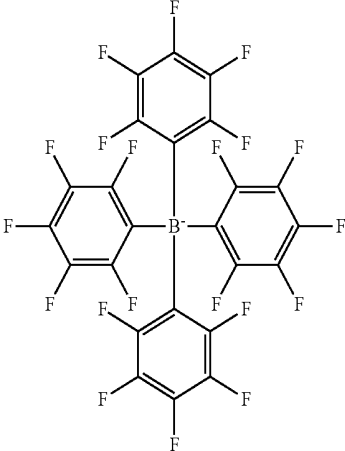
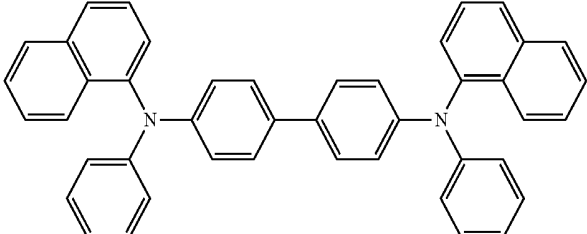
| MATERIAL  | EXAMPLES OF MATERIAL   | PUBLICATIONS  |
|---|--|---|
| Phosphonic acid and silane SAMs   | $\text{N} \left( \text{C}_6\text{H}_4\text{SiCl}_3 \right)_3$  | US20030162053   |
| Triarylamine or polythiophene polymers with conductivity dopants              |  <p style="text-align: center;">and</p>  <p style="text-align: center;">and</p>  | EA01725079A1  |
| Arylamines complexed with metal oxides such as molybdenum and tungsten oxides |  <p style="text-align: right;">+ MoO<sub>x</sub></p>   | SID Symposium<br>Digest, 37, 923 (2006)<br>WO2009018009 |

TABLE 1-continued

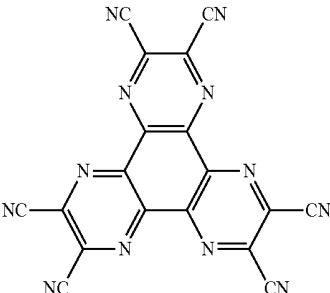
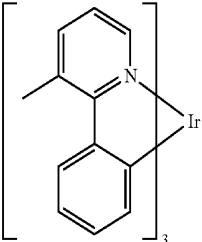
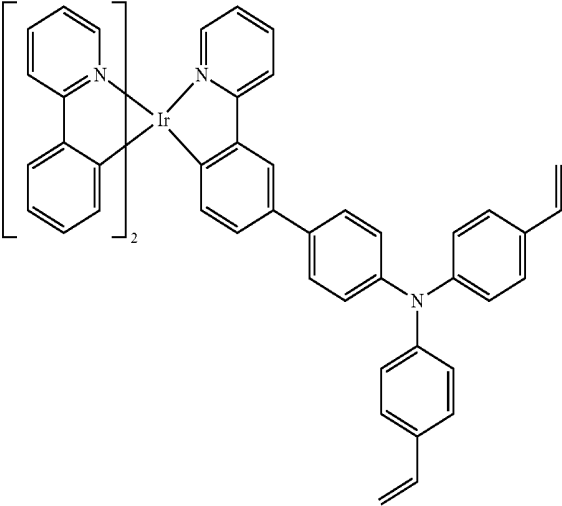
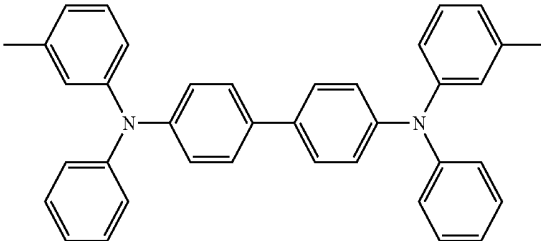
| MATERIAL                                    | EXAMPLES OF MATERIAL   | PUBLICATIONS                        |
|---|--|-------------------------------------|
| p-type semiconducting organic complexes     |     | US20020158242                       |
| Metal organometallic complexes              |    | US20060240279                       |
| Cross-linkable compounds                    |  | US20080220265                       |
| Hole transporting materials                 |  |                                     |
| Triarylamines<br>(e.g., TPD, $\alpha$ -NPD) |  | Appl. Phys. Lett.<br>51, 913 (1987) |

TABLE 1-continued

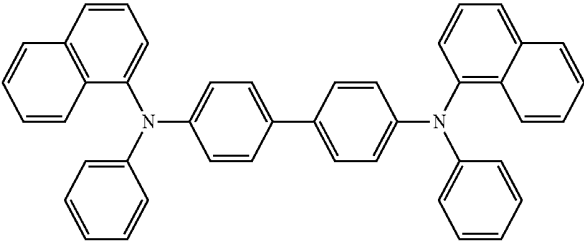
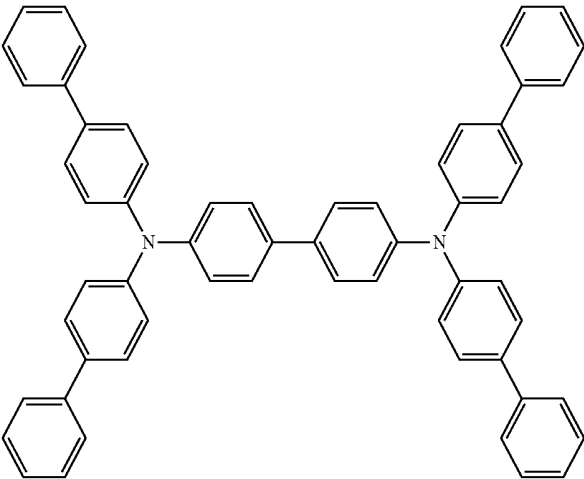
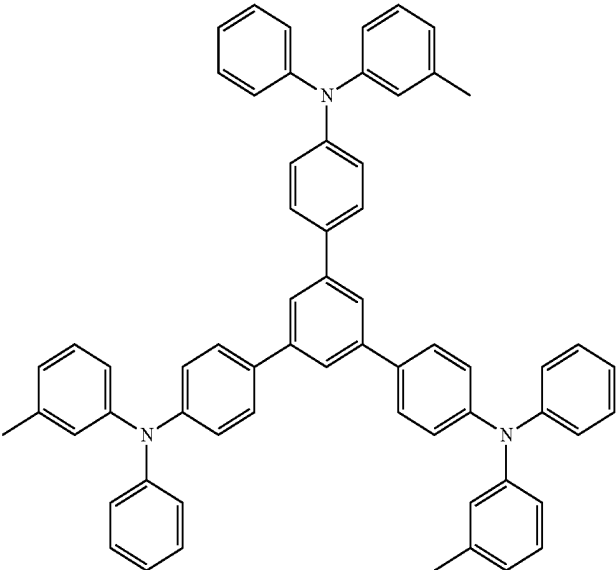
| MATERIAL | EXAMPLES OF MATERIAL   | PUBLICATIONS                     |
|----------|--|----------------------------------|
|          |    | U.S. Pat. No.<br>5,061,569       |
|          |   | EP650955                         |
|          |  | J. Mater. Chem. 3, 319<br>(1993) |

TABLE 1-continued

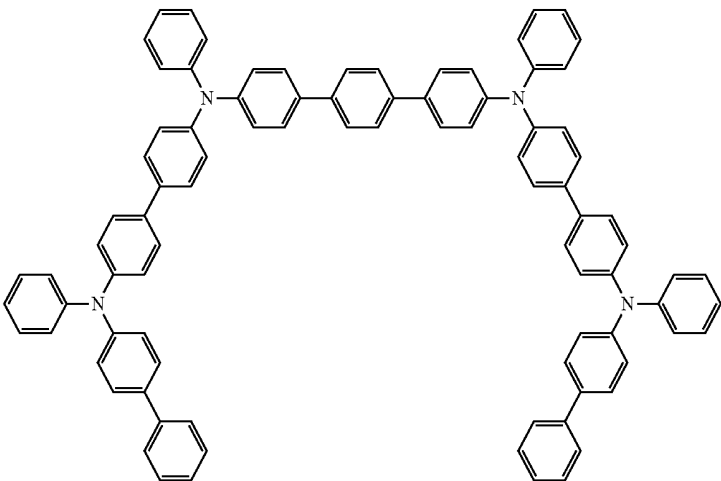
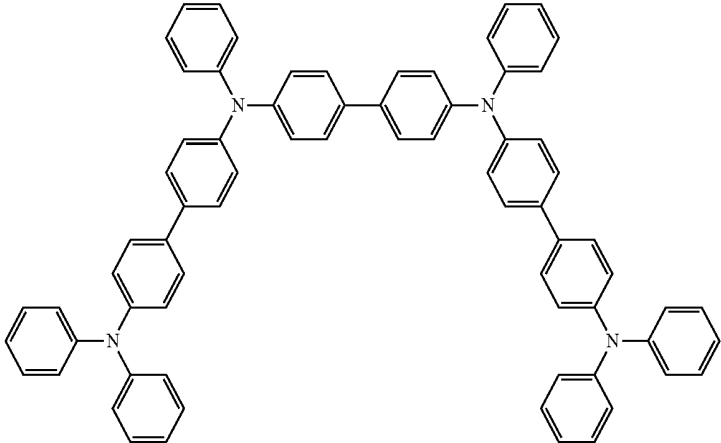
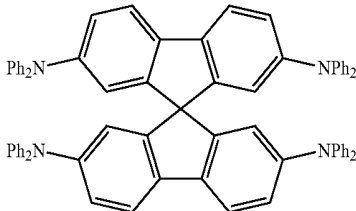
| MATERIAL                          | EXAMPLES OF MATERIAL   | PUBLICATIONS                        |
|-----------------------------------|--|-------------------------------------|
|                                   |    | Appl. Phys. Lett. 90, 183503 (2007) |
|                                   |  | Appl. Phys. Lett. 90, 183503 (2007) |
| Triaylamine on spirofluorene core |   | Synth. Met. 91, 209 (1997)          |

TABLE 1-continued

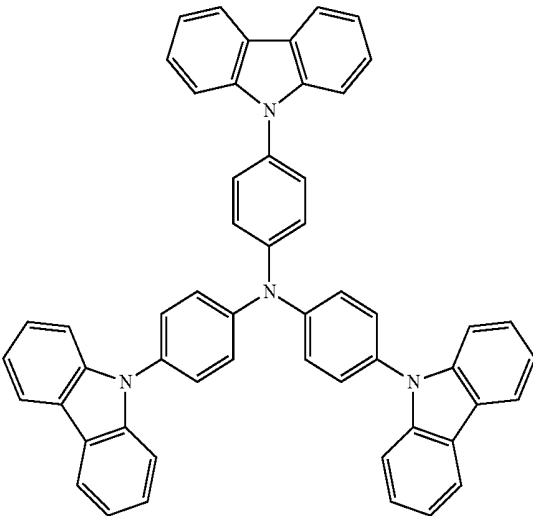
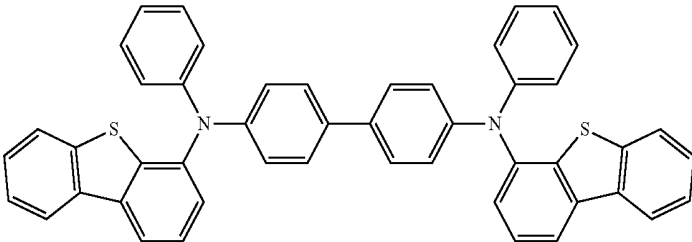
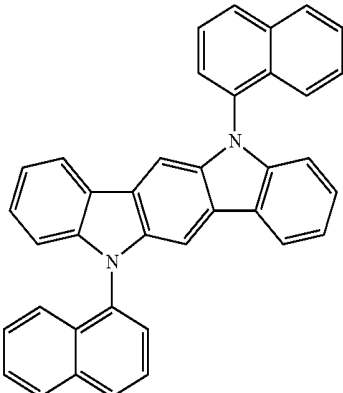
| MATERIAL   | EXAMPLES OF MATERIAL   | PUBLICATIONS                                |
|--|--|---|
| Arylamine carbazole compounds                        |    | Adv. Mater. 6, 677 (1994),<br>US20080124572 |
| Triarylamine with (di)benzothiophene/ (di)benzofuran |  | US20070278938,<br>US20080106190             |
| Indolocarbazoles                                     |   | Synth. Met. 111, 421 (2000)                 |

TABLE 1-continued

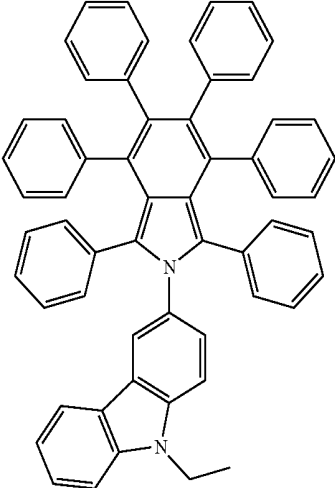
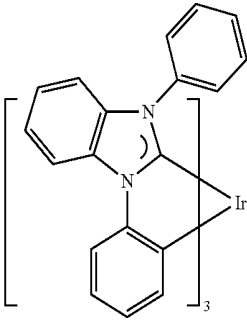
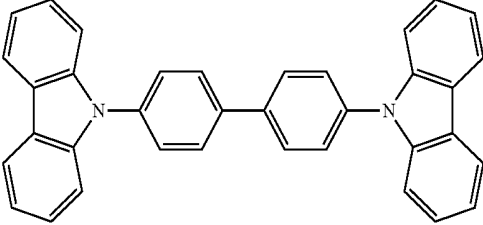
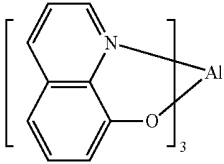
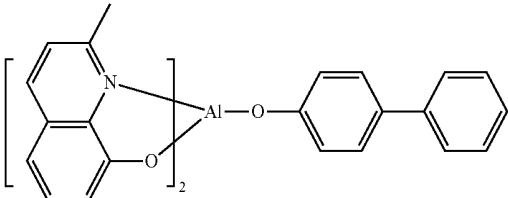
| MATERIAL   | EXAMPLES OF MATERIAL   | PUBLICATIONS                      |
|--|--|-----------------------------------|
| Isoindole compounds  |     | Chem. Mater. 15, 3148 (2003)      |
| Metal carbene complexes                                      |    | US20080018221                     |
| Phosphorescent OLED host materials<br>Red hosts              |  |                                   |
| Arylcarbazoles   |  | Appl. Phys. Lett. 78, 1622 (2001) |
| Metal 8-hydroxyquinolates<br>(e.g., Alq <sub>3</sub> , BAlq) |   | Nature 395, 151 (1998)            |
|  |  | US20060202194                     |



TABLE 1-continued

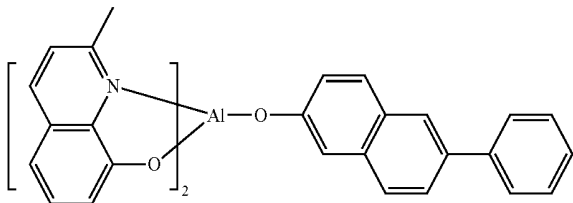
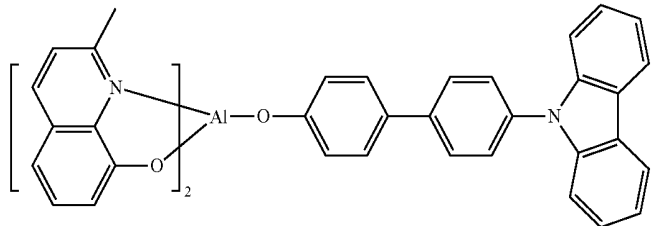
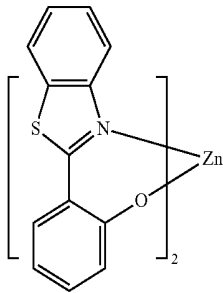
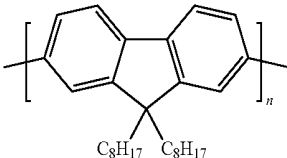
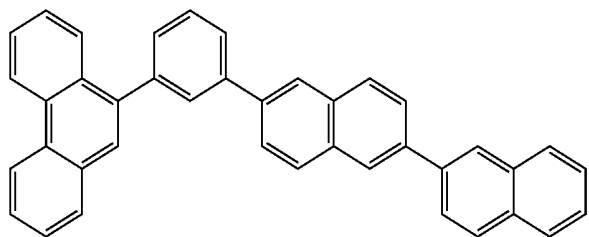
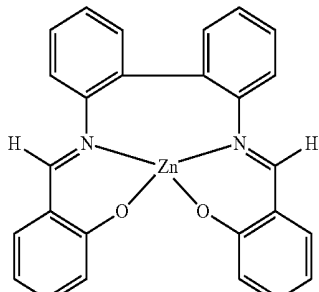
| MATERIAL   | EXAMPLES OF MATERIAL   | PUBLICATIONS   |
|--|--|--|
|  |    | WO2005014551   |
|  |    | WO2006072002   |
| Metal phenoxybenzothiazole compounds                   |    | Appl. Phys. Lett. 90, 123509 (2007)  |
| Conjugated oligomers and polymers (e.g., polyfluorene) |   | Org. Electron. 1, 15 (2000)  |
| Aromatic fused rings                                   |  | WO2009066779, WO2009066778, WO2009063833, US20090045730, US20090045731, WO2009008311, US20090008605, US20090009065 |
| Zinc complexes   |   | WO2009062578   |

TABLE 1-continued

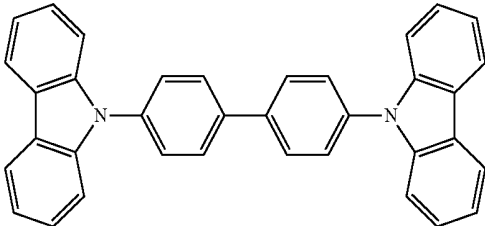
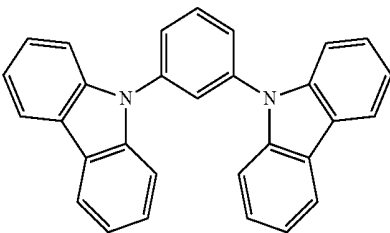
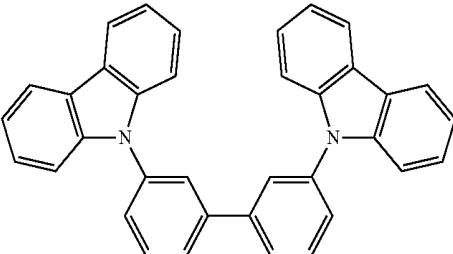
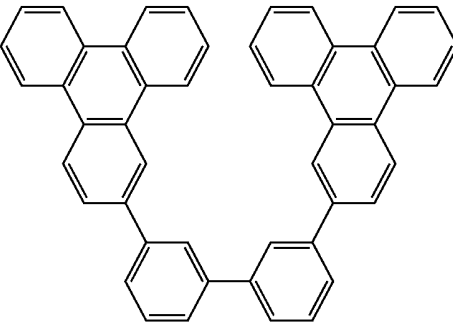
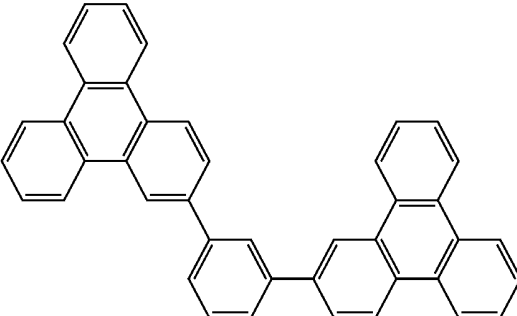
| MATERIAL                   | EXAMPLES OF MATERIAL   | PUBLICATIONS                      |
|----------------------------|--|-----------------------------------|
| Green hosts                |  |                                   |
| Arylcarbazoles             |    | Appl. Phys. Lett. 78, 1622 (2001) |
|                            |    | US20030175553                     |
|                            |  | WO2001039234                      |
| Aryltriphenylene compounds |  | US20060280965                     |
|                            |  | US20060280965                     |

TABLE 1-continued

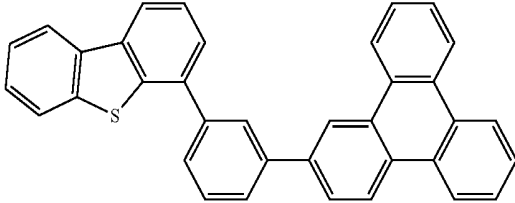
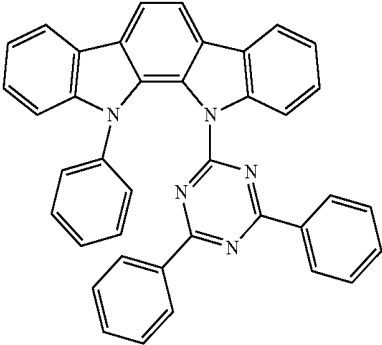
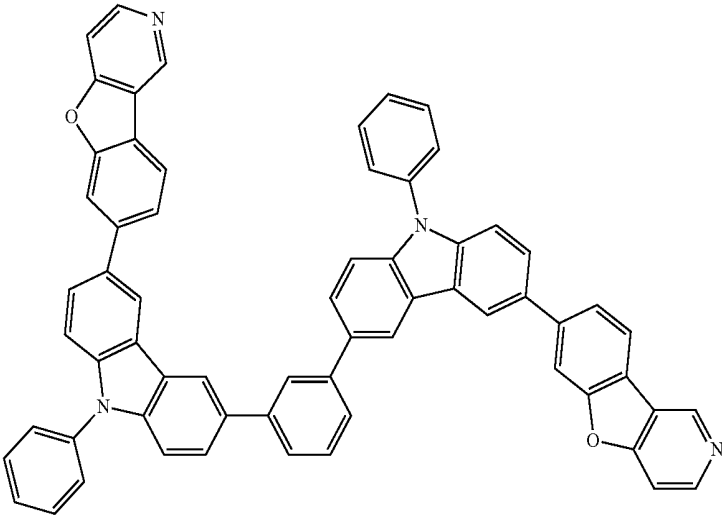
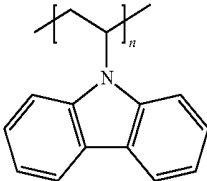
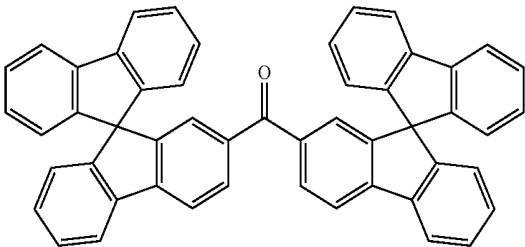
| MATERIAL                         | EXAMPLES OF MATERIAL   | PUBLICATIONS                         |
|----------------------------------|--|--------------------------------------|
|                                  |    | WO2009021126                         |
| Donor acceptor<br>type molecules |    | WO2008056746                         |
| Aza-carbazole/<br>DBT/DBF        |  | JP2008074939                         |
| Polymers (e.g., PVK)             |   | Appl. Phys. Lett. 77,<br>2280 (2000) |
| Spirofluorene compounds          |  | WO2004093207                         |

TABLE 1-continued

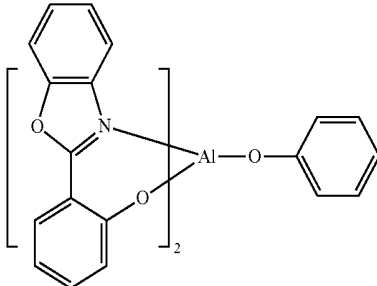
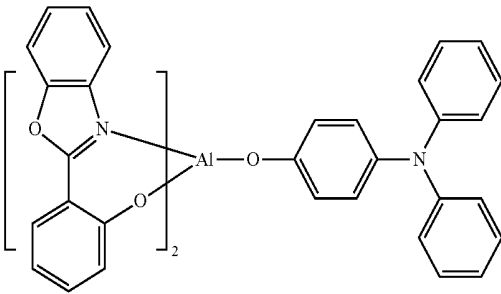
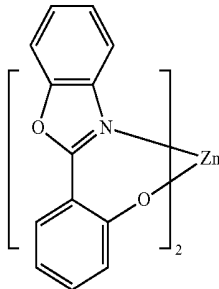
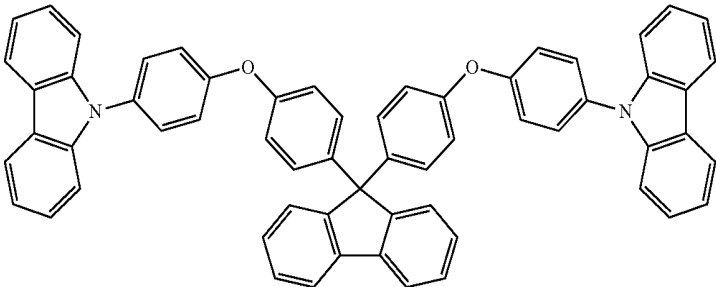
| MATERIAL                                 | EXAMPLES OF MATERIAL   | PUBLICATIONS |
|--|--|--------------|
| Metal<br>phenoxybenzoxazole<br>compounds |     | WO2005089025 |
|  |   | WO2006132173 |
|  |   | JP200511610  |
| Spirofluorene-carbazole<br>compounds     |  | JP2007254297 |

TABLE 1-continued

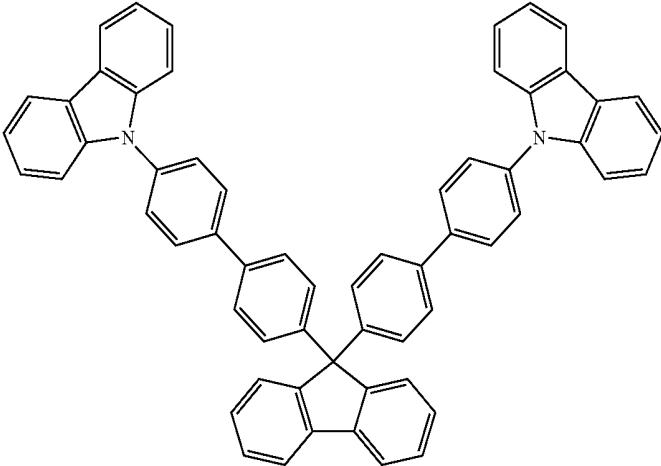
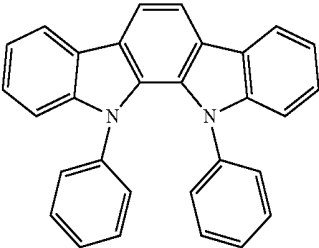
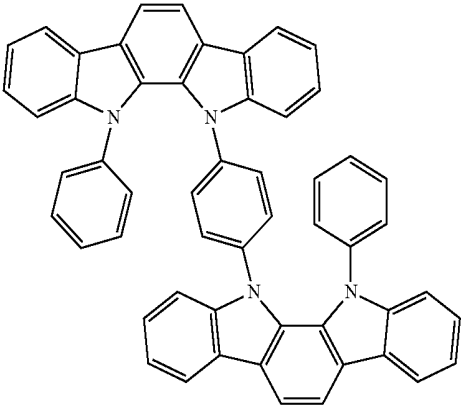
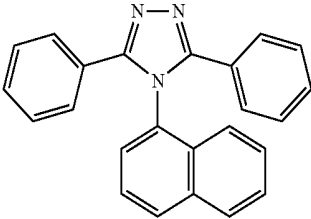
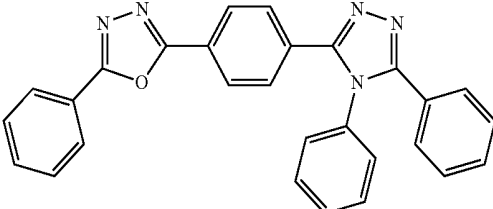
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|   |    | JP2007254297                      |
| Indolocarbazoles  |    | WO2007063796                      |
|   |  | WO2007063754                      |
| 5-member ring electron deficient heterocycles<br>(e.g., triazole, oxadiazole) |   | J. Appl. Phys. 90,<br>5048 (2001) |
|   |  | WO2004107822                      |

TABLE 1-continued

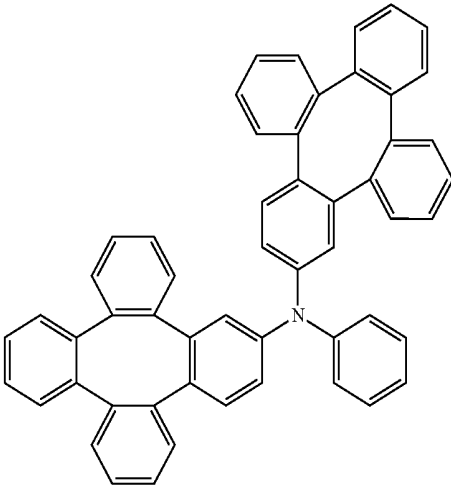
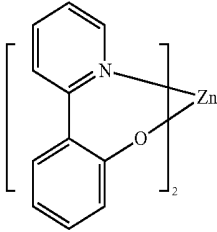
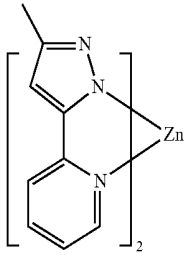
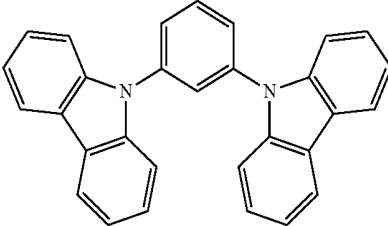
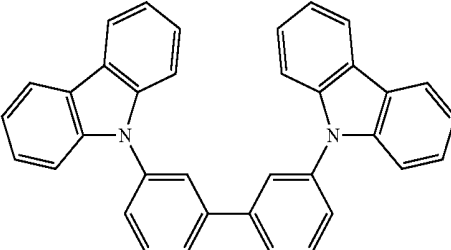
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| Tetraphenylene complexes                                     |   | US20050112407   |
| Metal phenoxypyridine compounds                              |   | WO2005030900  |
| Metal coordination complexes (e.g., Zn, Al with NAN ligands) |    | US20040137268,<br>US20040137267                           |
| Blue hosts   |   |   |
| Arylcarbazoles   | <br> | Appl. Phys. Lett, 82,<br>2422 (2003)<br><br>US20070190359 |

TABLE 1-continued

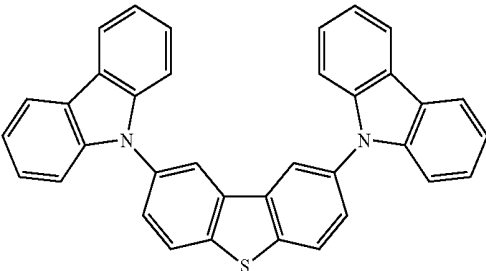
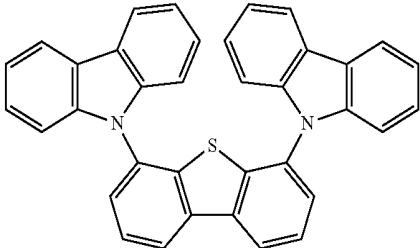
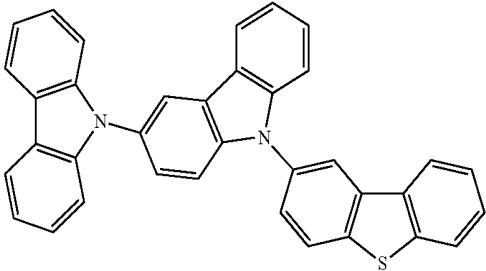
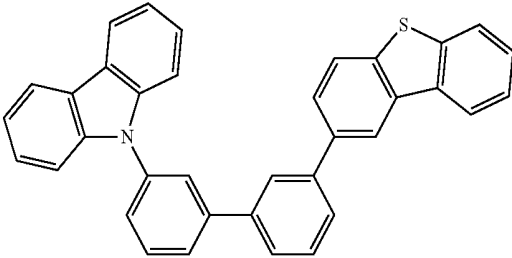
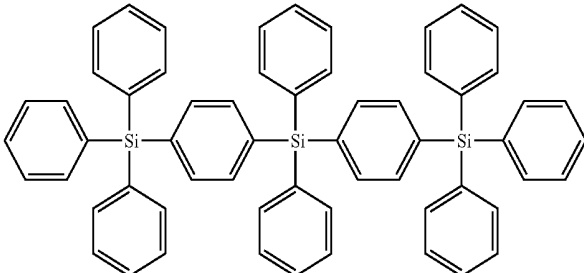
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|--|--|---------------------------------|
| Dibenzothiophene/<br>Dibenzofuran-carbazole<br>compounds |    | WO2006114966,<br>US20090167162  |
|  |    | US20090167162                   |
|  |  | WO2009086028                    |
|  |  | US20090030202,<br>US20090017330 |
| Silicon aryl compounds                                   |  | US20050238919                   |

TABLE 1-continued

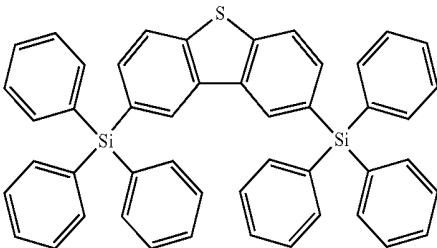
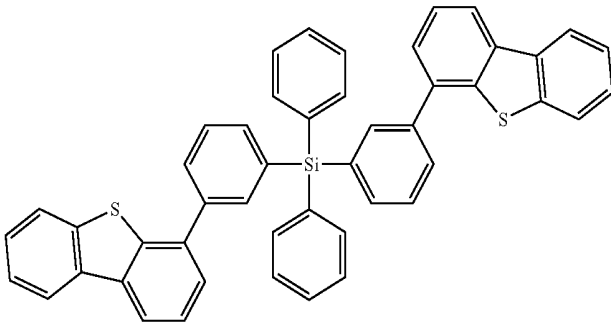
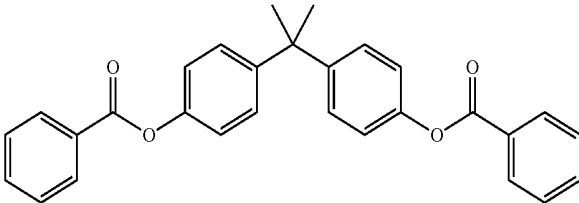
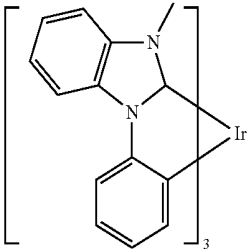
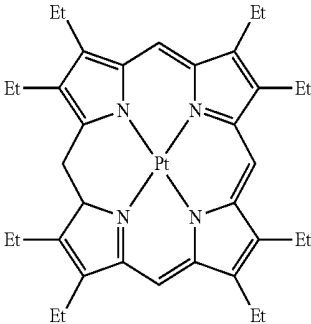
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|  |    | WO2009003898               |
| Silicon/Germanium<br>aryl compounds          |   | EP2034538A                 |
| Aryl benzoyl ester                           |  | WO2006100298               |
| High triplet metal<br>organometallic complex |   | U.S. Pat. No.<br>7,154,114 |
|  | Phosphorescent dopants<br>Red dopants  |                            |
| Heavy metal porphyrins<br>(e.g., PtOEP)      |   | Nature 395, 151<br>(1998)  |



TABLE 1-continued

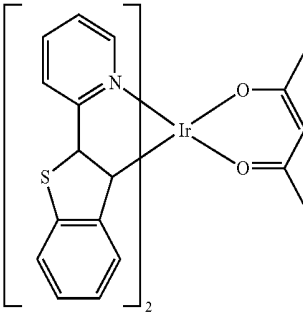
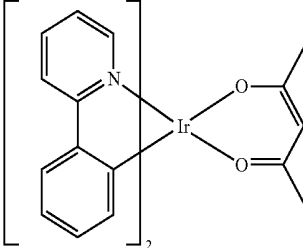
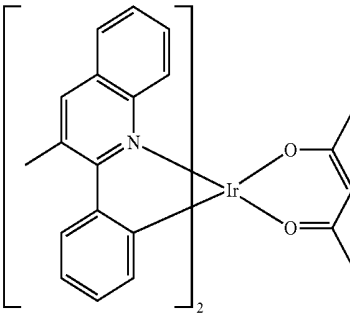
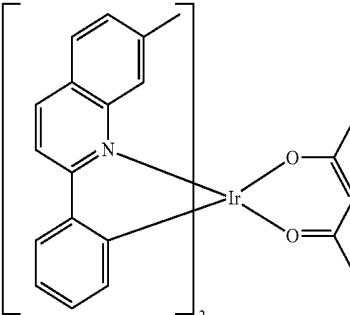
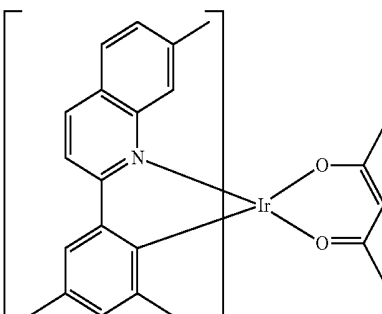
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| Iridium(III) organometallic complexes |    | Appl. Phys. Lett. 78, 1622 (2001) |
|                                       |   | US2006835469                      |
|                                       |  | US2006835469                      |
|                                       |  | US20060202194                     |
|                                       |  | US20060202194                     |

TABLE 1-continued

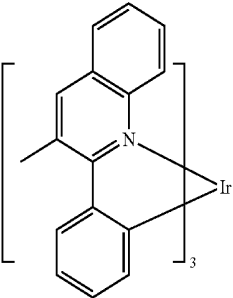
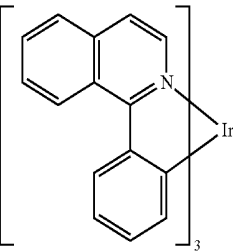
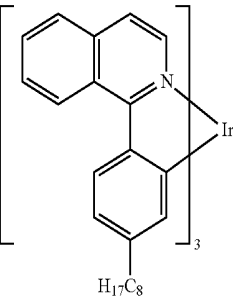
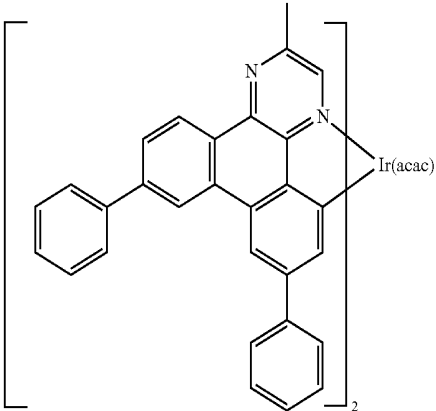
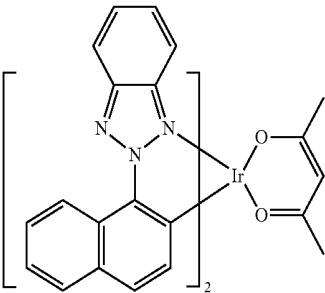
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|          |     | US20070087321              |
|          |     | US20070087321              |
|          |   | Adv. Mater. 19, 739 (2007) |
|          |  | WO2009100991               |
|          |   | WO2008101842               |

TABLE 1-continued

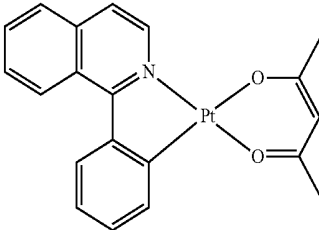
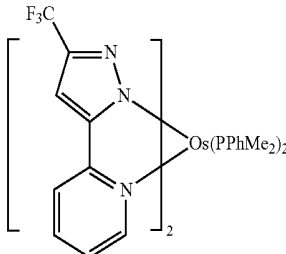
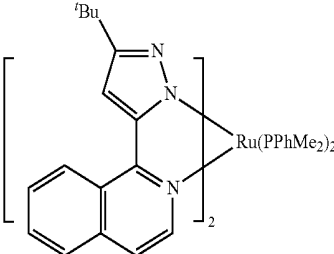
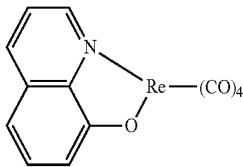
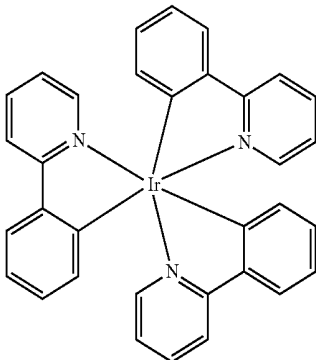
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| Platinum(II)<br>organometallic complexes  |    | WO2003040257                    |
| Osmium(III) complexes                     |   | Chem. Mater. 17, 3532<br>(2005) |
| Ruthenium(II) complexes                   |  | Adv. Mater. 17, 1059<br>(2005)  |
| Rhenium (I), (II), and (III)<br>complexes |  | US20050244673                   |
| Green dopants                             |   |                                 |
| Iridium(III) organometallic<br>complexes  |  | Inorg. Chem. 40, 1704<br>(2001) |
|   | and its derivatives   |                                 |

TABLE 1-continued

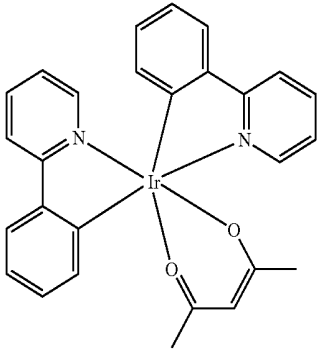
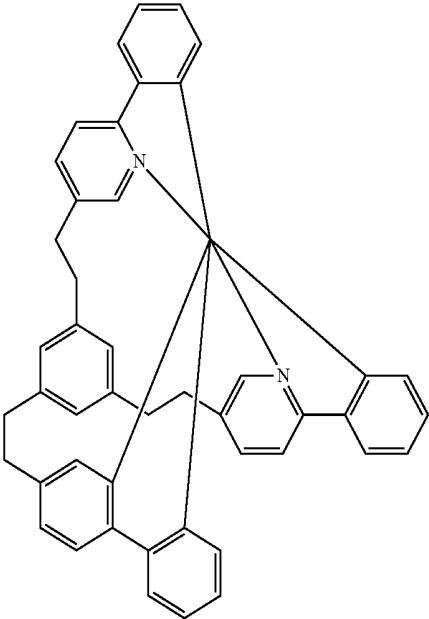
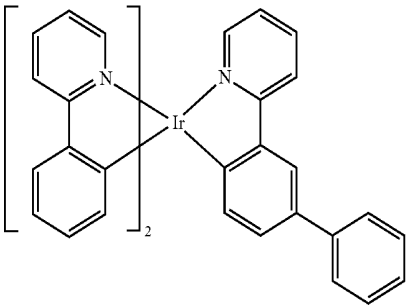
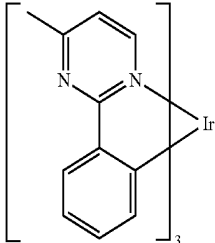
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|          |     | US20020034656              |
|          |    | U.S. Pat. No.<br>7,332,232 |
|          |  | US20090108737              |
|          |   | US20090039776              |

TABLE 1-continued

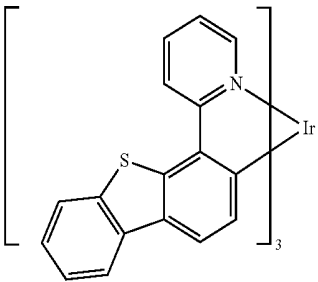
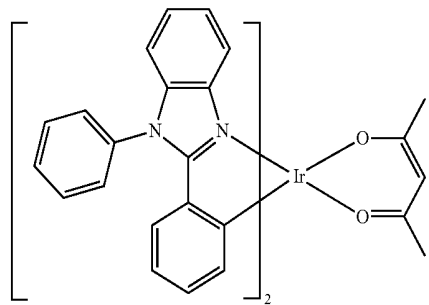
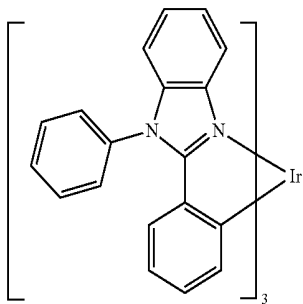
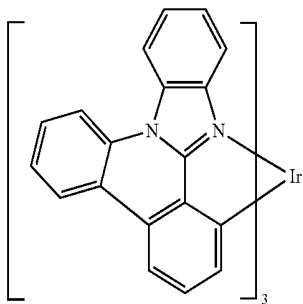
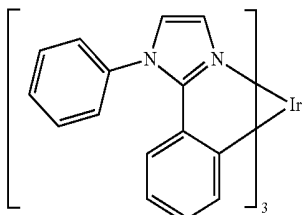
| MATERIAL | EXAMPLES OF MATERIAL  | PUBLICATIONS                    |
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|          |    | U.S. Pat. No.<br>6,921,915      |
|          |   | U.S. Pat. No.<br>6,687,266      |
|          |  | Chem. Mater. 16, 2480<br>(2004) |
|          |  | US20070190359                   |
|          |  | US20060008670<br>JP2007123392   |

TABLE 1-continued

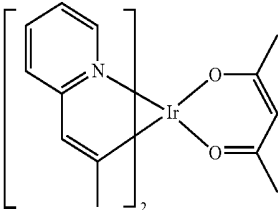
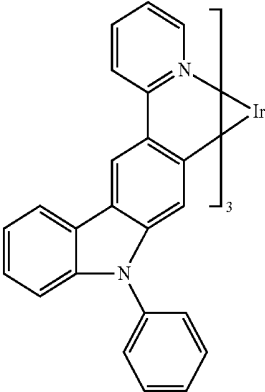
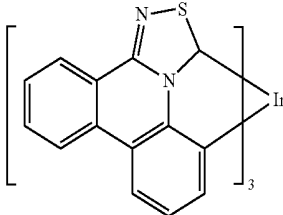
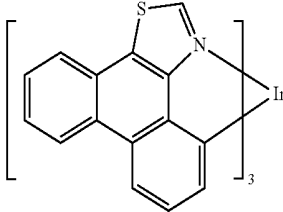
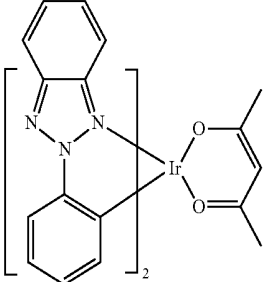
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|          |    | Adv. Mater. 16, 2003<br>(2004)          |
|          |   | Angew. Chem. Int. Ed.<br>2006, 45, 7800 |
|          |  | WO2009050290                            |
|          |  | US20090165846                           |
|          |  | US20080015355                           |

TABLE 1-continued

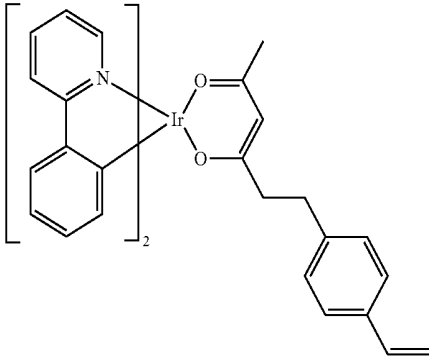
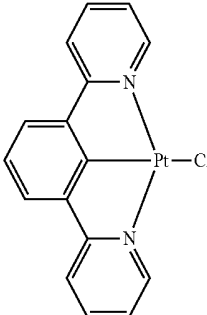
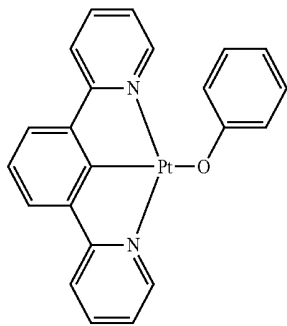
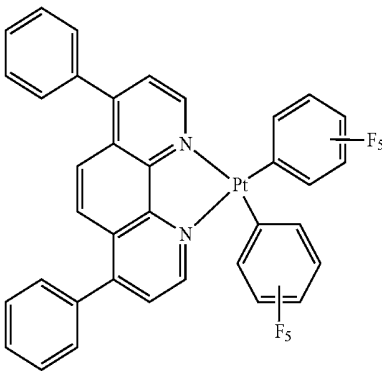
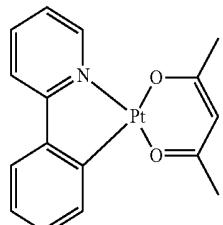
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|--|---|---|
| Monomer for polymeric metal organometallic compounds           |    | U.S. Pat. No. 7,250,226,<br>U.S. Pat. No. 7,396,598 |
| Pt(II) organometallic complexes, including polydentate ligands |   | Appl. Phys. Lett. 86, 153505 (2005)                 |
|  |  | Appl. Phys. Lett. 86, 153505 (2005)                 |
|  |  | Chem. Lett. 34, 592 (2005)                          |
|  |  | WO2002015645  |

TABLE 1-continued

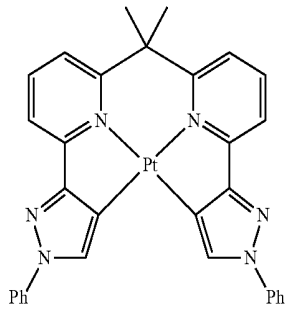
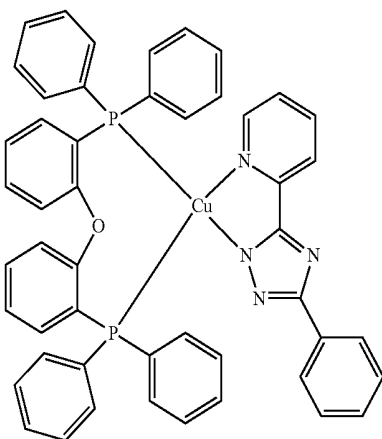
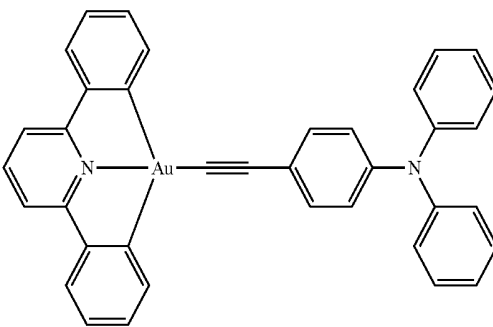
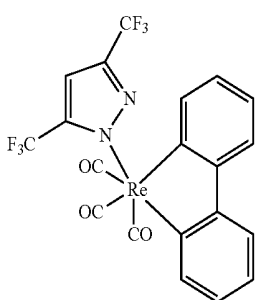
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|                        |     | US20060263635                |
| Cu complexes           |    | WO2009000673                 |
| Gold complexes         |  | Chem. Commun. 2906 (2005)    |
| Rhenium(III) complexes |   | Inorg. Chem. 42, 1248 (2003) |



TABLE 1-continued

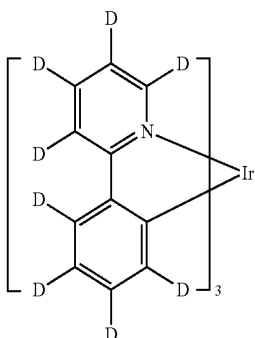
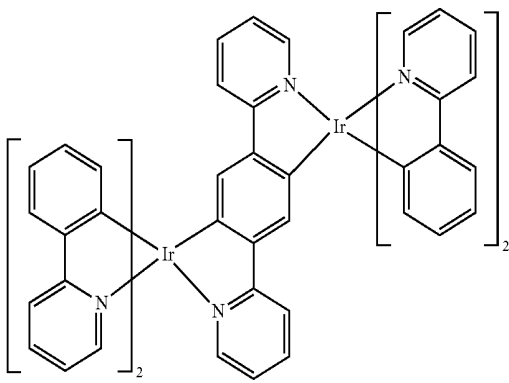
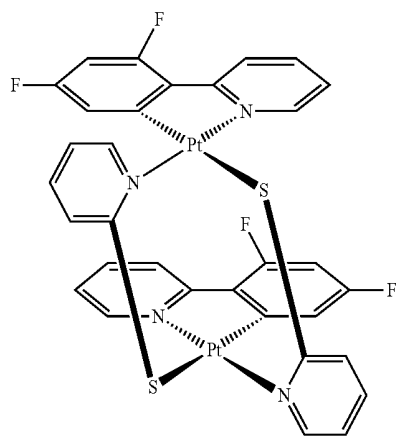
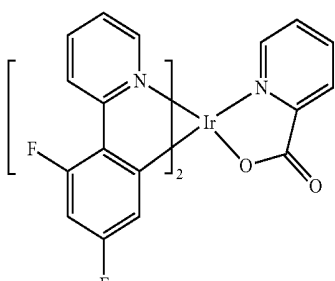
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|---|---|----------------------------|
| Deuterated organometallic complexes                     |    | US20030138657              |
| Organometallic complexes with two or more metal centers |  | US20030152802              |
|   |  | U.S. Pat. No.<br>7,090,928 |
| Blue dopants  |   |                            |
| Iridium(III) organometallic complexes                   |  | WO2002002714               |

TABLE 1-continued

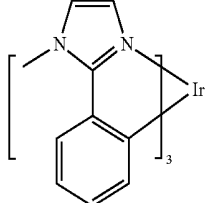
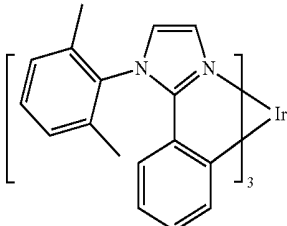
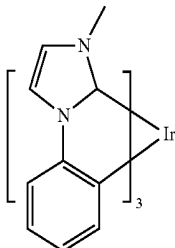
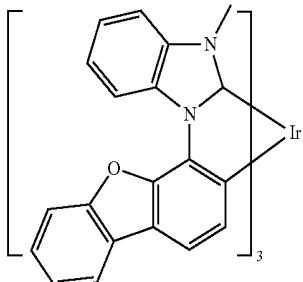
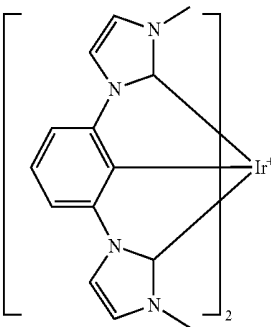
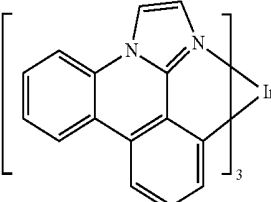
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|          |  | U.S. Pat. No.<br>7,534,505   |
|          |  | U.S. Pat. No.<br>7,445,855   |
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TABLE 1-continued

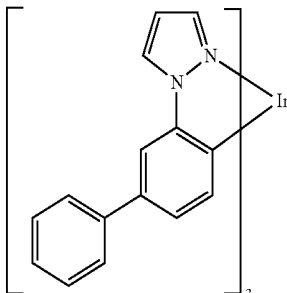
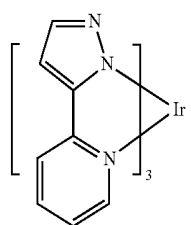
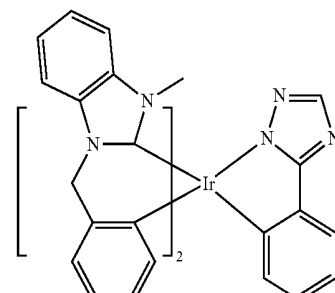
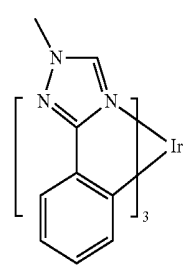
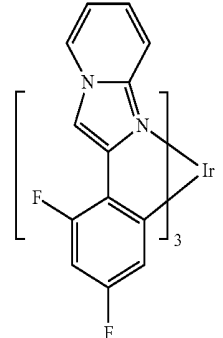
| MATERIAL | EXAMPLES OF MATERIAL  | PUBLICATIONS                          |
|----------|---|---------------------------------------|
|          |    | U.S. Pat. No.<br>7,338,722            |
|          |   | US20020134984                         |
|          |  | Angew. Chem. Int. Ed.<br>47, 1 (2008) |
|          |  | Chem. Mater. 18, 5119<br>(2006)       |
|          |  | Inorg. Chem. 46, 4308<br>(2007)       |

TABLE 1-continued

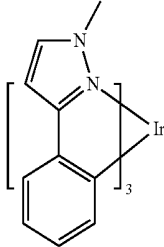
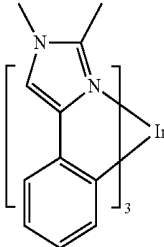
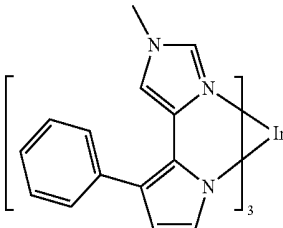
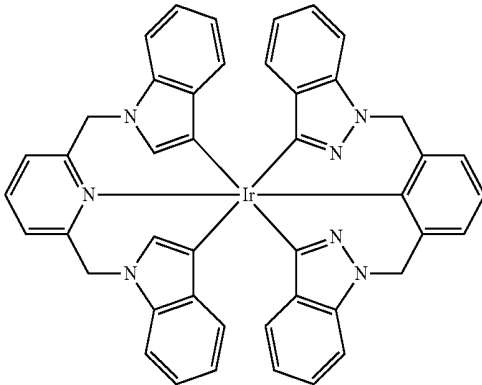
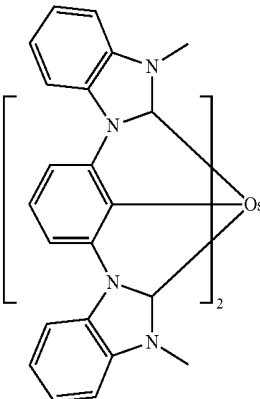
| MATERIAL             | EXAMPLES OF MATERIAL   | PUBLICATIONS               |
|----------------------|--|----------------------------|
|                      |     | WO2005123873               |
|                      |     | WO2005123873               |
|                      |    | WO2007004380               |
|                      |  | WO2006082742               |
| Osmium(II) complexes |   | U.S. Pat. No.<br>7,279,704 |

TABLE 1-continued

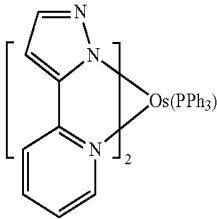
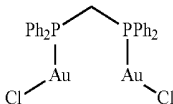
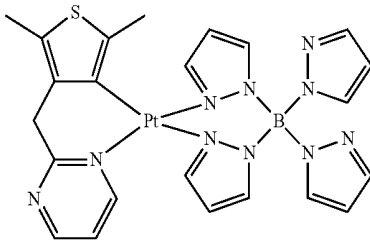
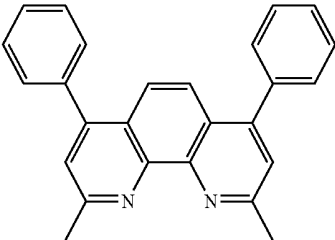
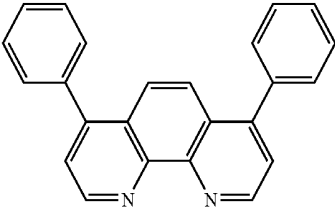
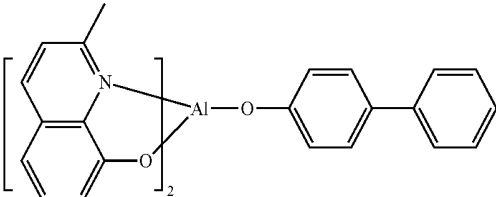
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|--|--|--------------------------------------|
|  |     | Organometallics<br>23, 3745 (2004)   |
| Gold complexes                               |     | Appl. Phys. Lett. 74,<br>1361 (1999) |
| Platinum(II) complexes                       |    | WO2006098120,<br>WO2006103874        |
| Exciton/hole blocking layer materials        |  |                                      |
| Bathocuprine compounds<br>(e.g., BCP, BPhen) |   | Appl. Phys. Lett. 75, 4<br>(1999)    |
|  |   | Appl. Phys. Lett. 79,<br>449 (2001)  |
| Metal 8-hydroxyquinolates<br>(e.g., BAlq)    |  | Appl. Phys. Lett. 81,<br>162 (2002)  |

TABLE 1-continued

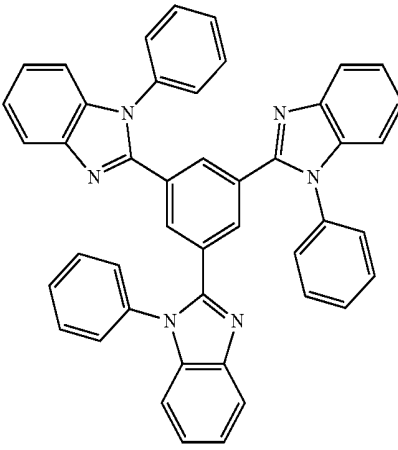
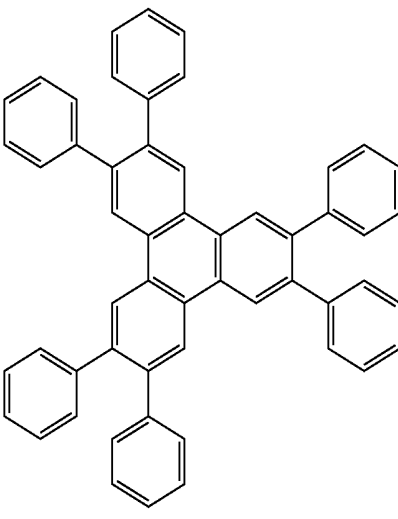
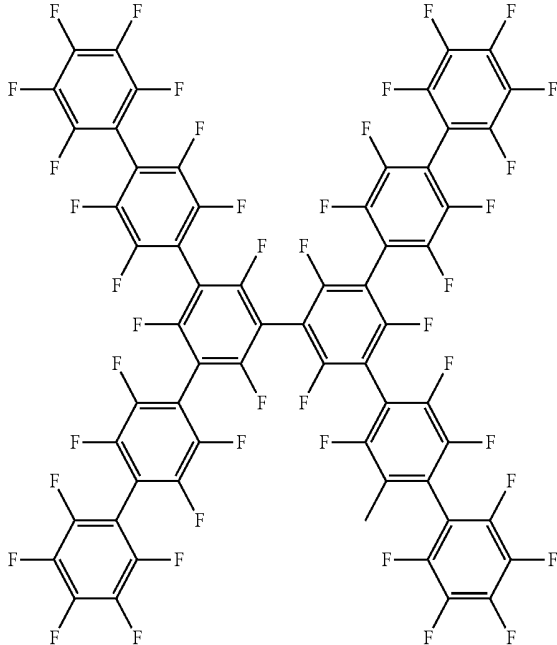
| MATERIAL  | EXAMPLES OF MATERIAL   | PUBLICATIONS                     |
|---|--|----------------------------------|
| 5-member ring electron deficient heterocycles such as triazole, oxadiazole, imidazole, benzoimidazole |     | Appl. Phys. Lett. 81, 162 (2002) |
| Triphenylene compounds  |    | US20050025993                    |
| Fluorinated aromatic compounds  |  | Appl. Phys. Lett. 79, 156 (2001) |

TABLE 1-continued

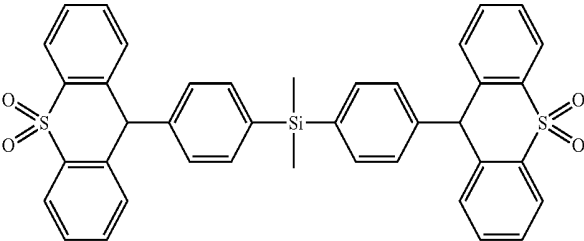
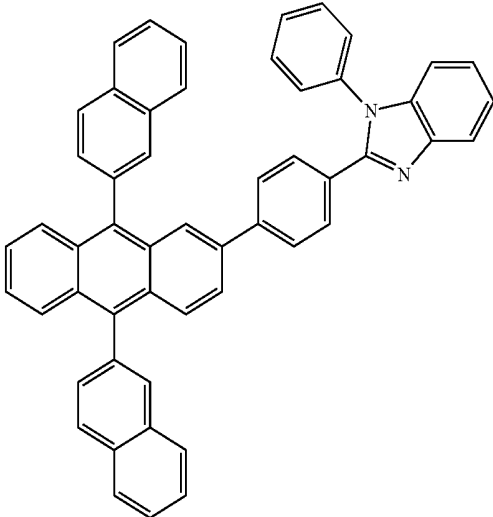
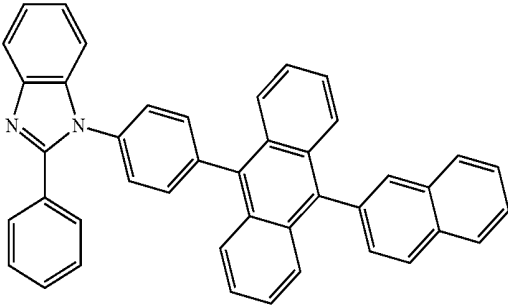
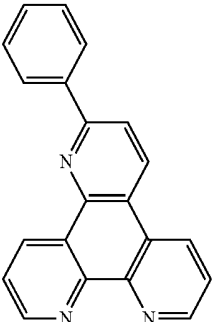
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|------------------------------------|--|---------------|
| Phenothiazine-S-oxide              |    | WO2008132085  |
| Electron transporting materials    |  |               |
| Anthracene-benzimidazole compounds |   | WO2003060956  |
|                                    |  | US20090179554 |
| Aza triphenylene derivatives       |   | US20090115316 |

TABLE 1-continued

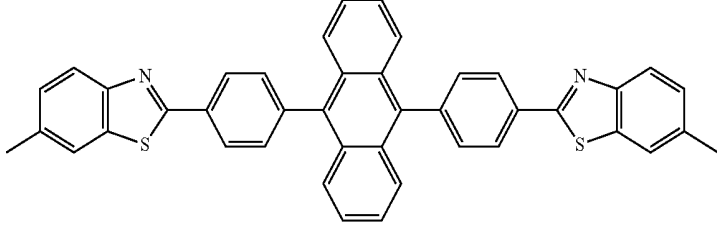
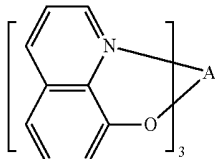
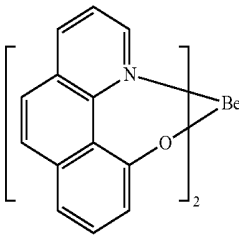
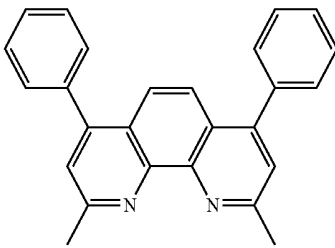
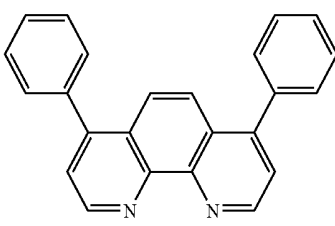
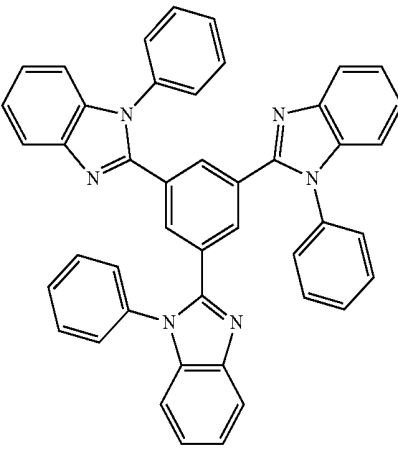
| MATERIAL  | EXAMPLES OF MATERIAL  | PUBLICATIONS  |
|---|---|---|
| Anthracene-benzothiazole compounds  |   | Appl. Phys. Lett. 89, 063504 (2006)                         |
| Metal 8-hydroxyquinolates (e.g., Alq <sub>3</sub> , Zrq <sub>4</sub> )                                |    | Appl. Phys. Lett. 51, 913 (1987)<br>U.S. Pat. No. 7,230,107 |
| Metal hydroxybenoquinolates   |   | Chem. Lett. 5, 905 (1993)                                   |
| Bathocuprine compounds such as BCP, BPhen, etc  |  | Appl. Phys. Lett. 91, 263503 (2007)                         |
|   |  | Appl. Phys. Lett. 79, 449 (2001)                            |
| 5-member ring electron deficient heterocycles (e.g., triazole, oxadiazole, imidazole, benzoimidazole) |  | Appl. Phys. Lett. 74, 865 (1999)                            |



TABLE 1-continued

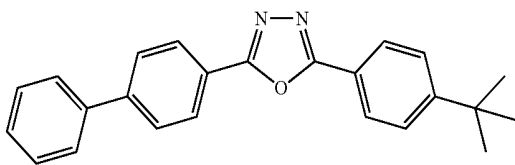
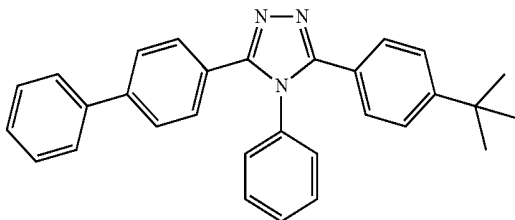
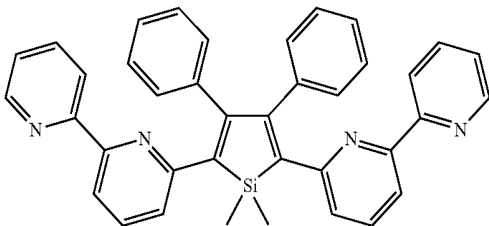
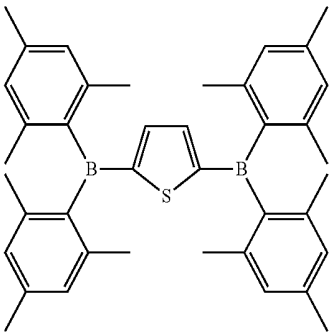
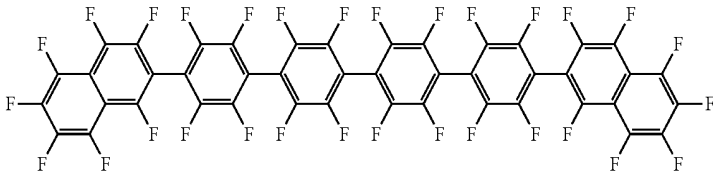
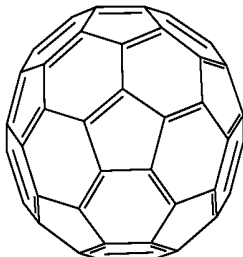
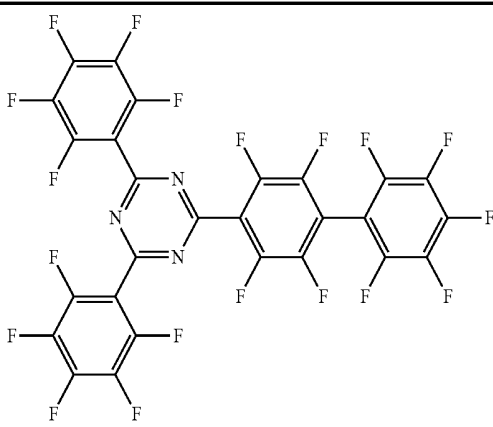
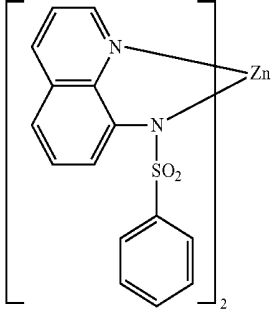
| MATERIAL                       | EXAMPLES OF MATERIAL   | PUBLICATIONS                         |
|--------------------------------|--|--------------------------------------|
|                                |    | Appl. Phys. Lett. 55, 1489 (1989)    |
|                                |    | Jpn. J. Apply. Phys. 32, L917 (1993) |
| Silole compounds               |   | Org. Electron. 4, 113 (2003)         |
| Arylborane compounds           |   | J. Am. Chem. Soc. 120, 9714 (1998)   |
| Fluorinated aromatic compounds |  | J. Am. Chem. Soc. 122, 1832 (2000)   |
| Fullerene (e.g., C60)          |   | US20090101870                        |

TABLE 1-continued

| MATERIAL           | EXAMPLES OF MATERIAL   | PUBLICATIONS            |
|--------------------|--|-------------------------|
| Triazine complexes |  | US20040036077           |
| Zn (NΔN) complexes |  | U.S. Pat. No. 6,528,187 |

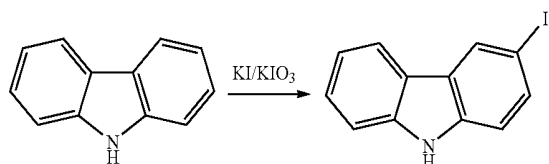
## EXPERIMENTAL

## Compound Examples

## Example 1

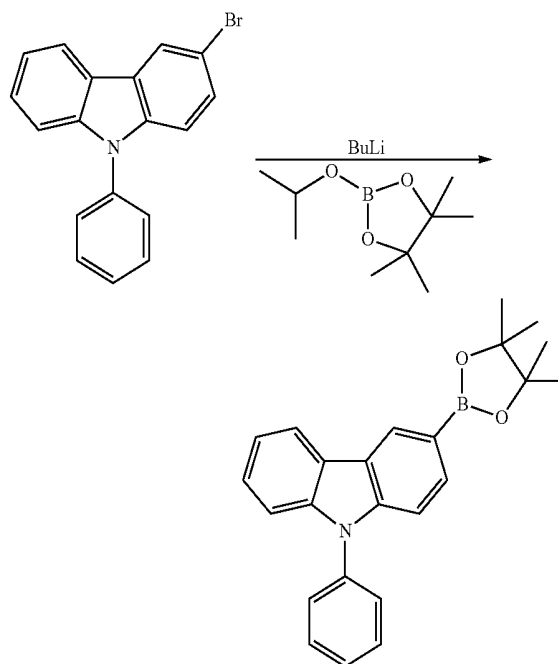
## Synthesis of Compound 1

[0100]



## Synthesis of 3-iodo-9H-carbazole

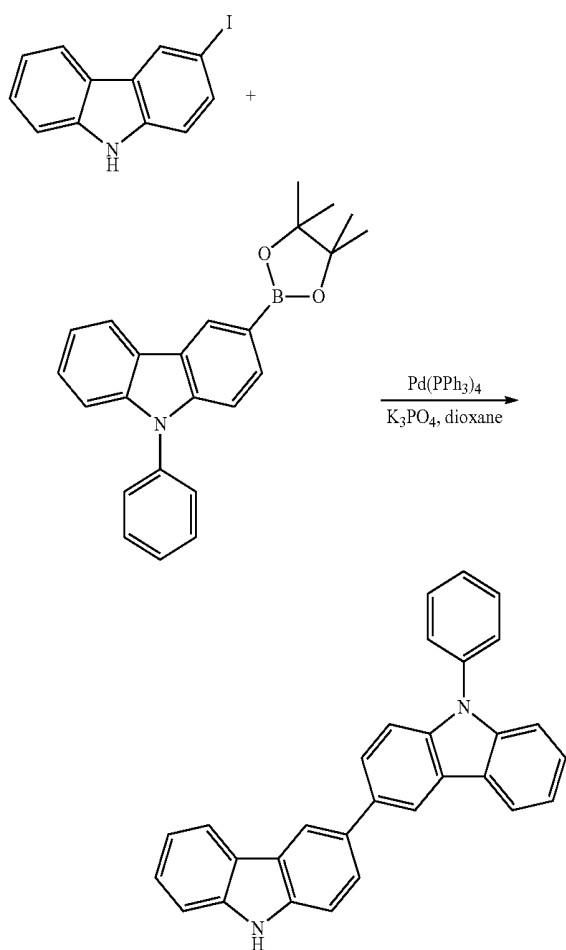
**[0101]** To a solution of 9H-carbazole (5.57 g, 33.3 mmol) and KI (3.68 g, 22.2 mmol) in AcOH (92 mL) was heated to 100° C. for 1 h. KIO<sub>3</sub> (3.57 g, 16.7 mmol) was added in portions to the solution, and the resulting mixture was stirred for another 2 h at 100° C. The mixture was poured into water (500 mL) and the precipitation was collected by filtration and washed with hot water. Recrystallization was made in DCM to afford 6.8 g (70%) of product as a white solid.



## Synthesis of 9-phenyl-3-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-9H-carbazole

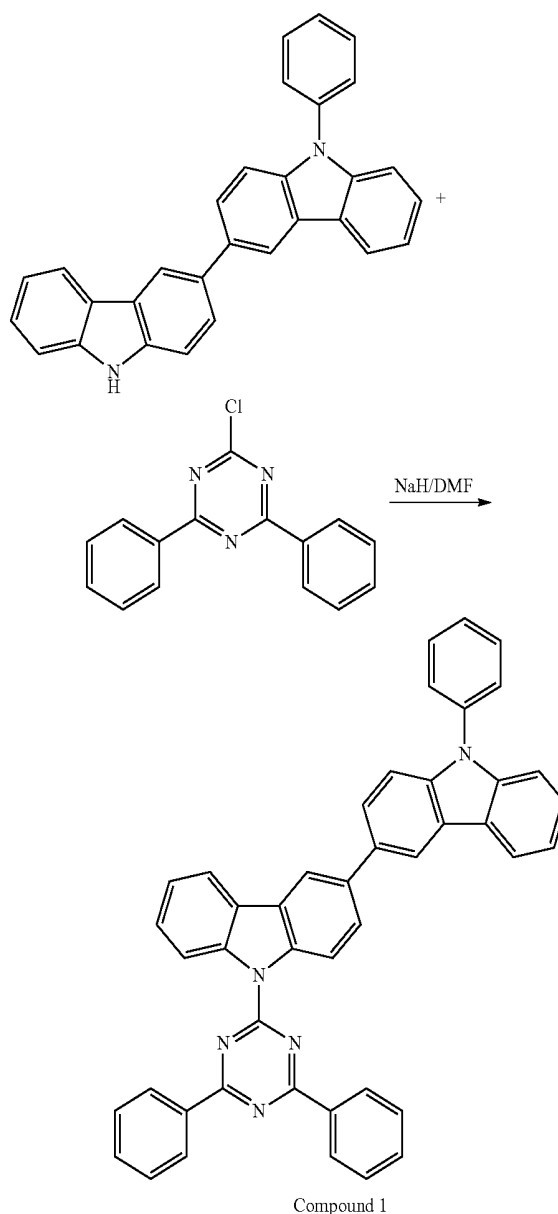
**[0102]** To a solution of 3-bromo-9-phenyl-9H-carbazole (20.3 g, 63 mmol) in THF (150 mL) at -78° C. was added

47.25 mL (75.8 mmol) of n-butyllithium (1.6 M in hexane). The mixture was stirred at  $-78^{\circ}\text{C}$ . for 1 h. 21 mL (100 mmol) of 2-isopropoxy-4,4,5,5-tetramethyl-[1,3,2]-dioxaborolane was added to the solution, and the resulting mixture was warmed to room temperature and stirred for 8 h. The mixture was poured into water and extracted with dichloromethane. The organic extracts were washed with brine and dried over magnesium sulfate. The solvent was removed by rotary evaporation, and recrystallization was made in hexane to afford 19.3 g (83%) of product as a white solid.



#### Synthesis of 3-(9-phenyl-9H-carbazol-3-yl)-9H-carbazole

**[0103]** To a solution of 3-iodo-9H-carbazole (879 mg, 3.0 mmol),  $\text{Pd}(\text{PPh}_3)_4$  (165 mg, 0.15 mmol), 9-phenyl-3-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-9H-carbazole (1.29 g, 4.5 mmol) and  $\text{K}_3\text{PO}_4$  (1.8 g, 18.0 mmol) in dioxane (5 mL). The solution was heated to  $85^{\circ}\text{C}$ . with vigorous stirring for 48 h under argon atmosphere. The mixture was poured into water and extracted with DCM. The organic extracts were washed with brine and dried over  $\text{MgSO}_4$ . The solvent was removed by rotary evaporation, and recrystallization was made in DCM to afford 900 mg (74%) of product.



#### Synthesis of 9-(4,6-diphenyl-1,3,5-triazin-2-yl)-3-(9-phenyl-9H-carbazol-3-yl)-9H-carbazole (Compound 1)

**[0104]** To a solution of sodium hydride (100 mg, 3.0 mmol) and 3-(9-phenyl-9H-carbazol-3-yl)-9H-carbazole (816 mg, 2.0 mmol) in dry DMF (40 mL) was stirred at room temperature for 1 h under argon atmosphere. 2-Chloro-4,6-diphenyl-1,3,5-triazine (448 mg, 1.67 mmol) was added to the solution at room temperature, then refluxed overnight. The mixture was poured into water and the precipitation was collected by filtration and washed with water, methanol and DCM to get 800 mg (75%) yellow solid.

#### Device Examples

**[0105]** All device examples were fabricated by high vacuum ( $<10^{-7}$  Torr) thermal evaporation. The anode electrode is 800 Å of indium tin oxide (ITO). The cathode con-



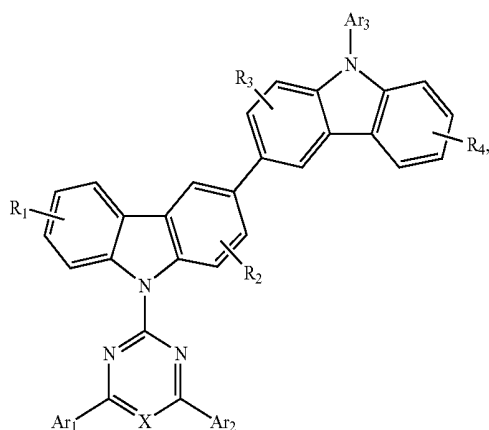
TABLE 3-continued

| VTE device data |       |          |                 |      |             |              |         |           |                       |                          |  |
|-----------------|-------|----------|-----------------|------|-------------|--------------|---------|-----------|-----------------------|--------------------------|--|
|                 |       | 1931 CIE |                 |      | FWHM        | At 1000 nits |         |           |                       | At 40 mA/cm <sup>2</sup> |  |
|                 | x     | y        | $\lambda_{max}$ | (nm) | Voltage (V) | LE (Cd/A)    | EQE (%) | PE (lm/W) | L <sub>0</sub> (nits) | LT80% (h)                |  |
| Example         |       |          |                 |      |             |              |         |           |                       |                          |  |
| Comp.           | 0.317 | 0.630    | 520             | 64   | 5.2         | 54.4         | 15.1    | 32.6      | 16,264                | 29                       |  |
| Example 2       |       |          |                 |      |             |              |         |           |                       |                          |  |

[0110] Device Examples 1 and 2 showed green PHOLEDs with Compound 1 as host with different E1 doping concentrations. The comparative examples used H3 (i.e., CBP, a commonly used PHOLED host) as the host. As can be seen from the table, devices with Compound 1 as host had comparative operating voltage, slightly lower efficiency than devices with H3 as the host. However, the device operating lifetime was much higher than comparative examples. Device Example 1 almost doubled the lifetime of Comparative Example 1 (86 h vs 46 h) and Device Example 2 almost tripled the lifetime of Comparative Example 2 (83 h vs. 29 h). Therefore, Compound 1 is an excellent host material for phosphorescent OLEDs.

[0111] It is understood that the various embodiments described herein are by way of example only, and are not intended to limit the scope of the invention. For example, many of the materials and structures described herein may be substituted with other materials and structures without deviating from the spirit of the invention. The present invention as claimed may therefore include variations from the particular examples and preferred embodiments described herein, as will be apparent to one of skill in the art. It is understood that various theories as to why the invention works are not intended to be limiting.

1. A compound having the formula:



wherein R<sub>1</sub>, R<sub>2</sub>, R<sub>3</sub>, and R<sub>4</sub> may represent mono, di, tri, or tetra substitutions;

wherein R<sub>1</sub>, R<sub>2</sub>, R<sub>3</sub>, and R<sub>4</sub> are independently selected from the group consisting of hydrogen, alkyl, alkoxy, amino, alkenyl, alkynyl, aryl and heteroaryl;

wherein Ar<sub>1</sub>, Ar<sub>2</sub>, and Ar<sub>3</sub> are independently selected from aryl or heteroaryl; and

wherein X is C or N.

2. The compound of claim 1, wherein Ar<sub>1</sub>, Ar<sub>2</sub>, and Ar<sub>3</sub> are further substituted.

3. The compound of claim 1, wherein Ar<sub>1</sub>, Ar<sub>2</sub>, and Ar<sub>3</sub> are independently selected from the group consisting of phenyl, pyridine, naphthalene, biphenyl, terphenyl, fluorene, dibenzofuran, dibenzothiophene, phenanthrene, and triphenylene; and

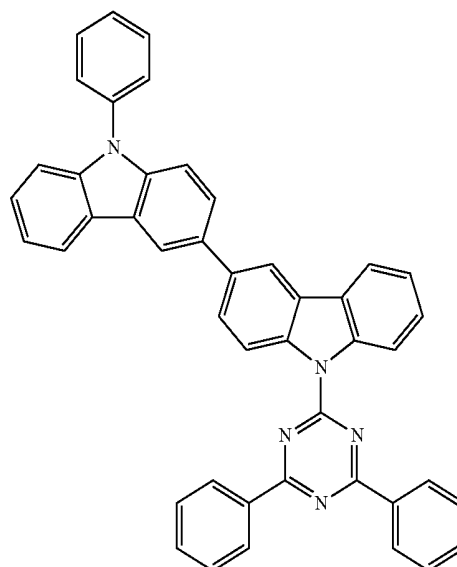
wherein Ar<sub>1</sub>, Ar<sub>2</sub>, and Ar<sub>3</sub> are independently further substituted with a substituent selected from the group consisting of hydrogen, alkyl, alkoxy, amino, alkenyl, alkynyl, aryl and heteroaryl, wherein the substituent is not an aryl or heteroaryl fused directly to Ar<sub>1</sub>, Ar<sub>2</sub>, and Ar<sub>3</sub>.

4. The compound of claim 1, wherein Ar<sub>1</sub> and Ar<sub>2</sub> are independently selected from the group consisting of phenyl, pyridine, and naphthalene.

5. The compound of claim 1, wherein Ar<sub>3</sub> is selected from the group consisting of phenyl, biphenyl, dibenzofuran, and dibenzothiophene.

6. The compound of claim 1, wherein R<sub>1</sub>, R<sub>2</sub>, R<sub>3</sub>, and R<sub>4</sub> are hydrogen.

7. The compound of claim 1, wherein the compound is selected from the group consisting of:

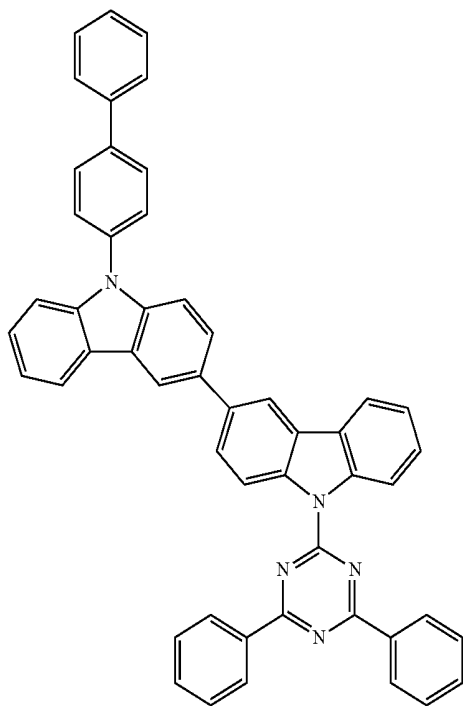


Compound 1

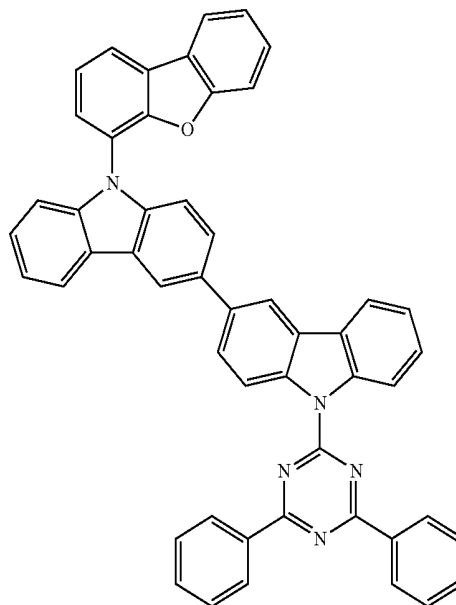
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Compound 2

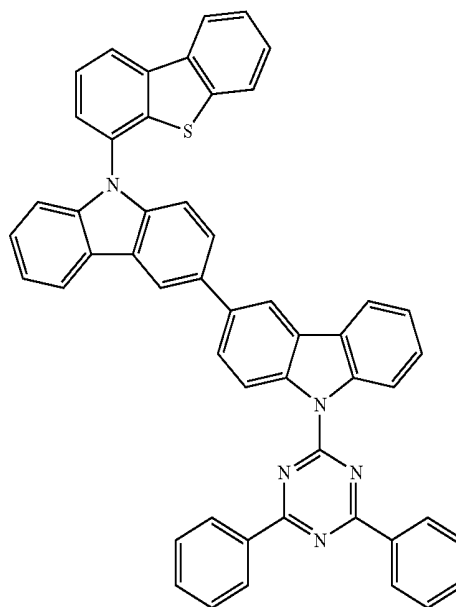
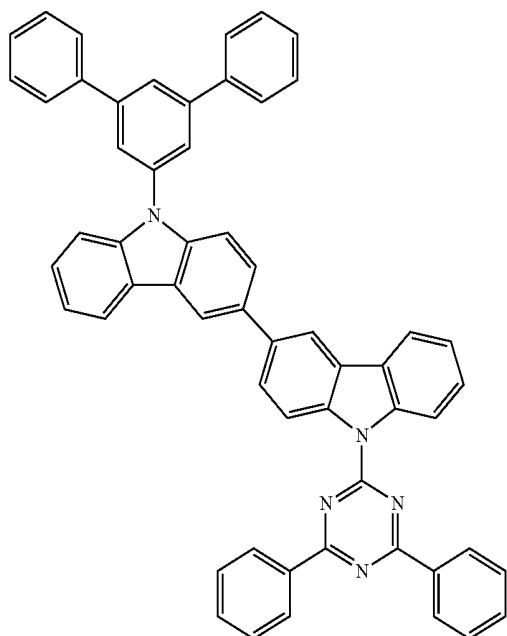


Compound 4



Compound 5

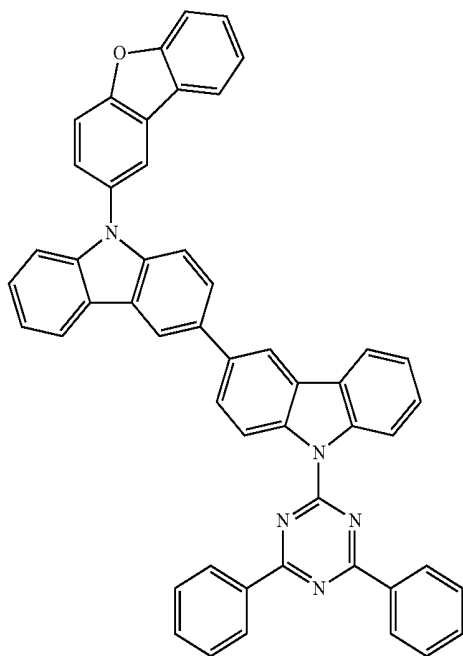
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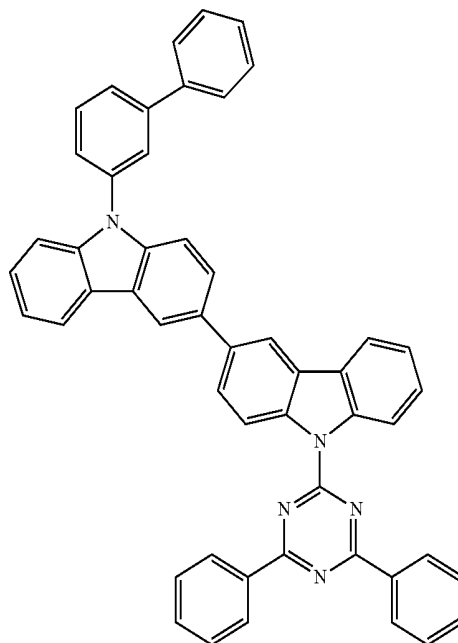
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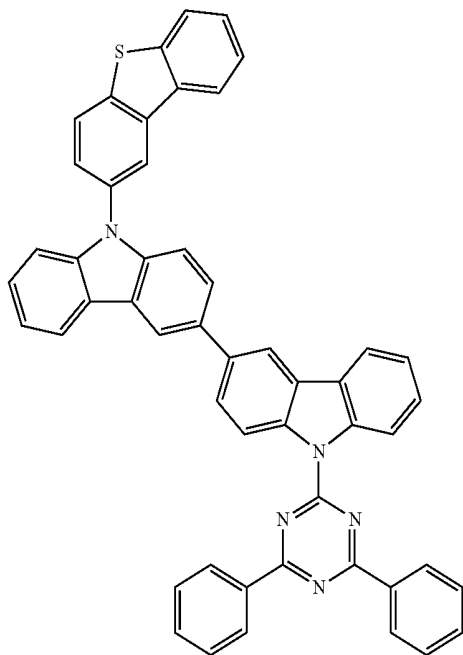
Compound 6



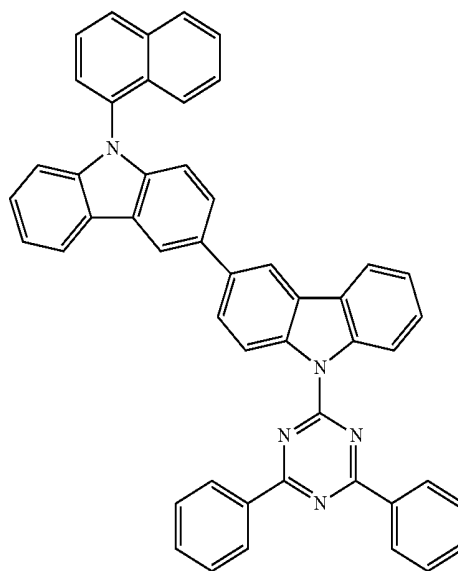
Compound 8



Compound 7



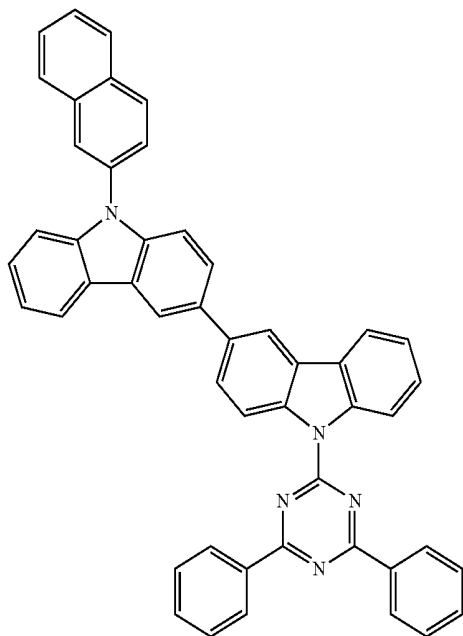
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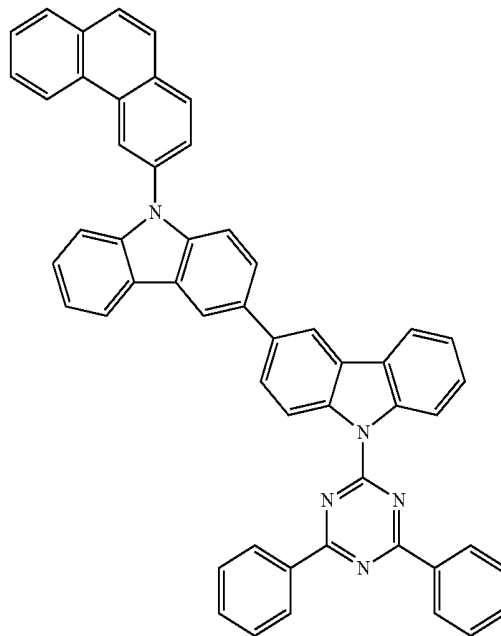
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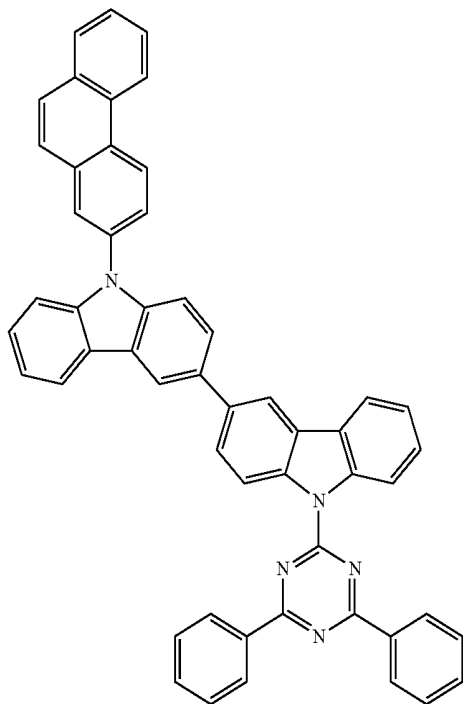
Compound 10



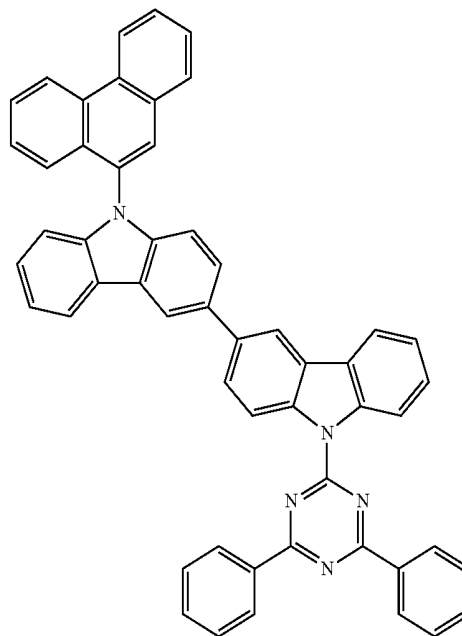
Compound 12



Compound 11



Compound 13

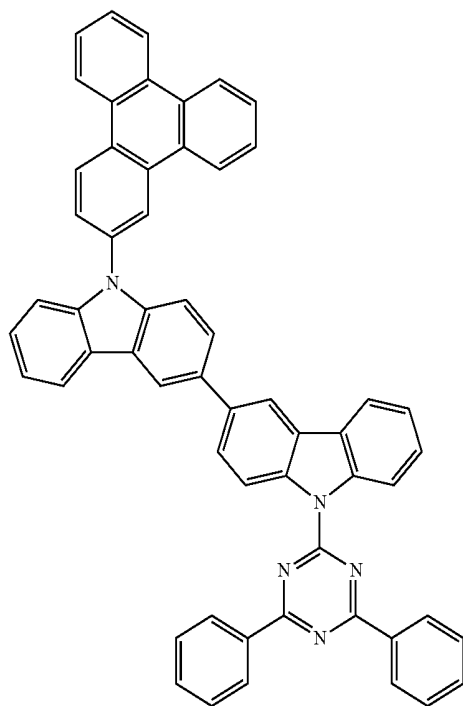




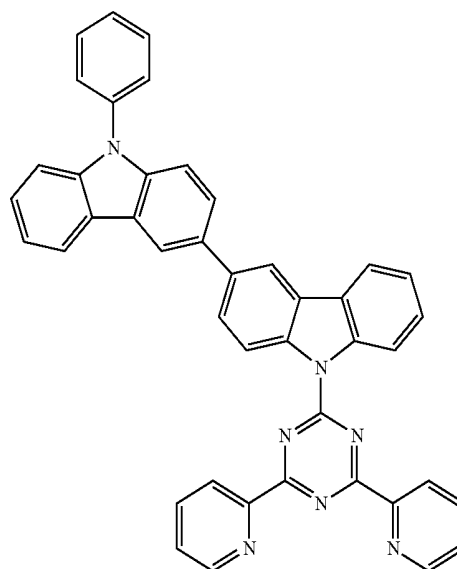
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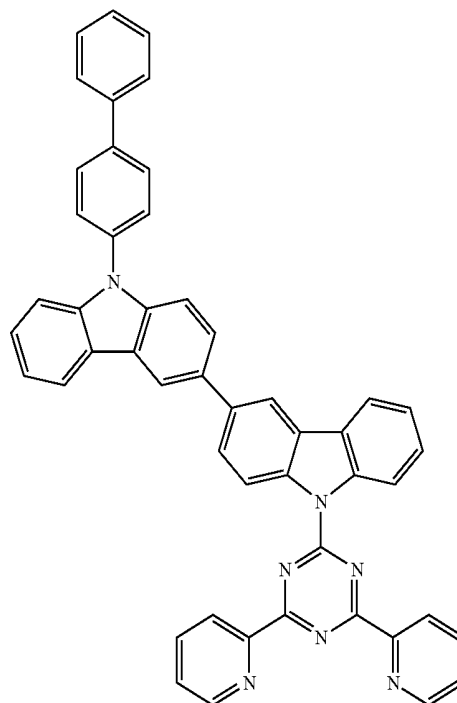
Compound 14



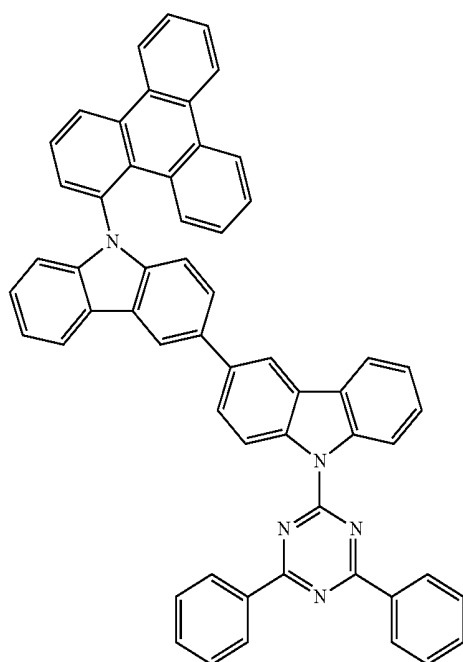
Compound 16



Compound 17



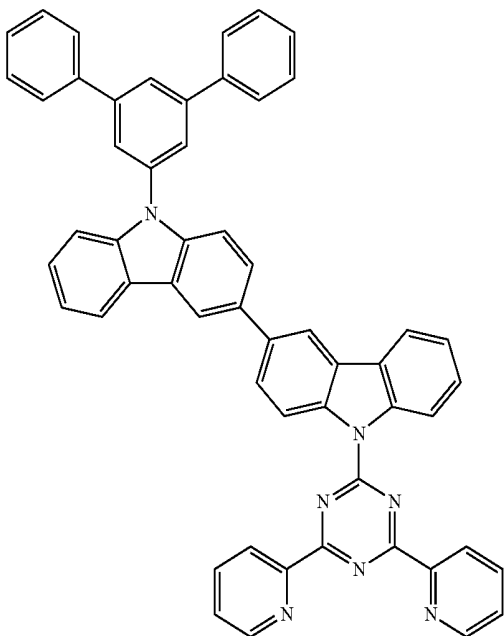
Compound 15



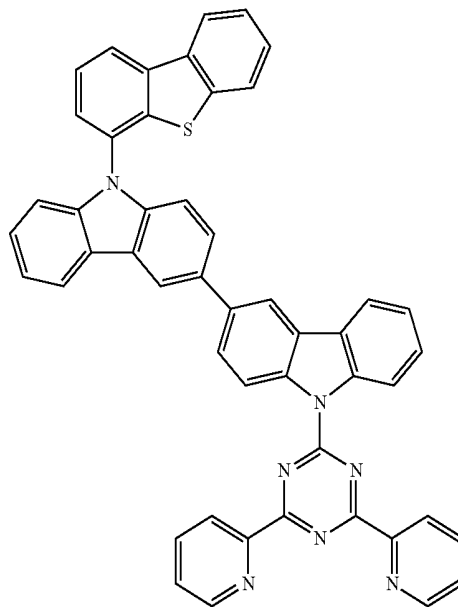
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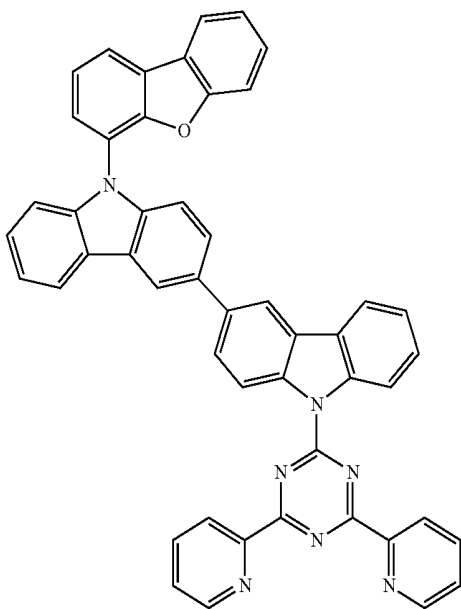
Compound 18



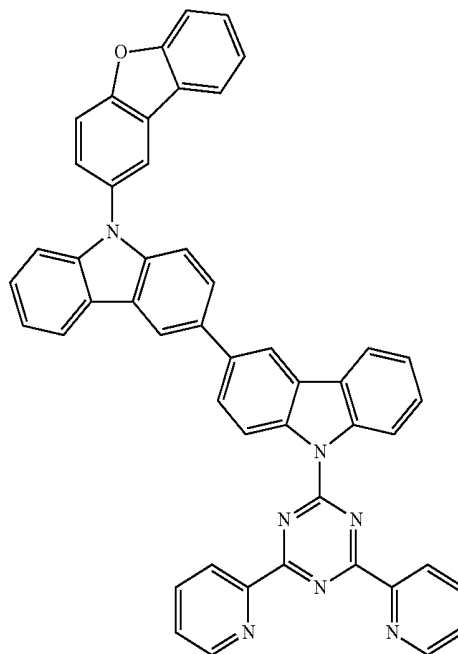
Compound 20



Compound 19



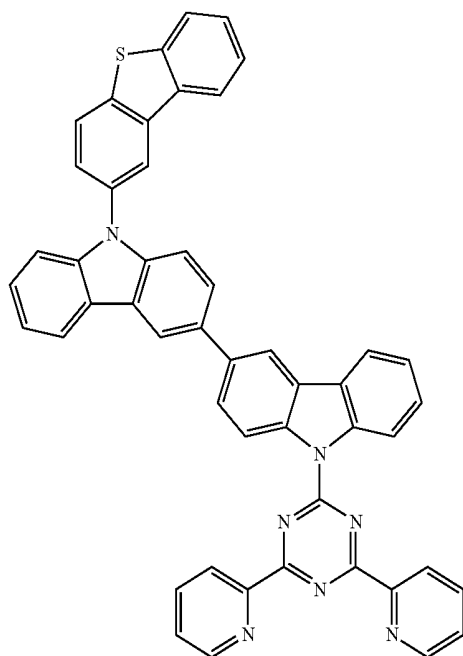
Compound 21



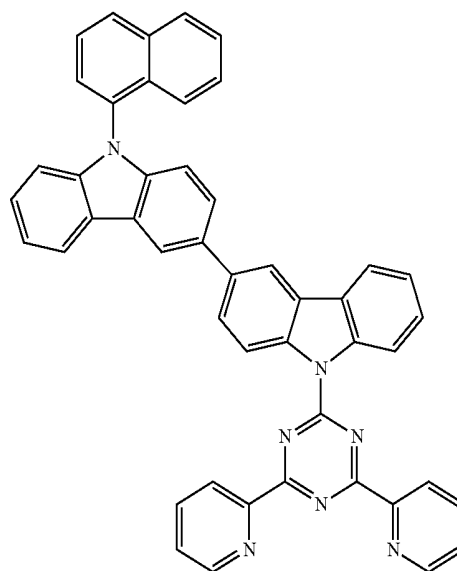
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Compound 22

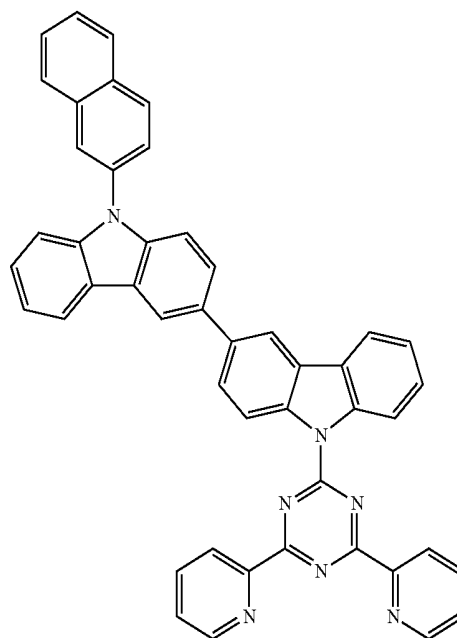
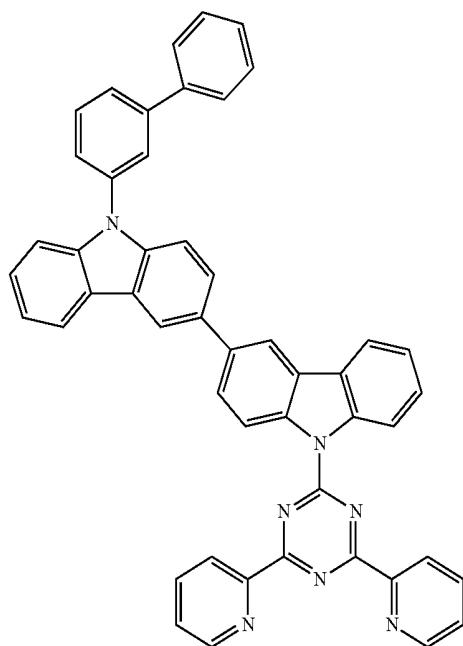


Compound 24



Compound 25

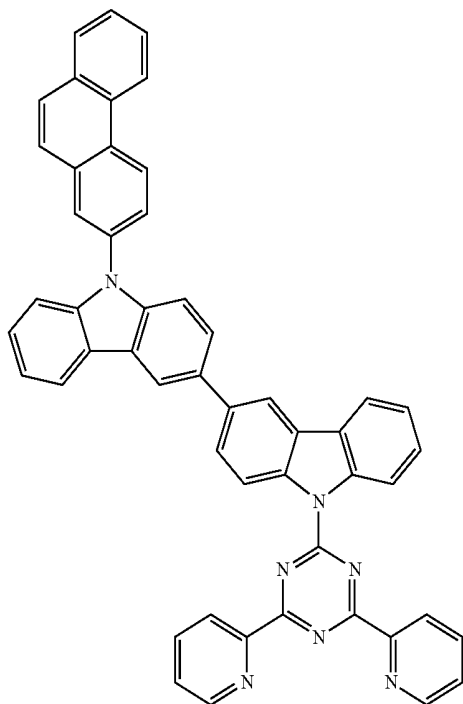
Compound 23



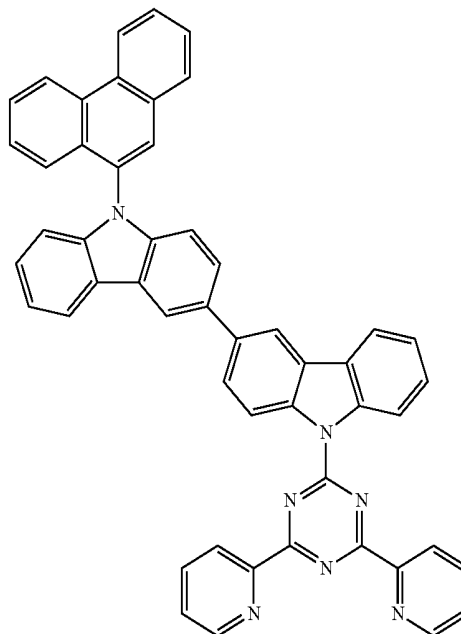
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Compound 26

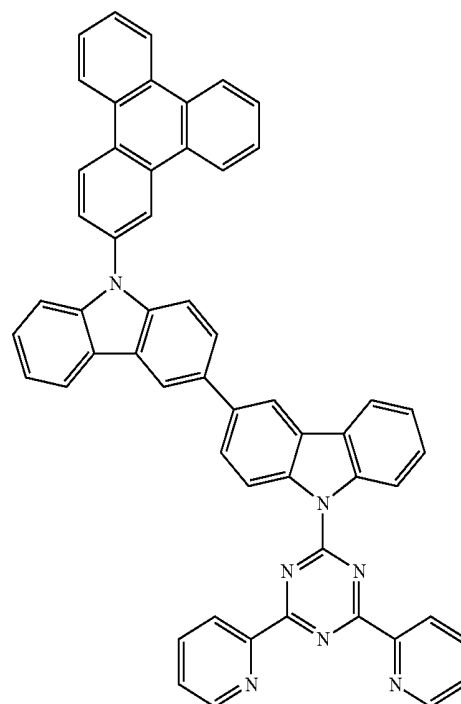
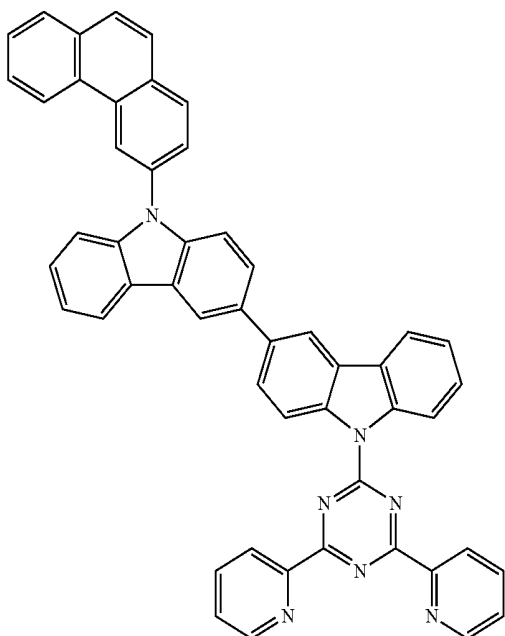


Compound 28



Compound 29

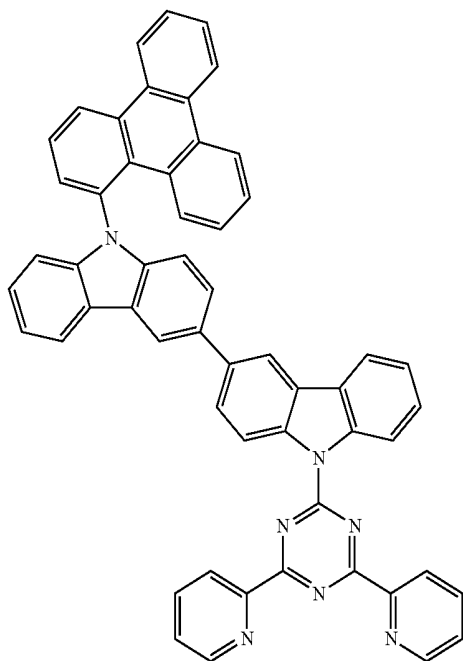
Compound 27



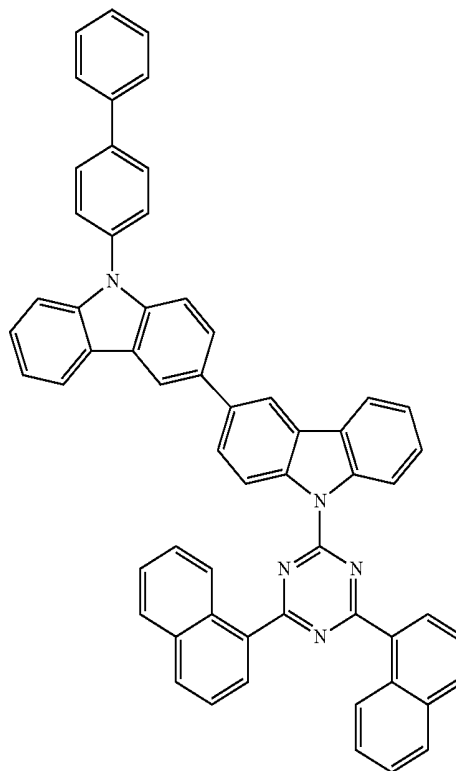
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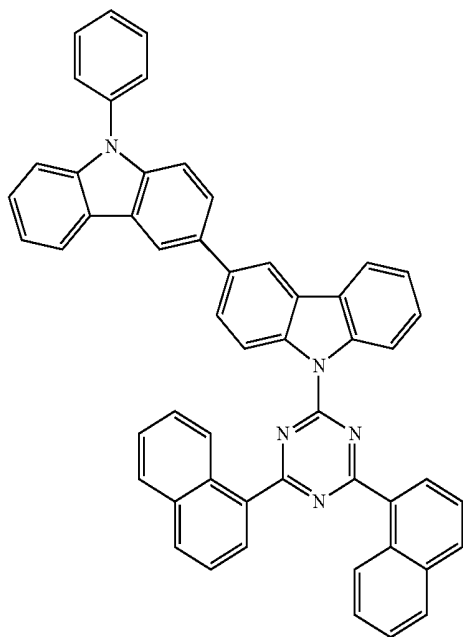
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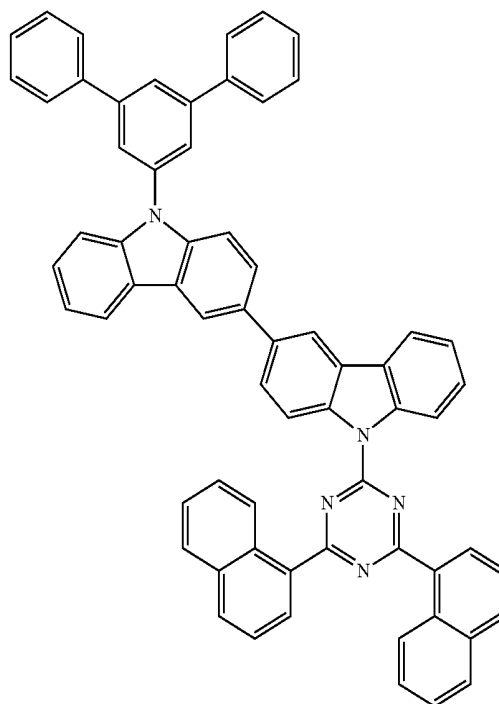
Compound 32



Compound 31

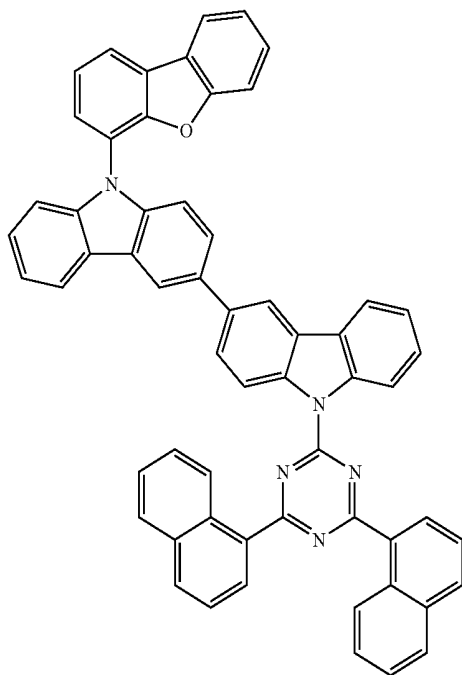


Compound 33



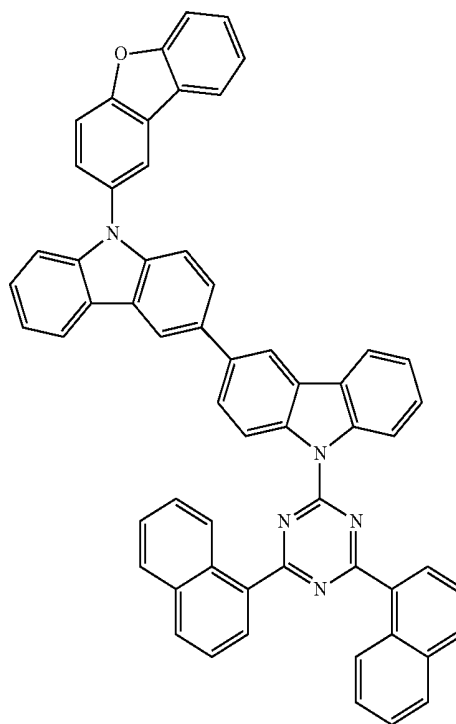
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Compound 34

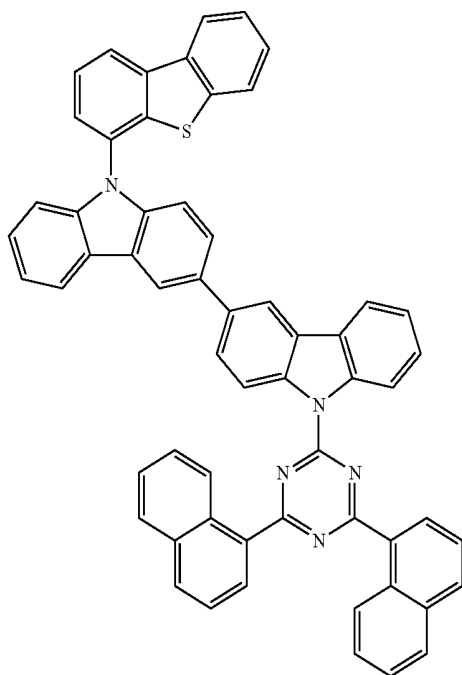


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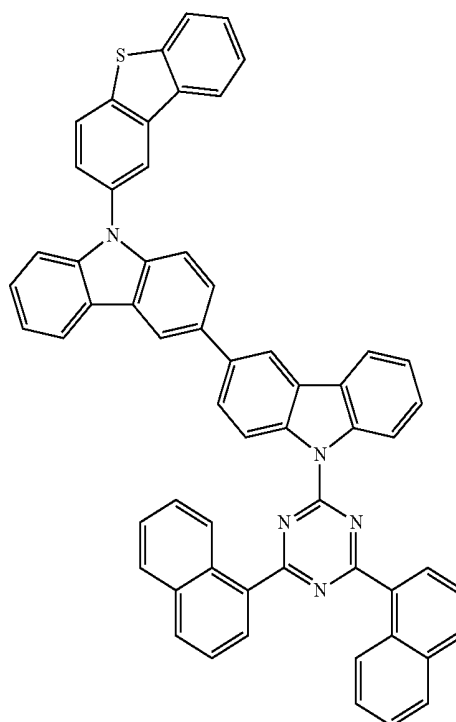
Compound 36



Compound 35



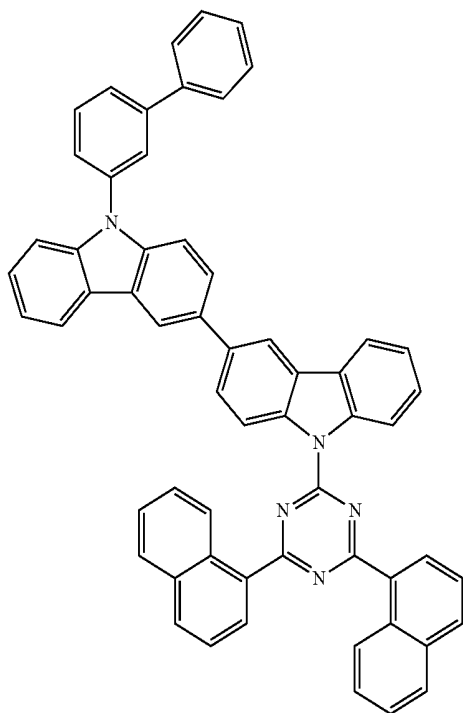
Compound 37



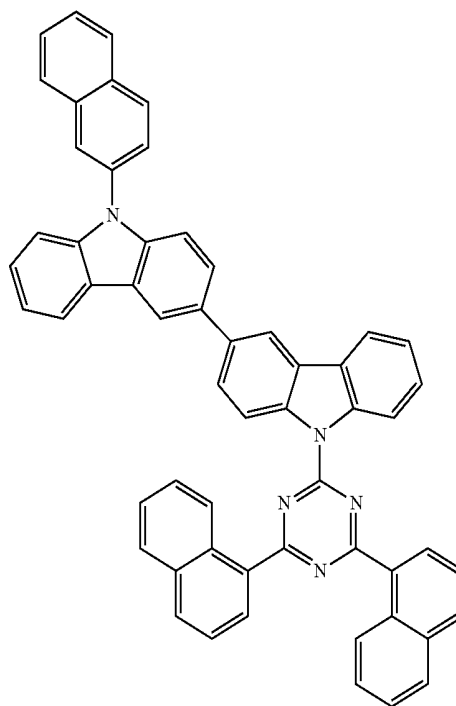
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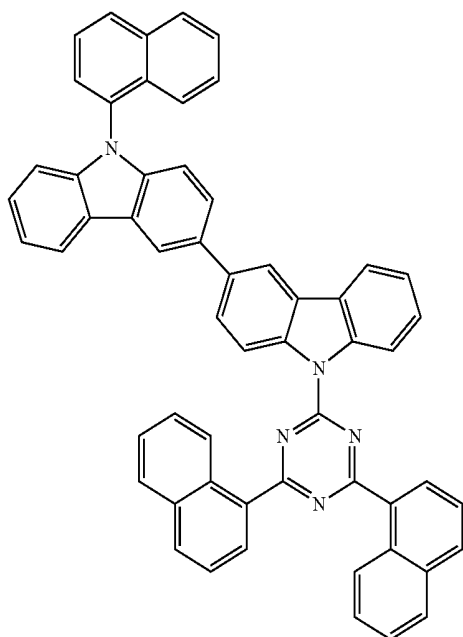
Compound 38



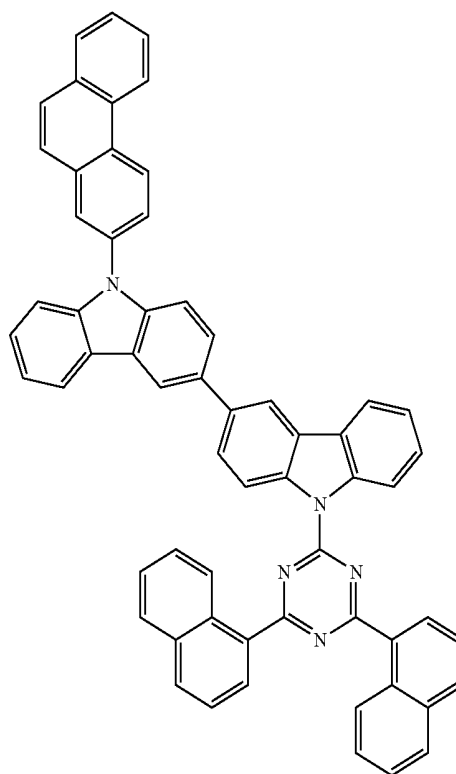
Compound 40



Compound 39



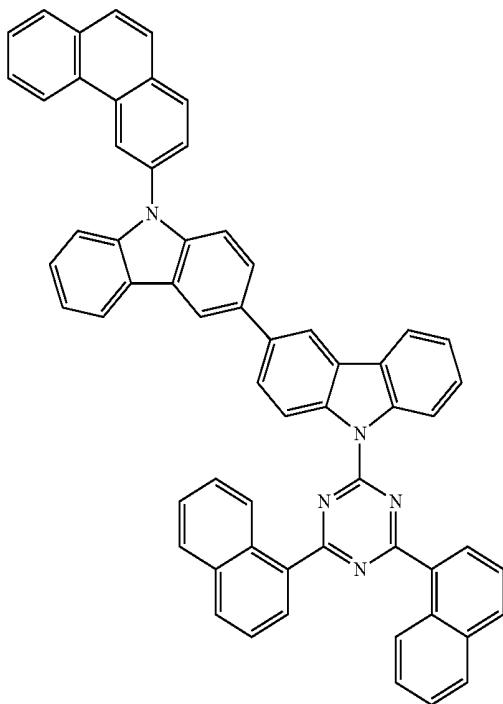
Compound 41



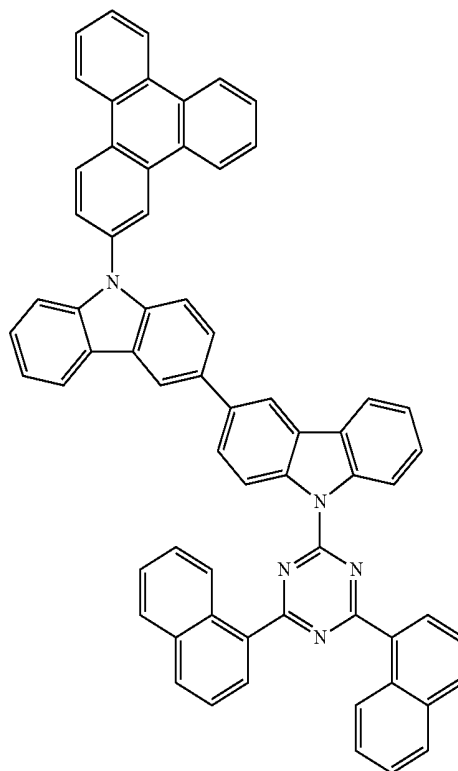
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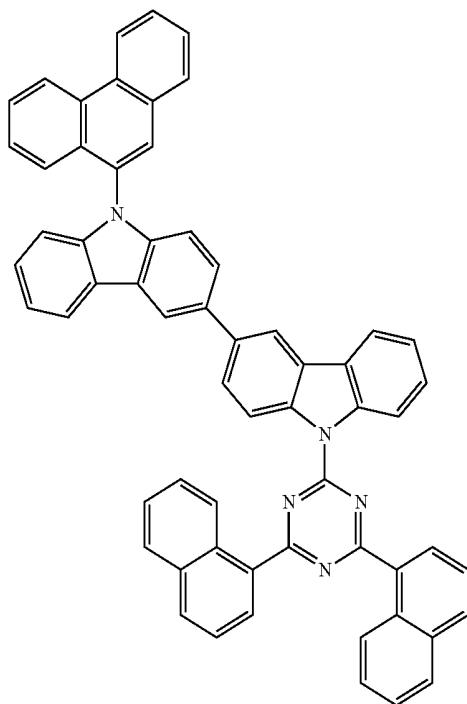
Compound 42



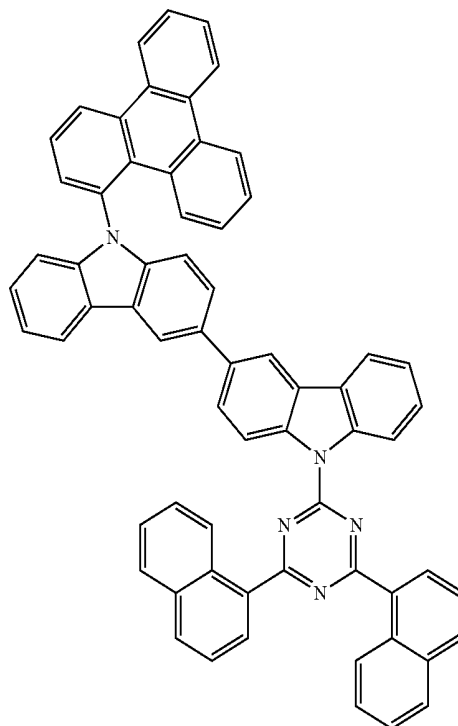
Compound 44



Compound 43



Compound 45

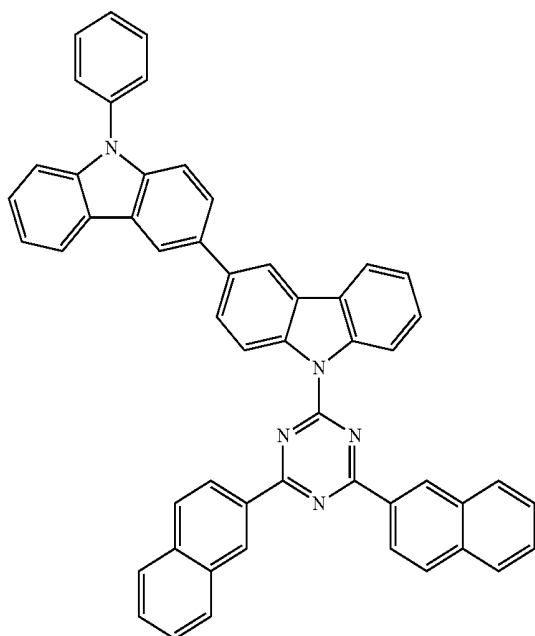




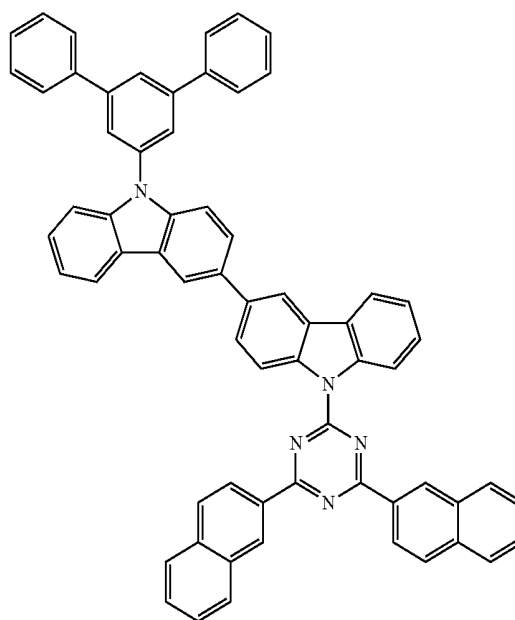
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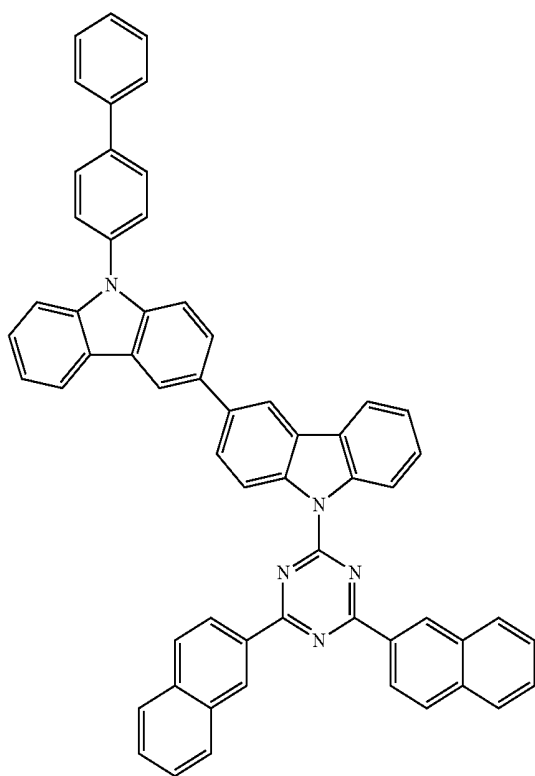
Compound 46



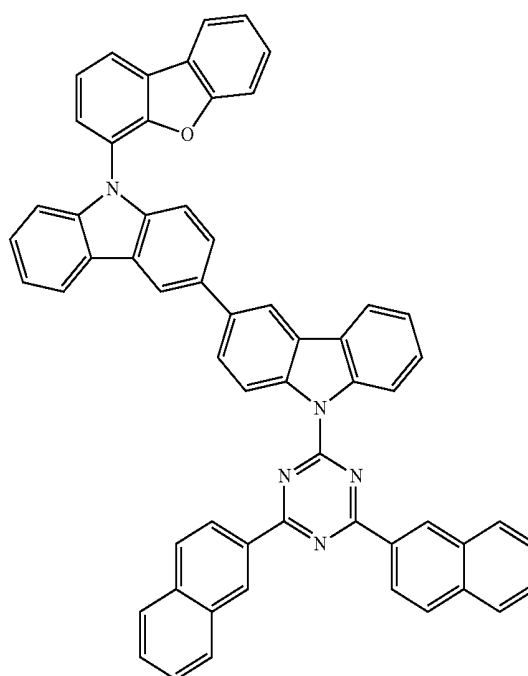
Compound 48



Compound 47



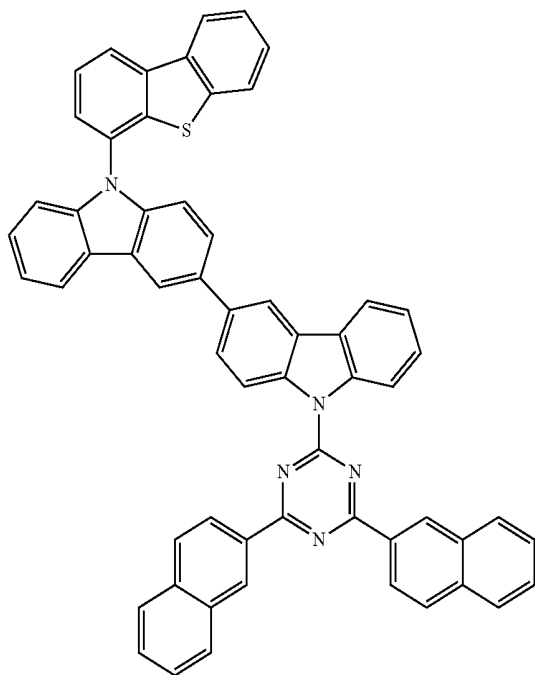
Compound 49



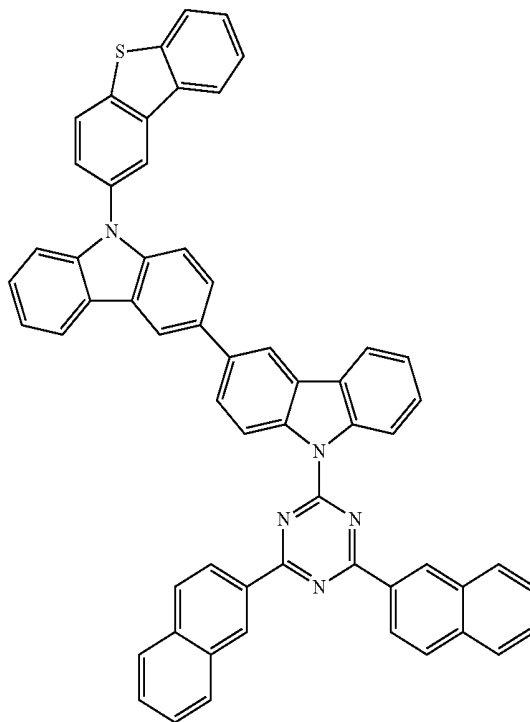
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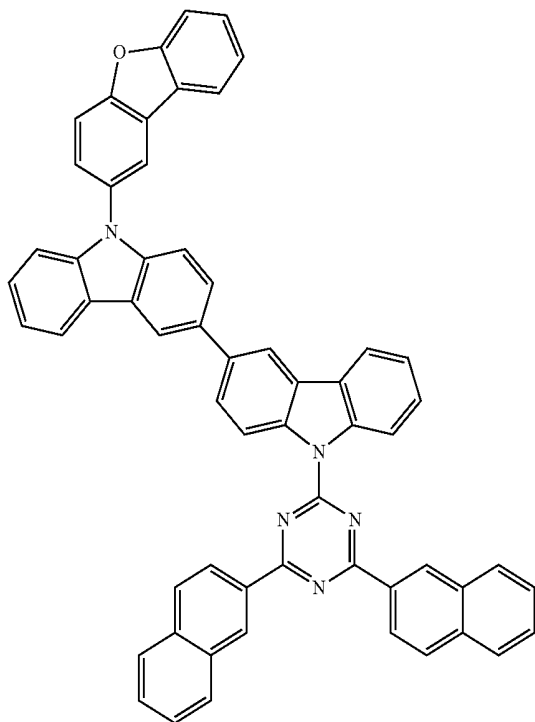
Compound 50



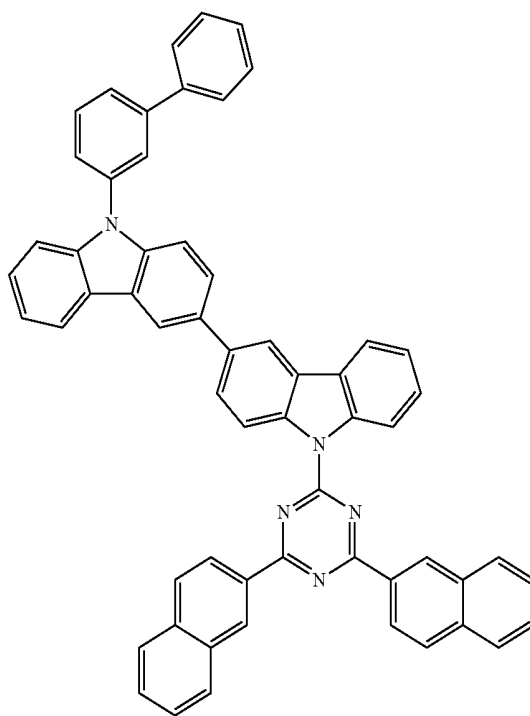
Compound 52



Compound 51



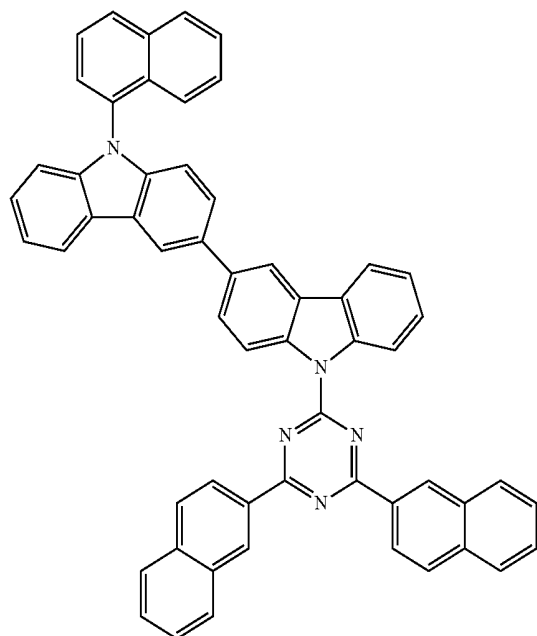
Compound 53



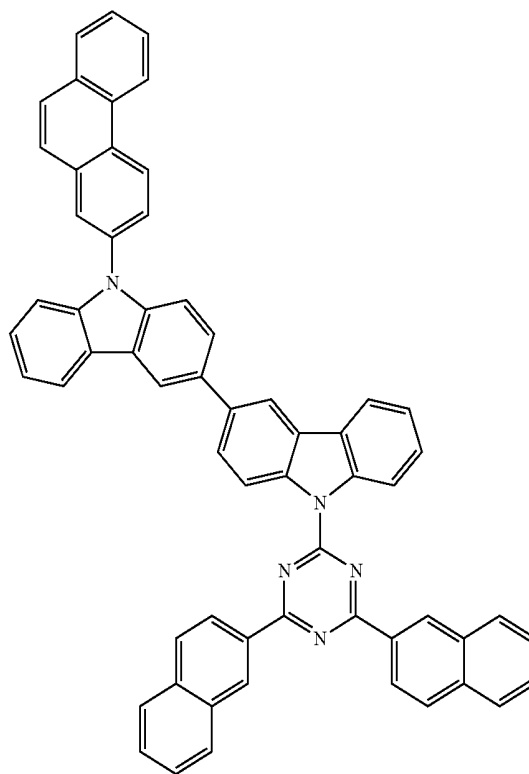
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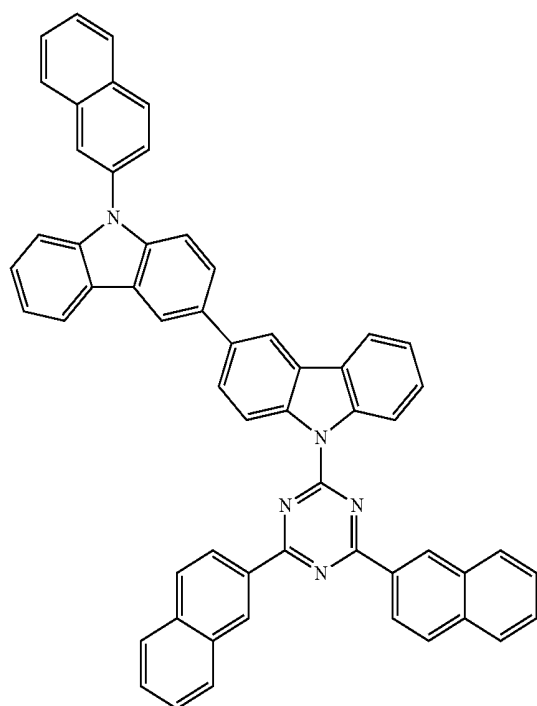
Compound 54



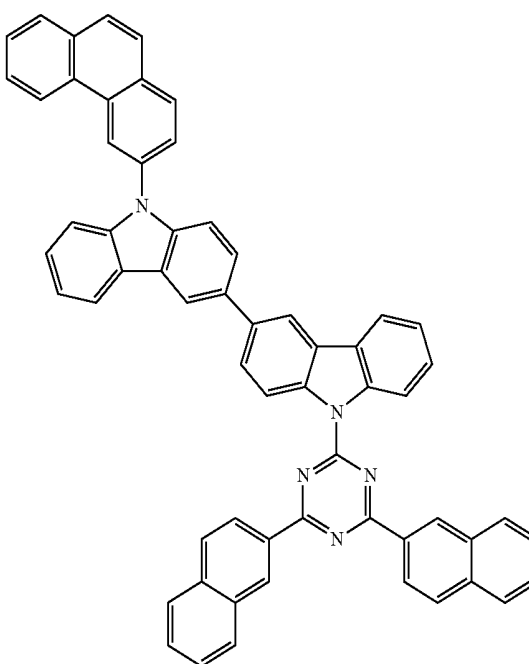
Compound 56



Compound 55



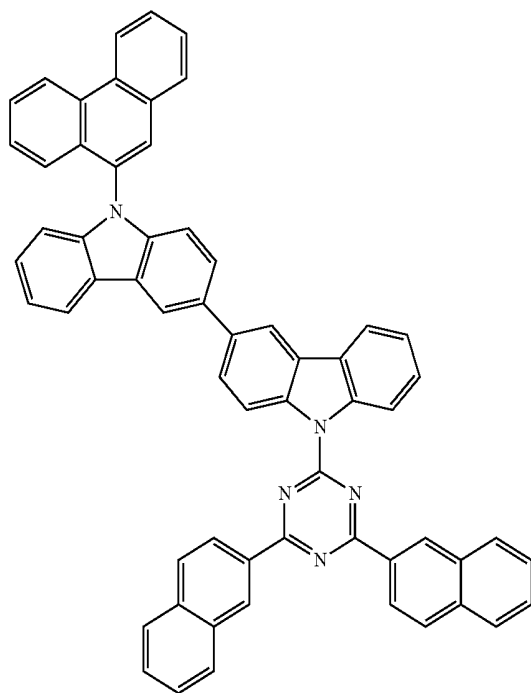
Compound 57



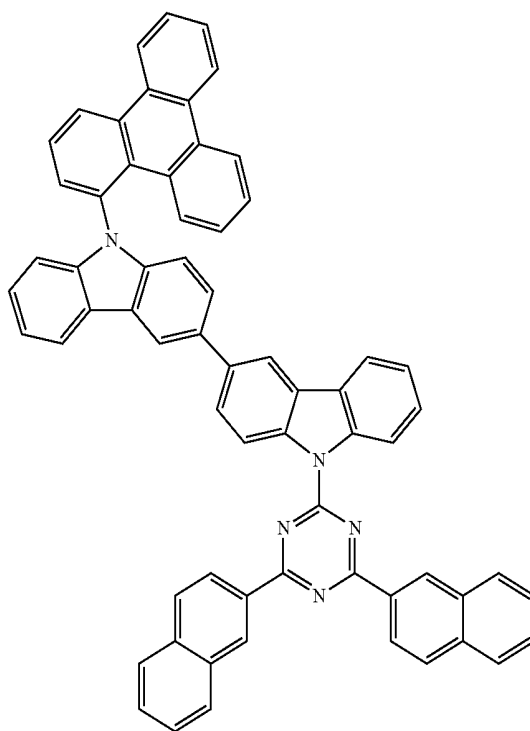
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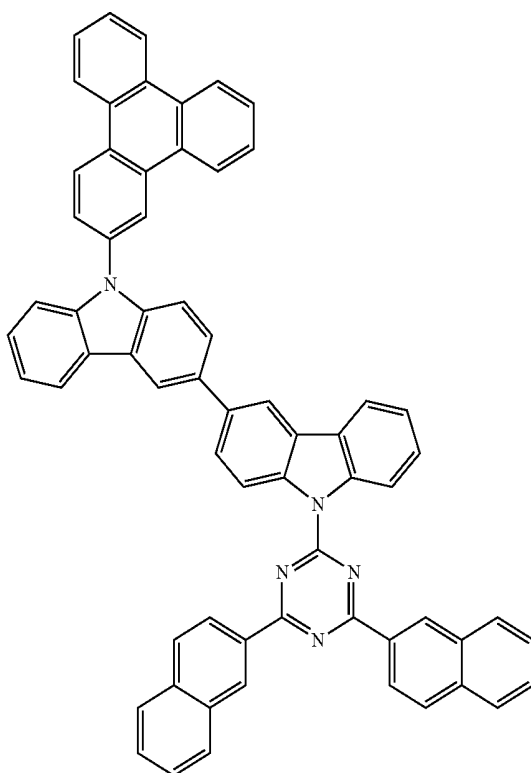
Compound 58



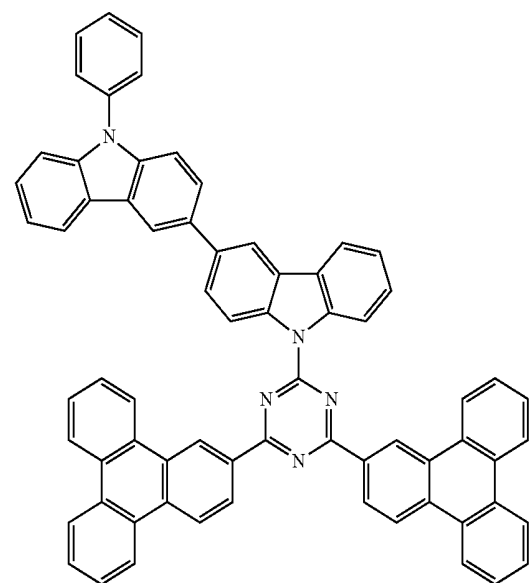
Compound 60



Compound 59



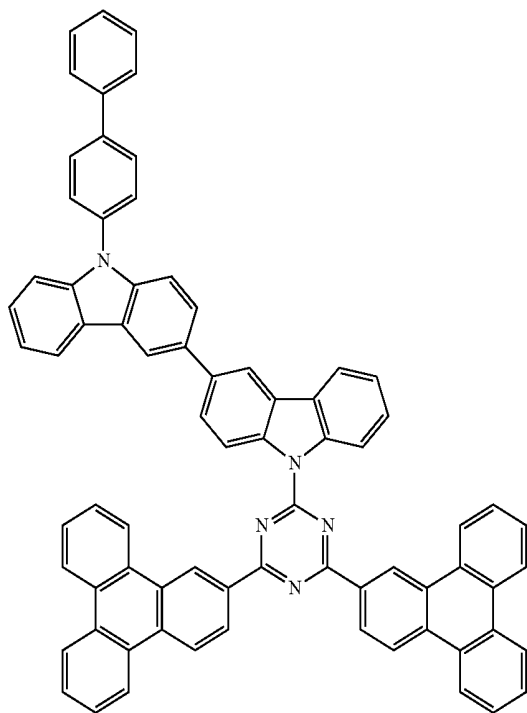
Compound 61



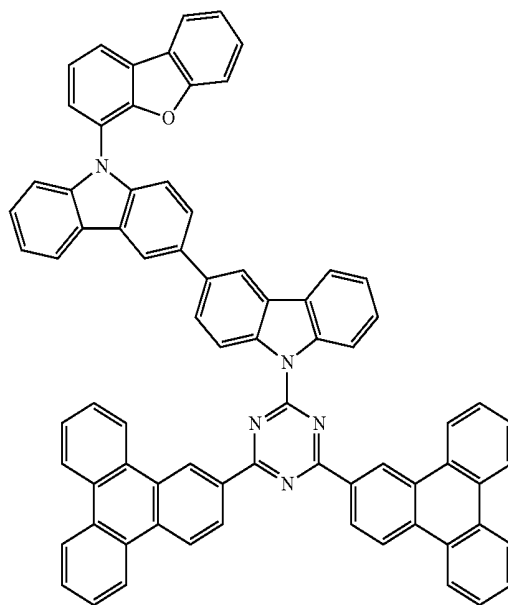
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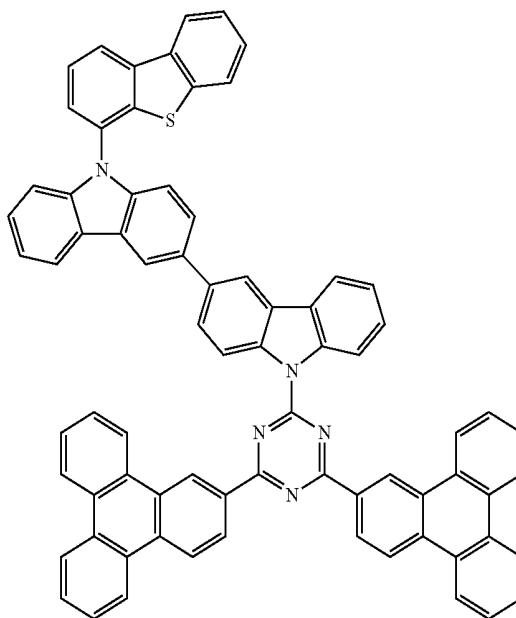
Compound 62



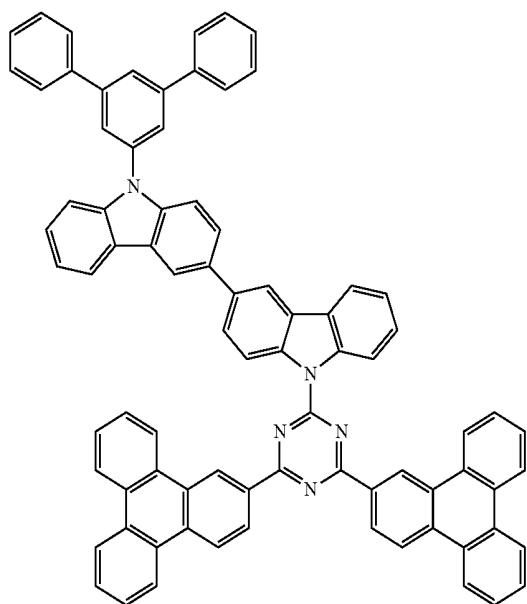
Compound 64



Compound 65



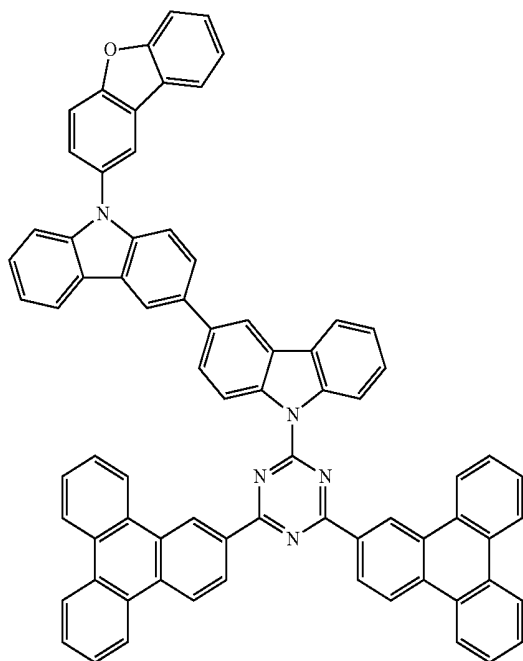
Compound 63



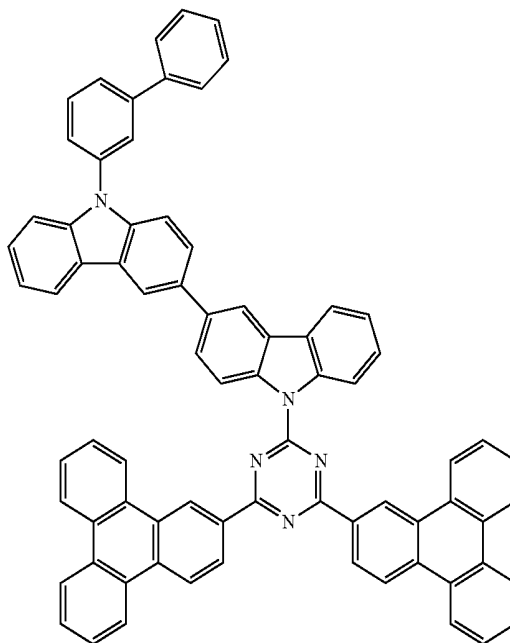
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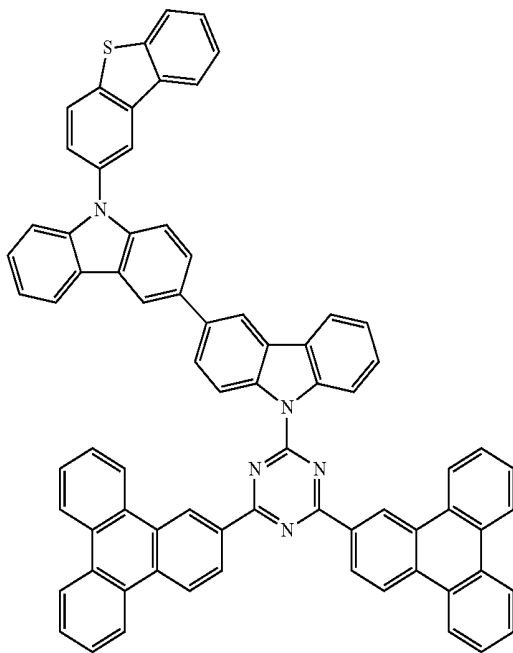
Compound 66



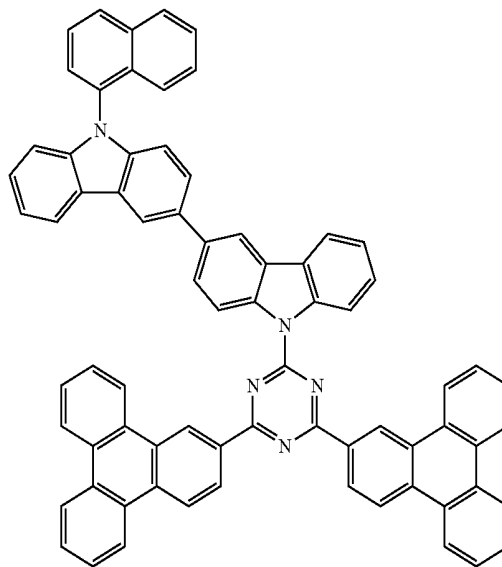
Compound 68



Compound 67



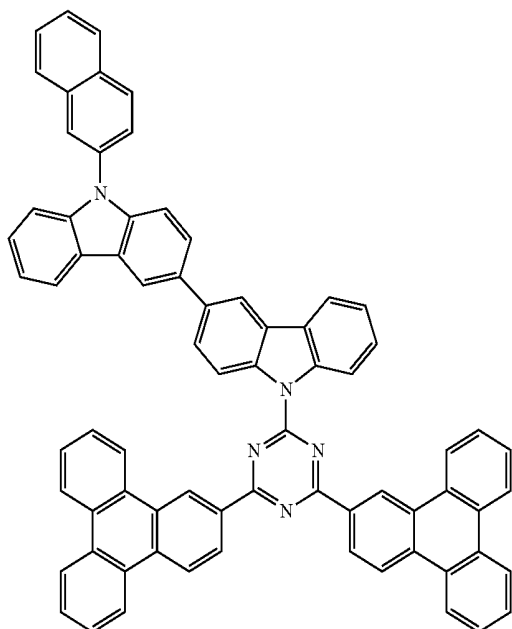
Compound 69



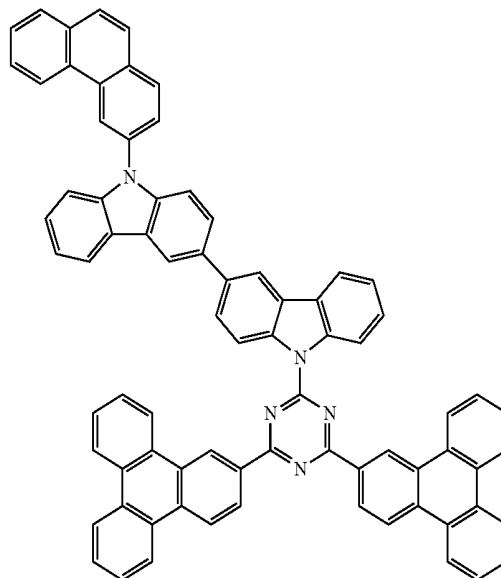
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Compound 70

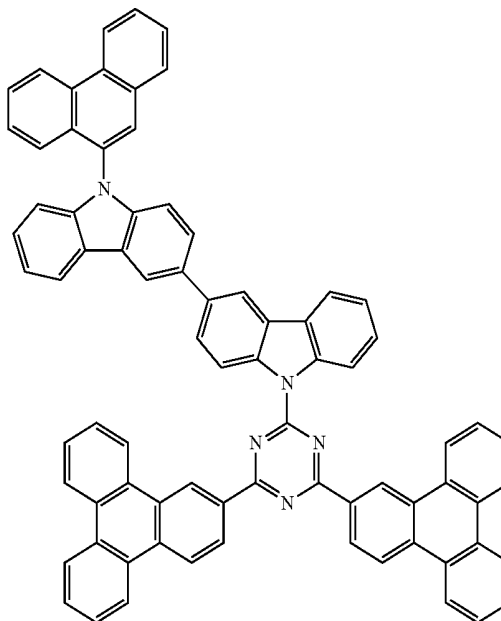
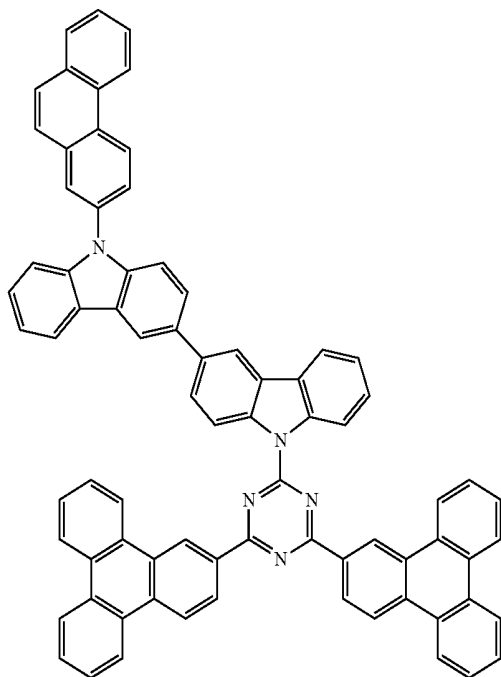


Compound 72



Compound 73

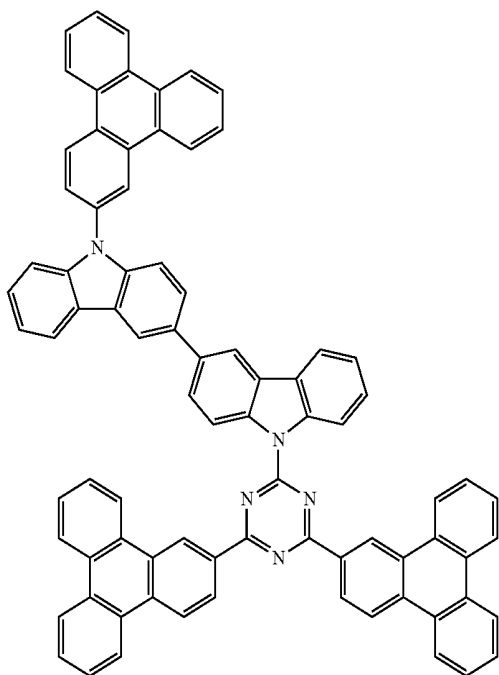
Compound 71



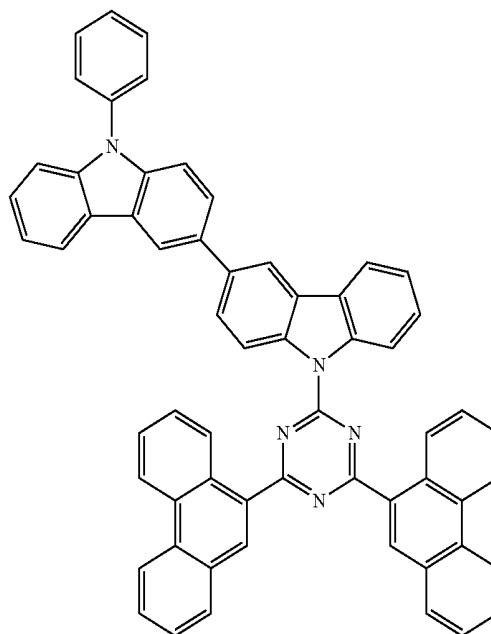
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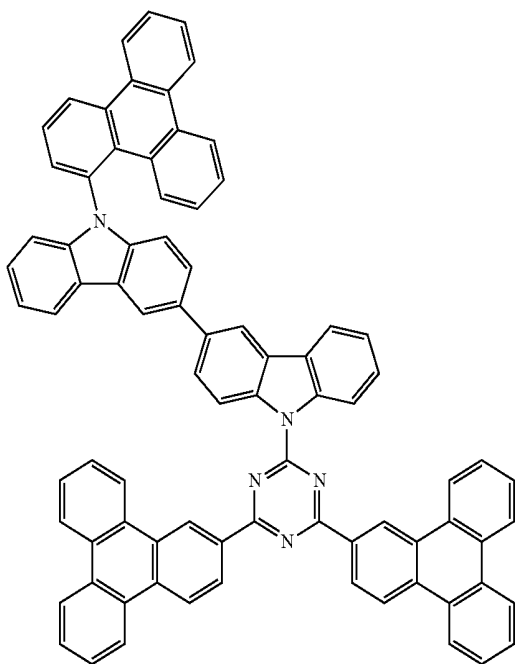
Compound 74



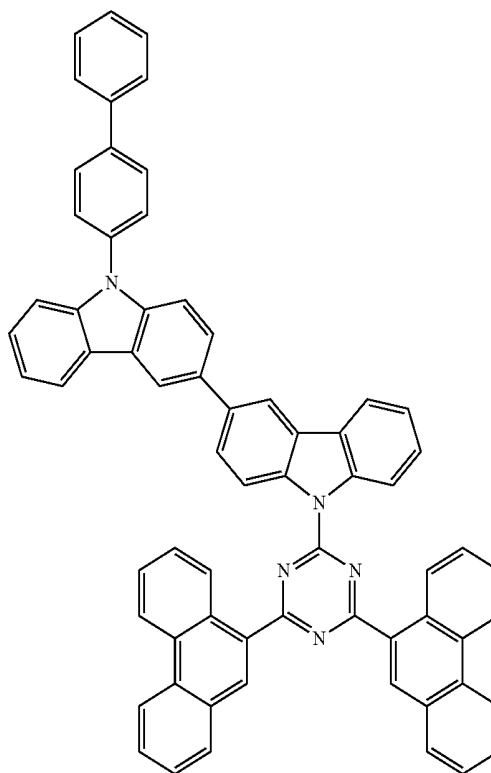
Compound 76



Compound 75



Compound 77

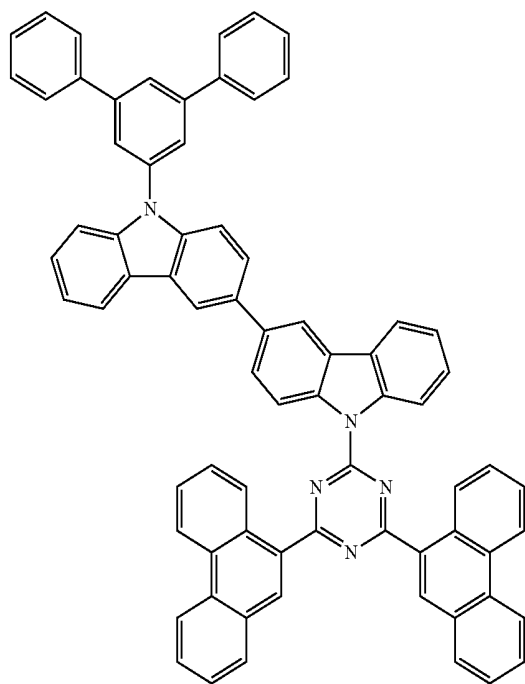




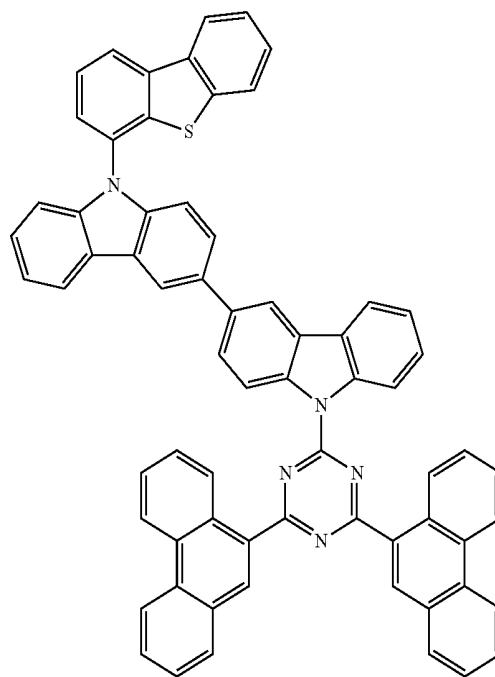
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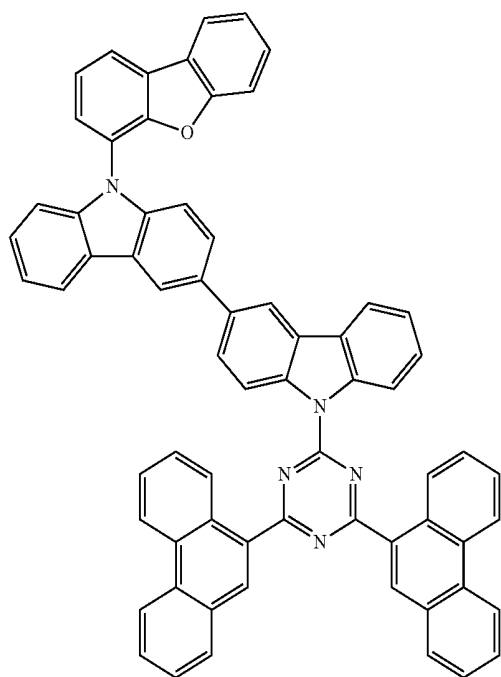
Compound 78



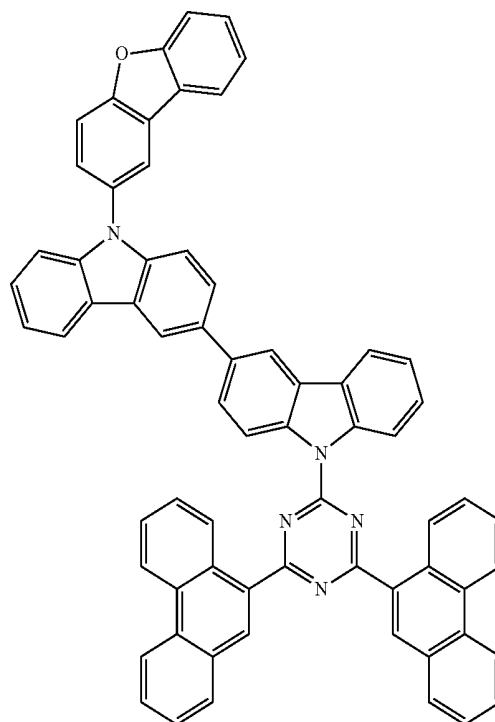
Compound 80



Compound 79



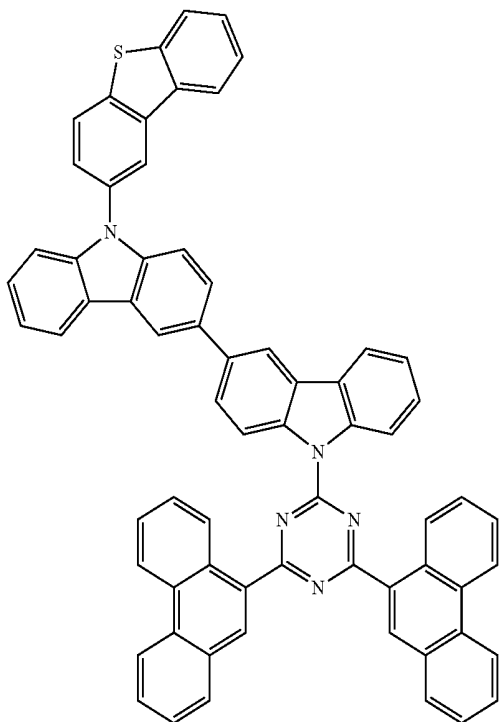
Compound 81



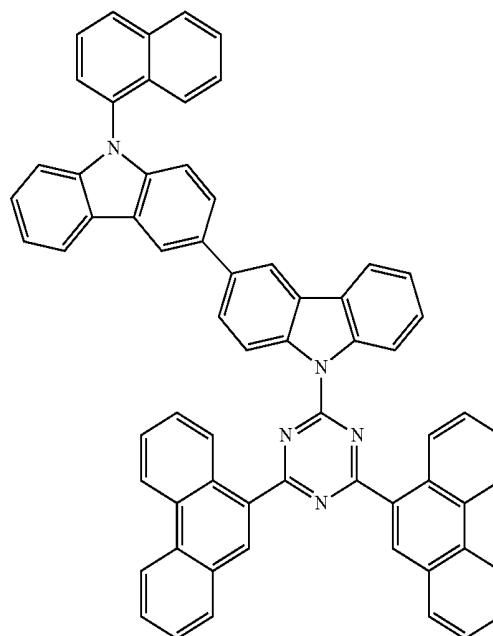
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Compound 82

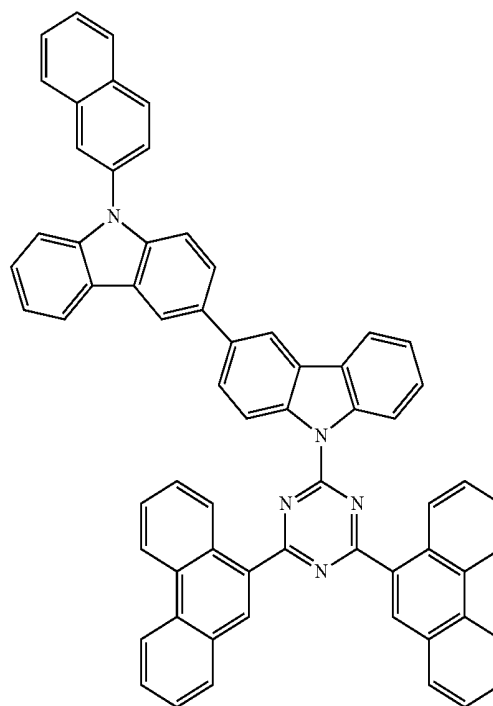
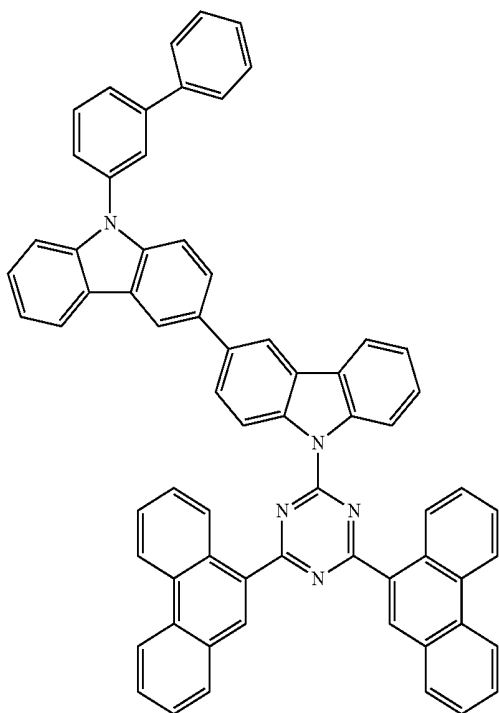


Compound 84



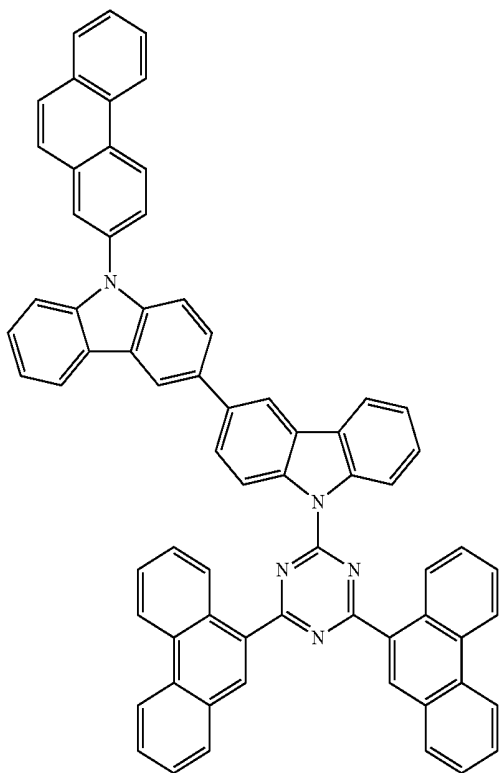
Compound 85

Compound 83



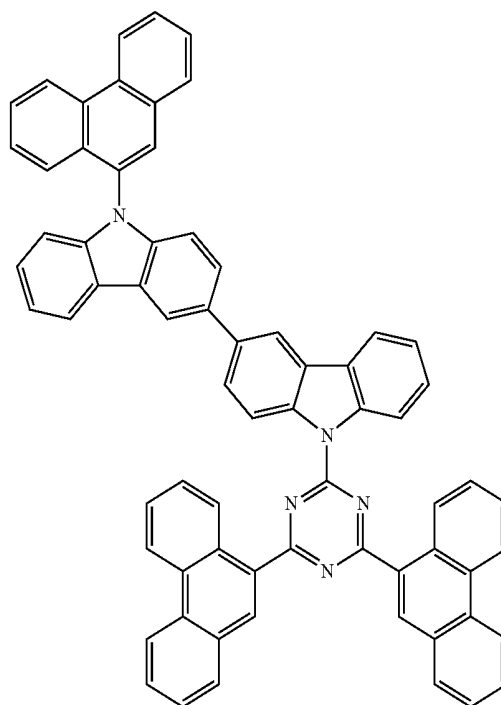
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Compound 86



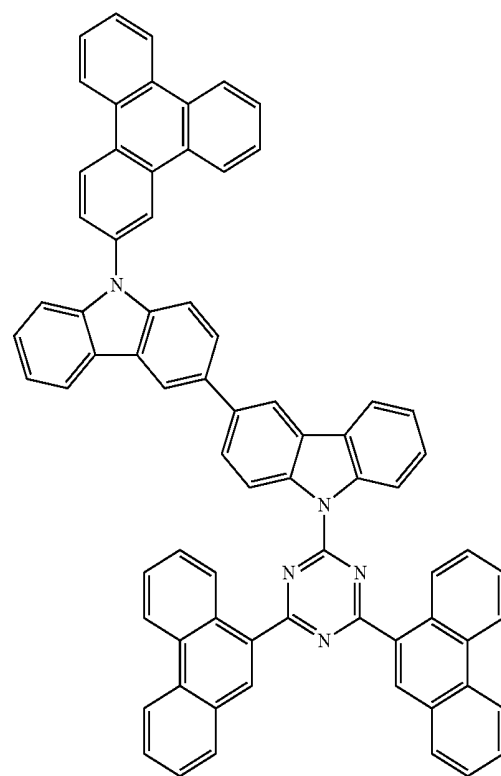
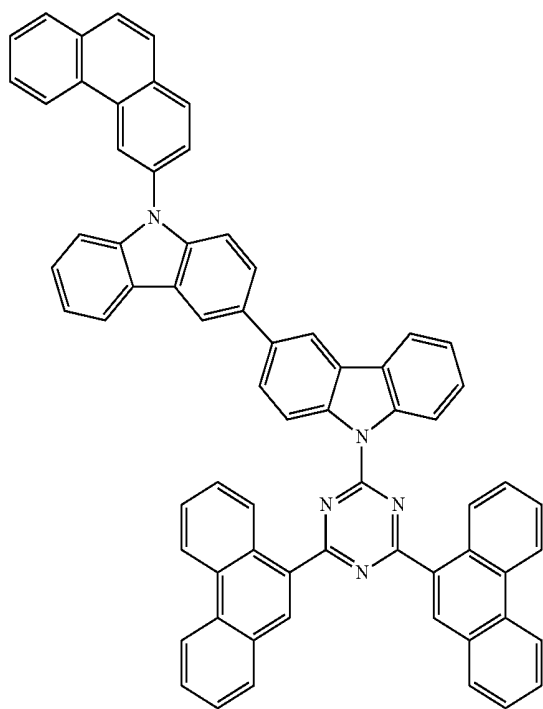
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Compound 88



Compound 89

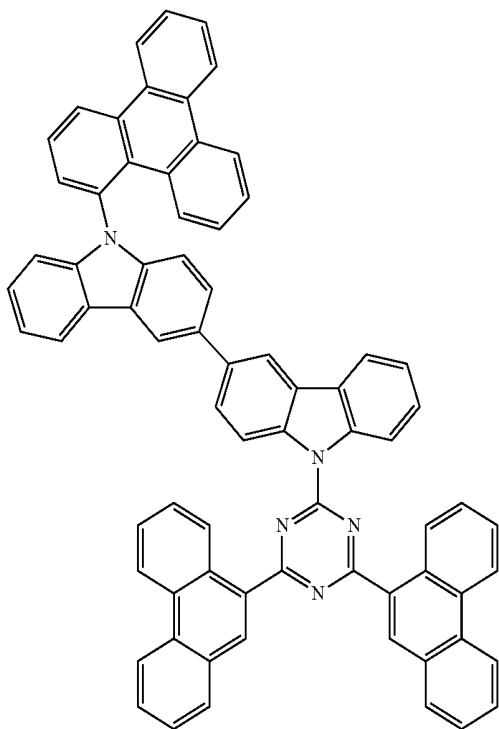
Compound 87



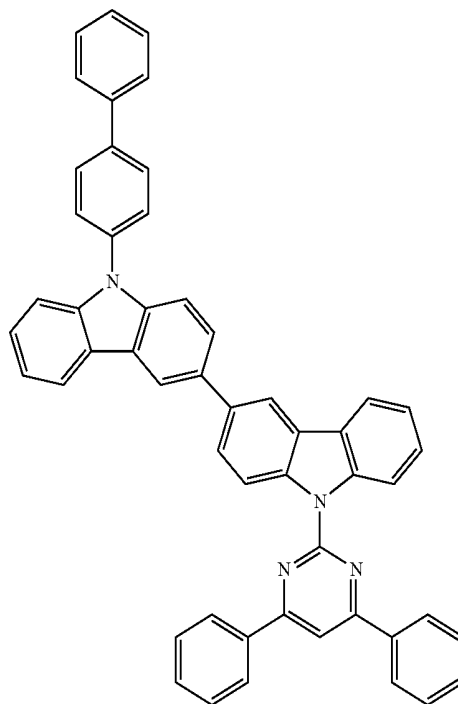
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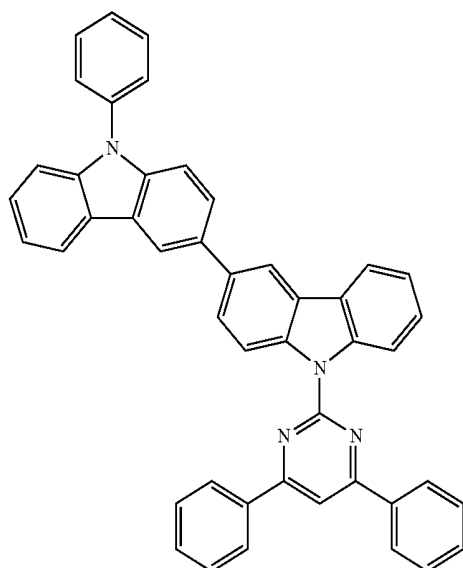
Compound 90



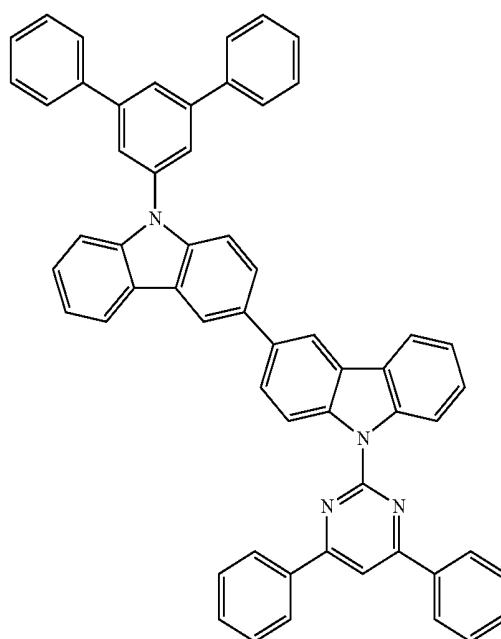
Compound 92



Compound 91



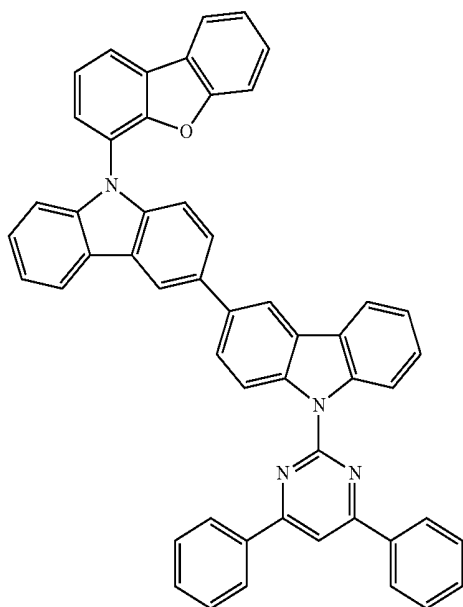
Compound 93



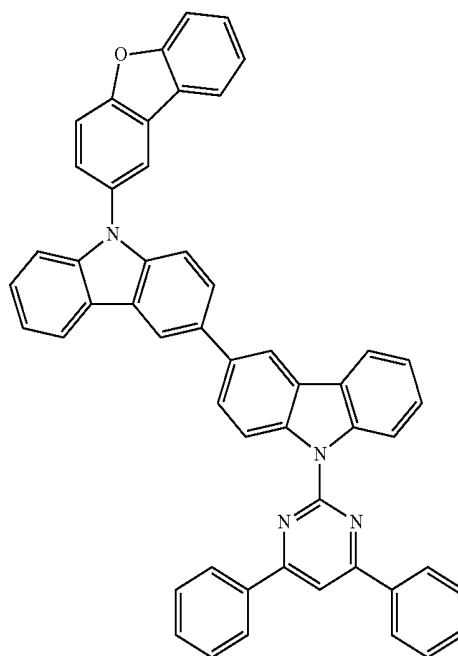
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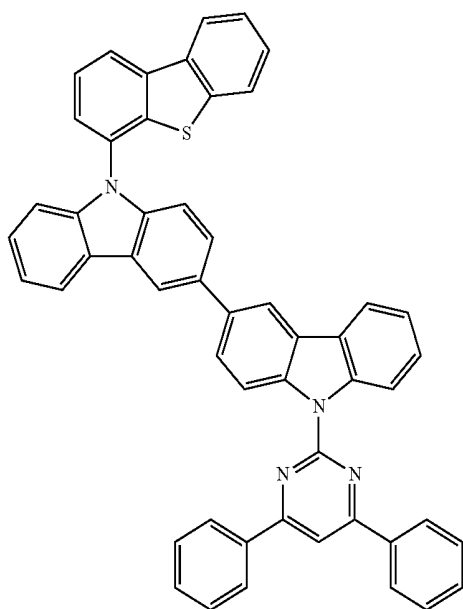
Compound 94



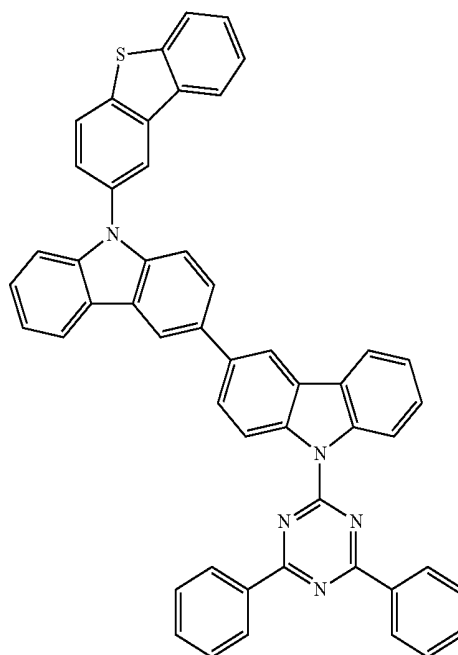
Compound 96



Compound 95



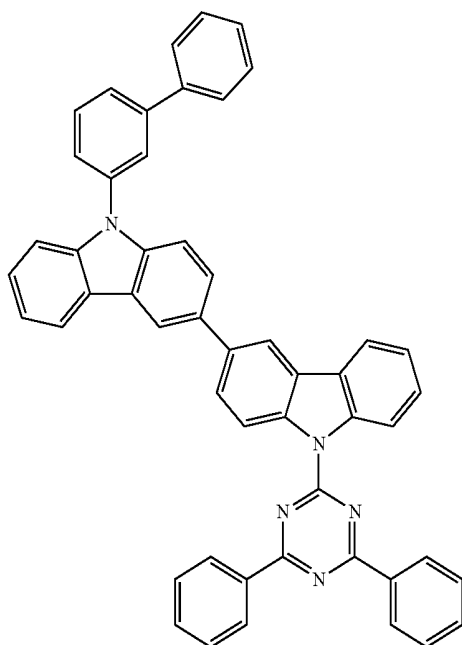
Compound 97



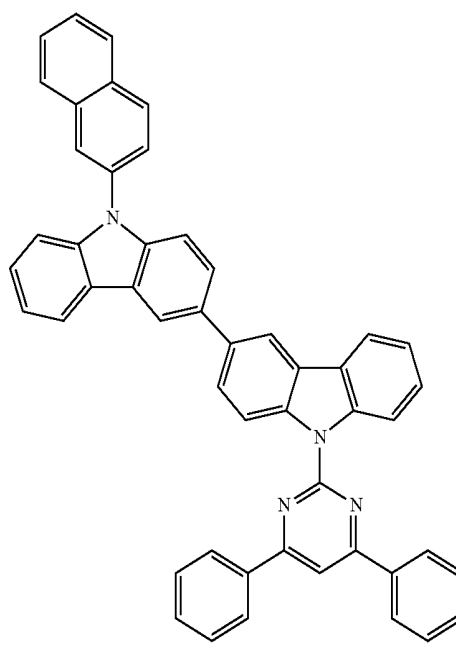
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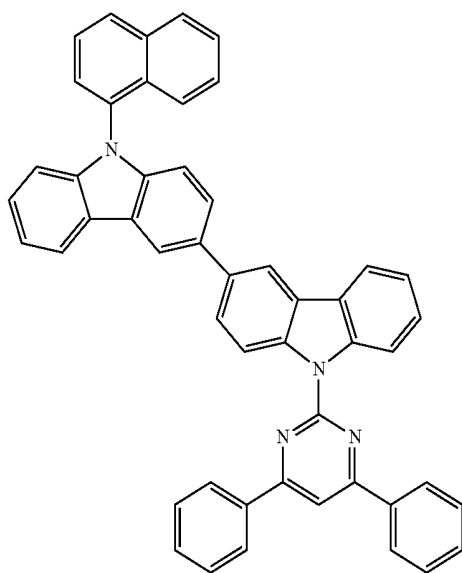
Compound 98



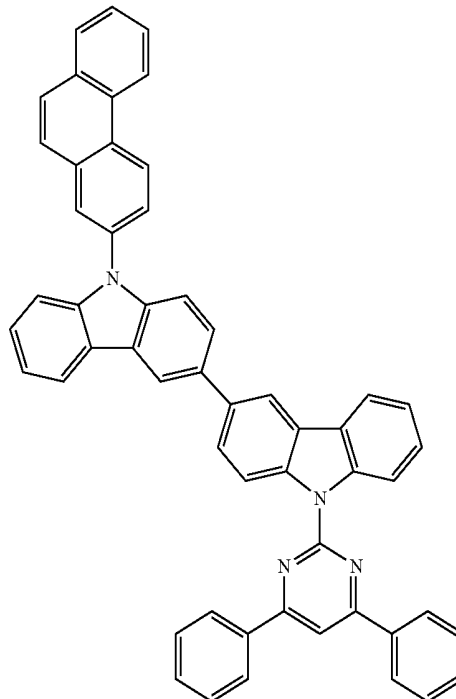
Compound 100



Compound 99



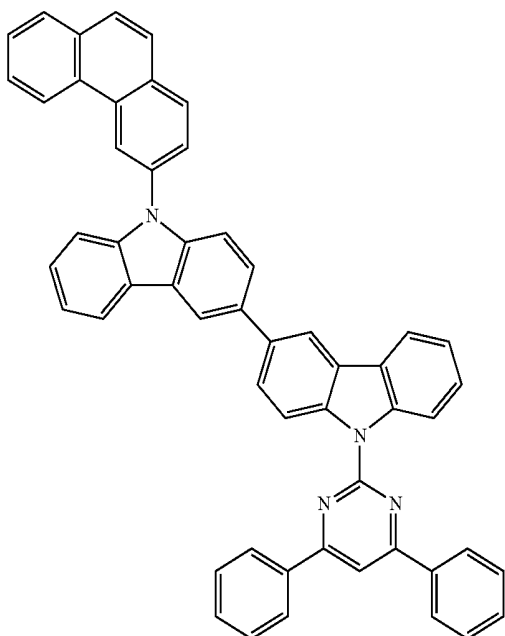
Compound 101



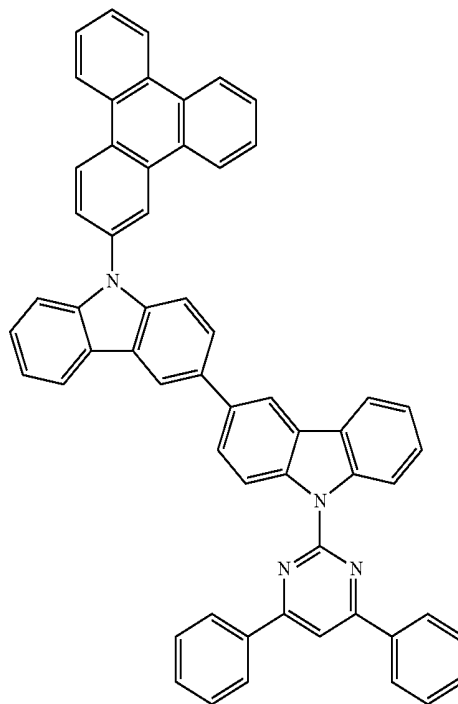
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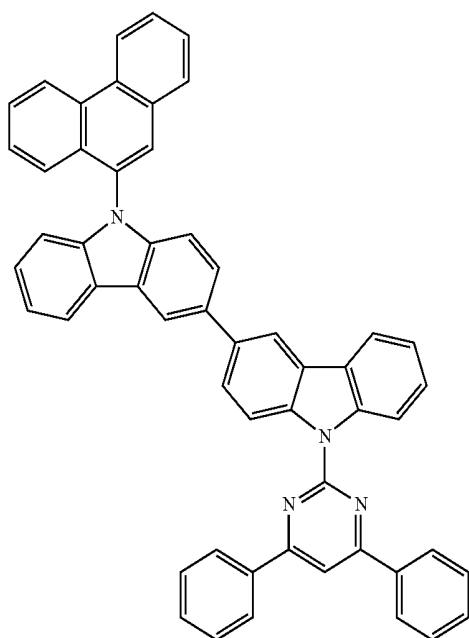
Compound 102



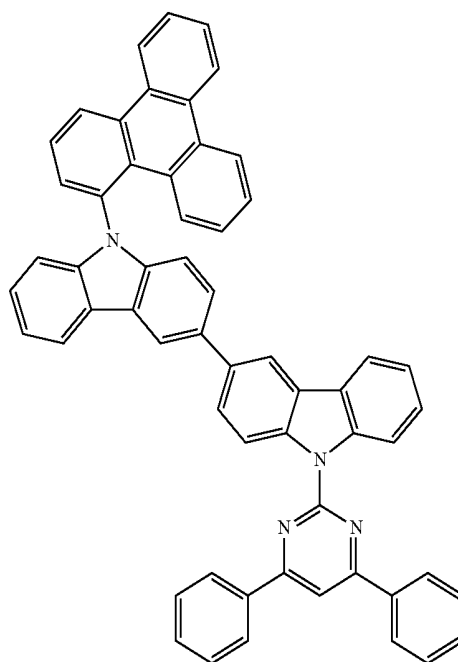
Compound 104



Compound 103



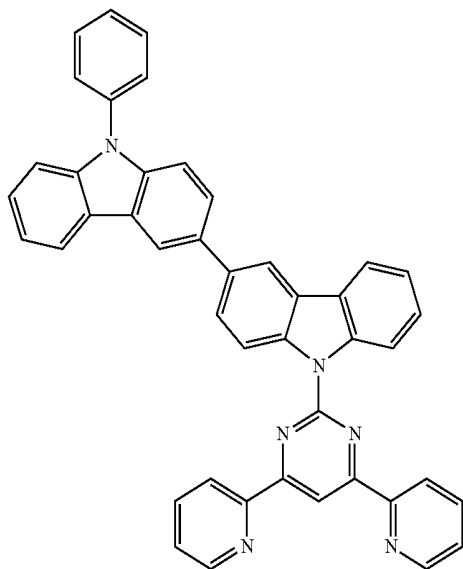
Compound 105



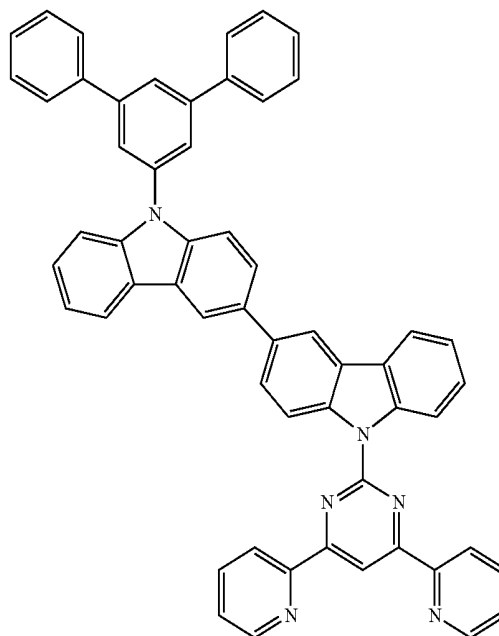
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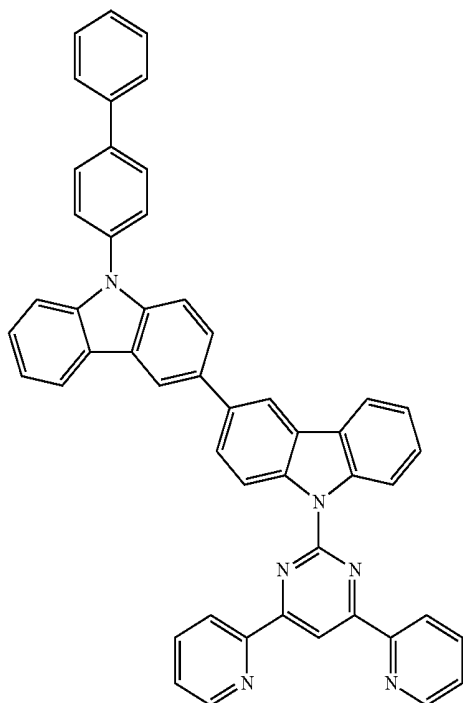
Compound 106



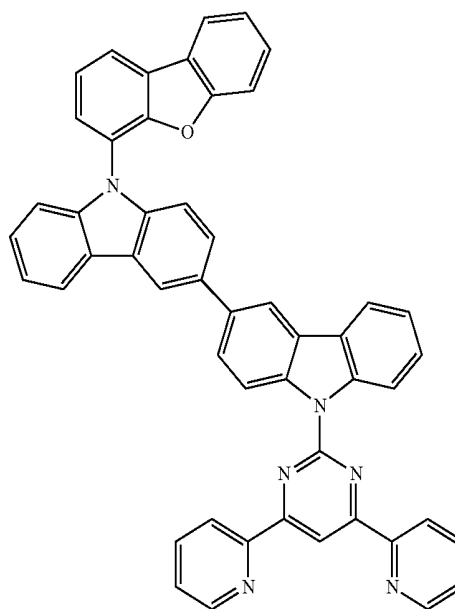
Compound 108



Compound 107



Compound 109

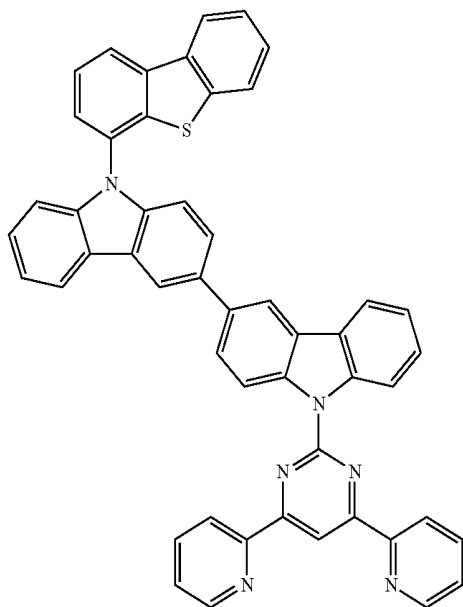




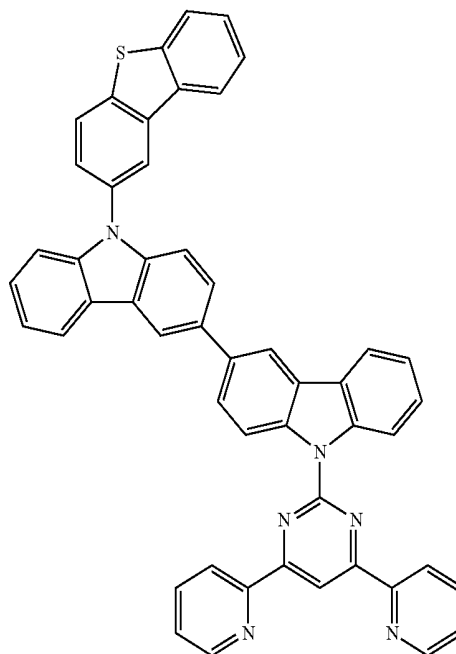
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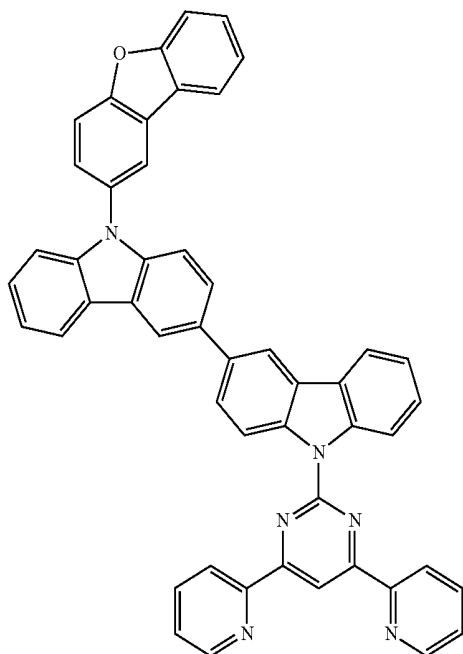
Compound 110



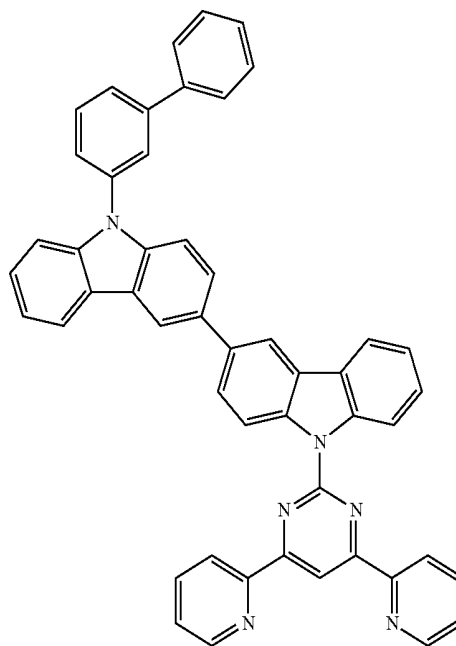
Compound 112



Compound 111



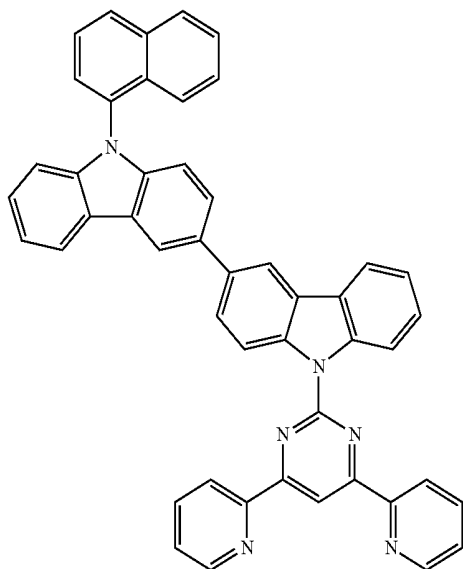
Compound 113



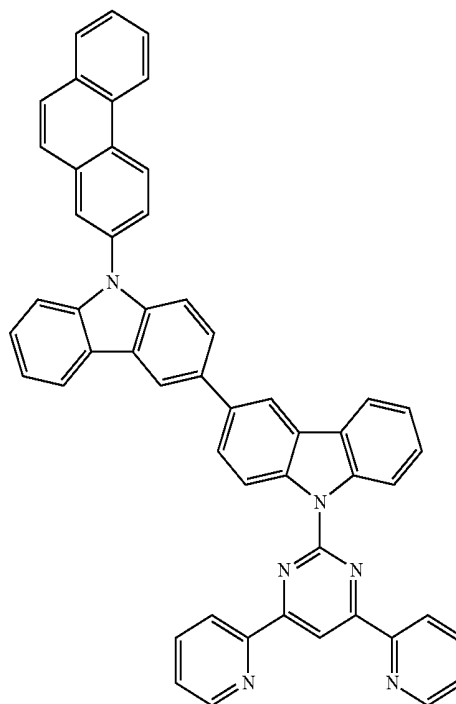
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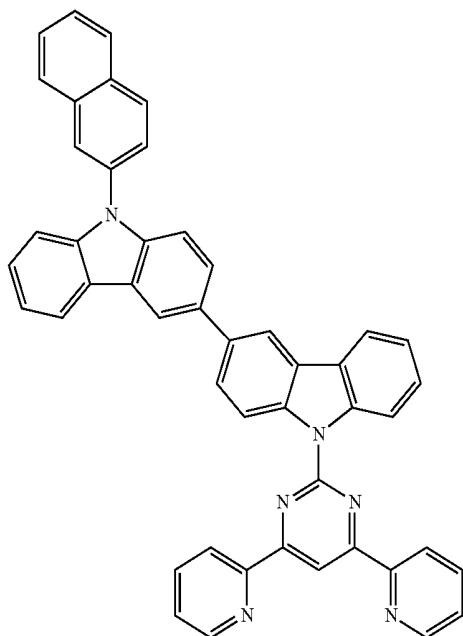
Compound 114



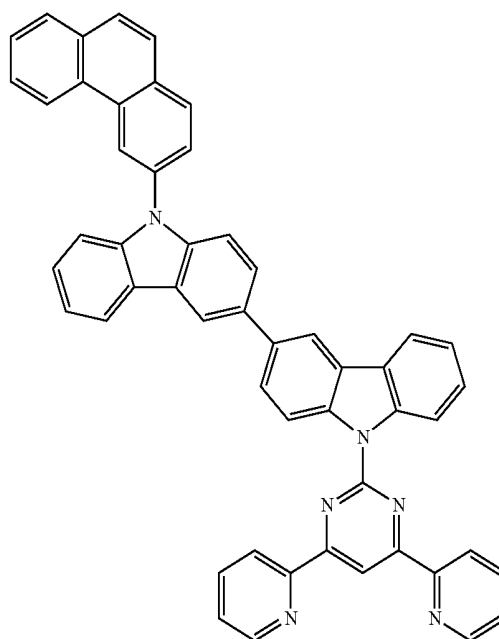
Compound 116



Compound 115



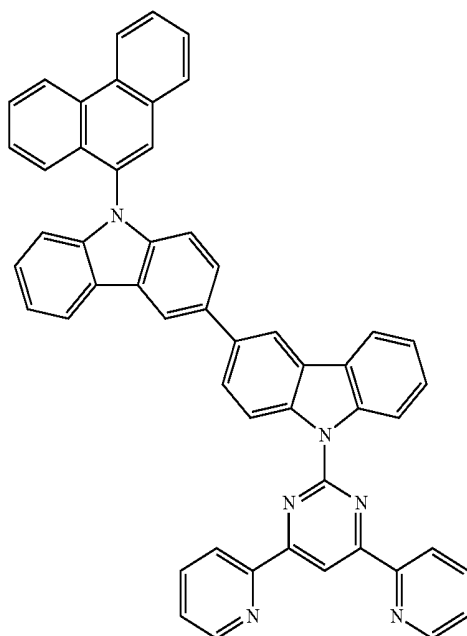
Compound 117



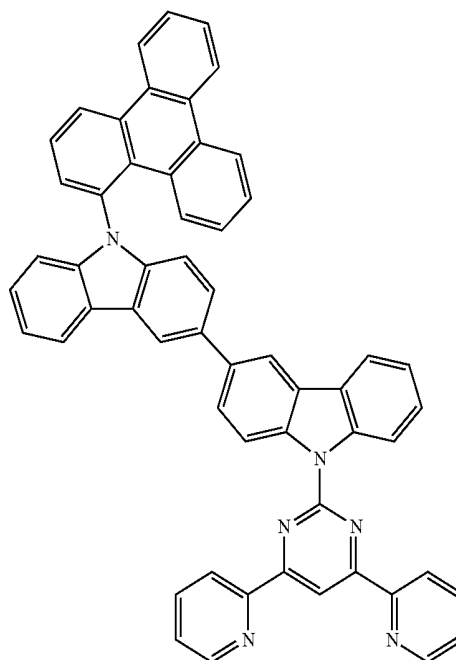
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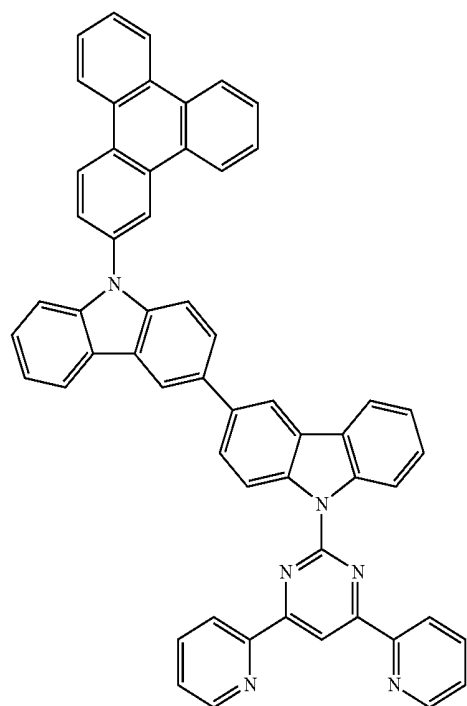
Compound 118



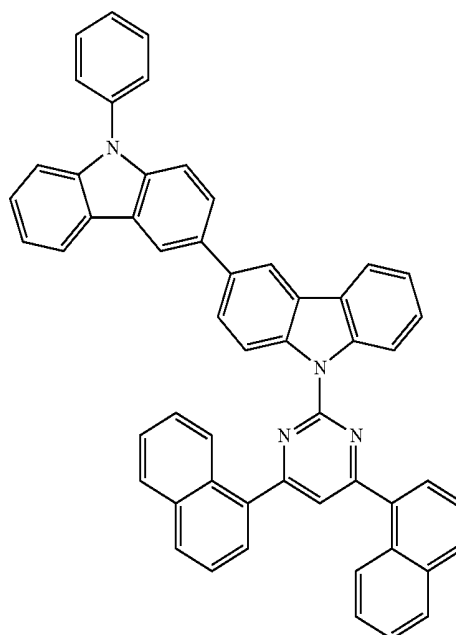
Compound 120



Compound 119



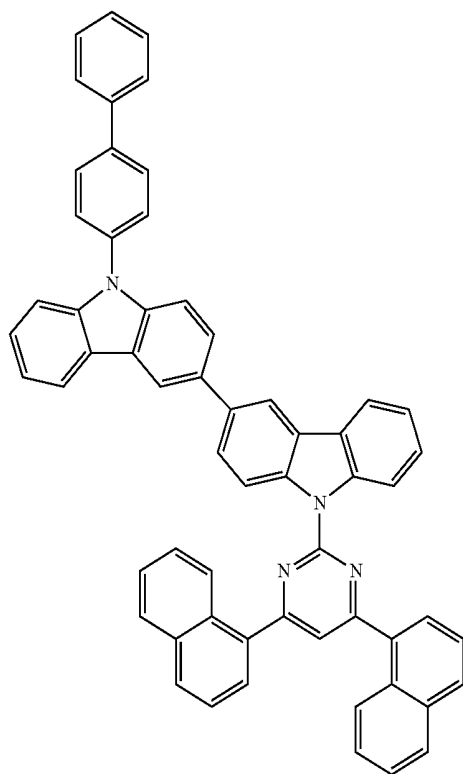
Compound 121



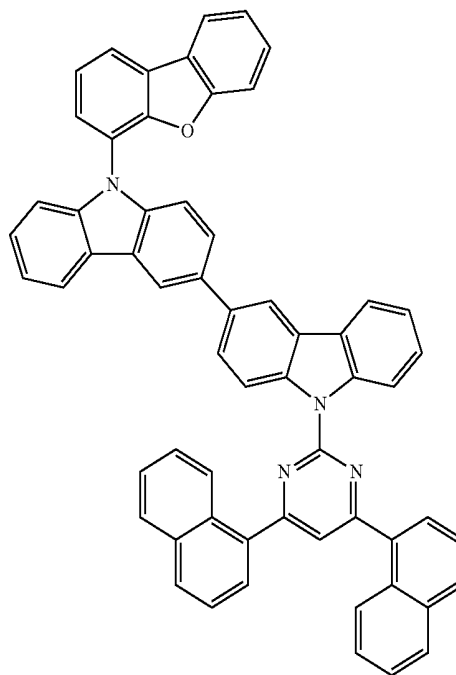
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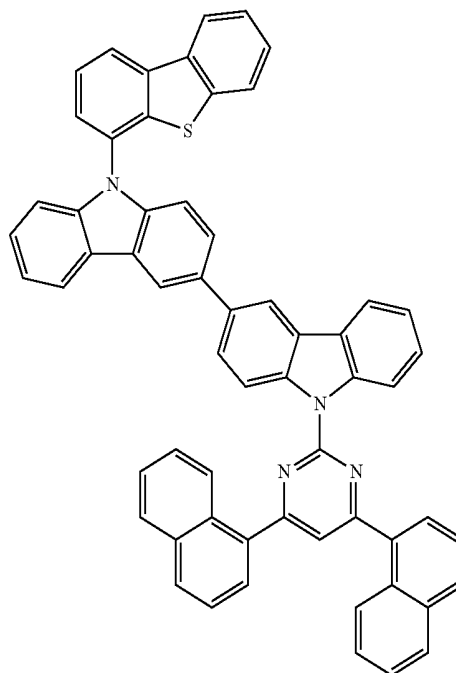
Compound 122



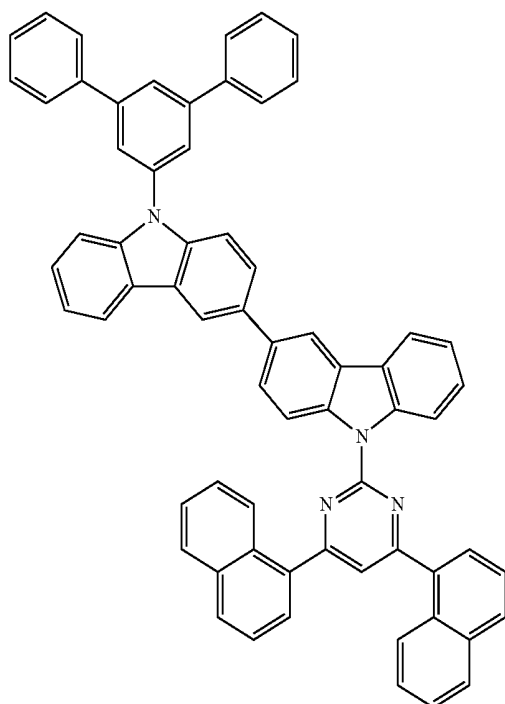
Compound 124



Compound 125



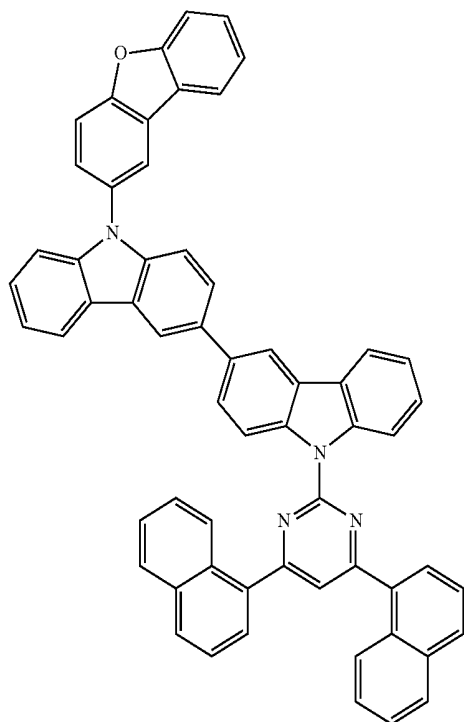
Compound 123



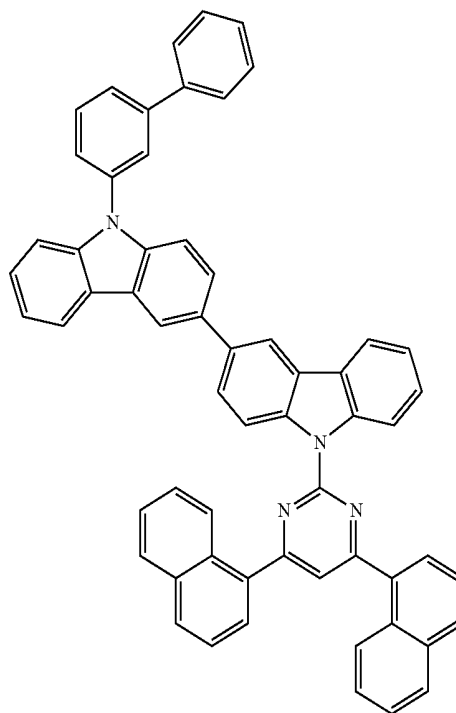
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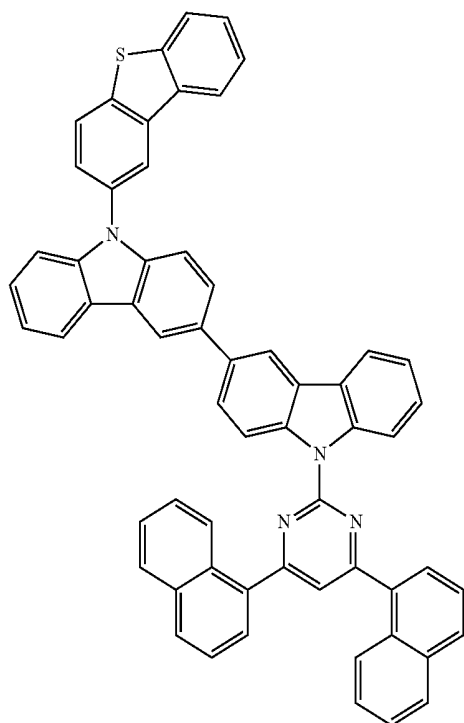
Compound 126



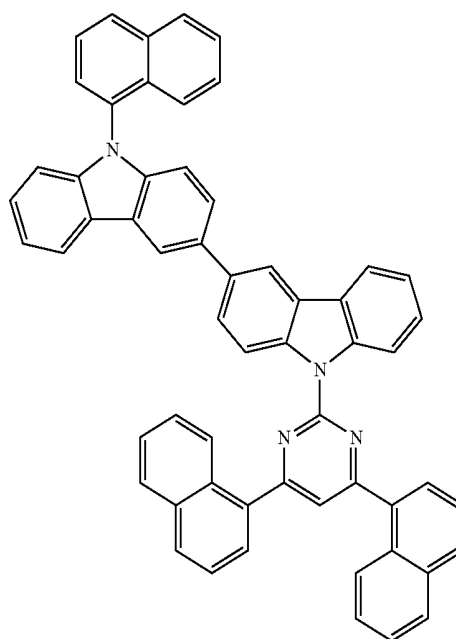
Compound 128



Compound 127



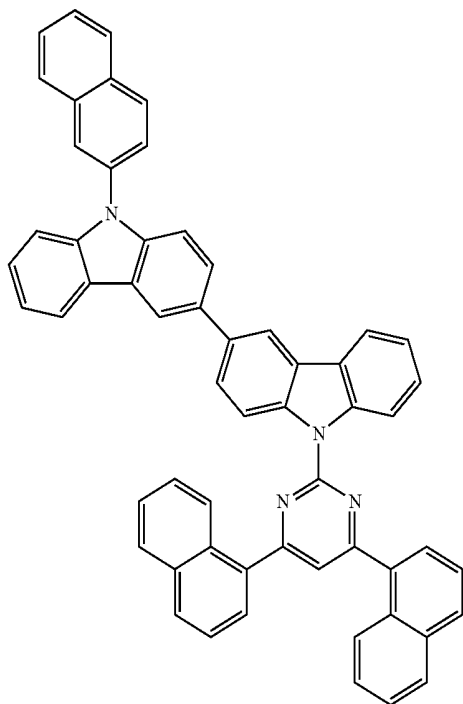
Compound 129



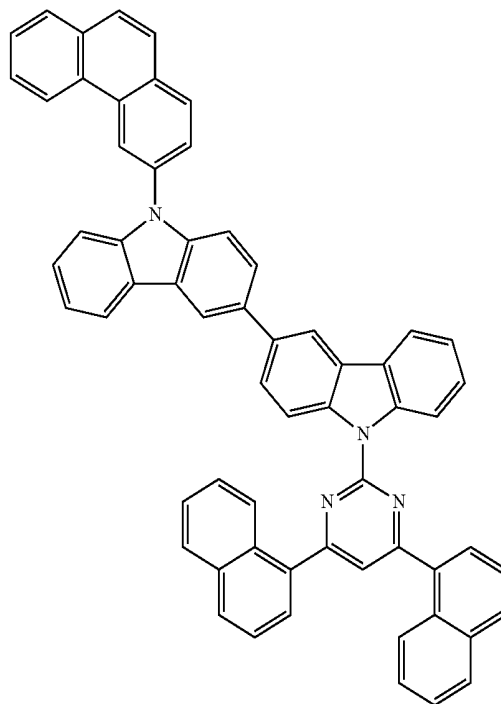
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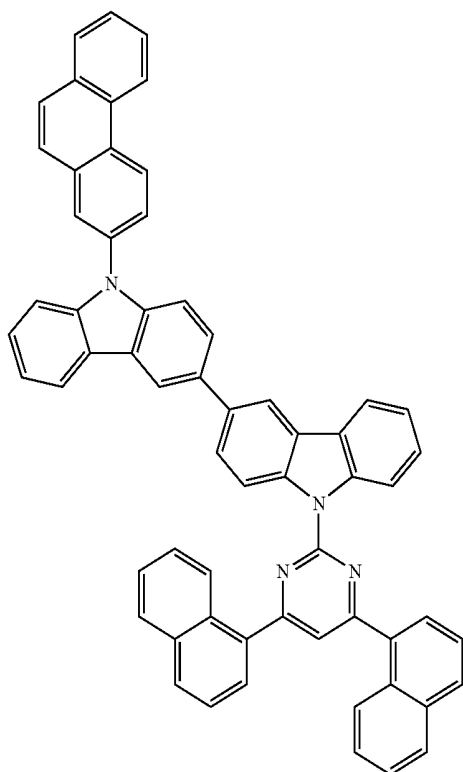
Compound 130



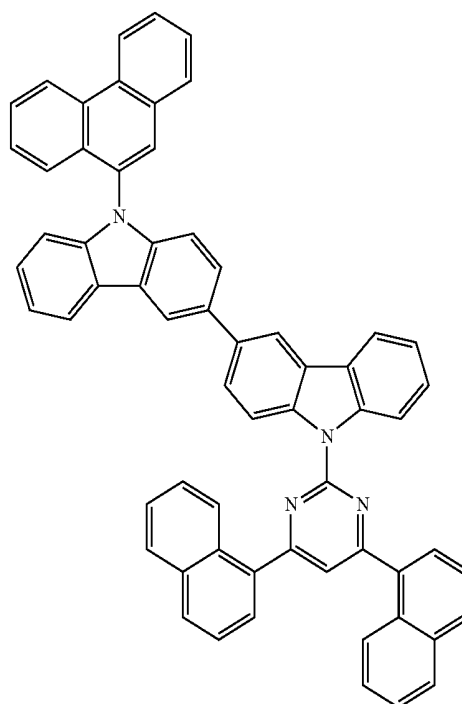
Compound 132



Compound 131



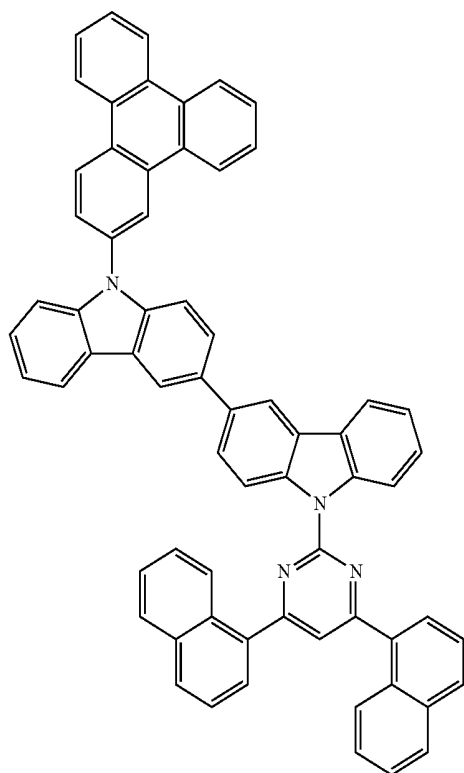
Compound 133



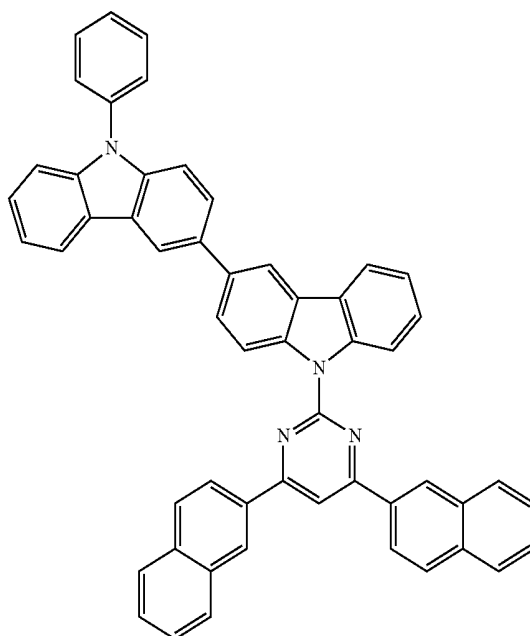
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Compound 134

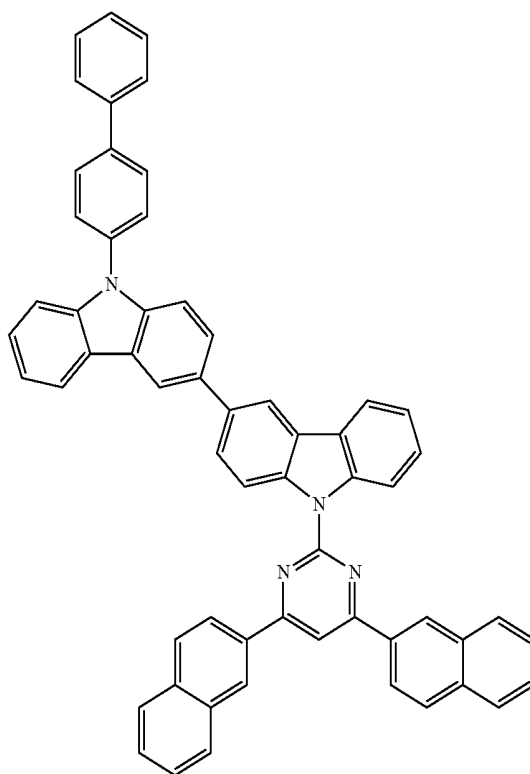
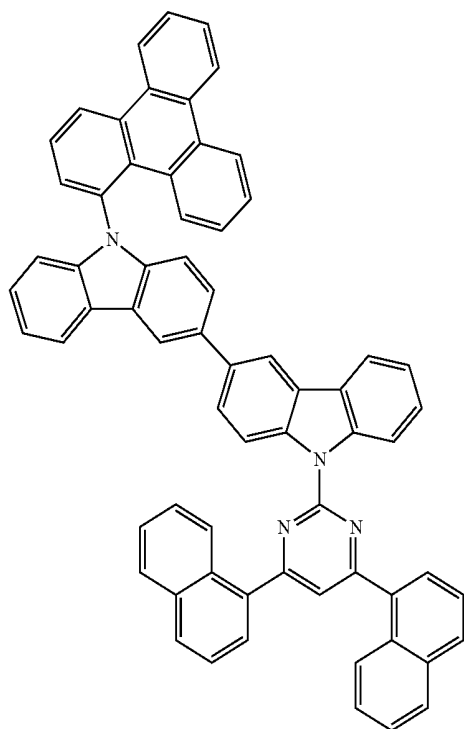


Compound 136



Compound 137

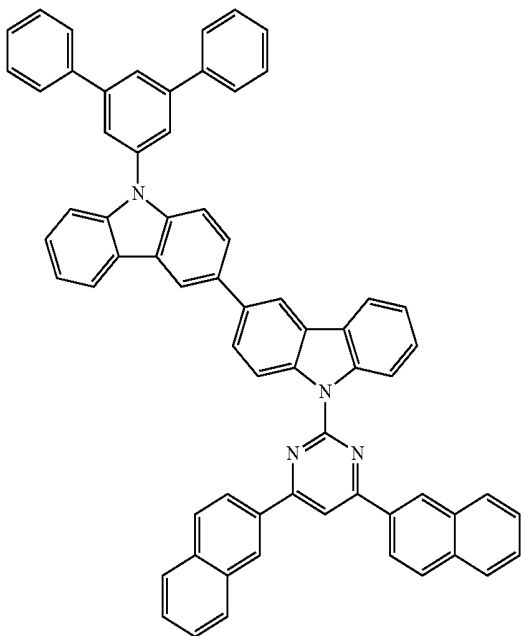
Compound 135



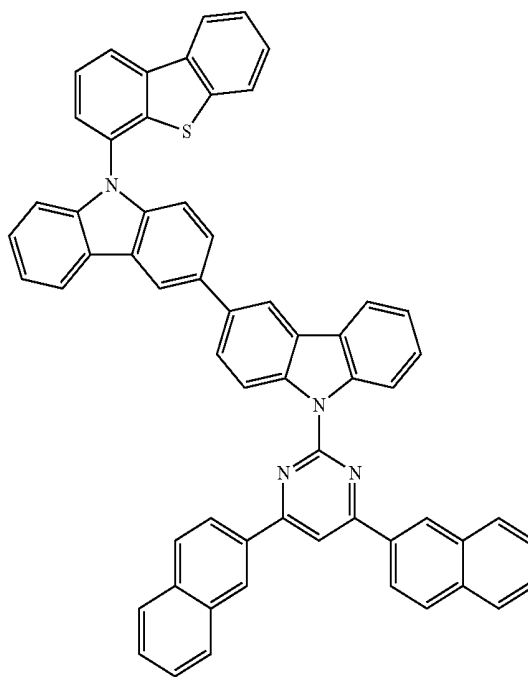
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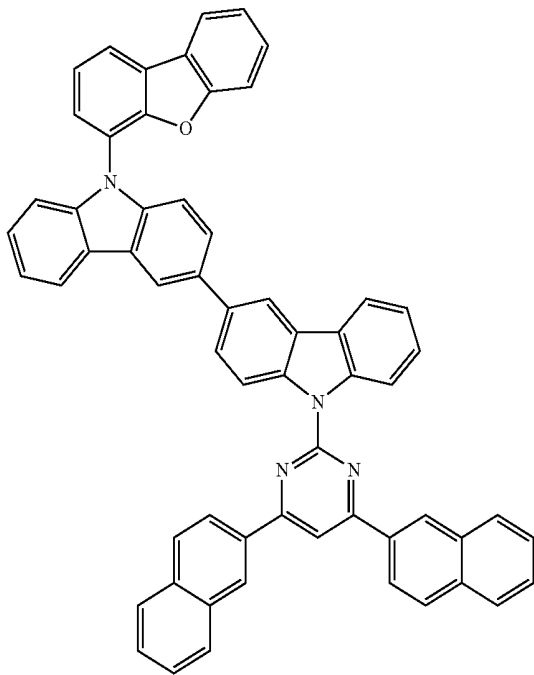
Compound 138



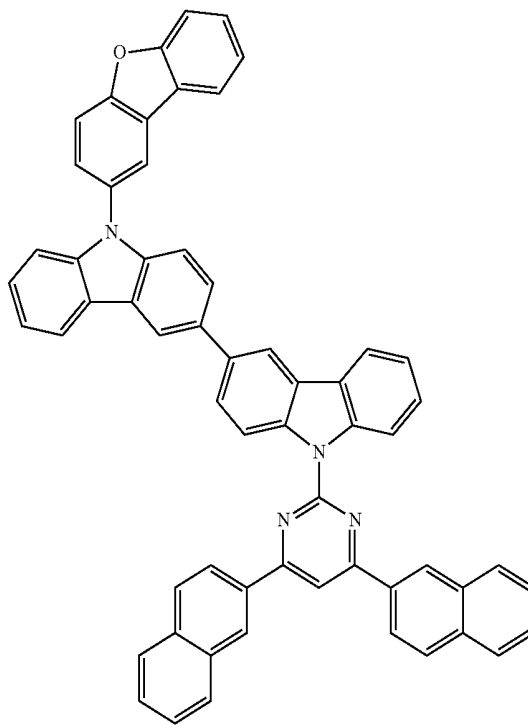
Compound 140



Compound 139



Compound 141

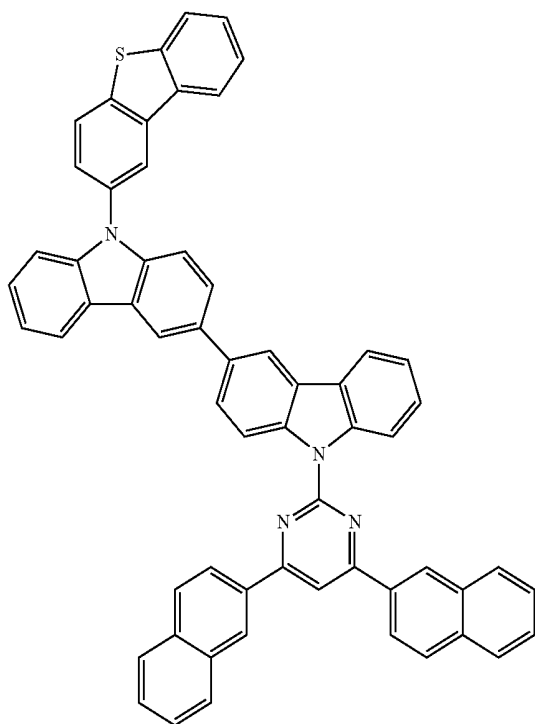




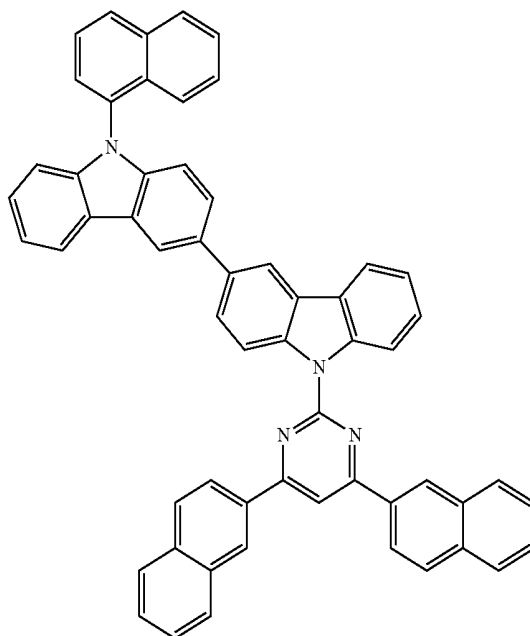
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Compound 142

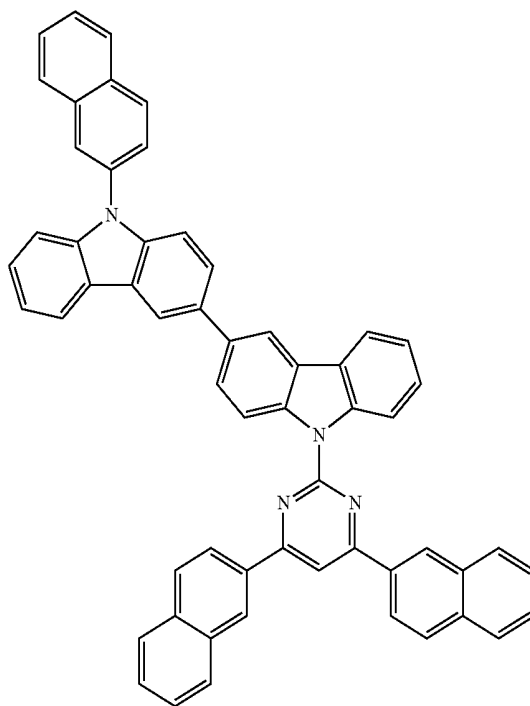
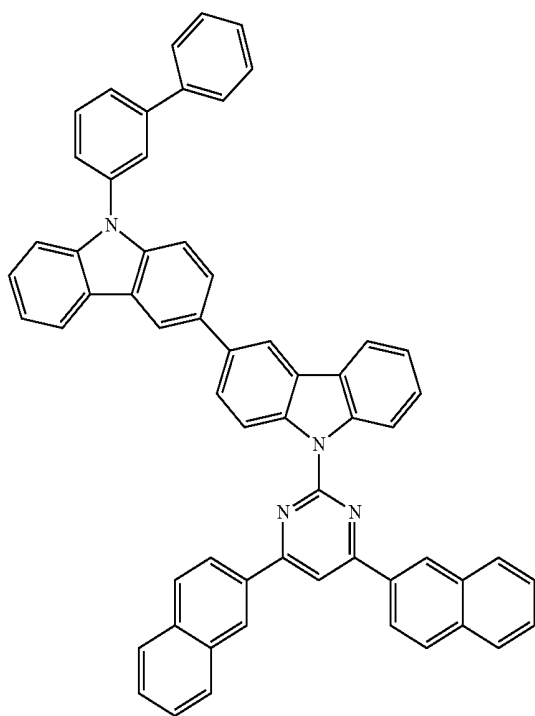


Compound 144



Compound 145

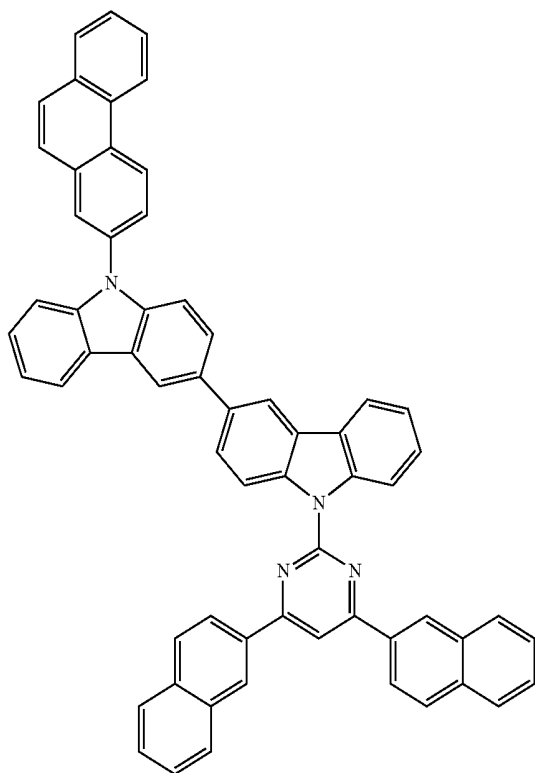
Compound 143



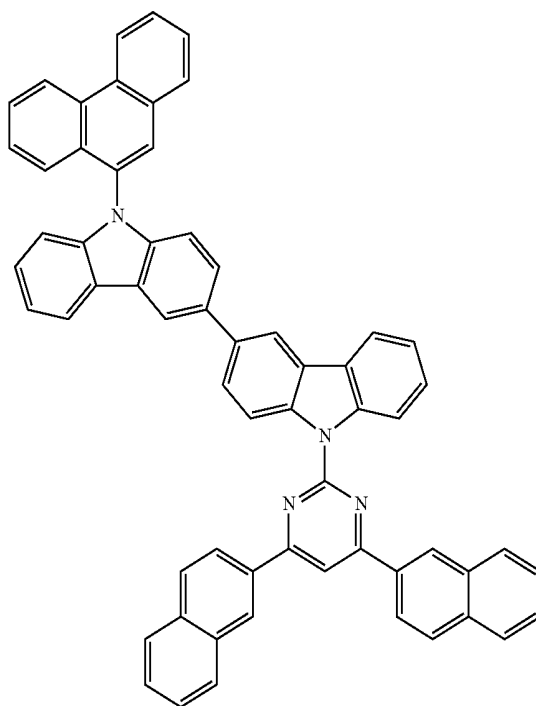
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Compound 146

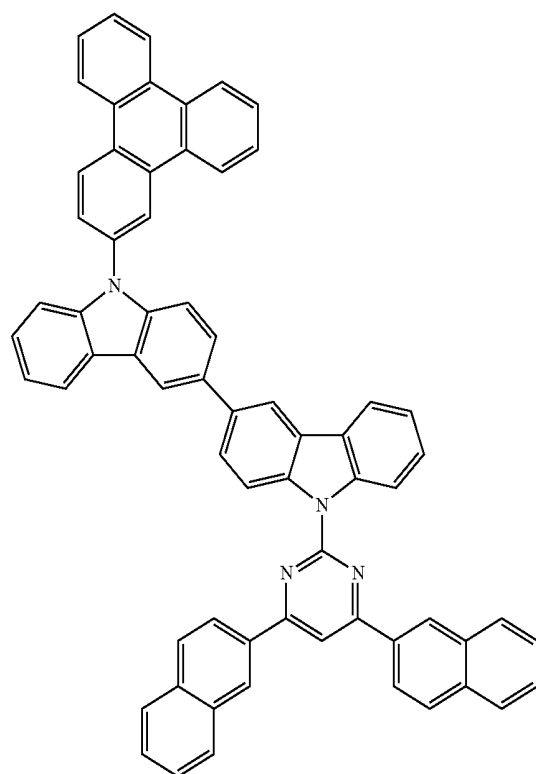
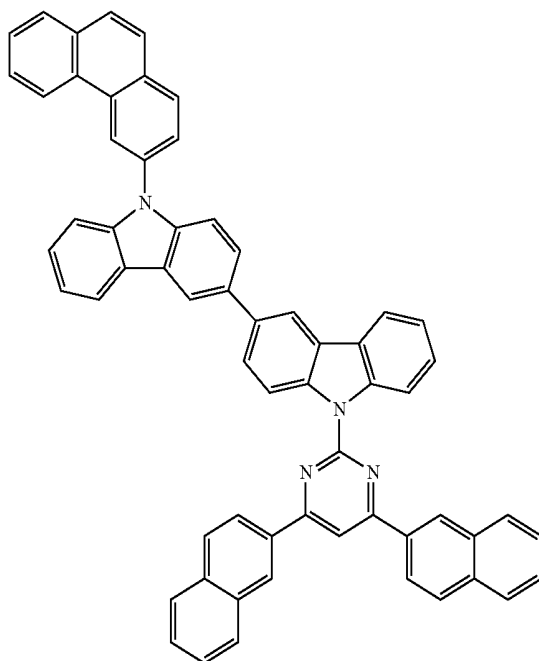


Compound 148



Compound 149

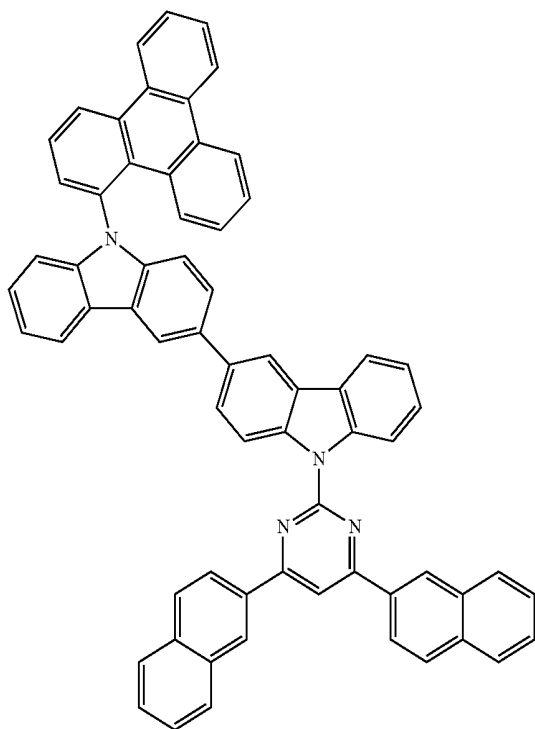
Compound 147



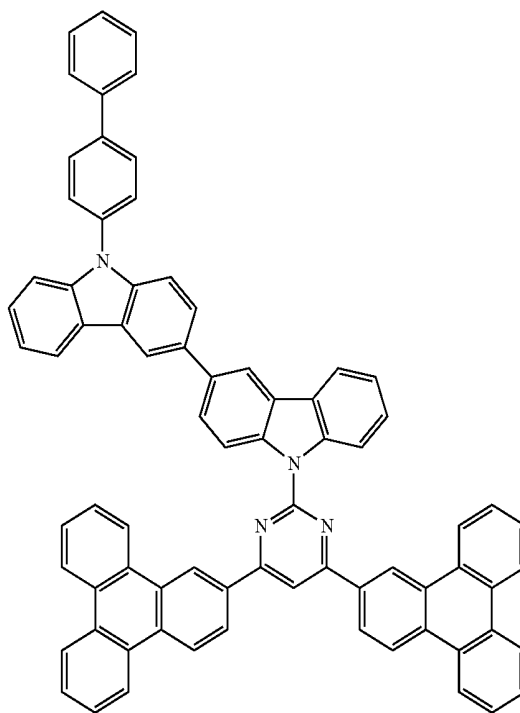
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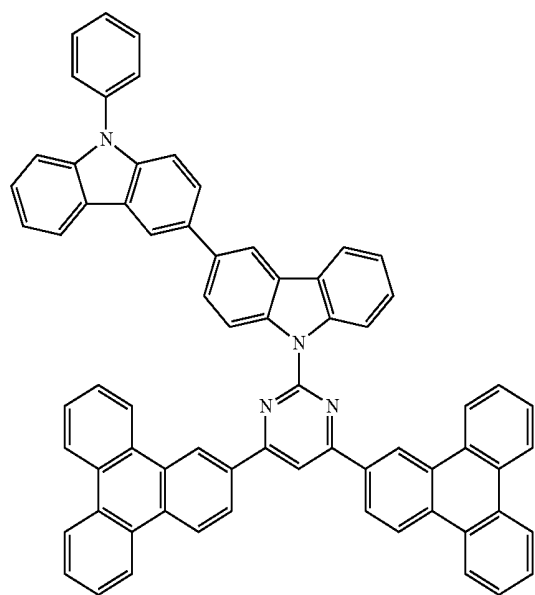
Compound 150



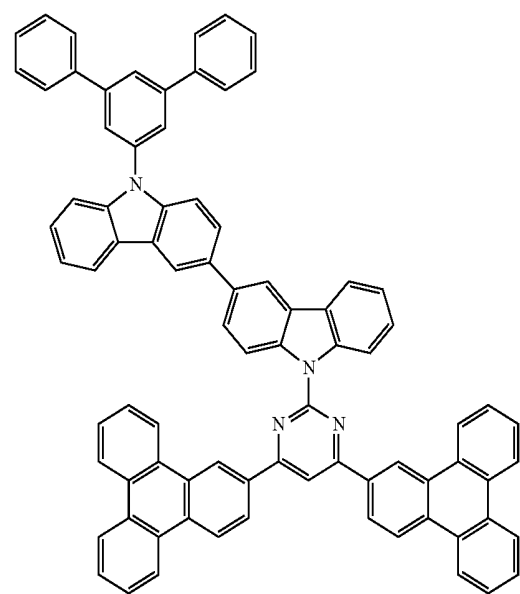
Compound 152



Compound 151



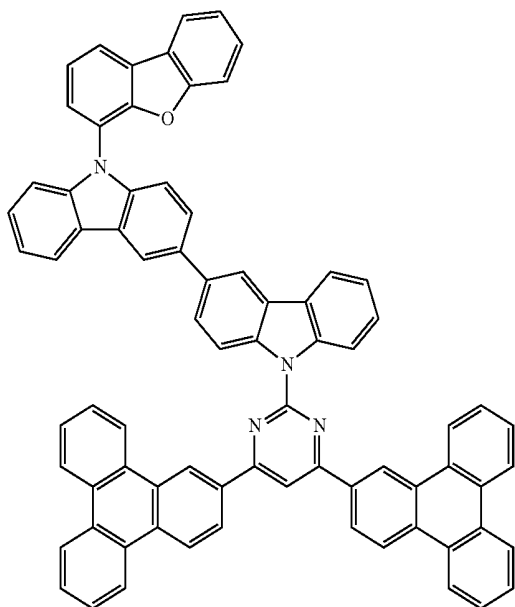
Compound 153



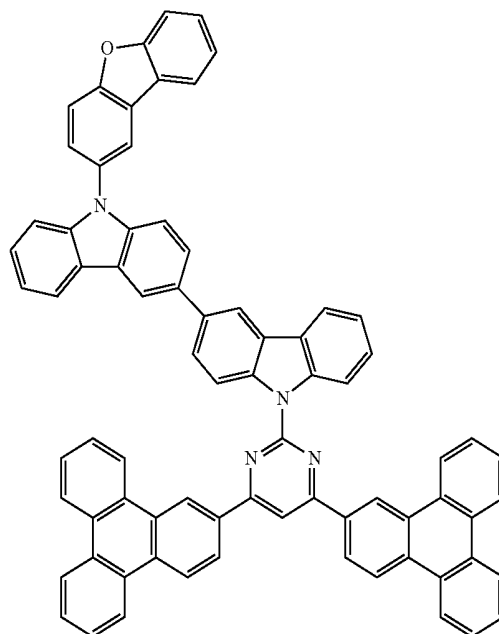
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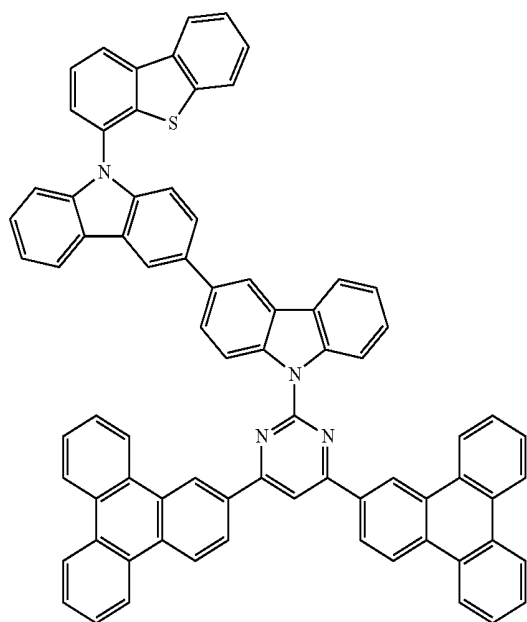
Compound 154



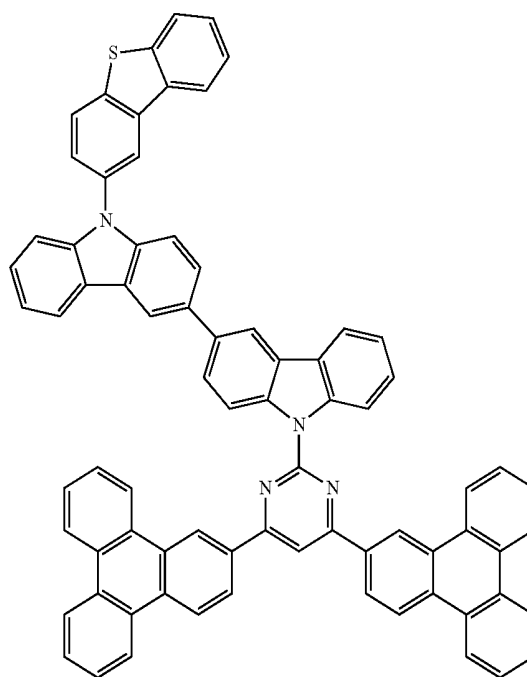
Compound 156



Compound 155



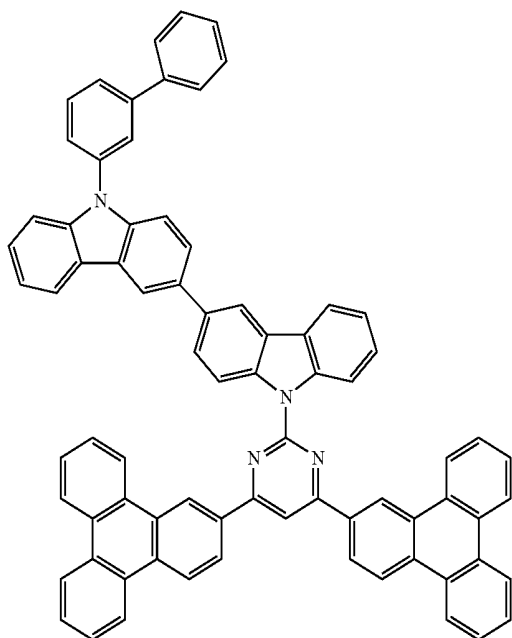
Compound 157



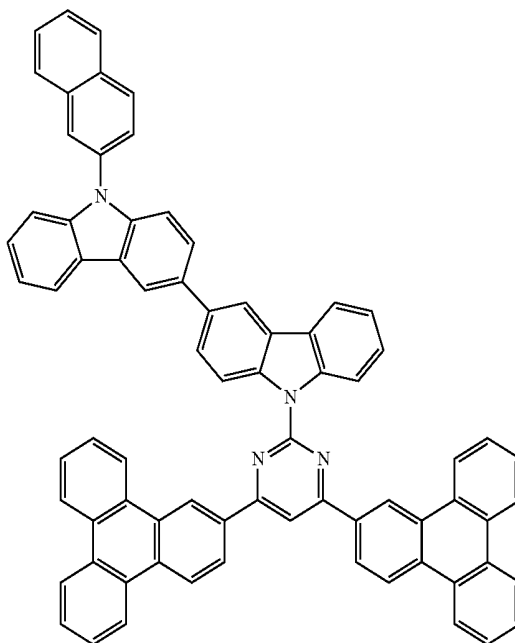
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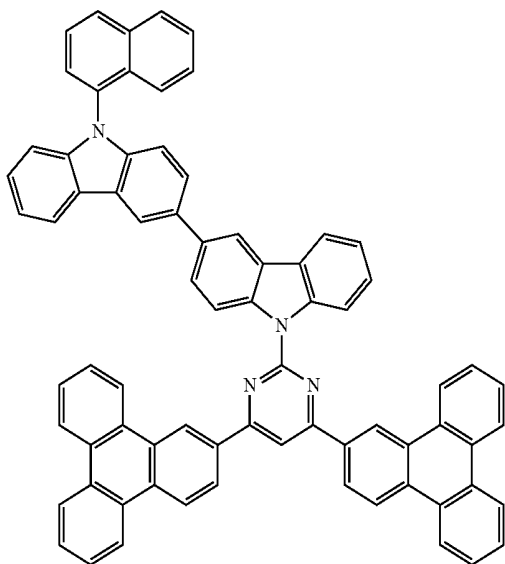
Compound 158



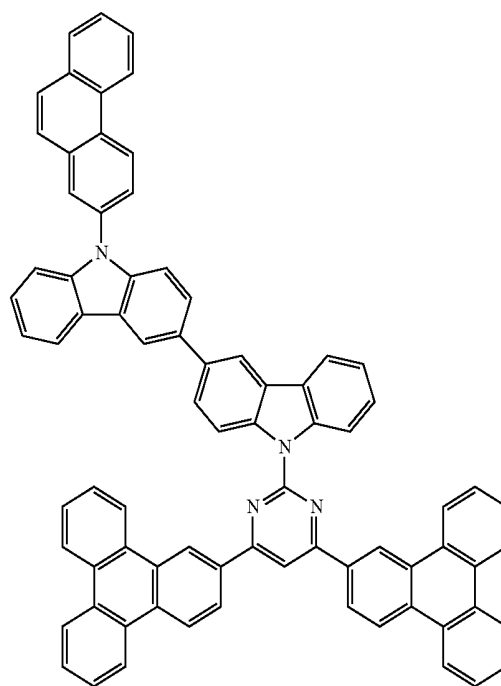
Compound 160



Compound 159



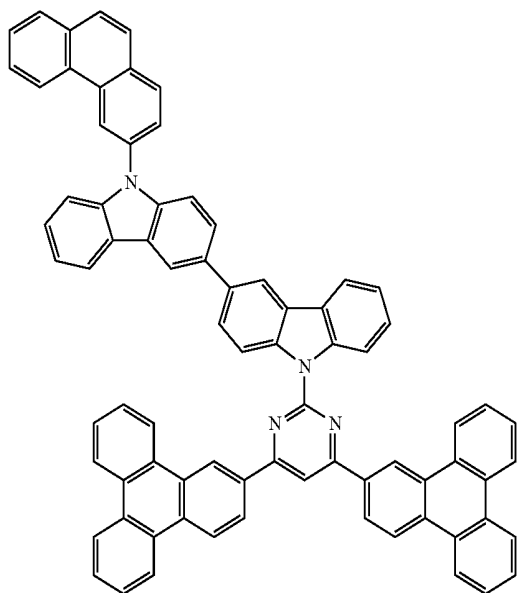
Compound 161



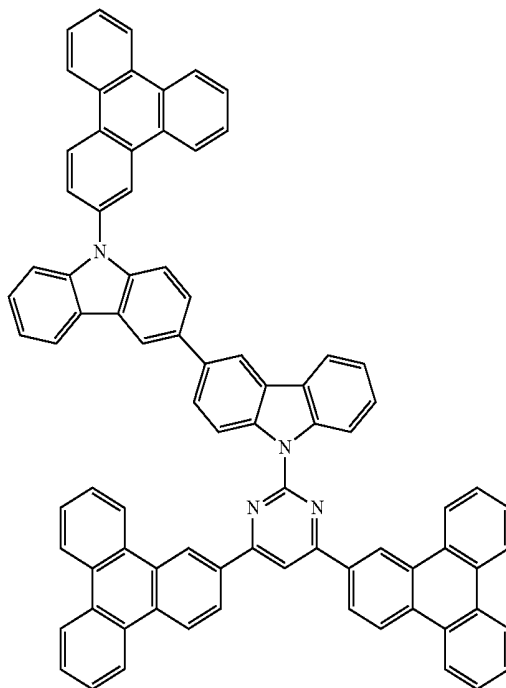
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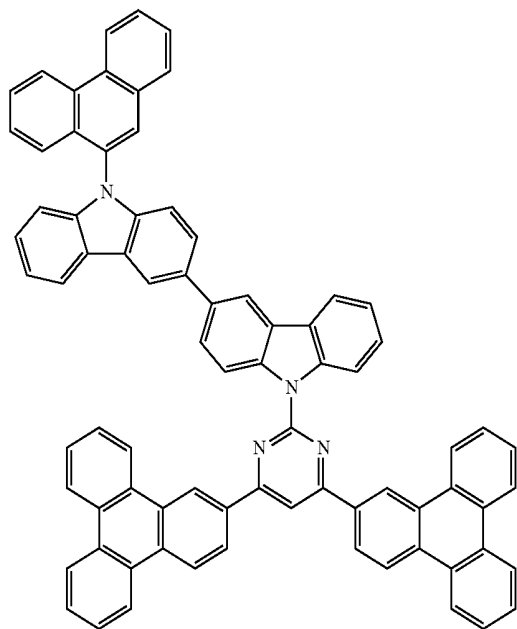
Compound 162



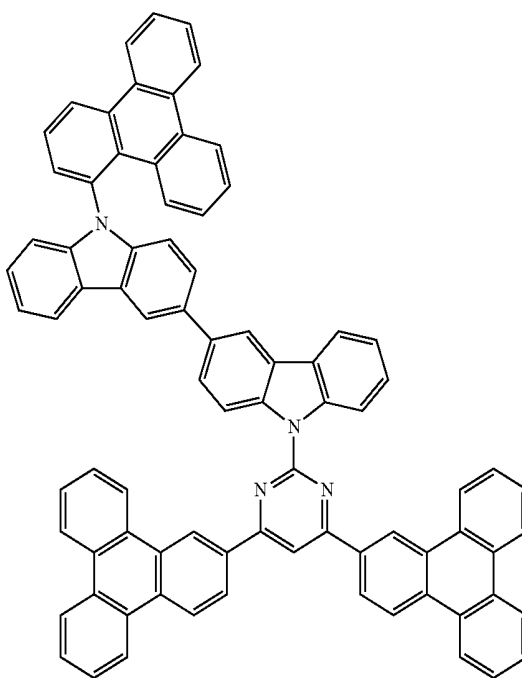
Compound 164



Compound 163



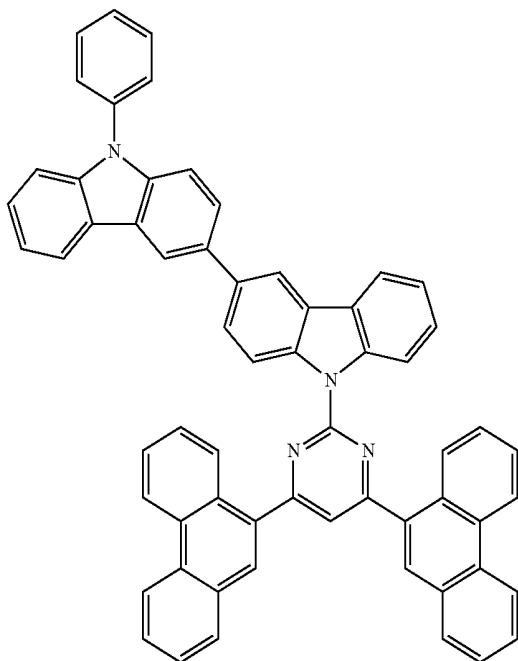
Compound 165



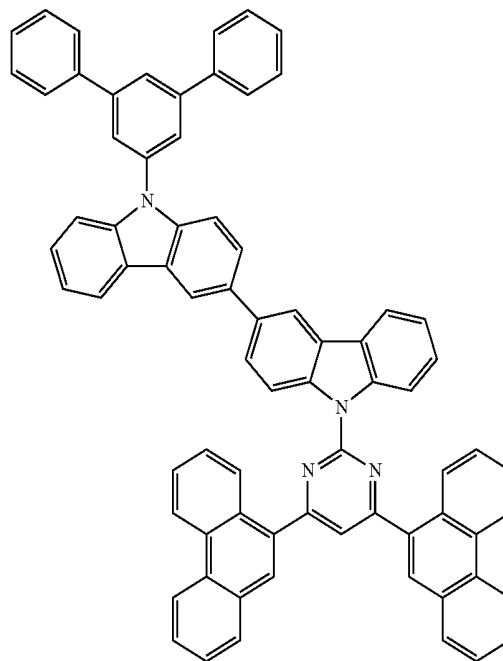
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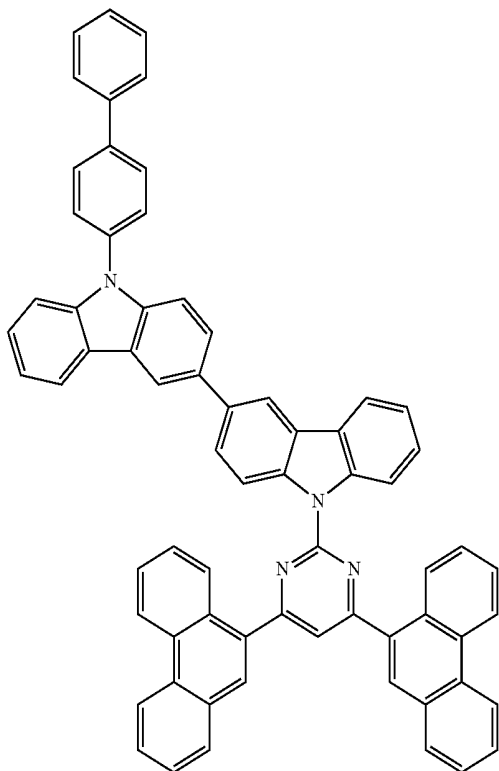
Compound 166



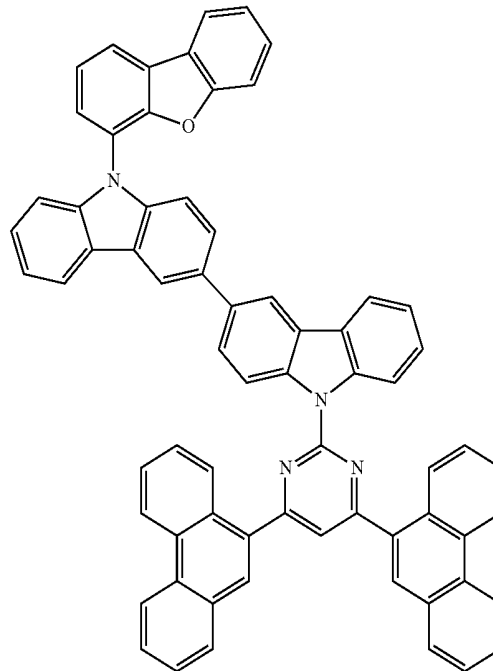
Compound 168



Compound 167

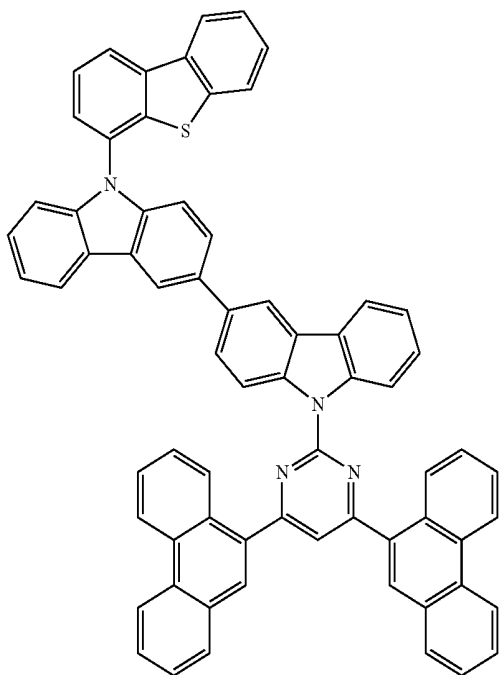


Compound 169



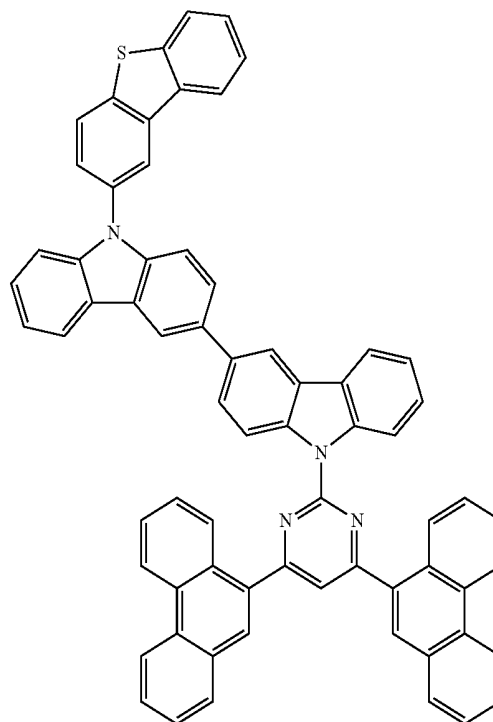
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Compound 170

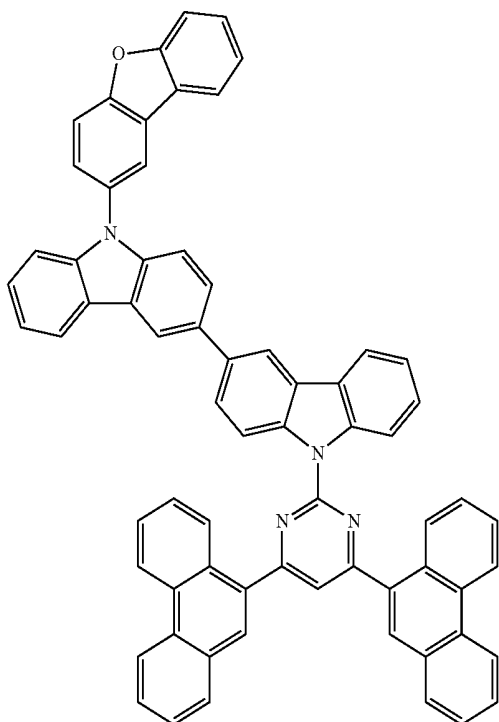


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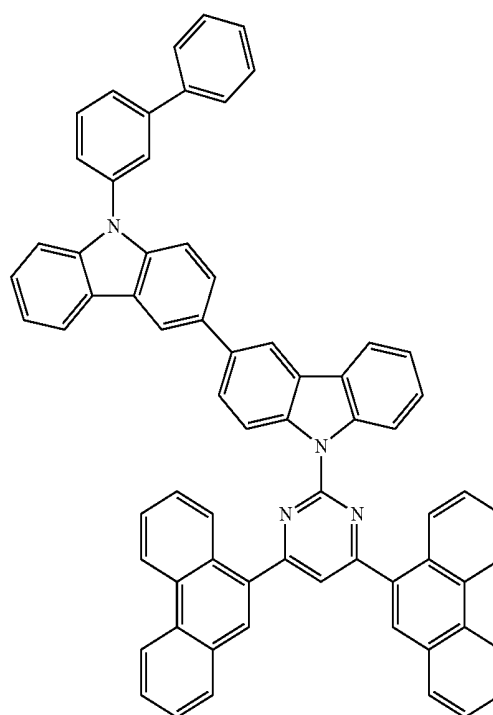
Compound 172



Compound 171



Compound 173

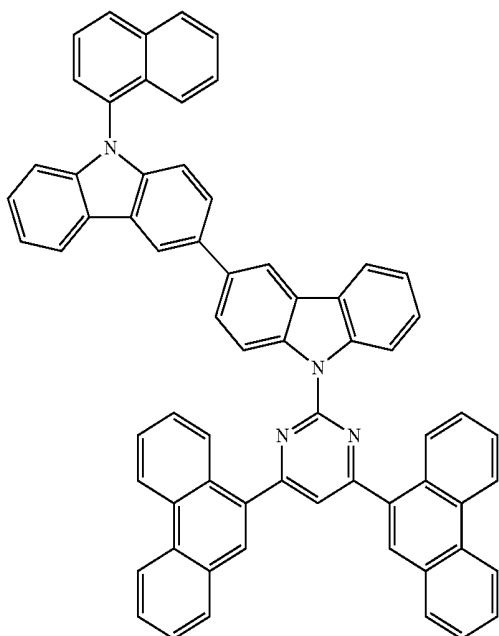




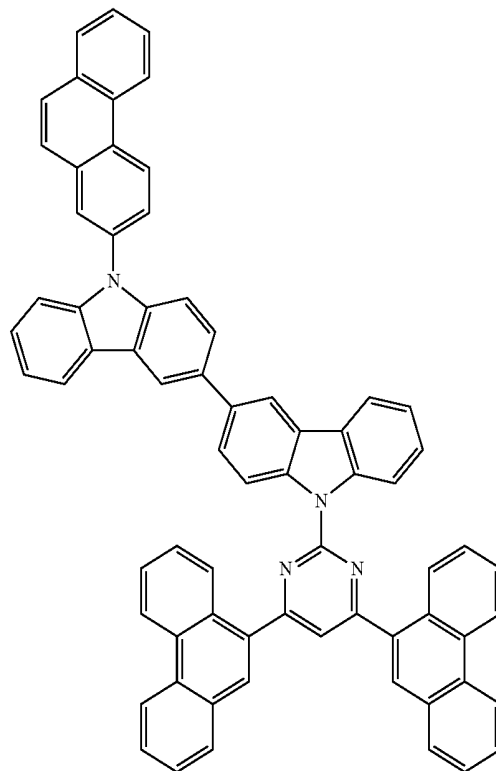
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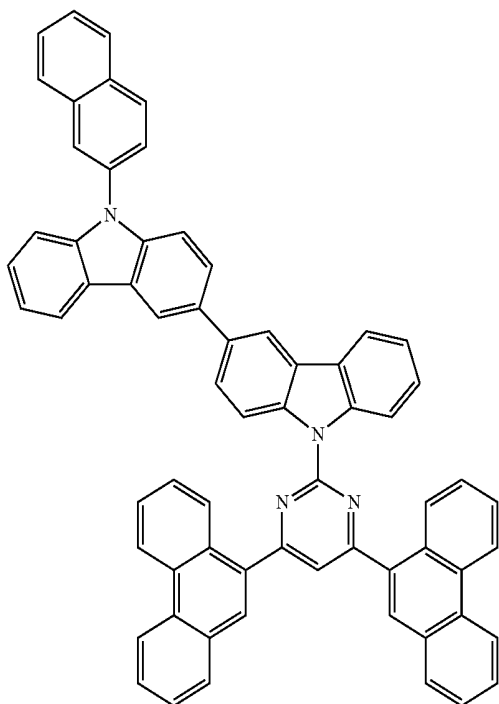
Compound 174



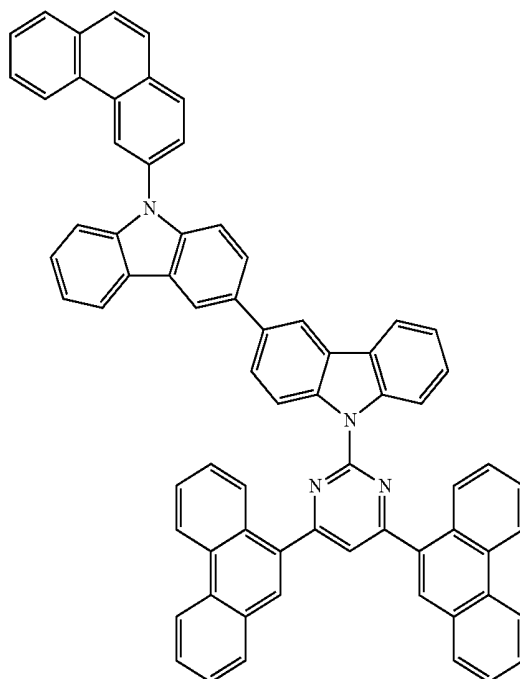
Compound 176



Compound 175



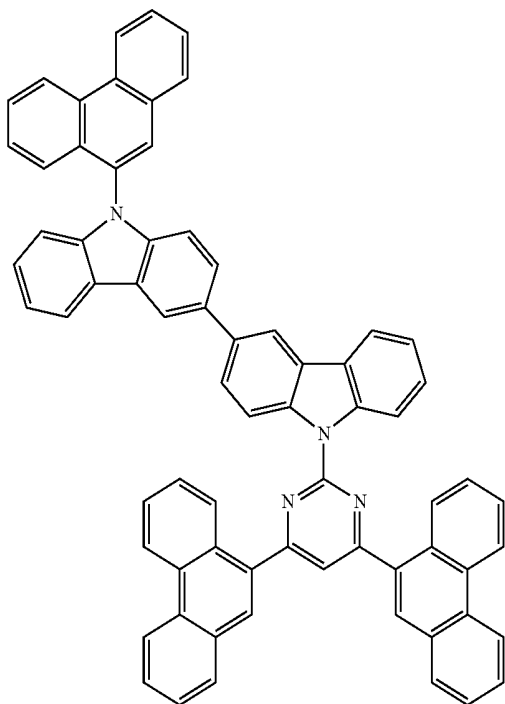
Compound 177



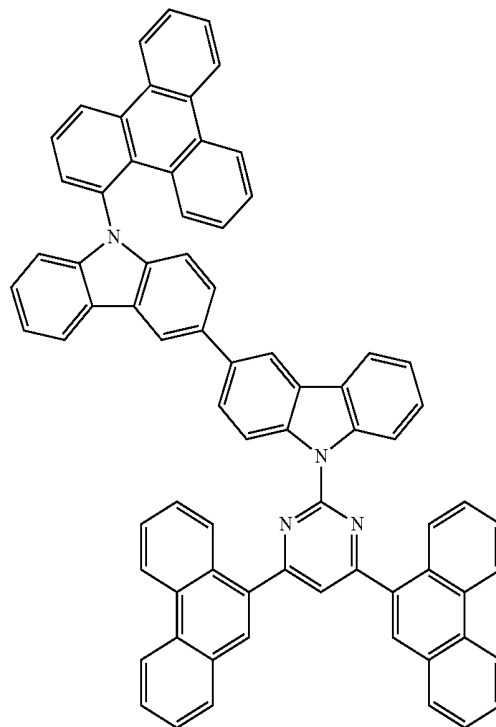
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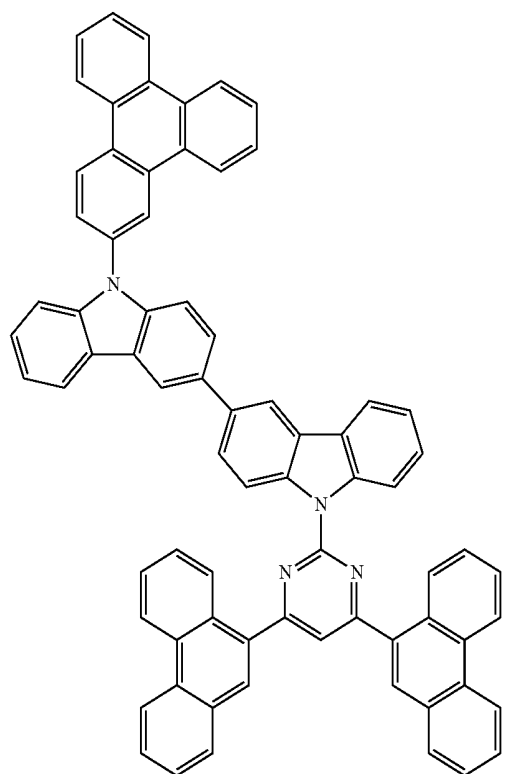
Compound 178



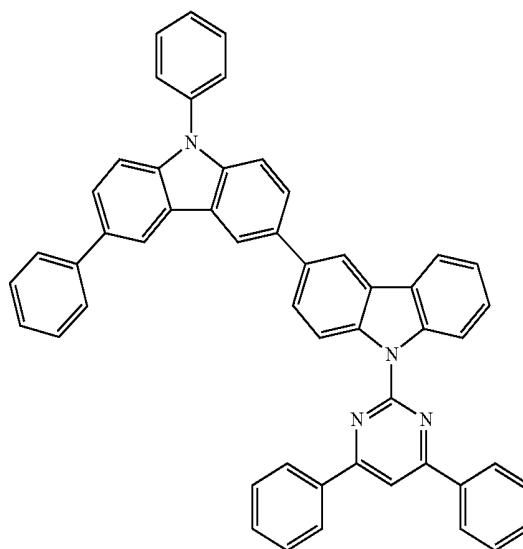
Compound 180



Compound 179

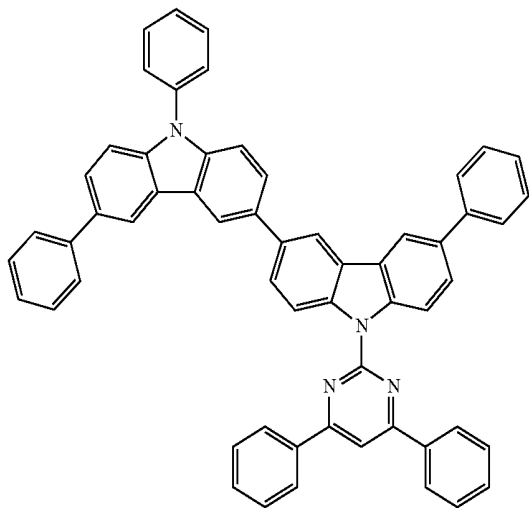


Compound 181

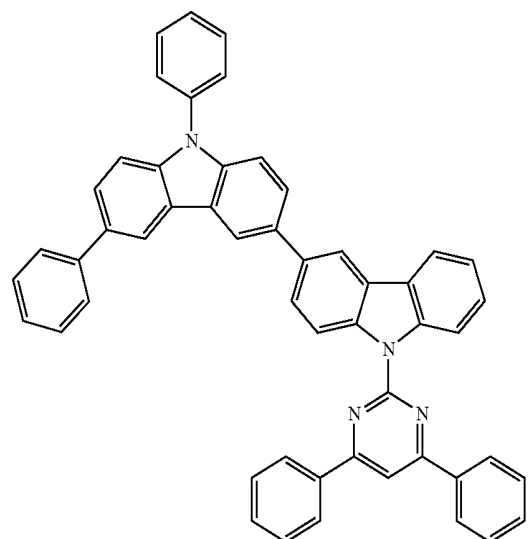


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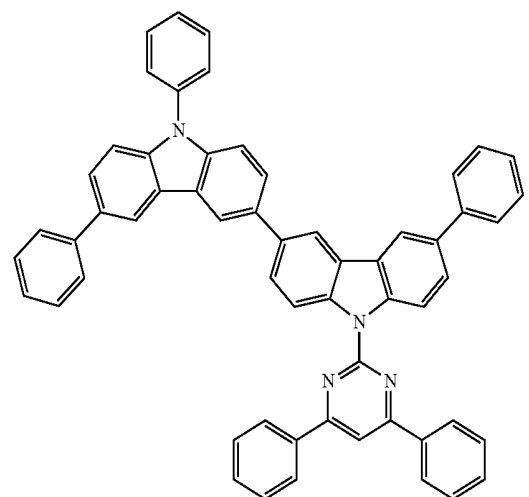
Compound 182



Compound 183



Compound 184



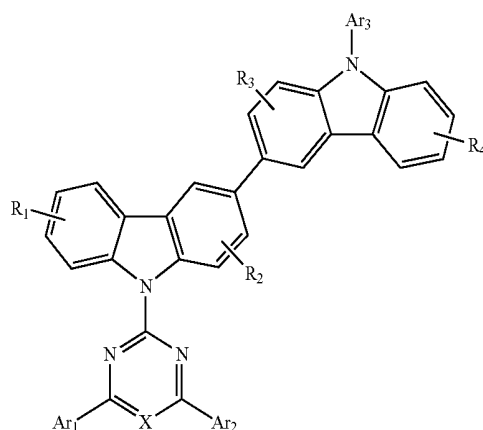
8. A first device comprising an organic light emitting device, further comprising:

an anode;

a cathode; and

an organic layer, disposed between the anode and the cathode, wherein the organic layer comprises a compound having the formula:

Formula I



wherein  $R_1$ ,  $R_2$ ,  $R_3$ , and  $R_4$  may represent mono, di, tri, or tetra substitutions;

wherein  $R_1$ ,  $R_2$ ,  $R_3$ , and  $R_4$  are independently selected from the group consisting of hydrogen, alkyl, alkoxy, amino, alkenyl, alkynyl, aryl and heteroaryl;

wherein  $Ar_1$ ,  $Ar_2$ , and  $Ar_3$  are independently selected from aryl or heteroaryl; and

wherein X is C or N.

9. The device of claim 8, wherein  $Ar_1$ ,  $Ar_2$ , and  $Ar_3$  are further substituted.

10. The device of claim 8, wherein  $Ar_1$ ,  $Ar_2$ , and  $Ar_3$  are independently selected from the group consisting of phenyl, pyridine, naphthalene, biphenyl, terphenyl, fluorene, dibenzofuran, dibenzothiophene, phenanthrene, and triphenylene; and

wherein  $Ar_1$ ,  $Ar_2$ , and  $Ar_3$  are independently further substituted with a substituent selected from the group consisting of hydrogen, alkyl, alkoxy, amino, alkenyl, alkynyl, aryl and heteroaryl, wherein the substituent is not an aryl or heteroaryl fused directly to  $Ar_1$ ,  $Ar_2$ , and  $Ar_3$ .

11. The device of claim 8, wherein  $Ar_1$  and  $Ar_2$  are independently selected from the group consisting of phenyl, pyridine, and naphthalene.

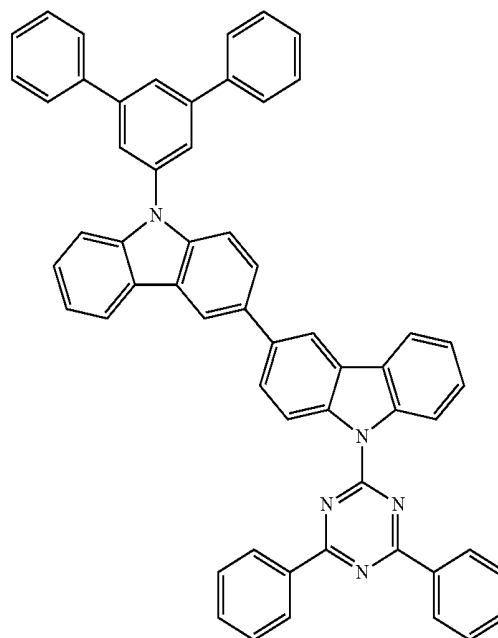
12. The device of claim 8, wherein  $Ar_3$  is selected from the group consisting of phenyl, biphenyl, dibenzofuran, and dibenzothiophene.

13. The device of claim 8, wherein  $R_1$ ,  $R_2$ ,  $R_3$ , and  $R_4$  are hydrogen.

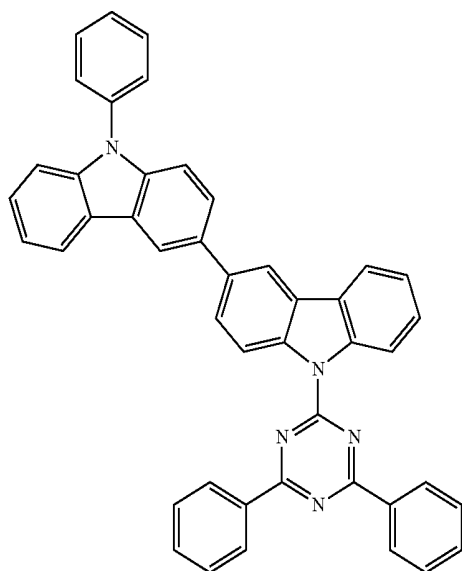
14. The device of claim 8, wherein the compound is selected from the group consisting of:

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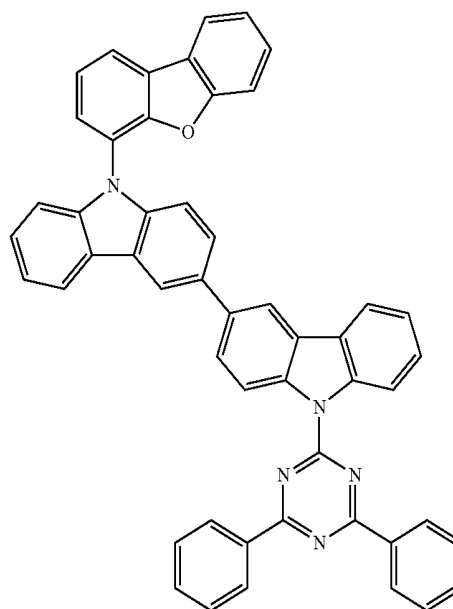
Compound 3



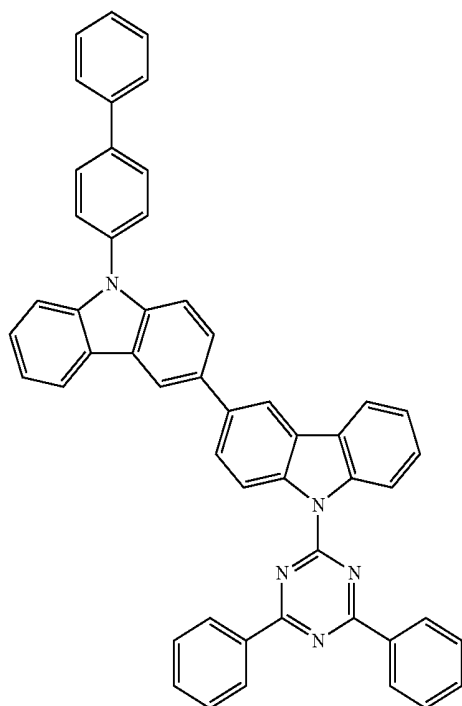
Compound 1



Compound 4



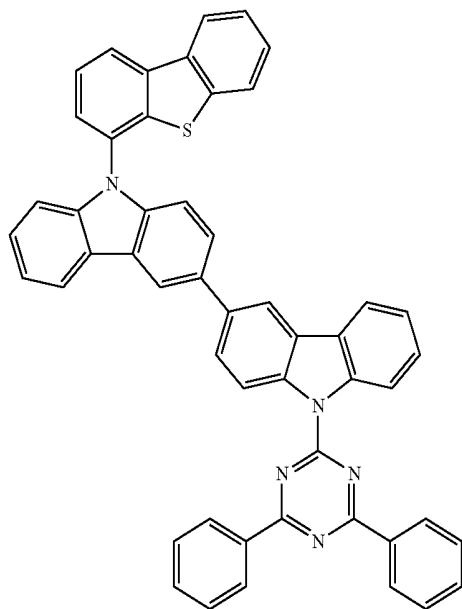
Compound 2



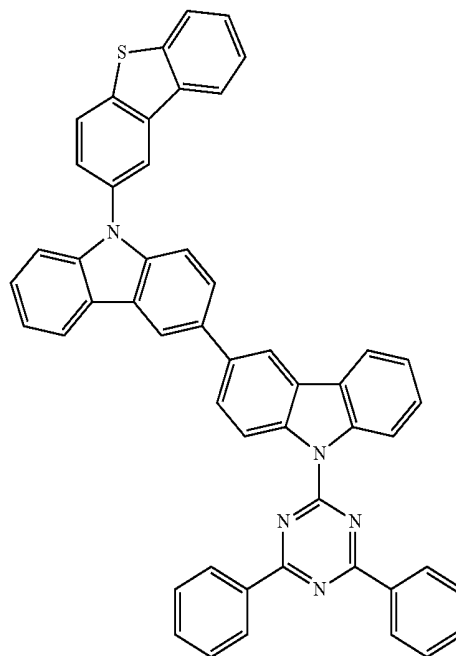
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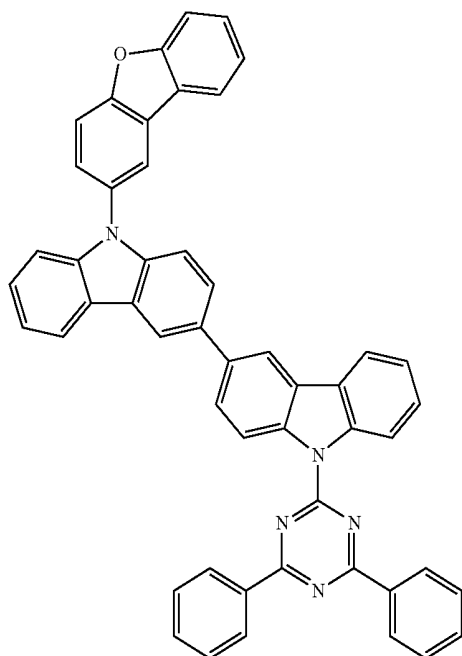
Compound 5



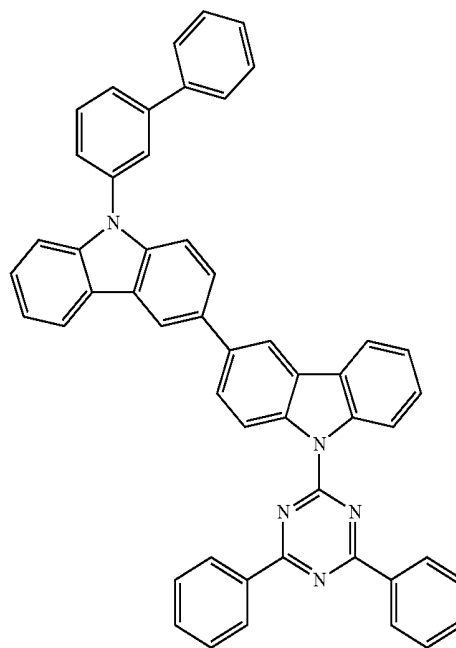
Compound 7



Compound 6



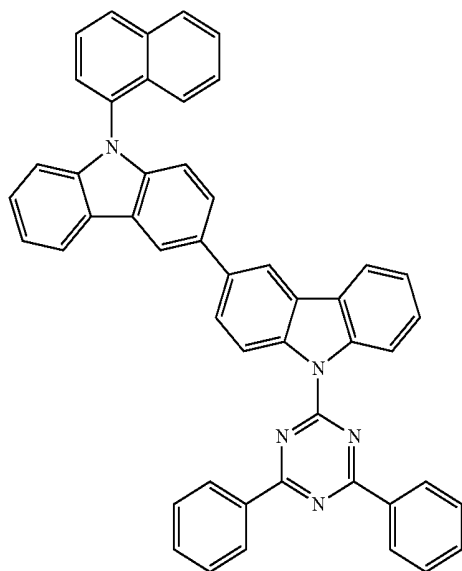
Compound 8



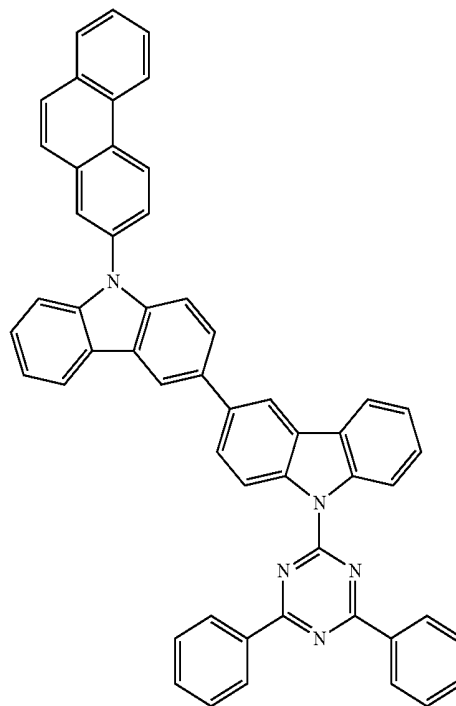
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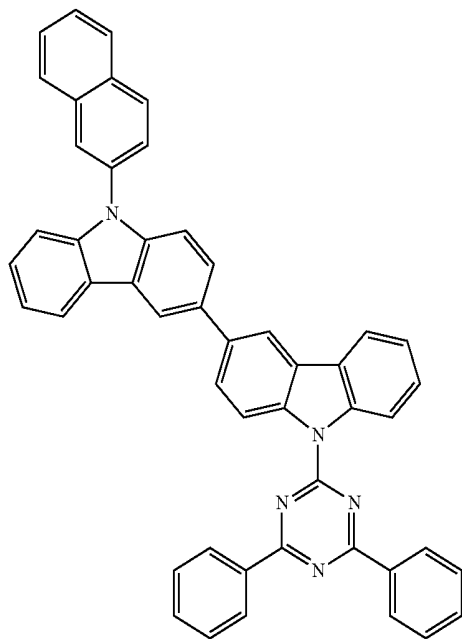
Compound 9



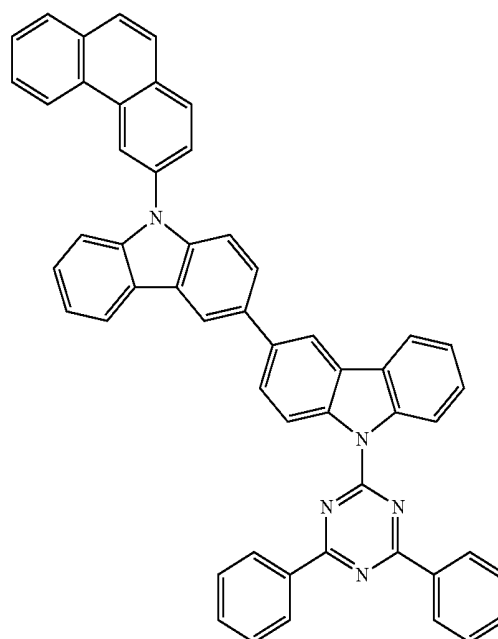
Compound 11



Compound 10



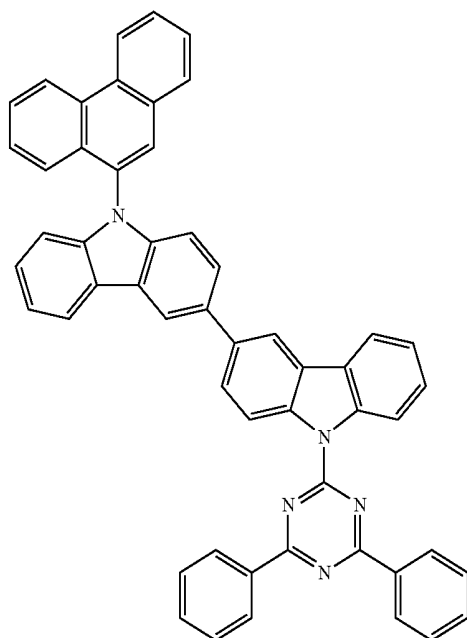
Compound 12



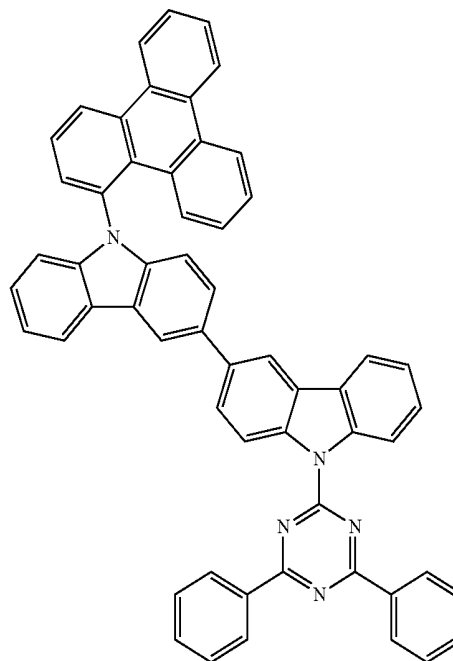
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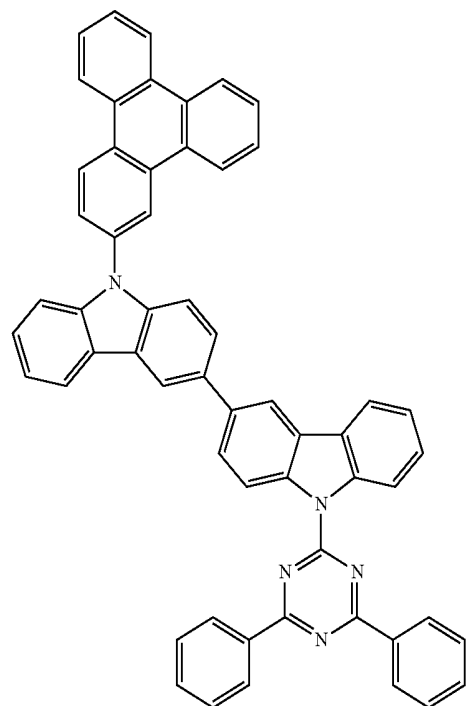
Compound 13



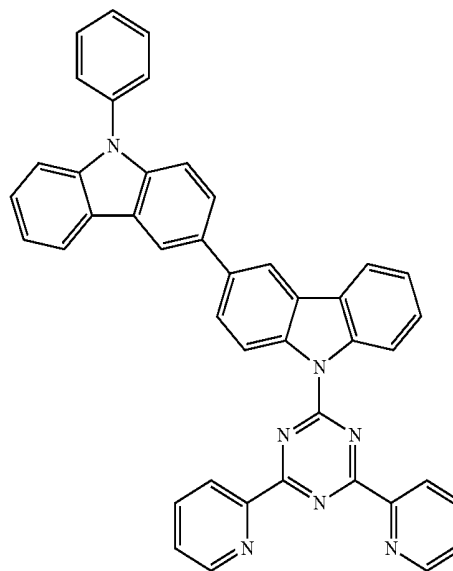
Compound 15



Compound 14



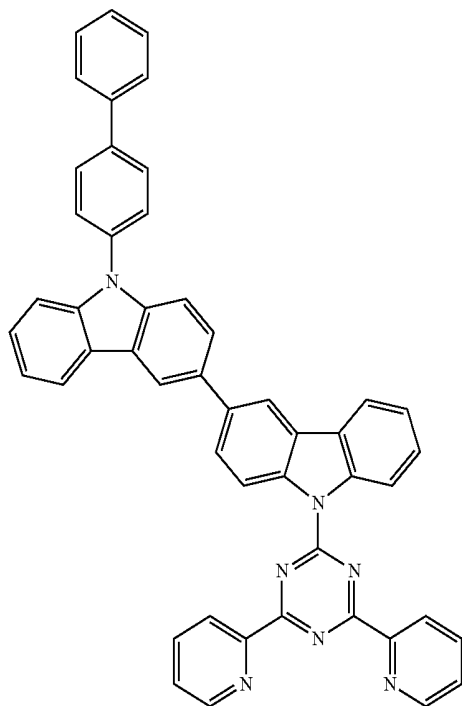
Compound 16



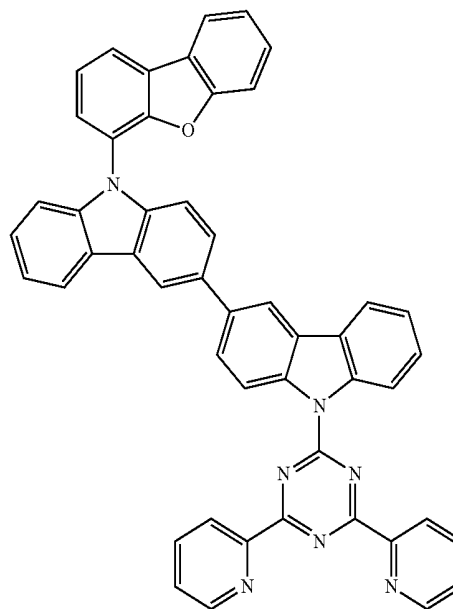
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Compound 17

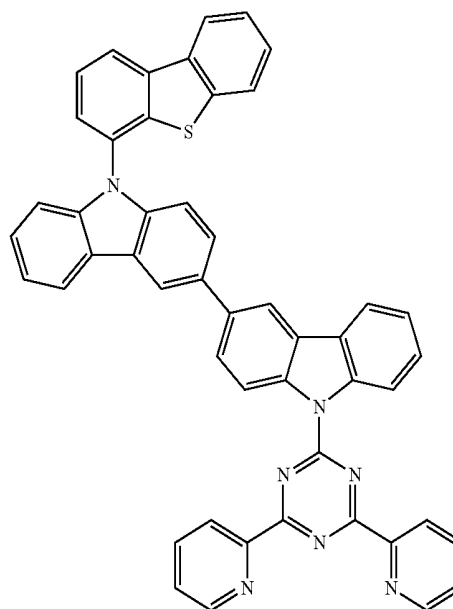
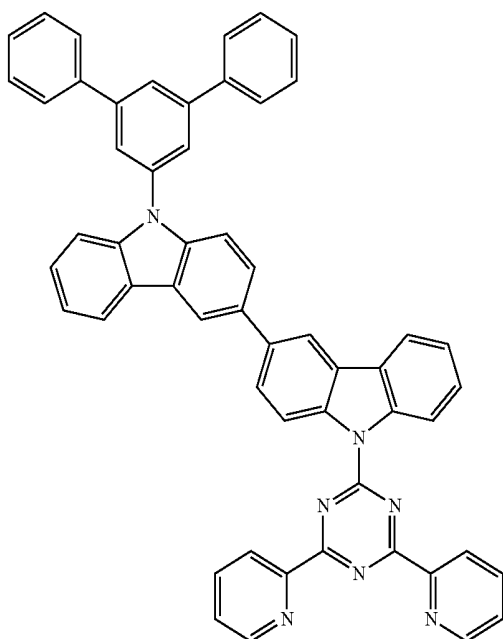


Compound 19



Compound 20

Compound 18

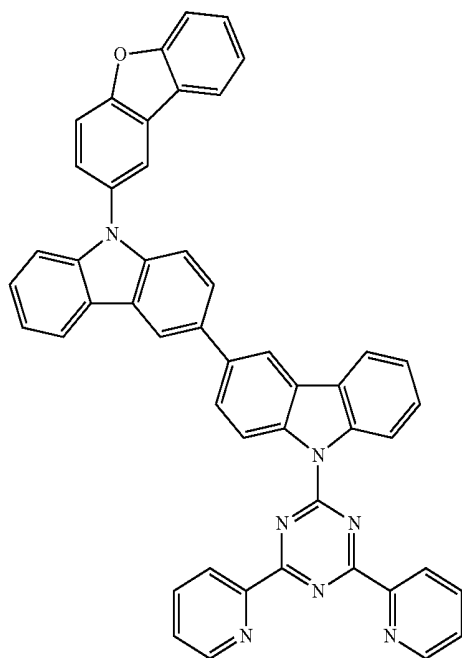




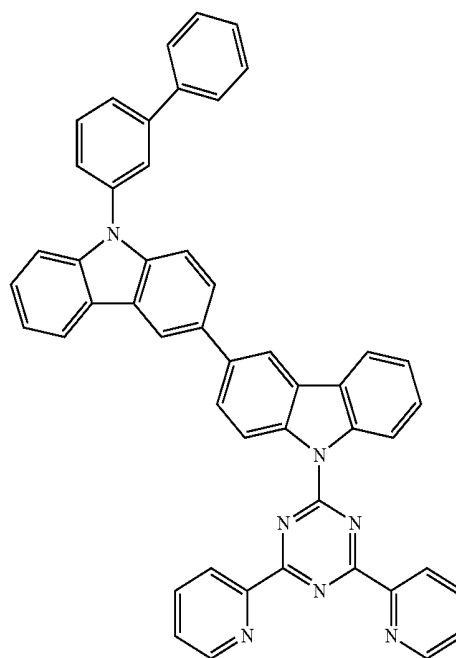
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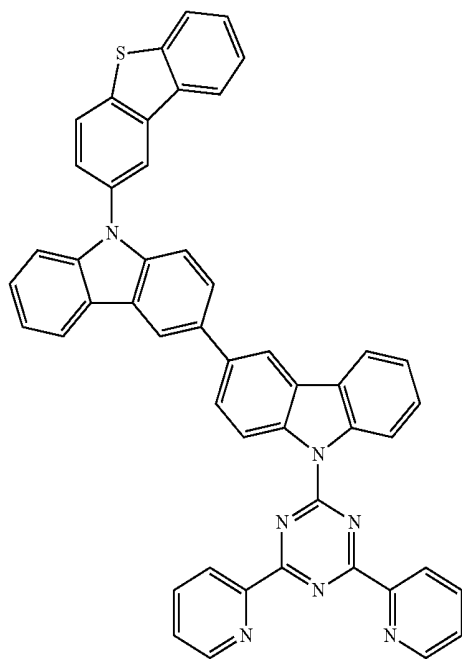
Compound 21



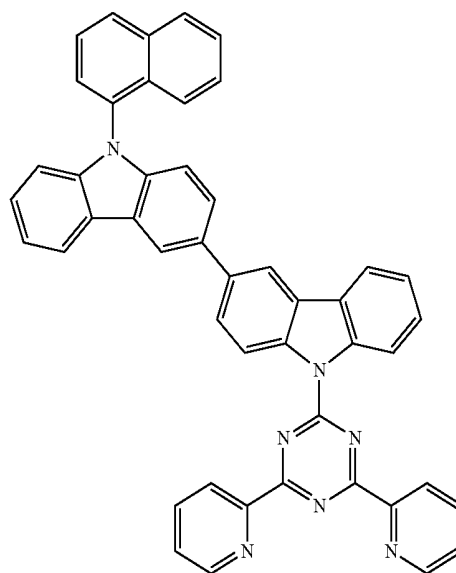
Compound 23



Compound 22



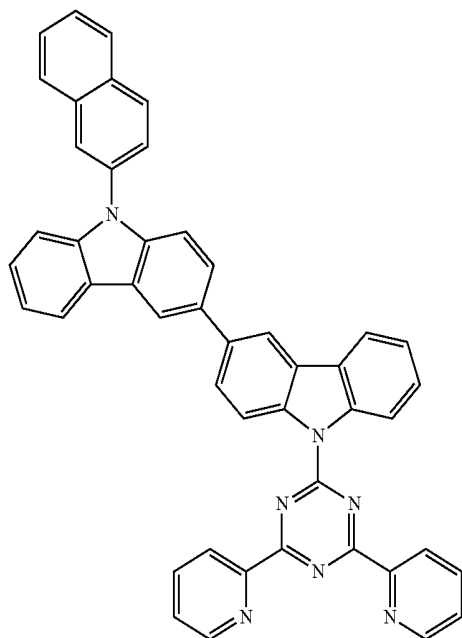
Compound 24



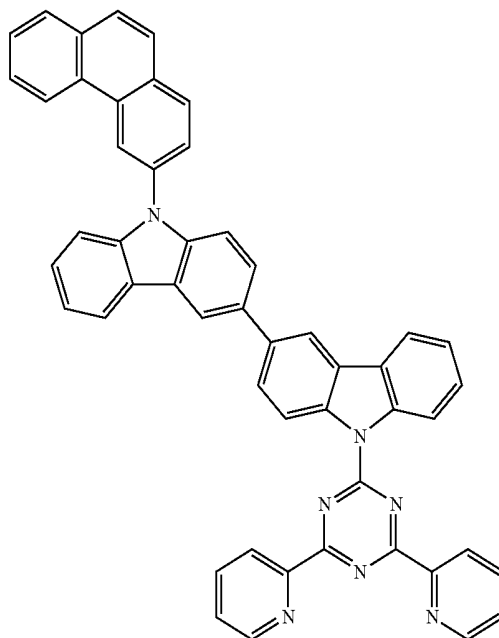
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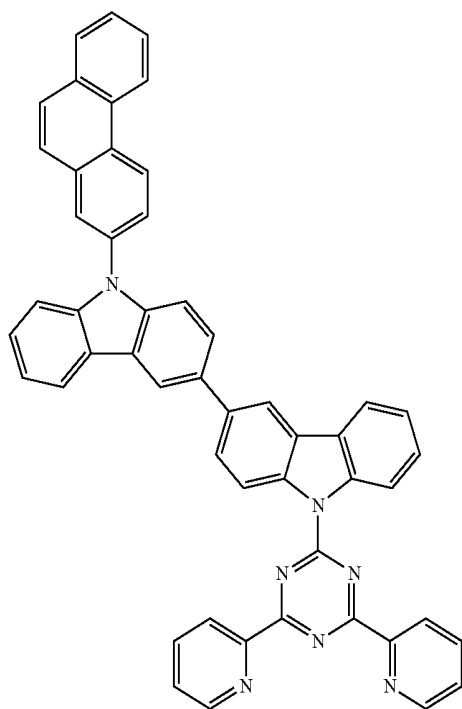
Compound 25



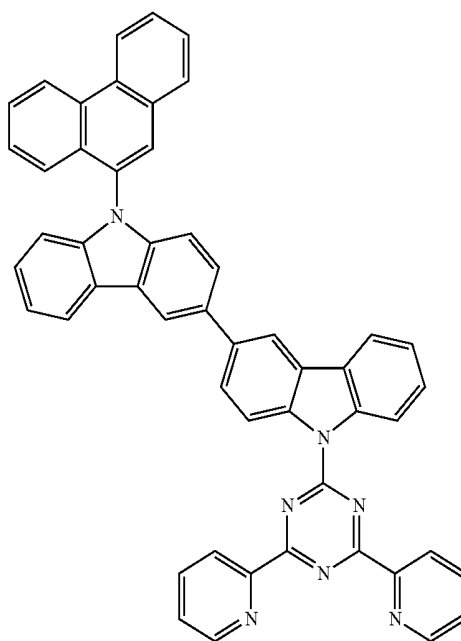
Compound 27



Compound 26



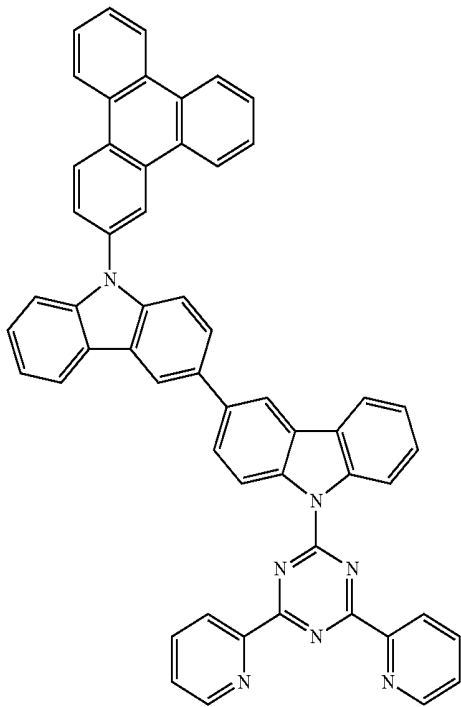
Compound 28



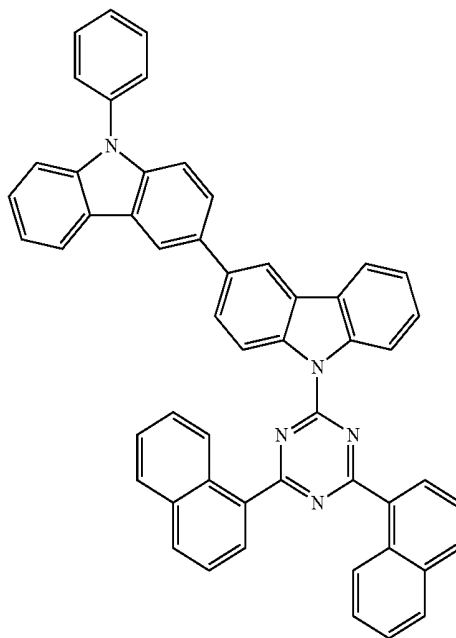
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Compound 29

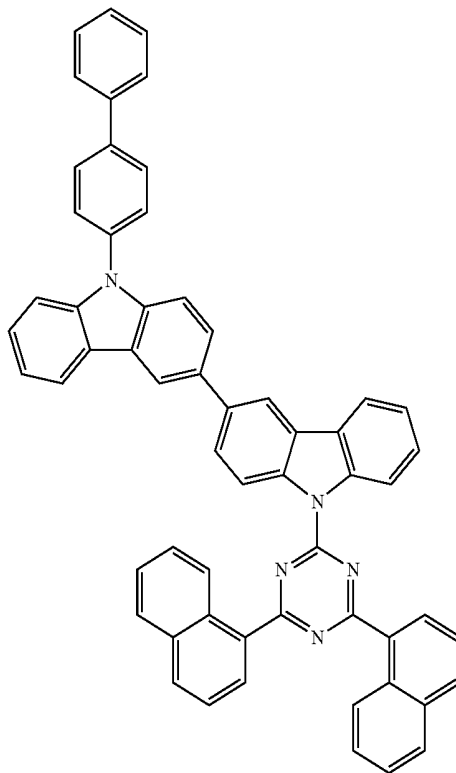
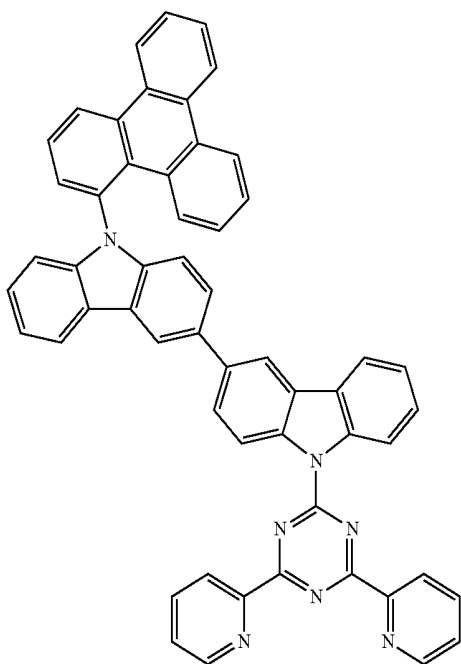


Compound 31



Compound 32

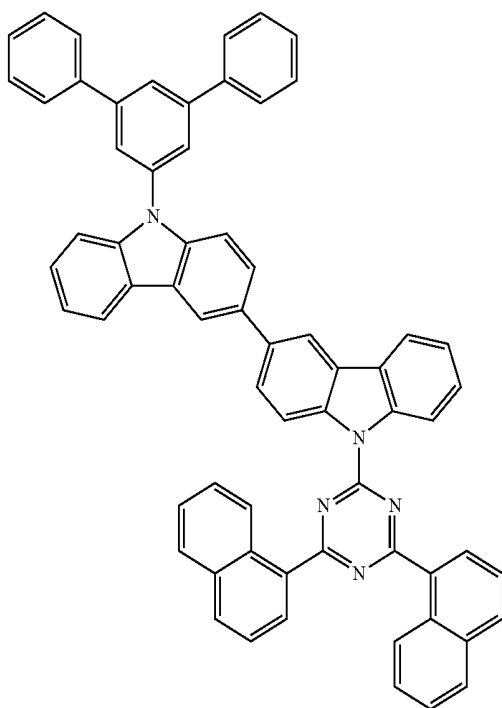
Compound 30



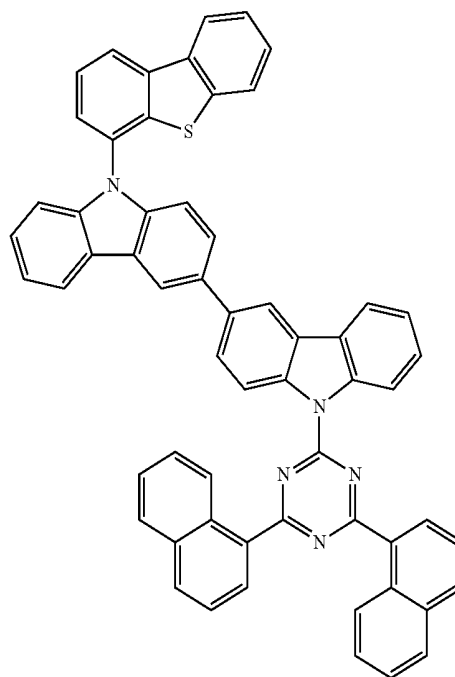
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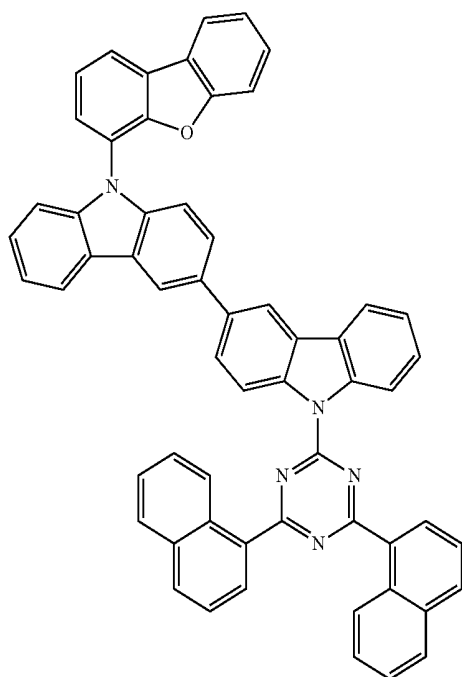
Compound 33



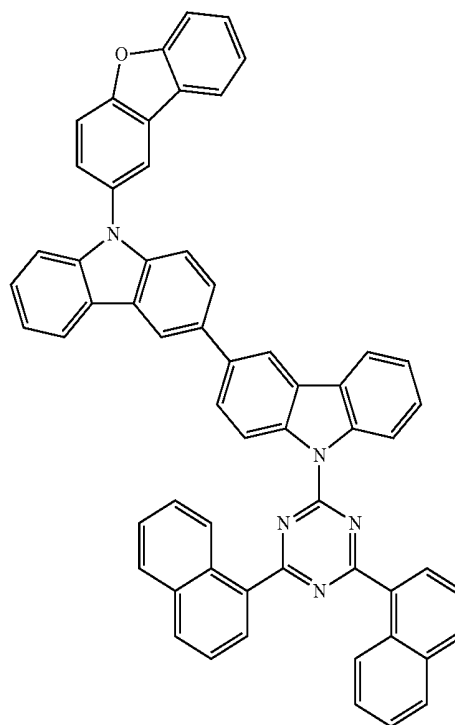
Compound 35



Compound 34

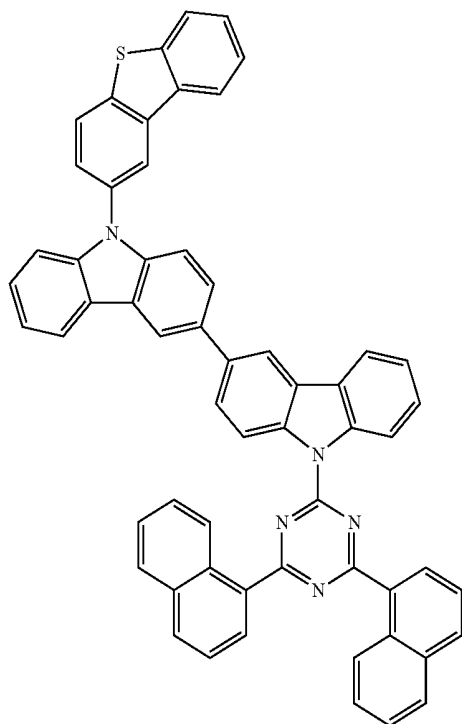


Compound 36



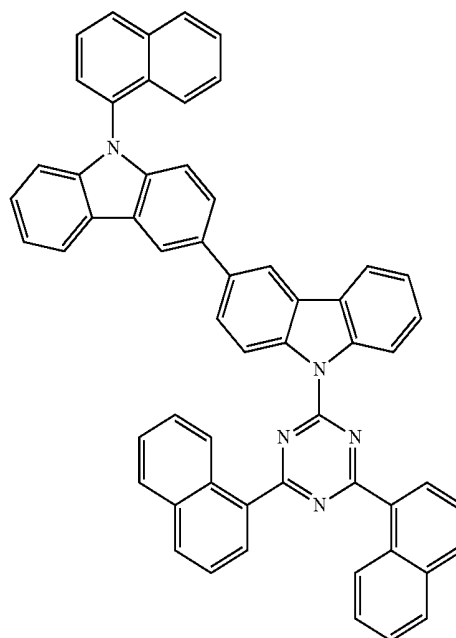
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Compound 37



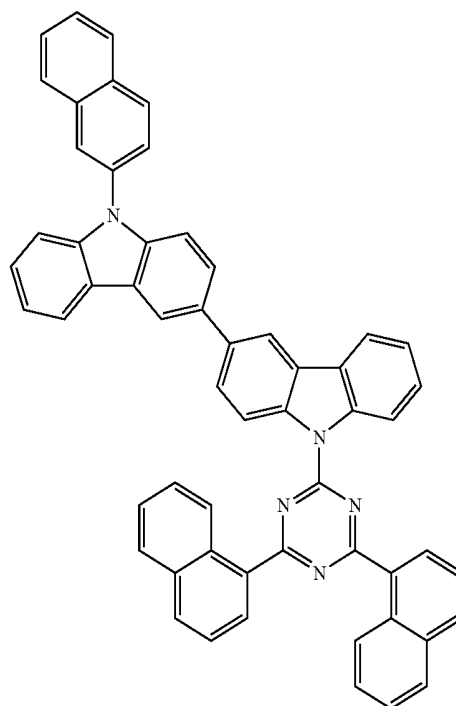
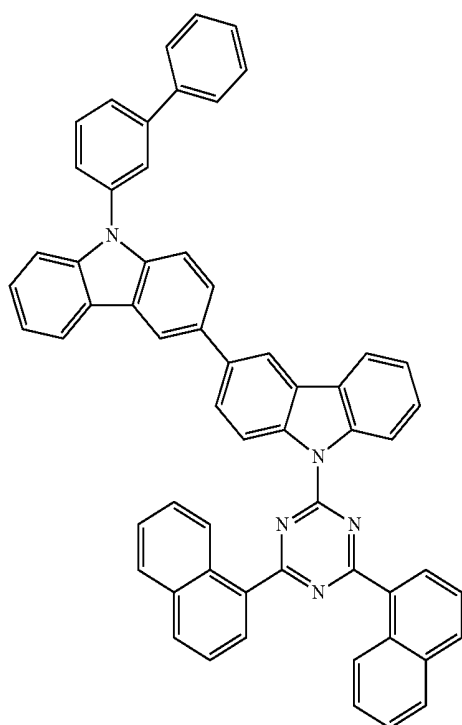
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Compound 39



Compound 40

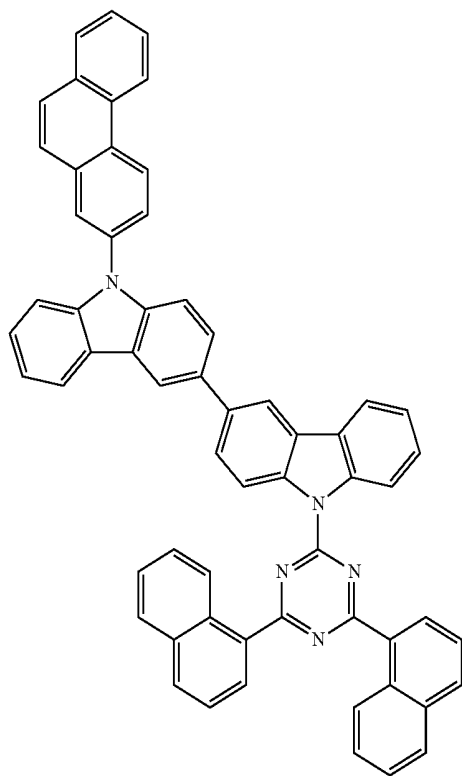
Compound 38



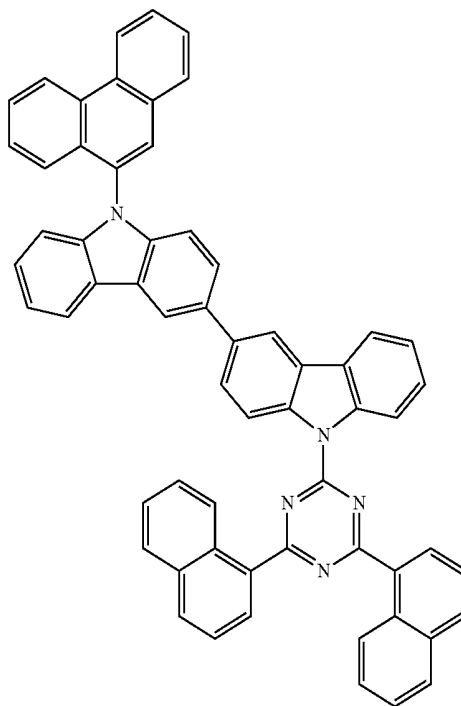
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Compound 41

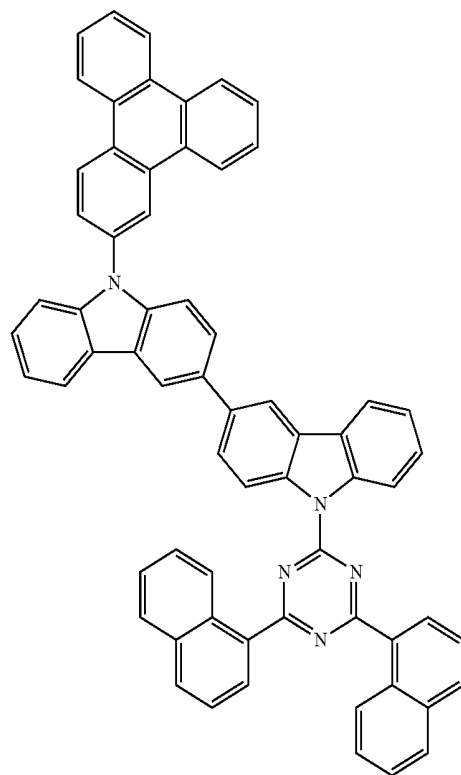
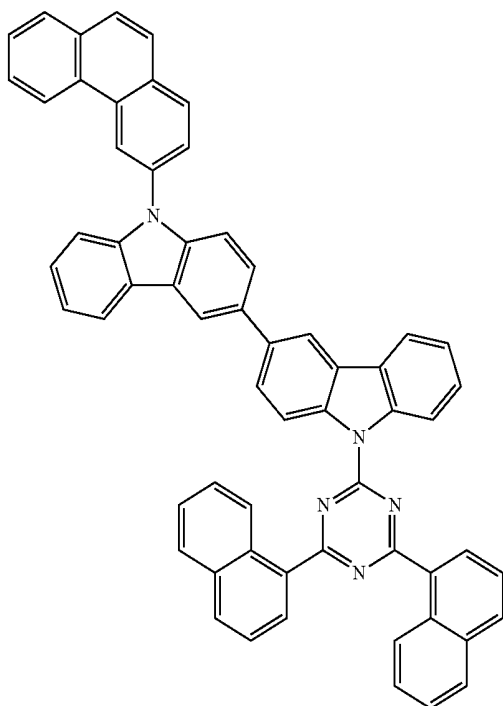


Compound 43



Compound 44

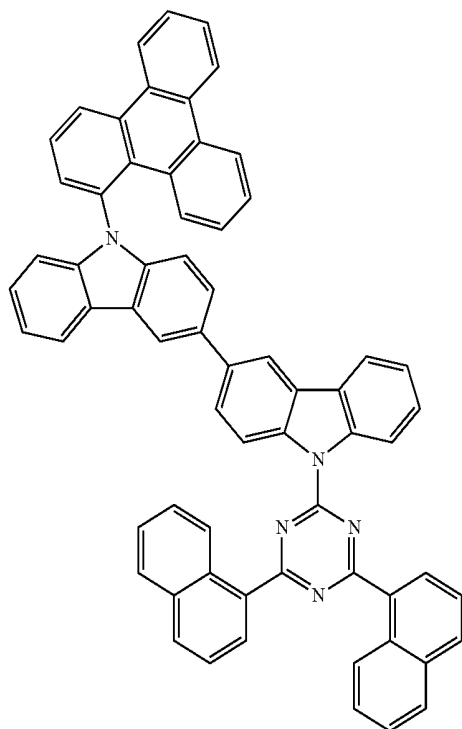
Compound 42



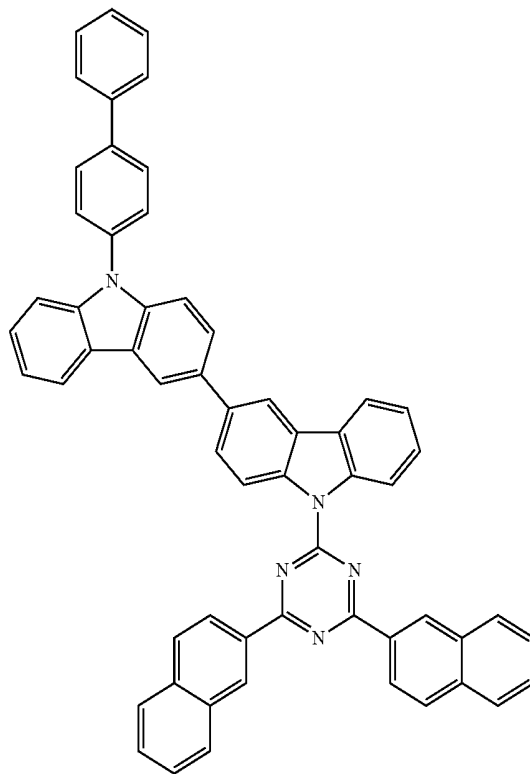
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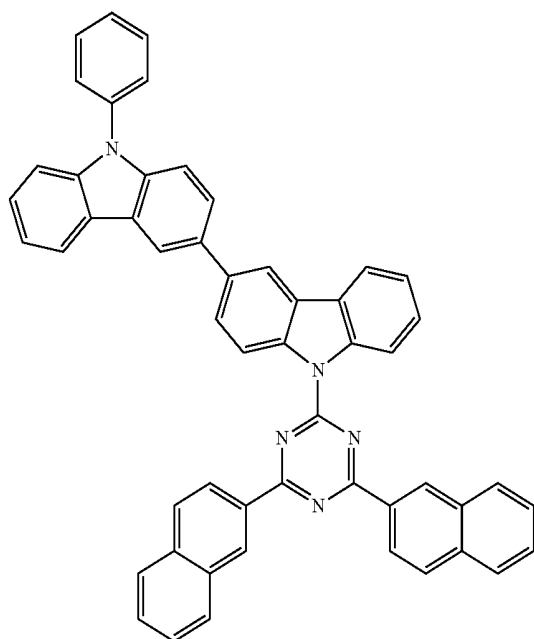
Compound 45



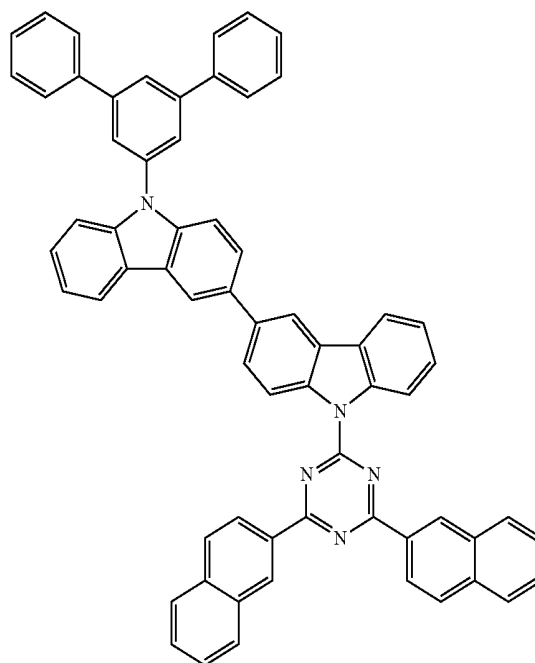
Compound 47



Compound 46



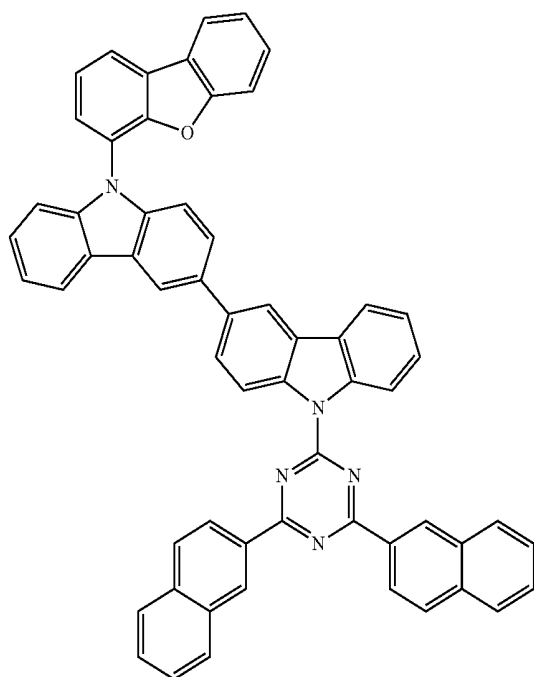
Compound 48



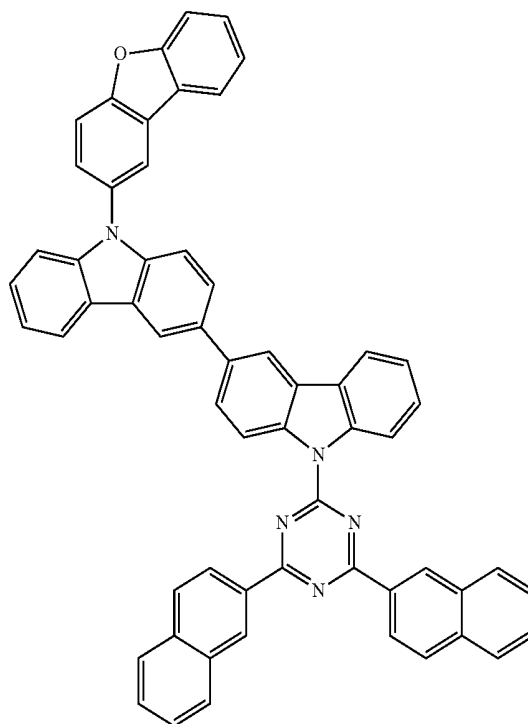
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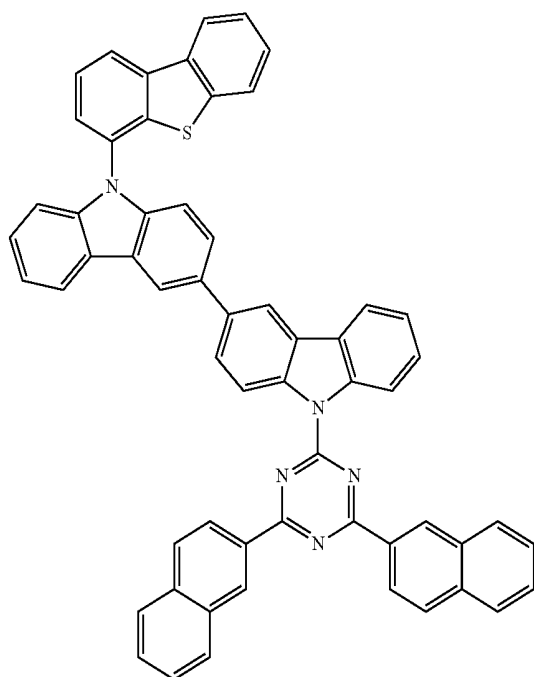
Compound 49



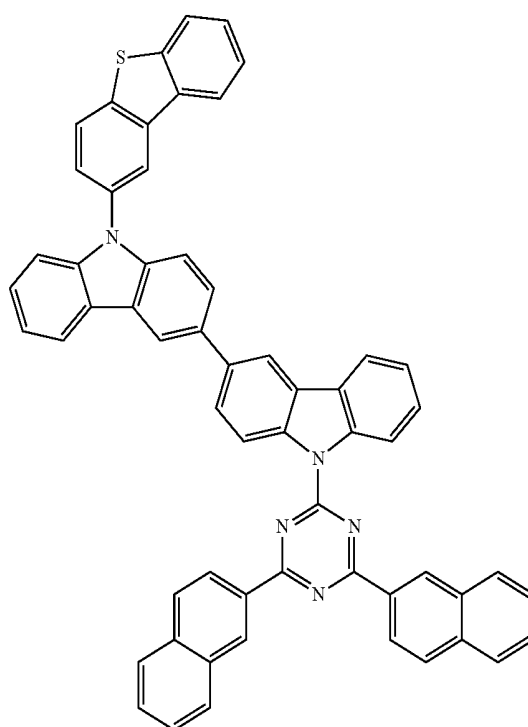
Compound 51



Compound 50



Compound 52

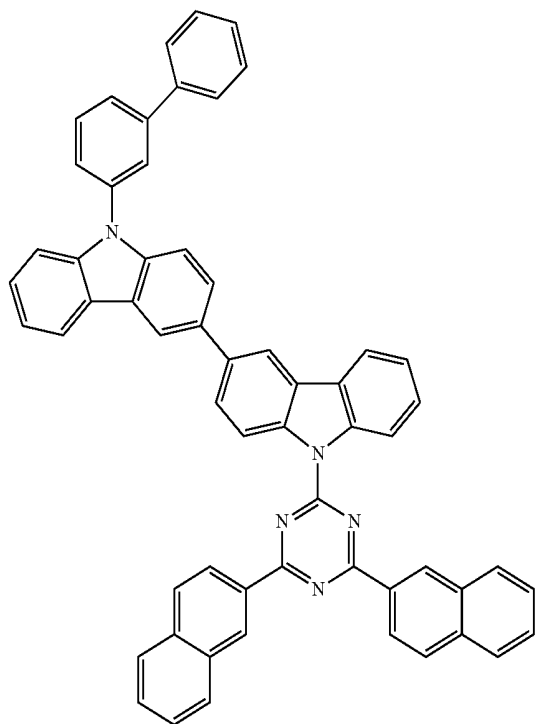




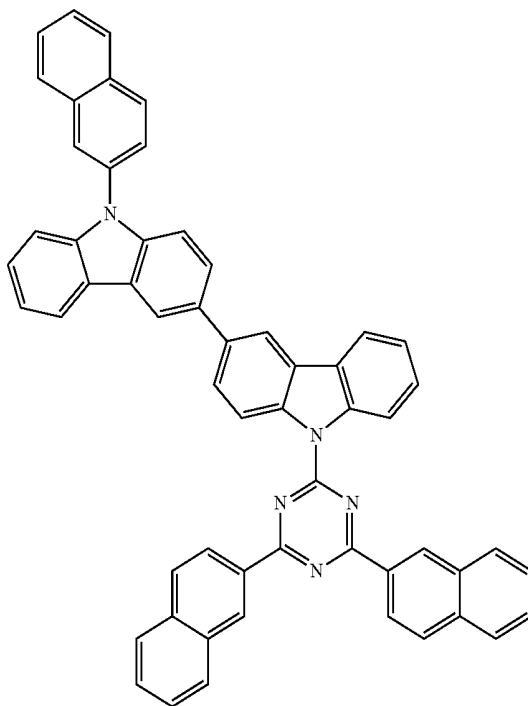
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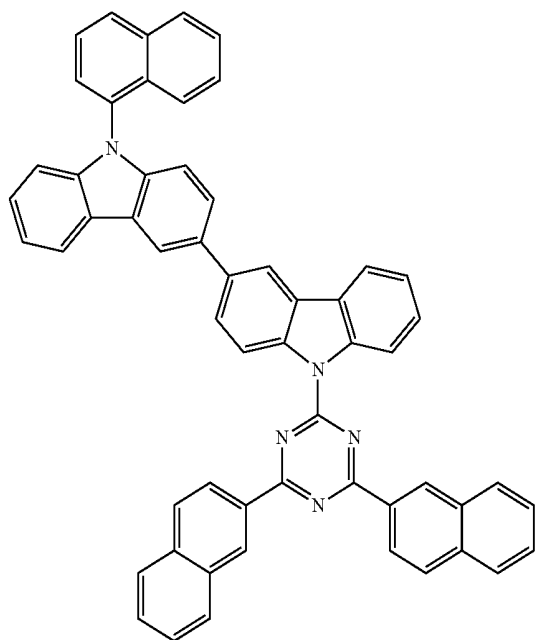
Compound 53



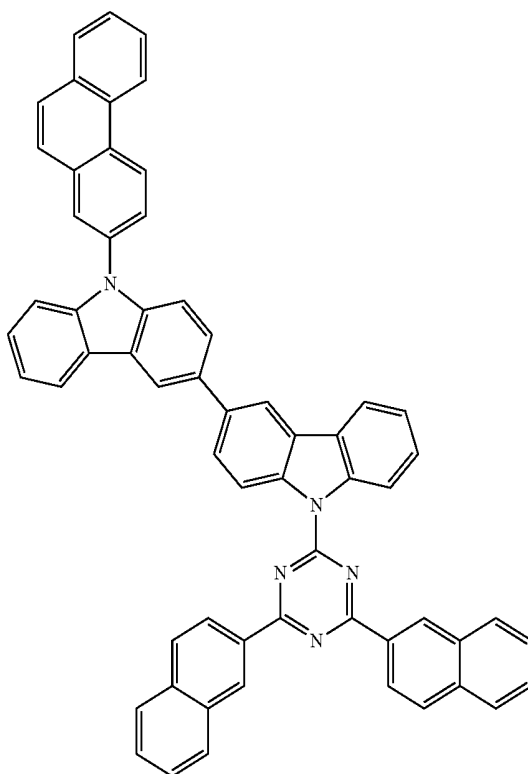
Compound 55



Compound 54



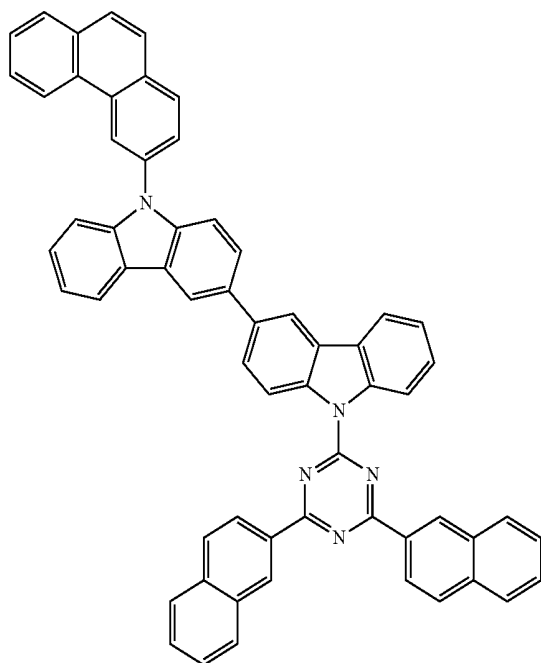
Compound 56



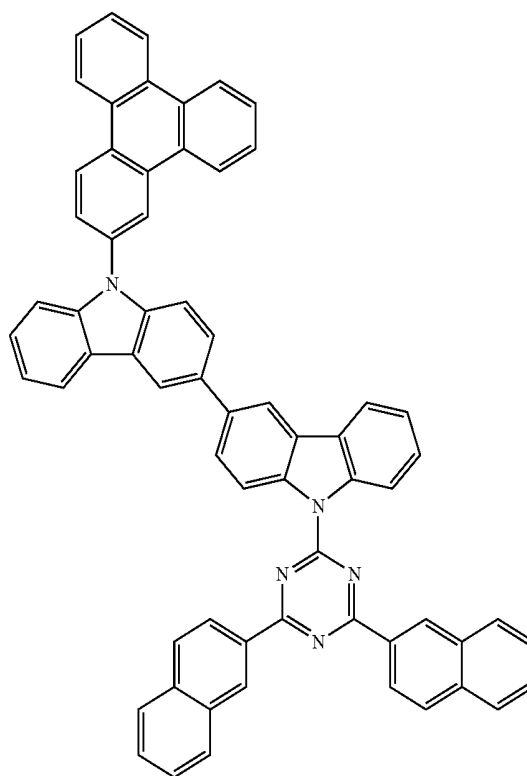
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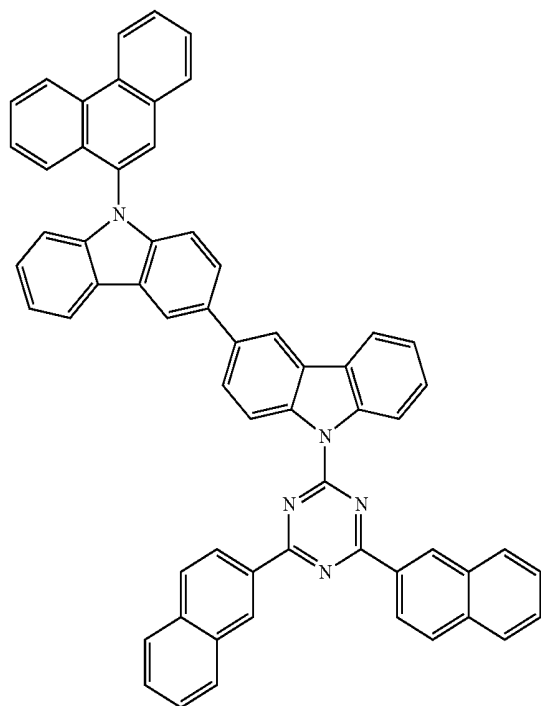
Compound 57



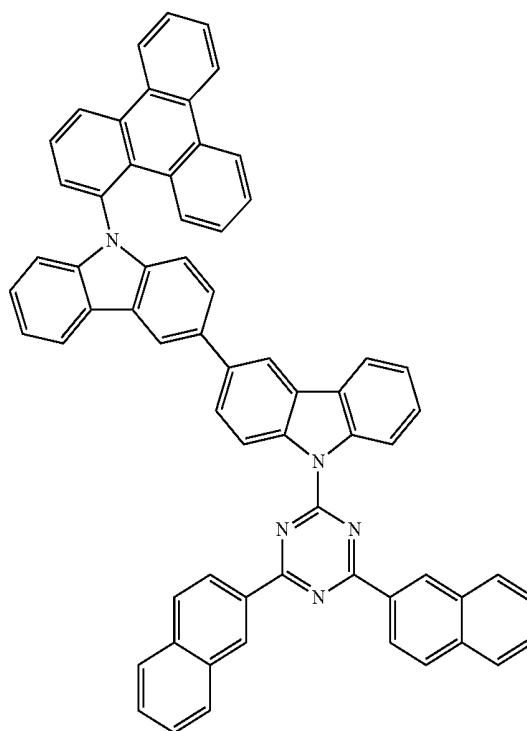
Compound 59



Compound 58



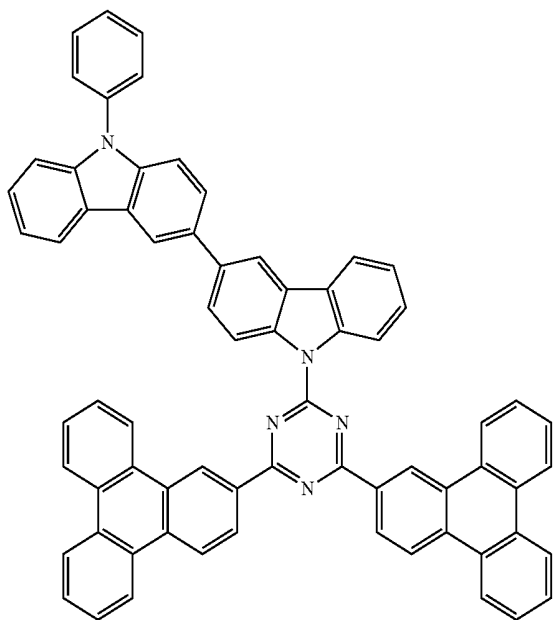
Compound 60



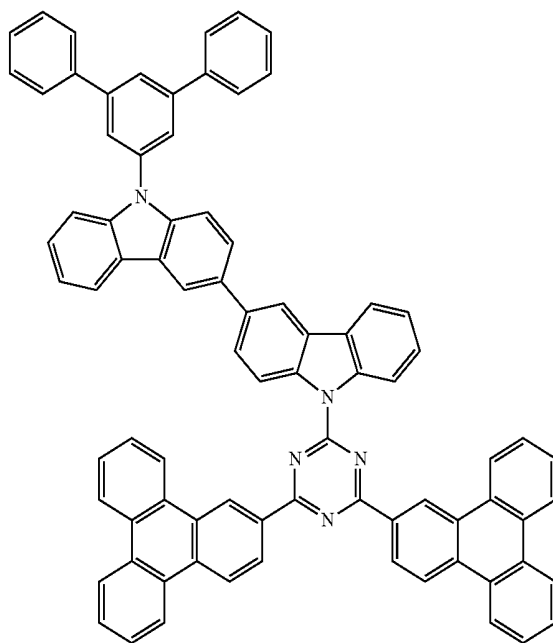
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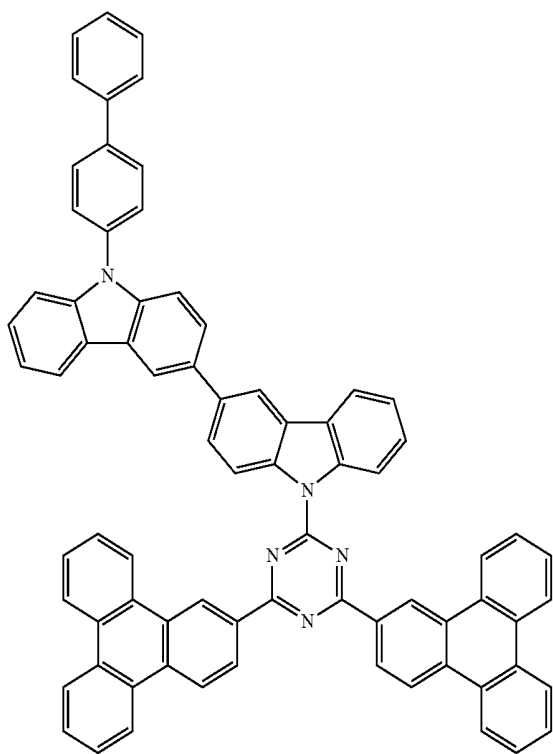
Compound 61



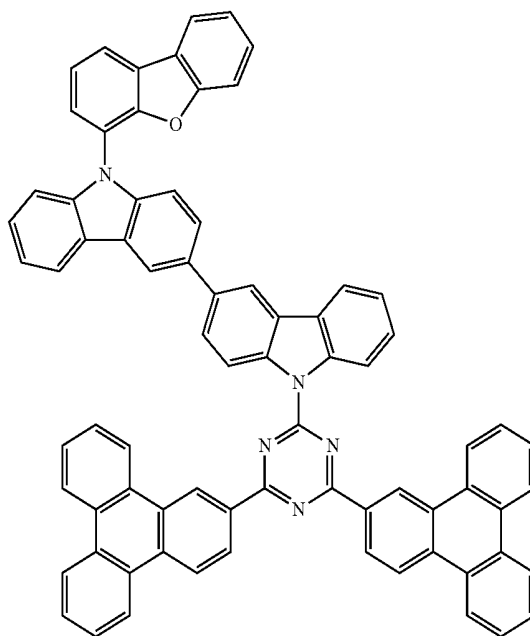
Compound 63



Compound 62



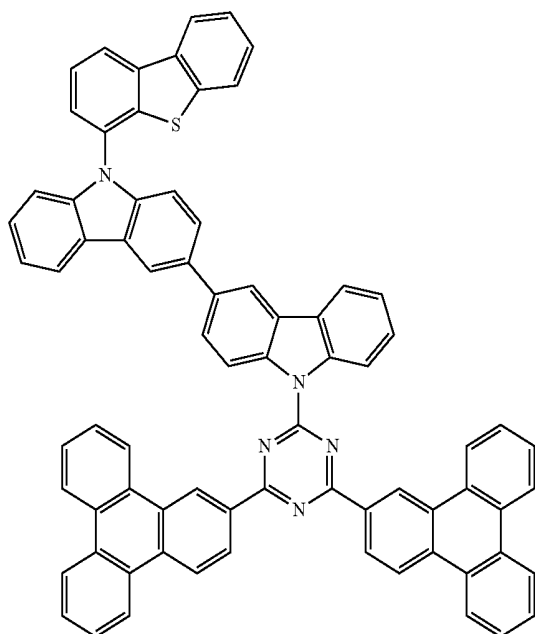
Compound 64



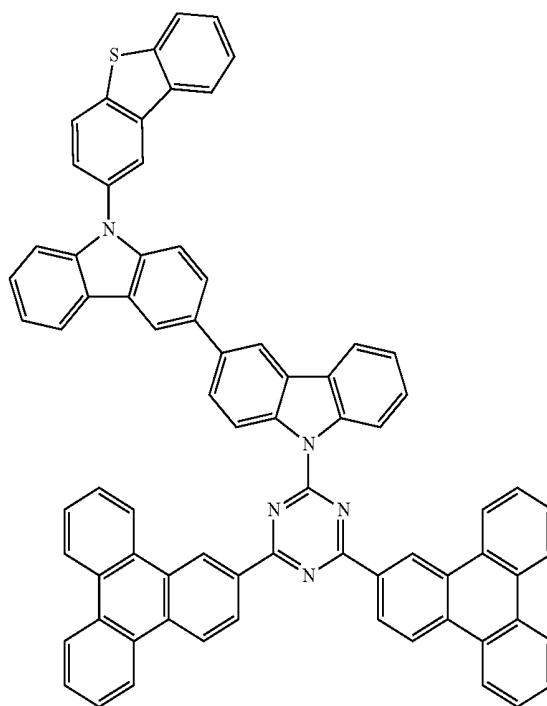
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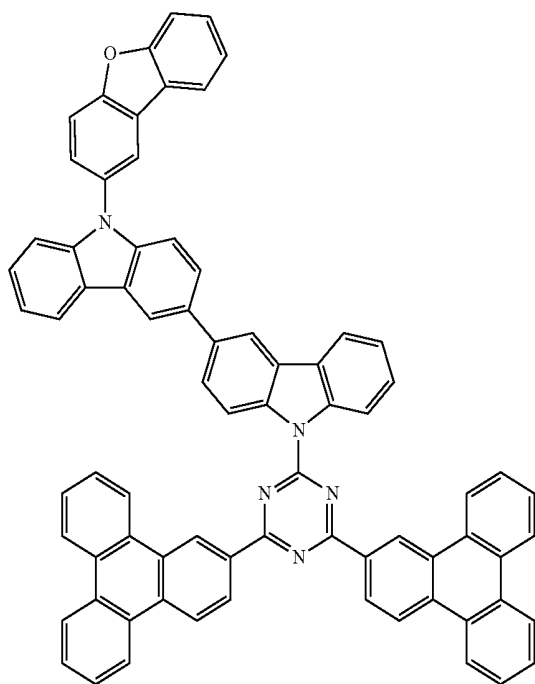
Compound 65



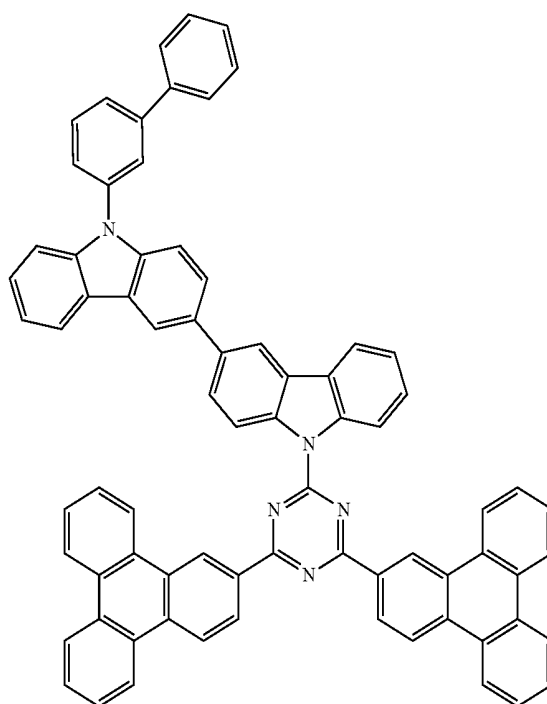
Compound 67



Compound 66



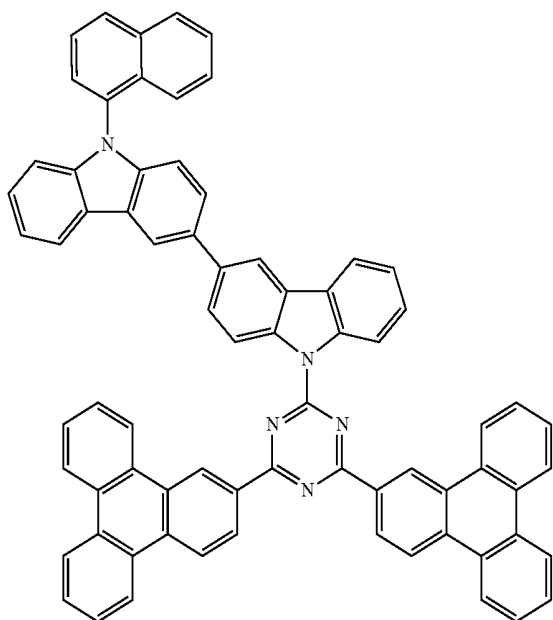
Compound 68



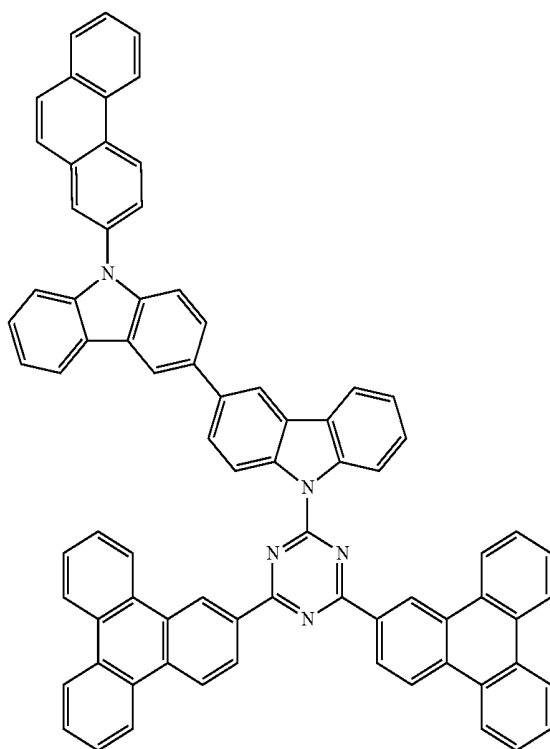
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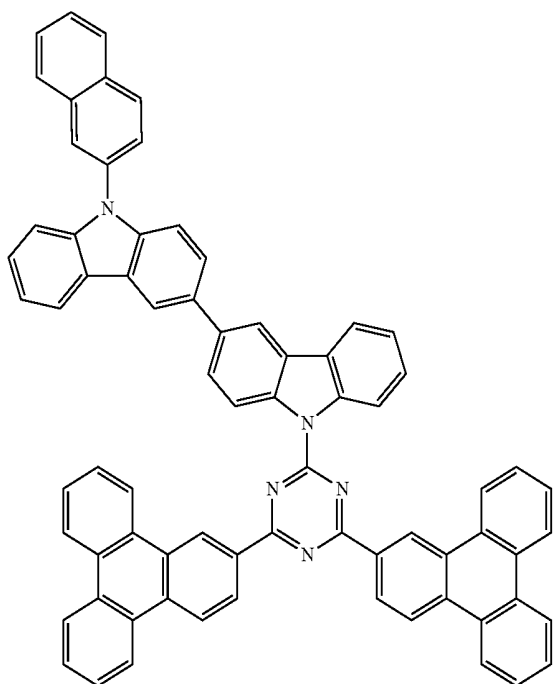
Compound 69



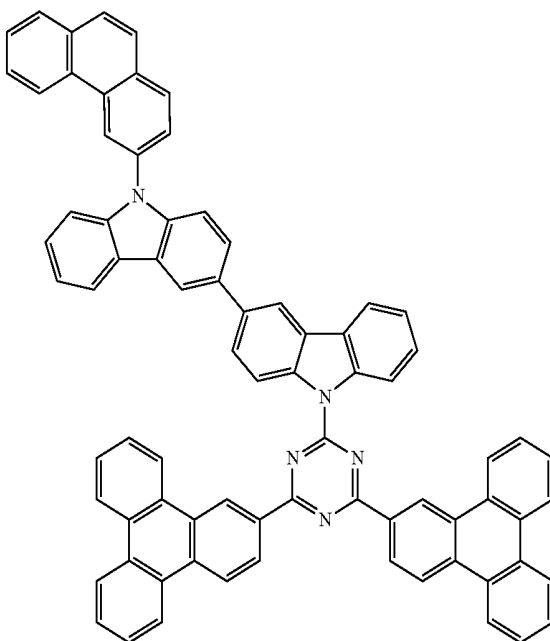
Compound 71



Compound 70



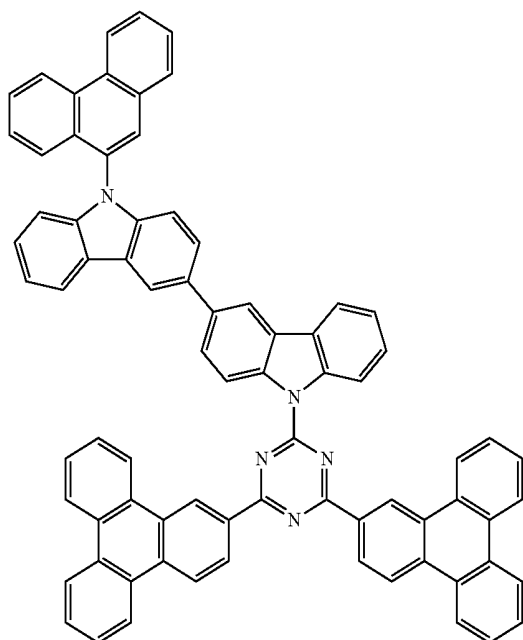
Compound 72



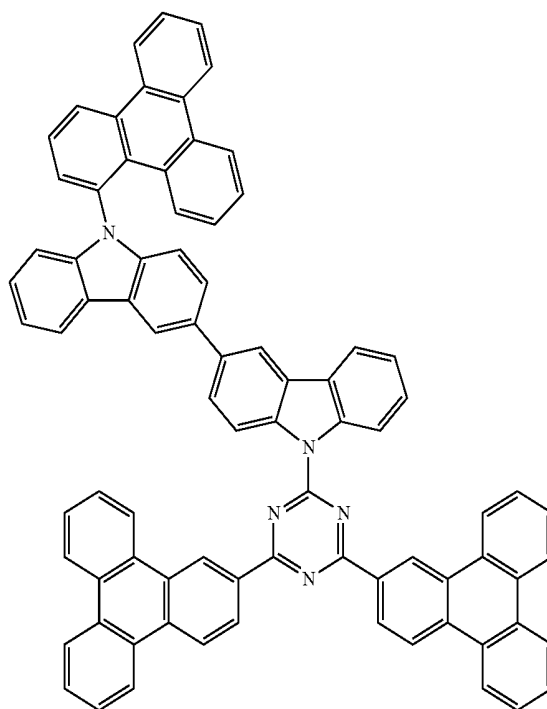
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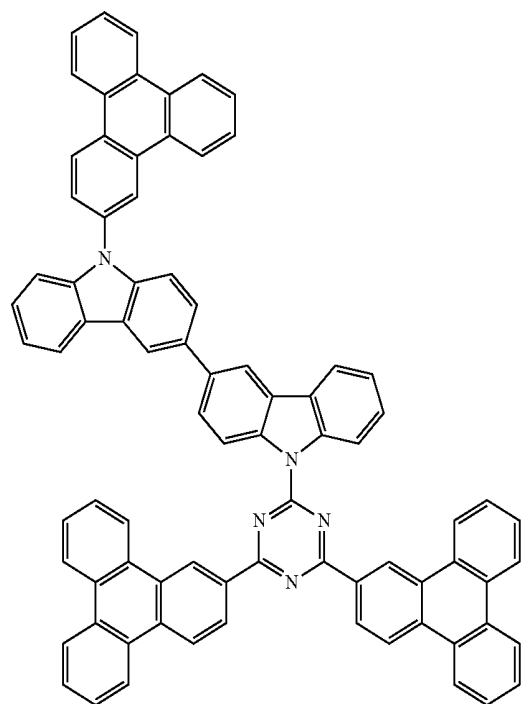
Compound 73



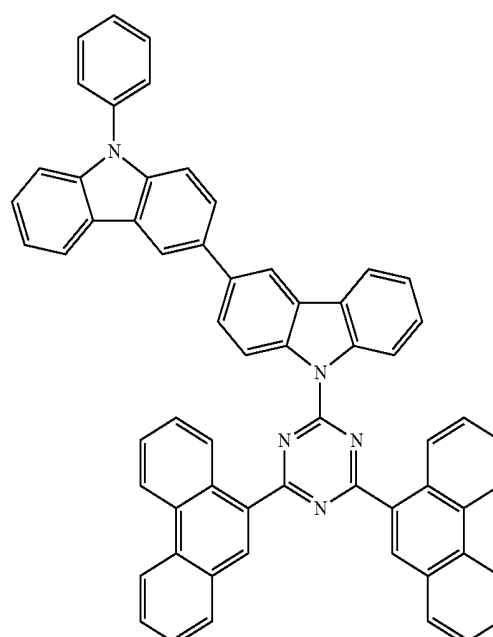
Compound 75



Compound 74



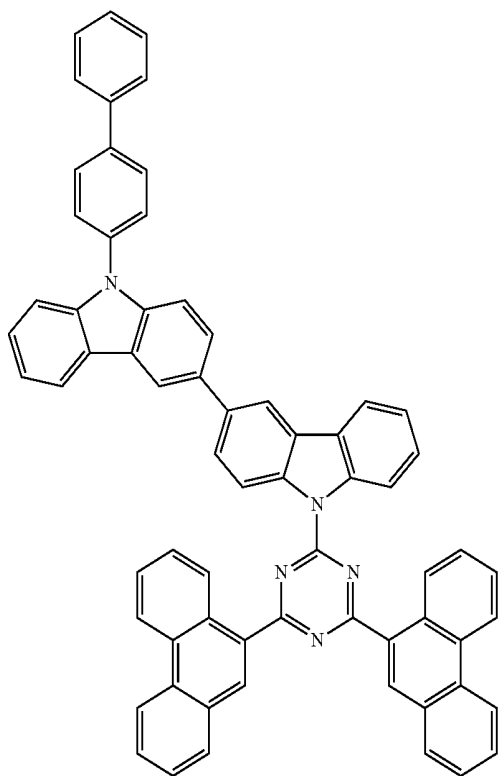
Compound 76



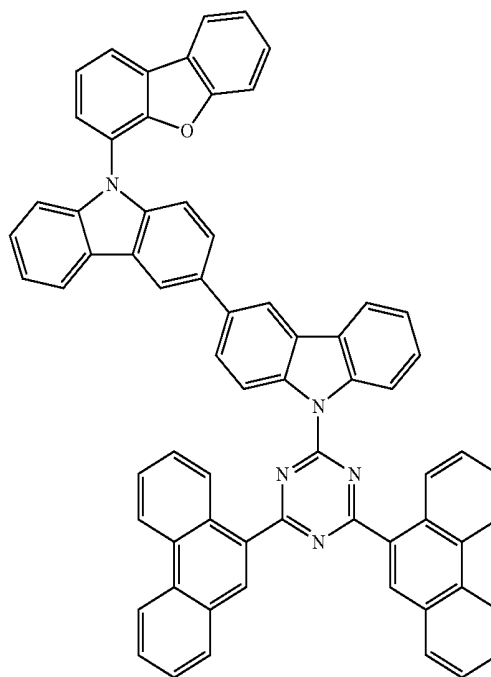
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Compound 77

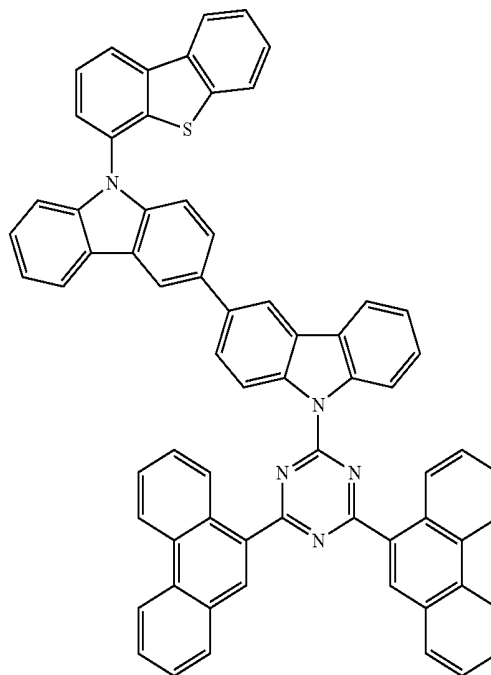
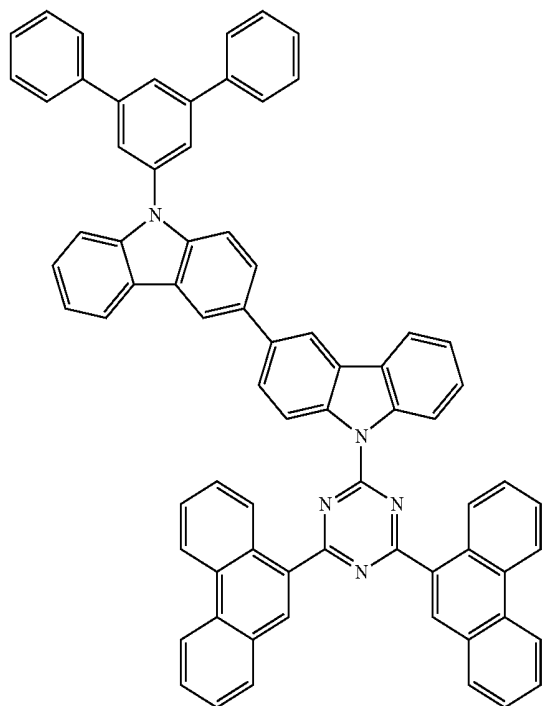


Compound 79



Compound 80

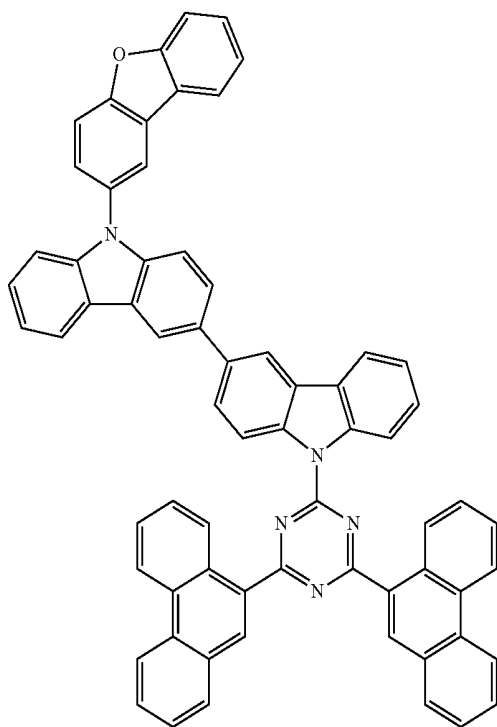
Compound 78



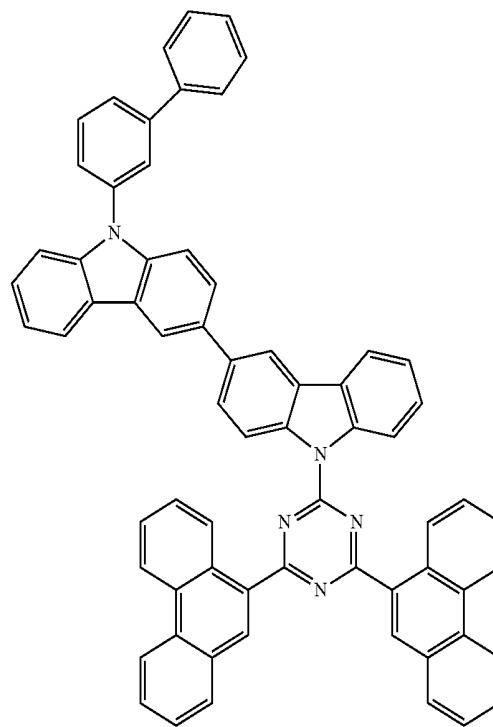
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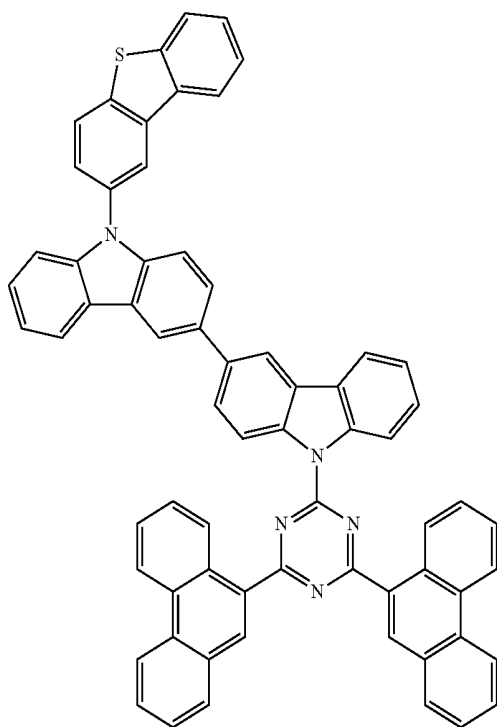
Compound 81



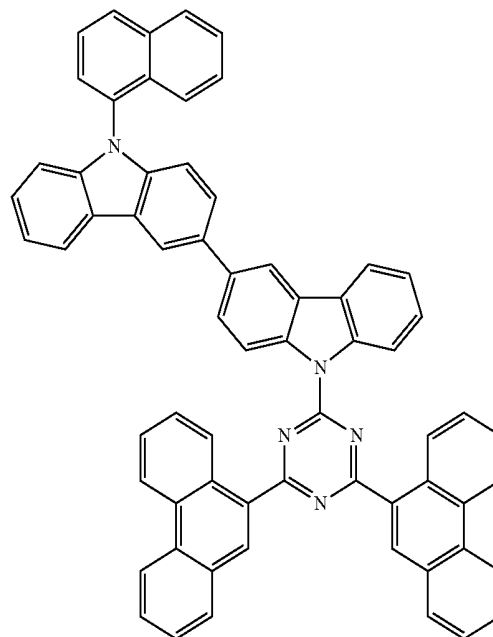
Compound 83



Compound 82



Compound 84

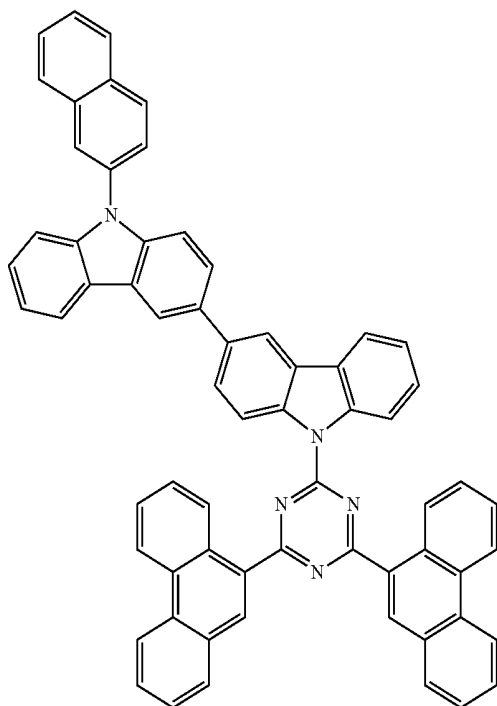




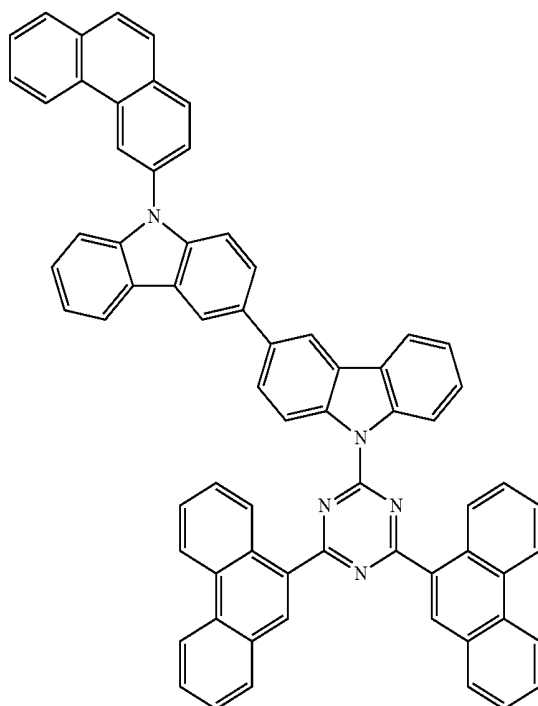
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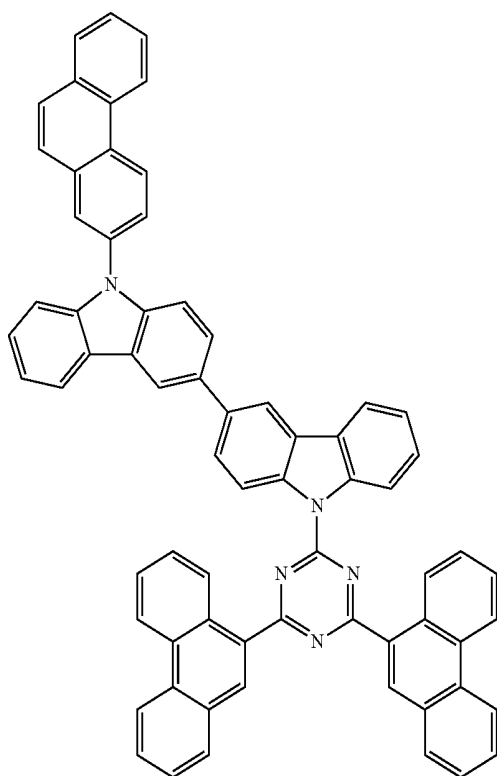
Compound 85



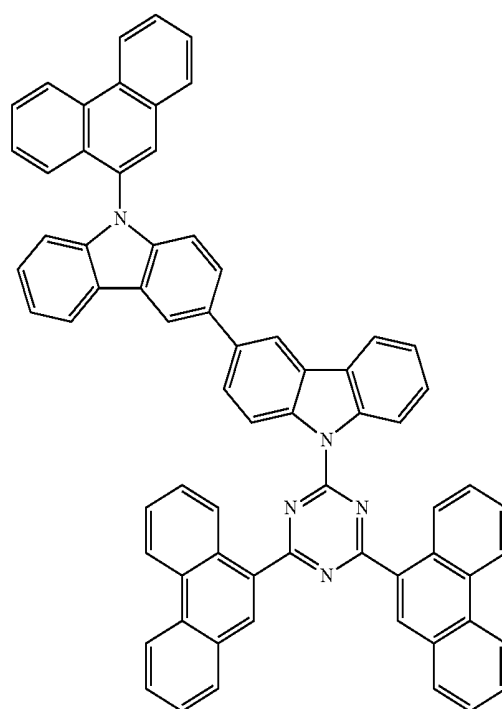
Compound 87



Compound 86

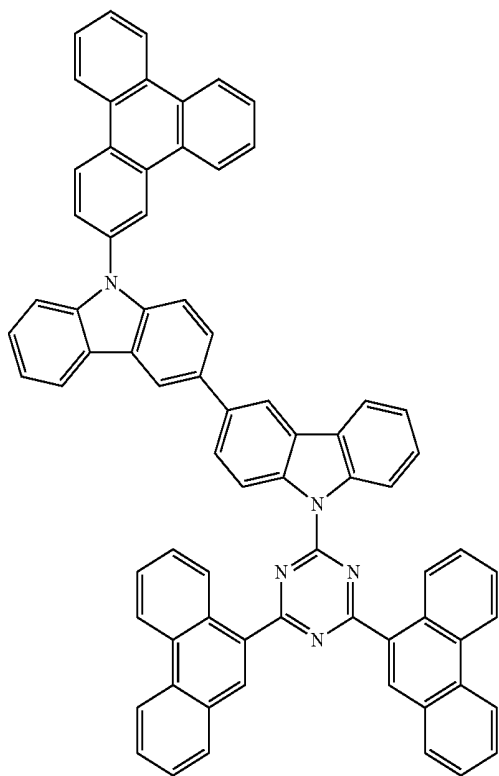


Compound 88

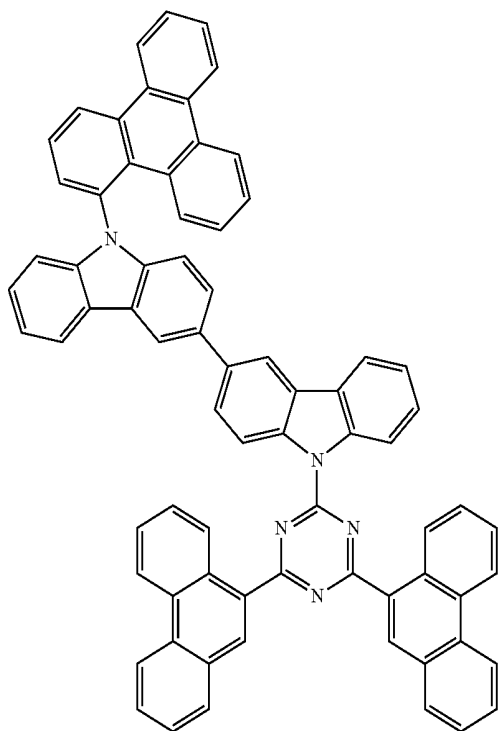


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Compound 89

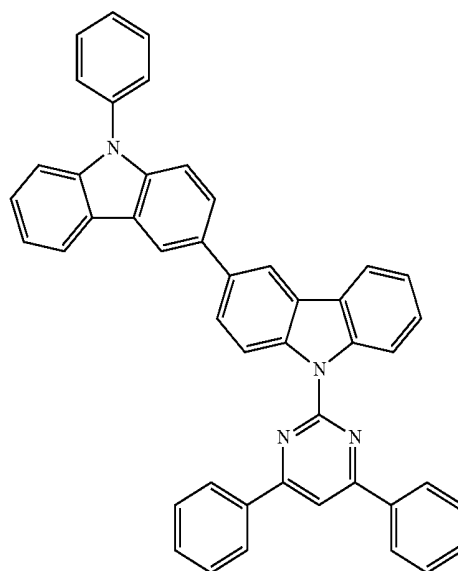


Compound 90

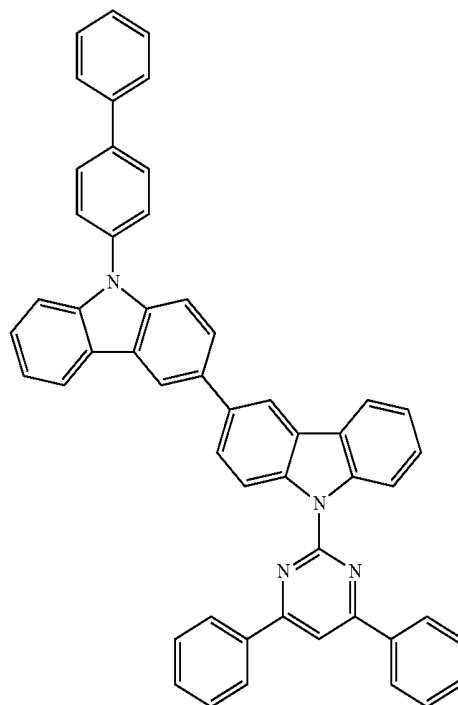


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Compound 91

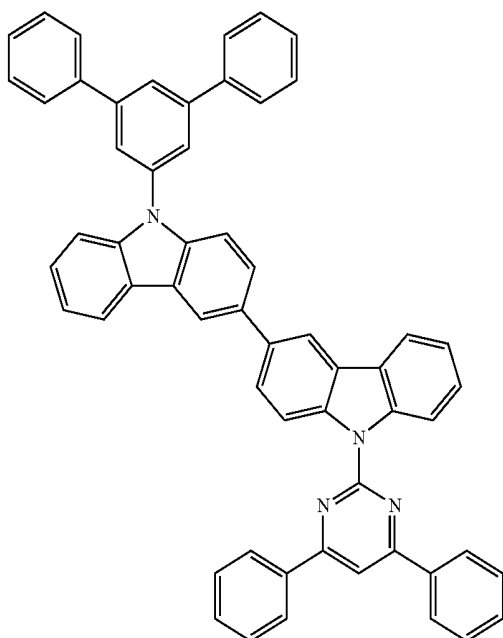


Compound 92



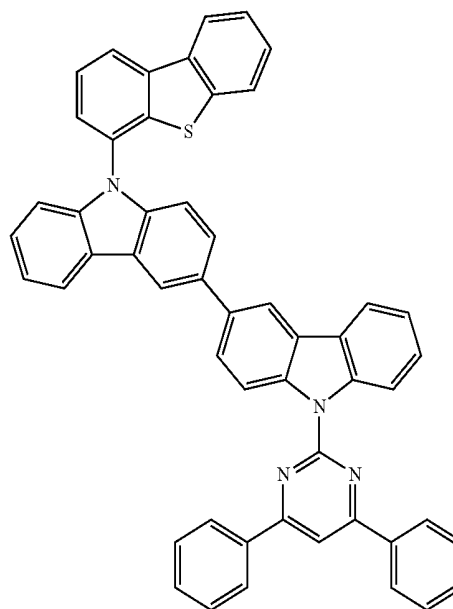
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Compound 93

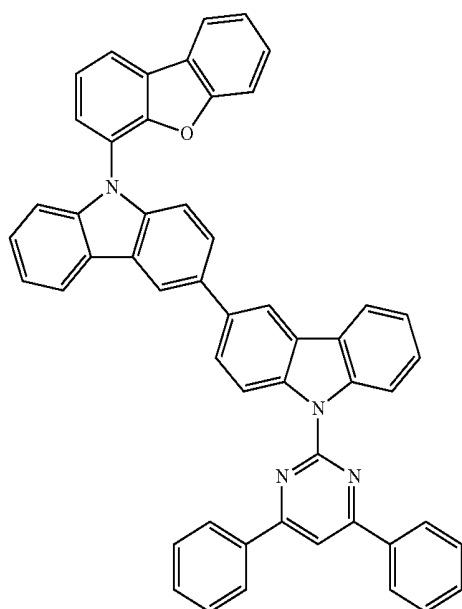


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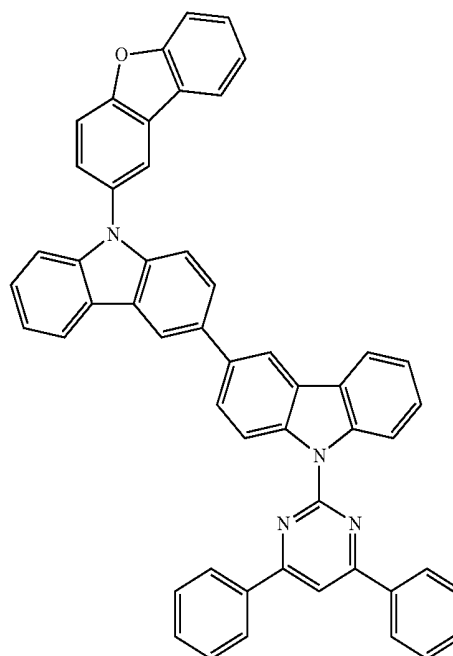
Compound 95



Compound 94



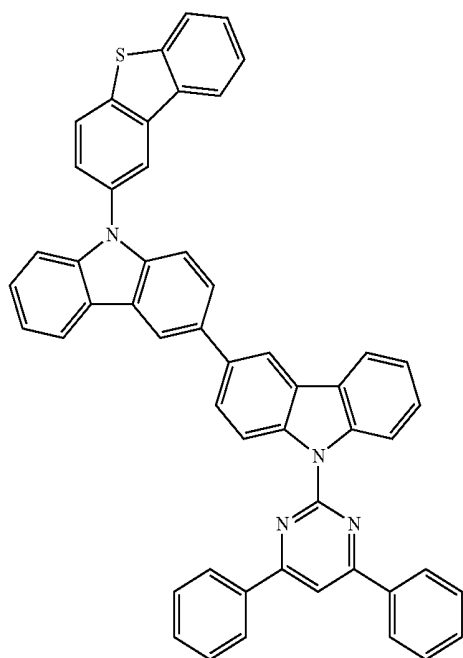
Compound 96



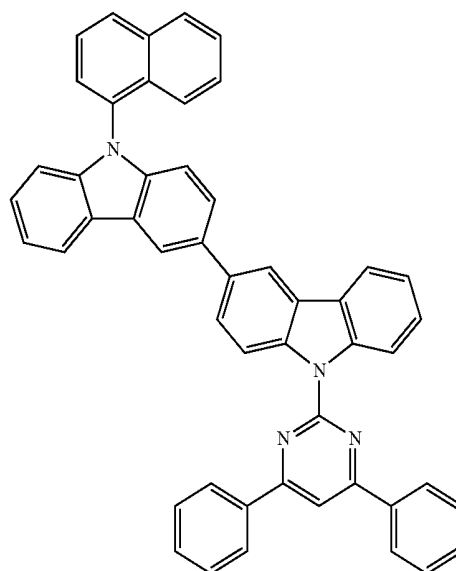
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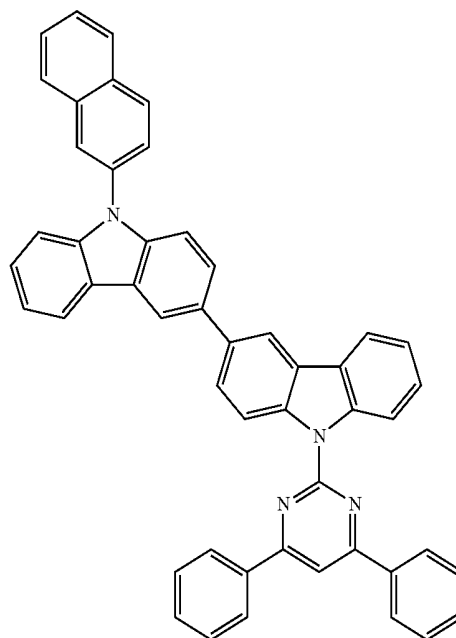
Compound 97



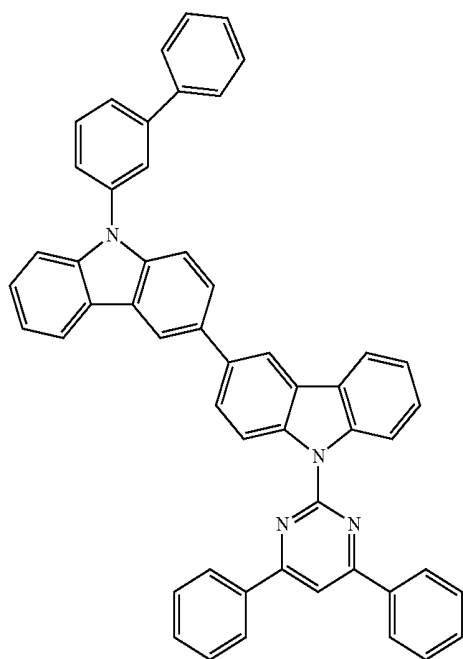
Compound 99



Compound 100



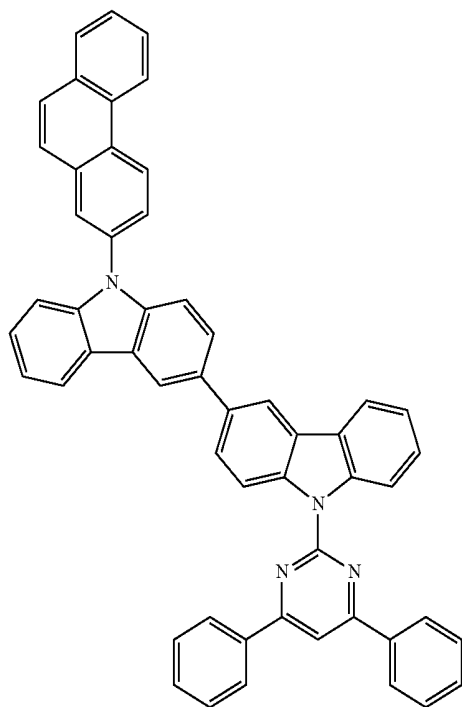
Compound 98



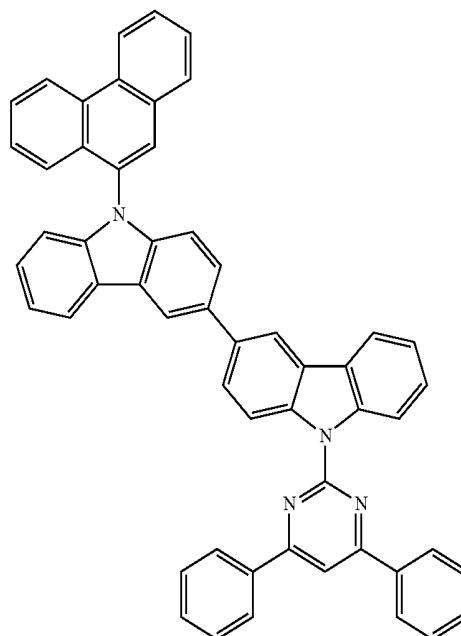
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Compound 101

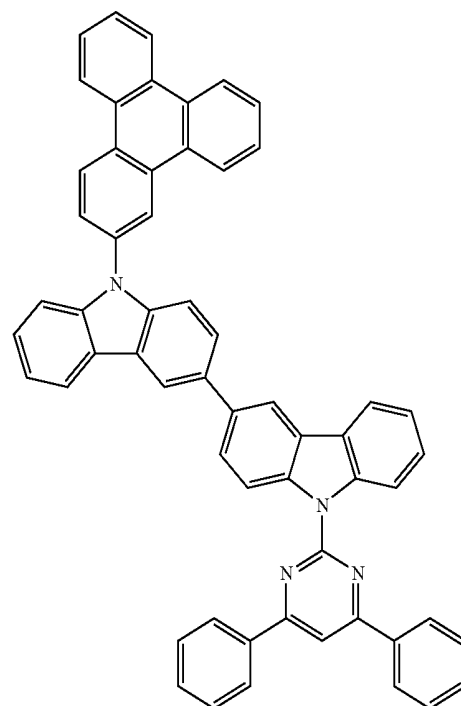
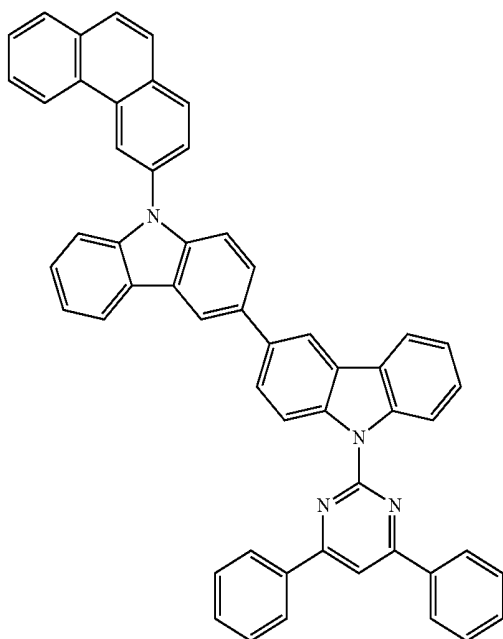


Compound 103



Compound 104

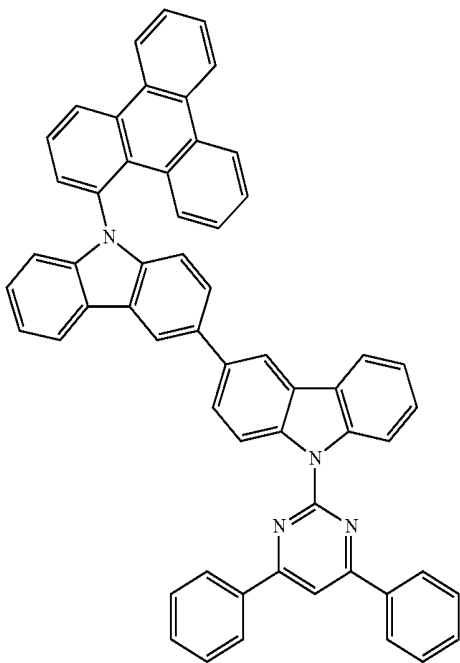
Compound 102



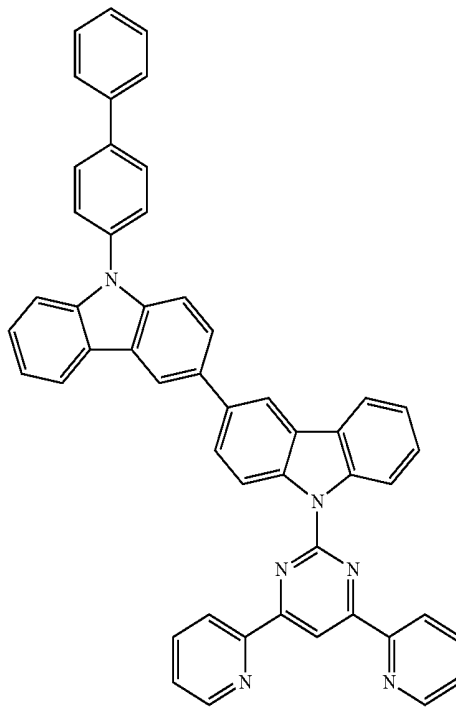
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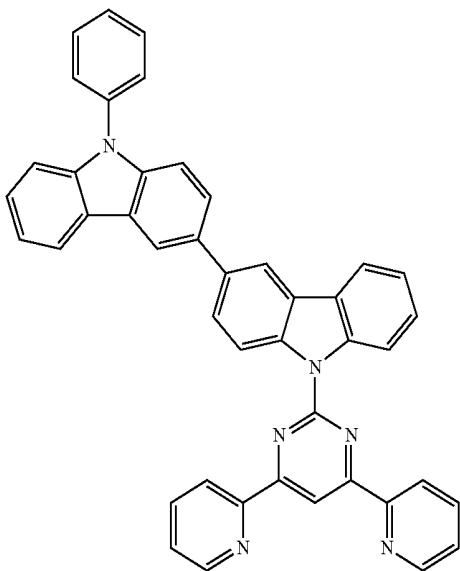
Compound 105



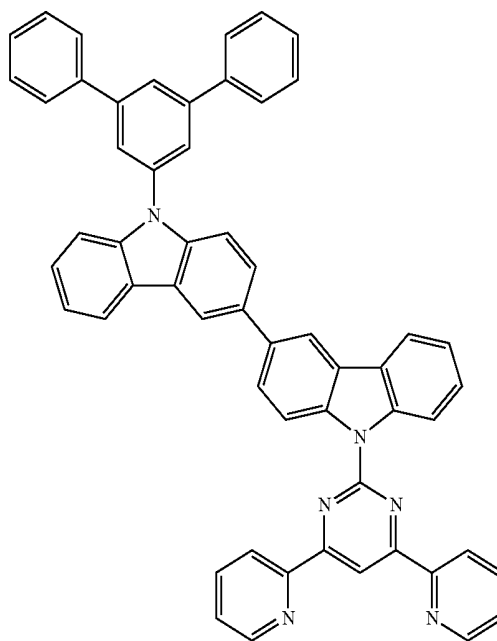
Compound 107



Compound 106



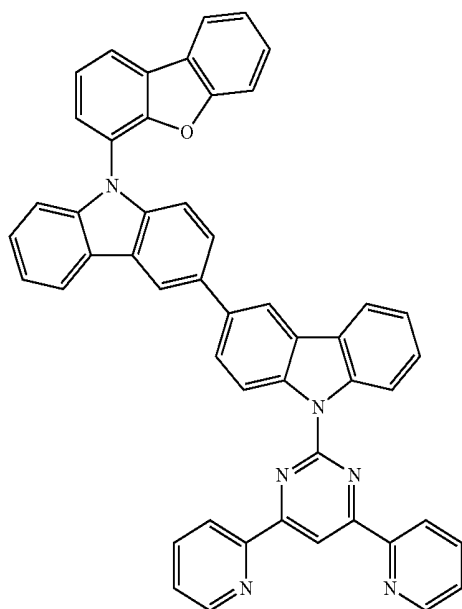
Compound 108



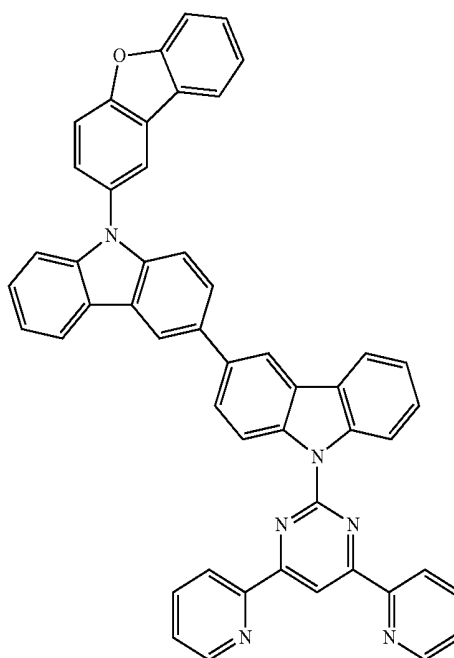
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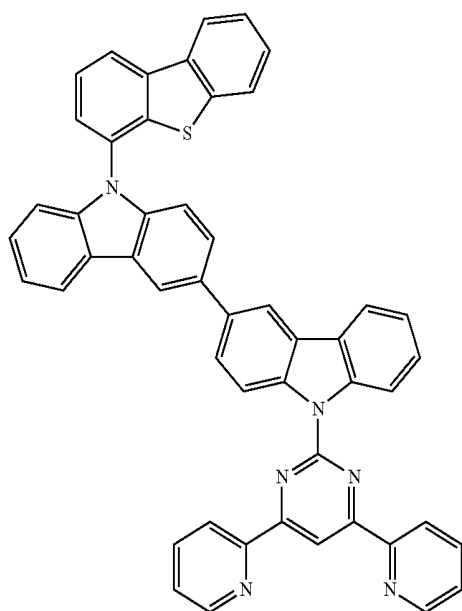
Compound 109



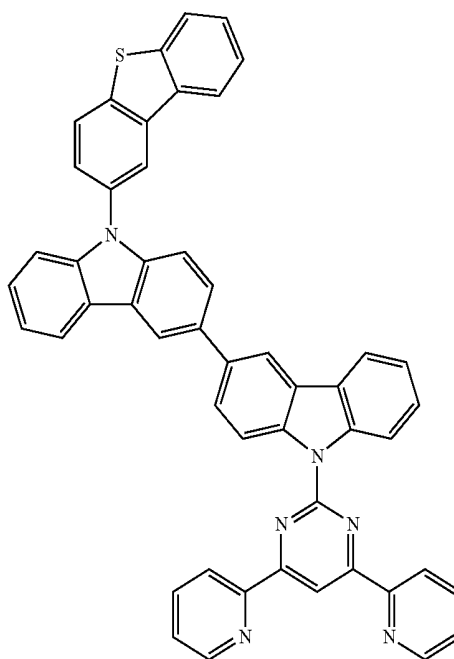
Compound 111



Compound 110



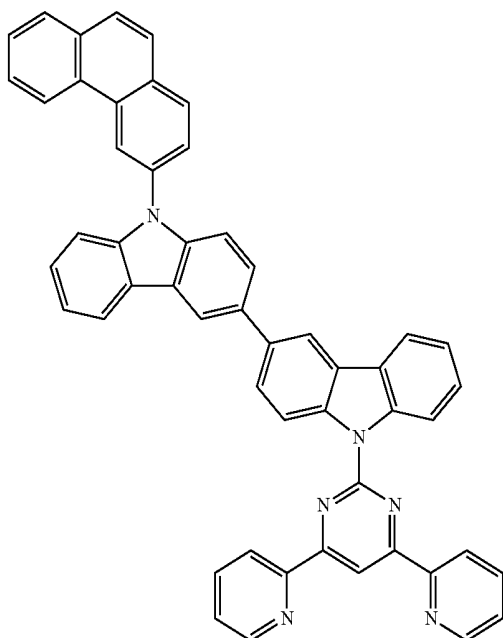
Compound 112



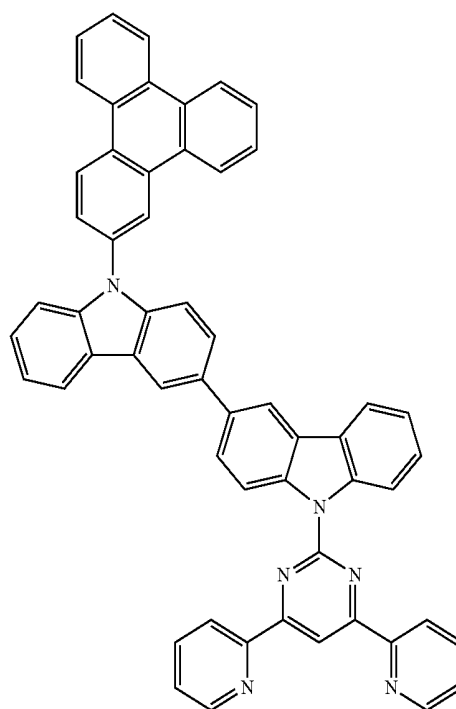
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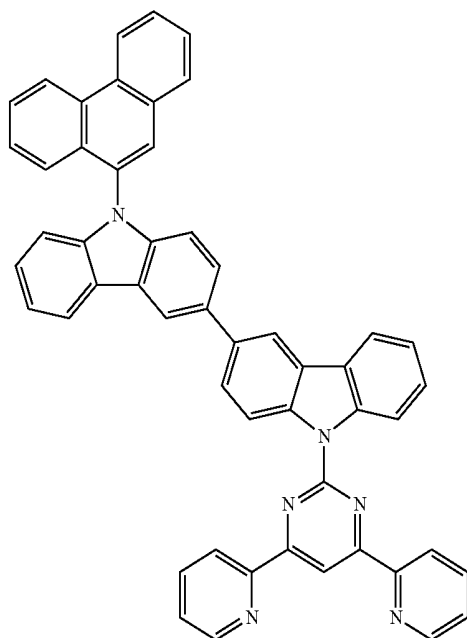
Compound 117



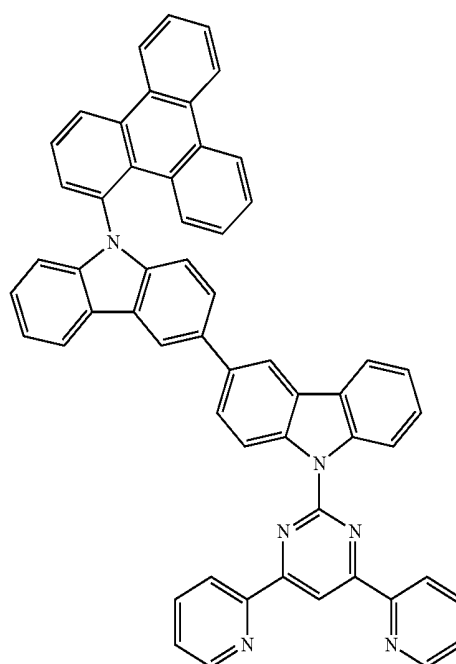
Compound 119



Compound 118



Compound 120

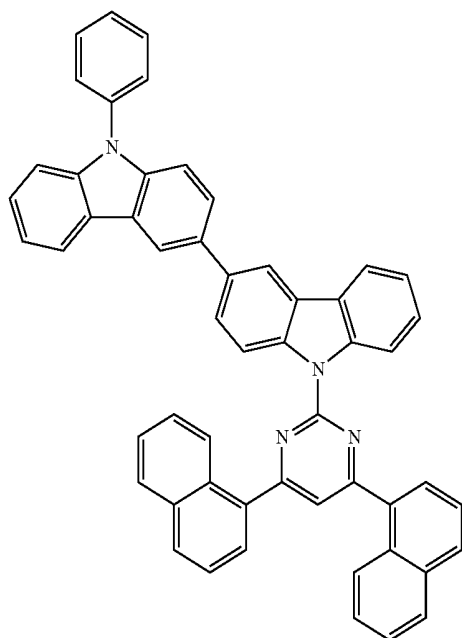




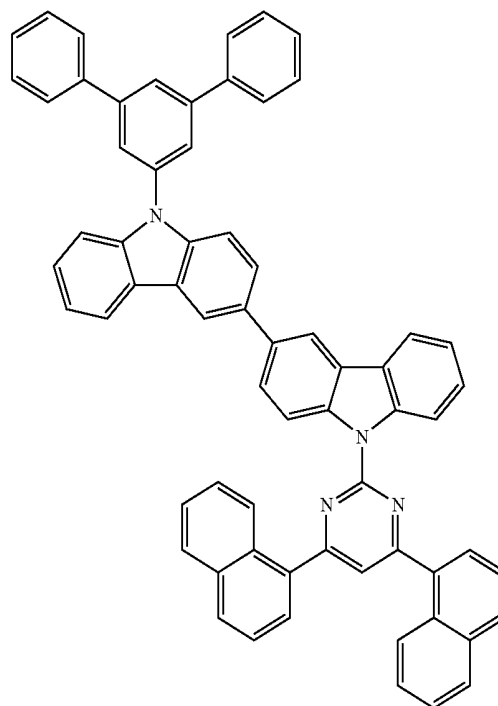
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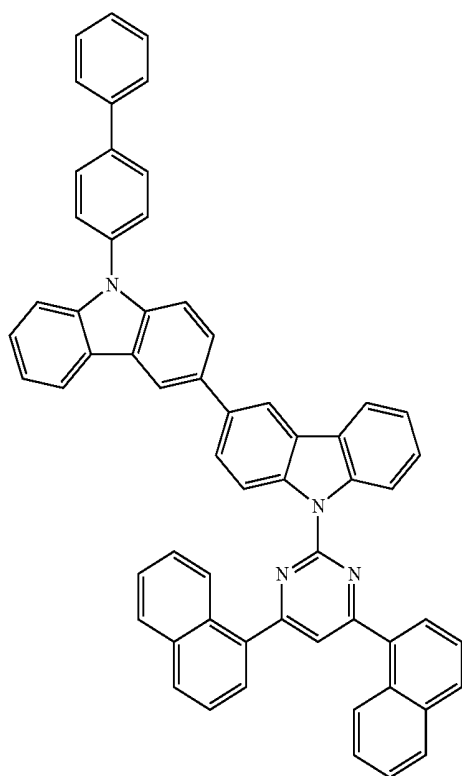
Compound 121



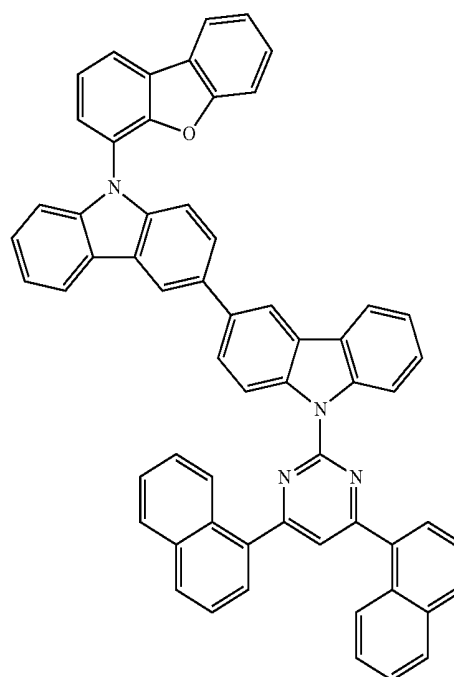
Compound 123



Compound 122



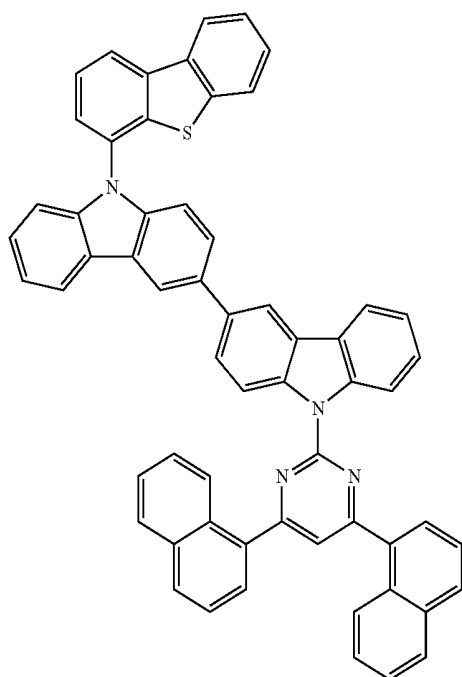
Compound 124



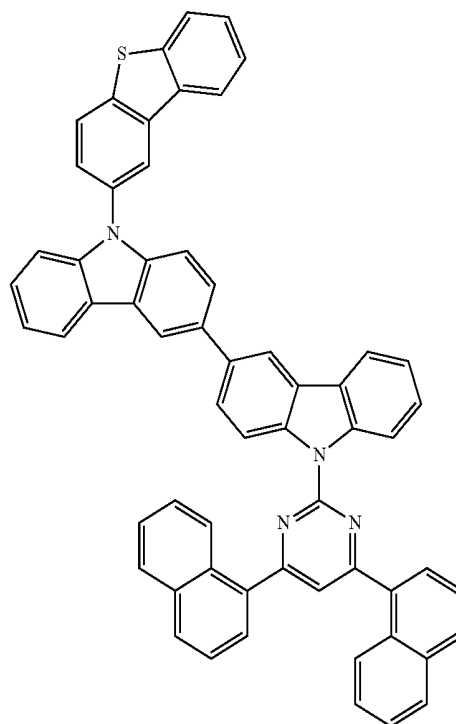
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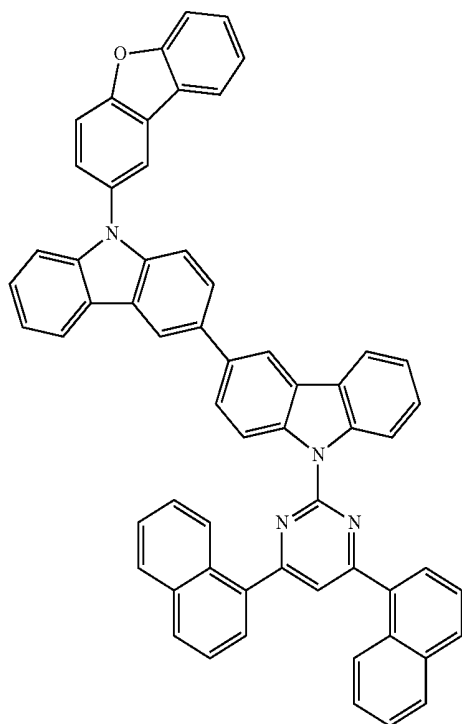
Compound 125



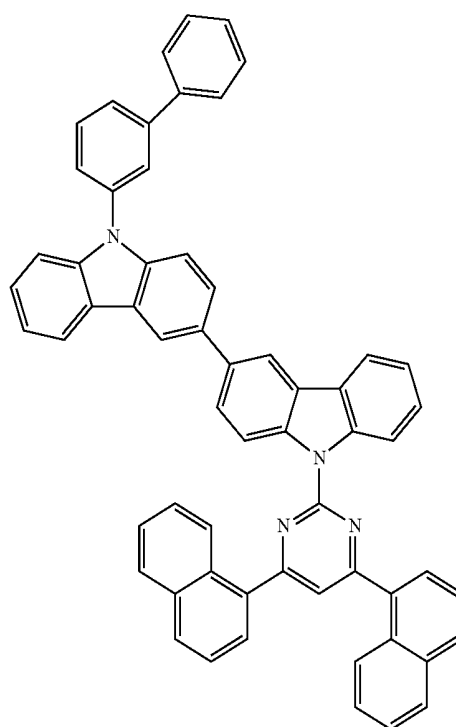
Compound 127



Compound 126



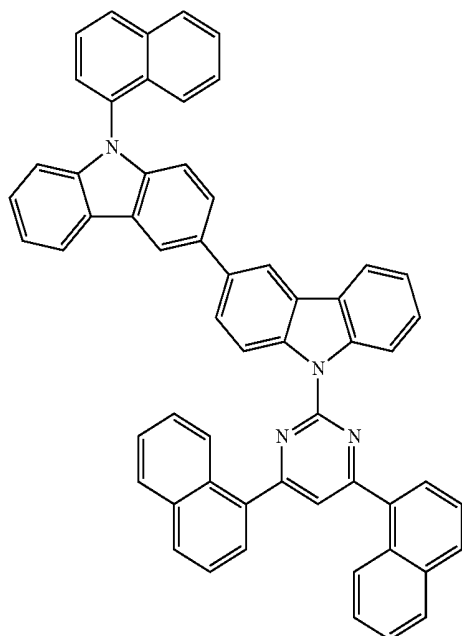
Compound 128



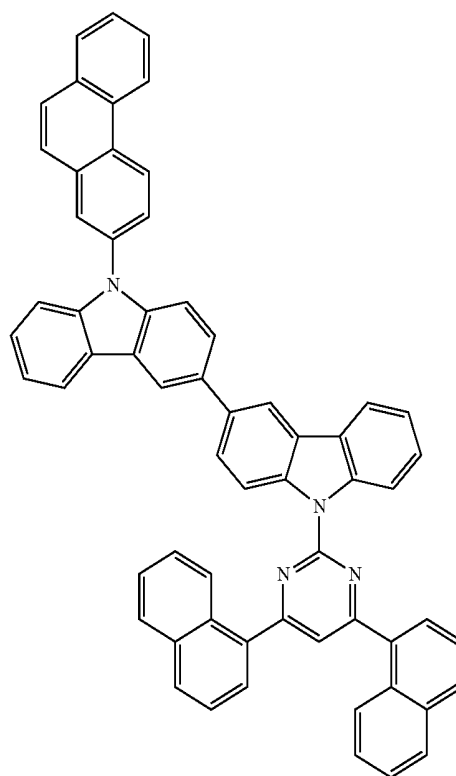
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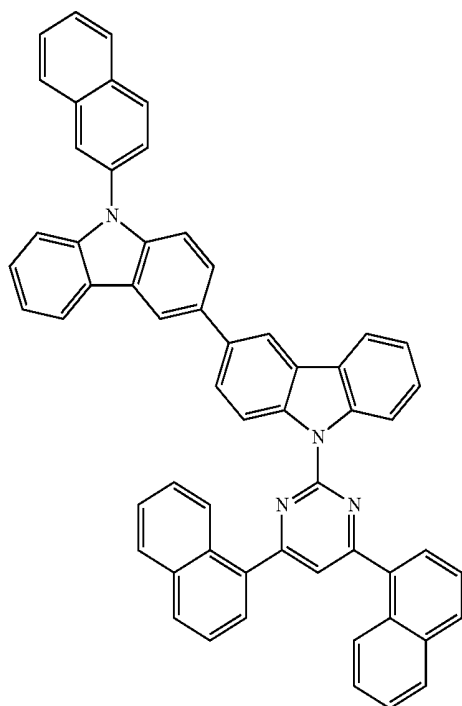
Compound 129



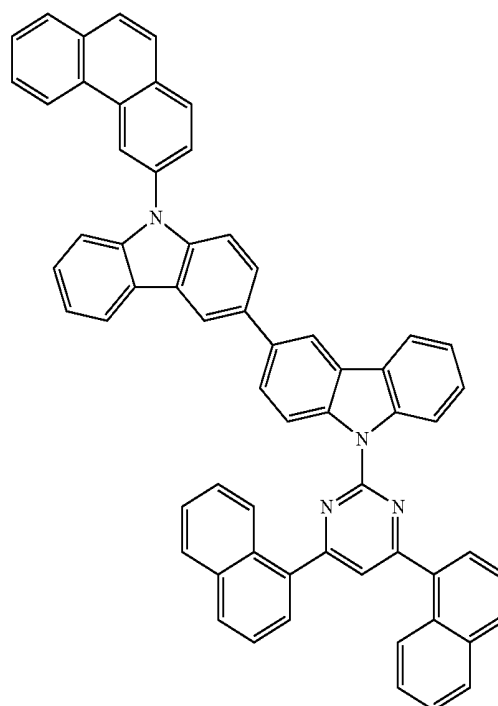
Compound 131



Compound 130



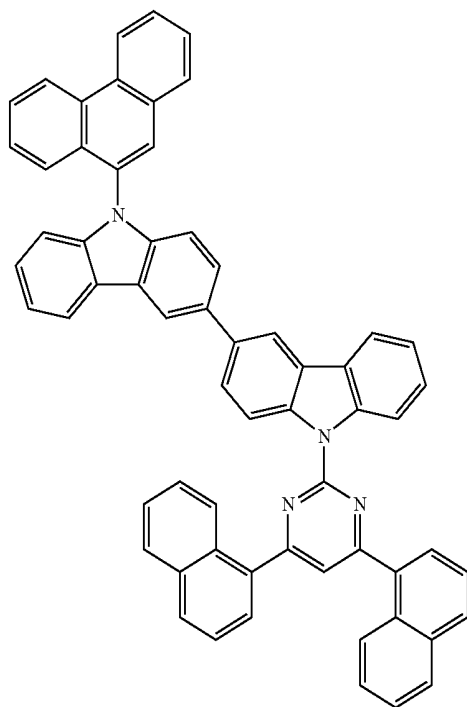
Compound 132



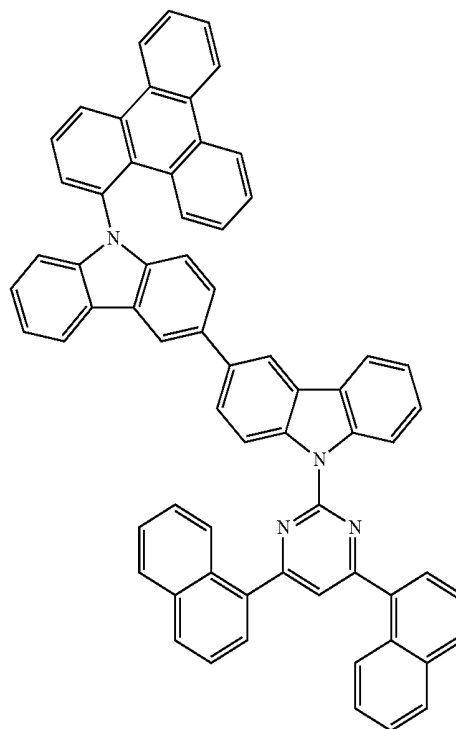
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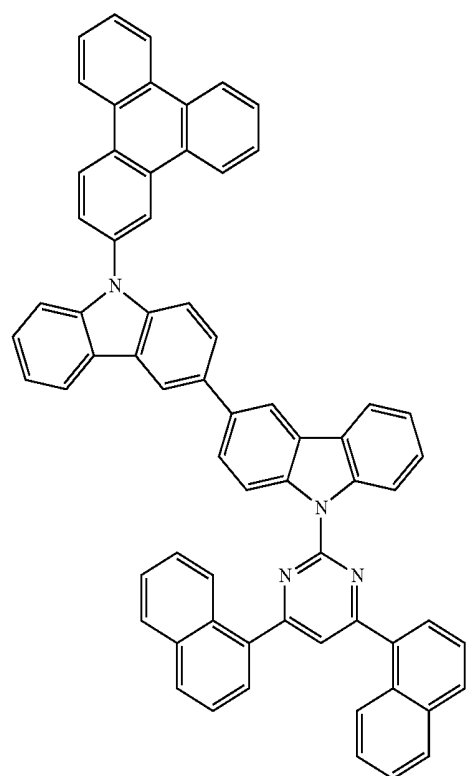
Compound 133



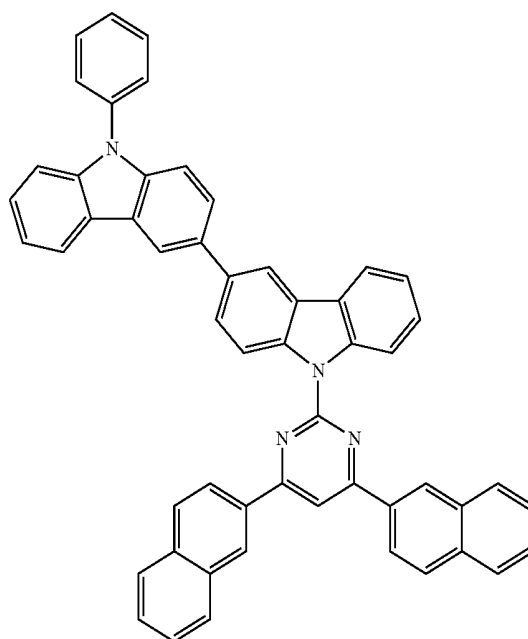
Compound 135



Compound 134



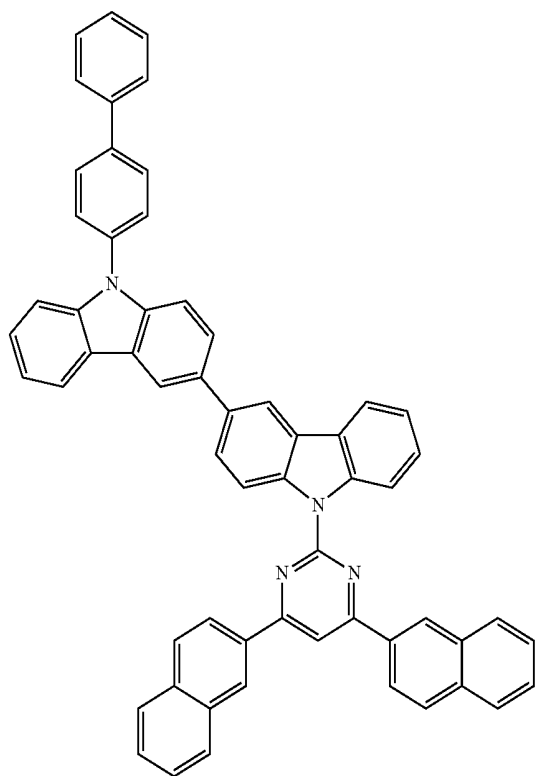
Compound 136



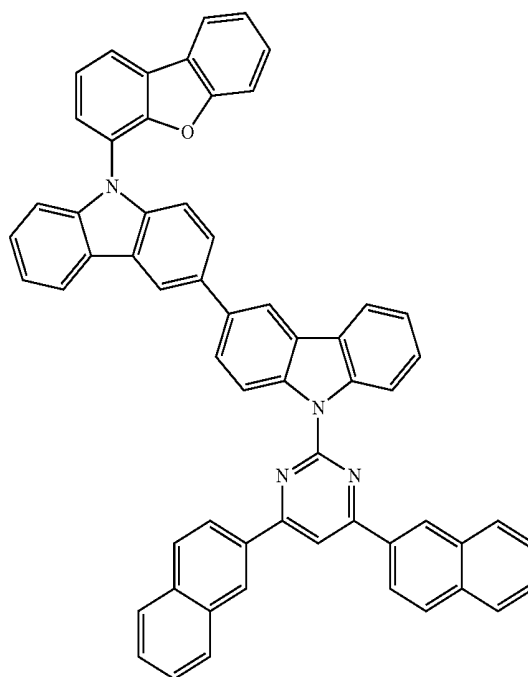
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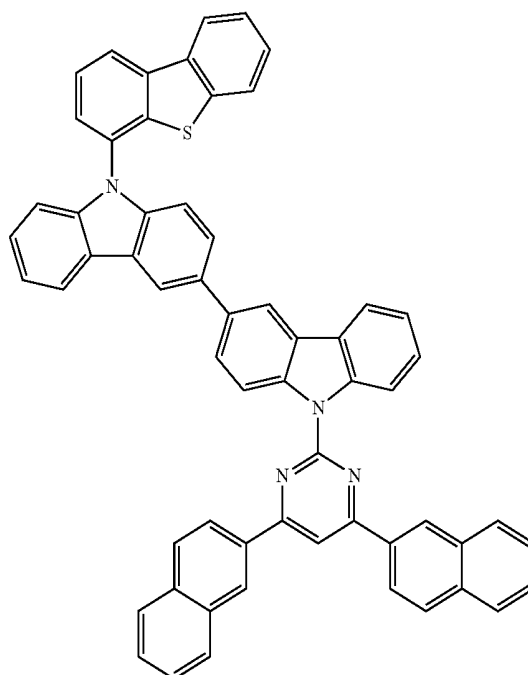
Compound 137



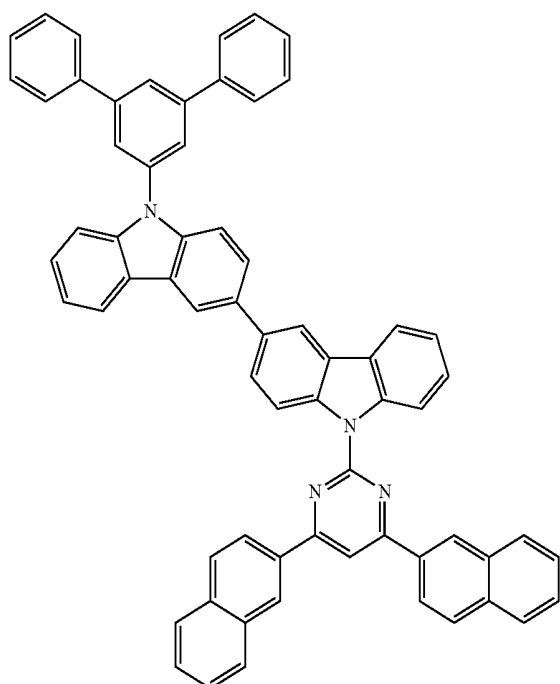
Compound 139



Compound 140



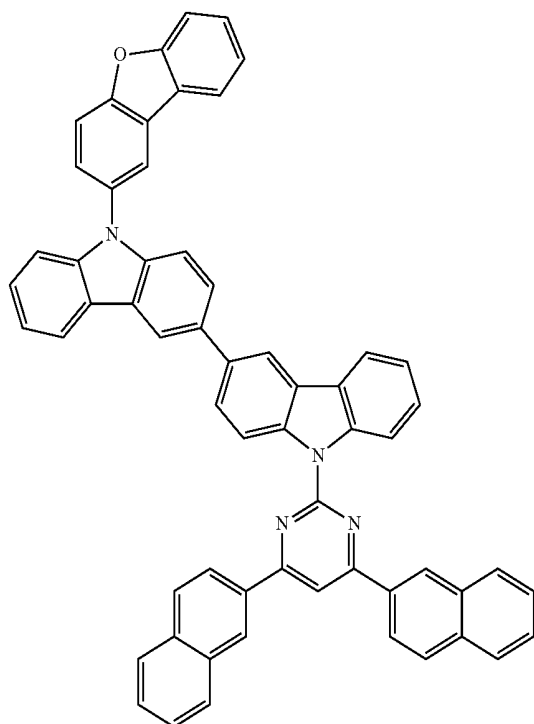
Compound 138



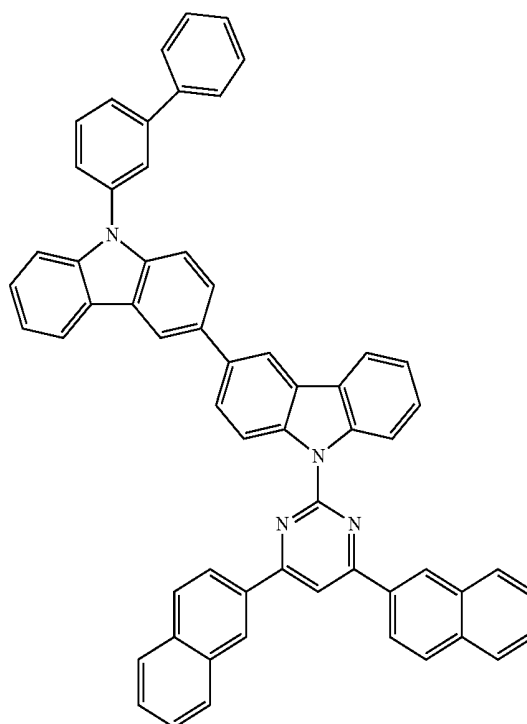
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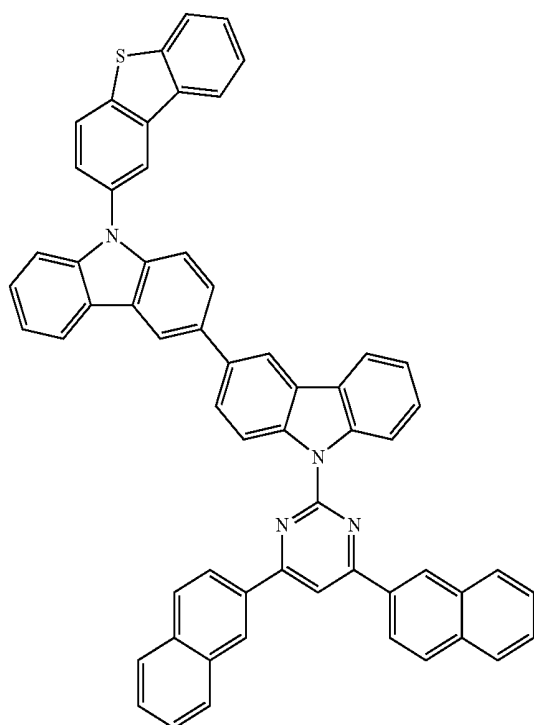
Compound 141



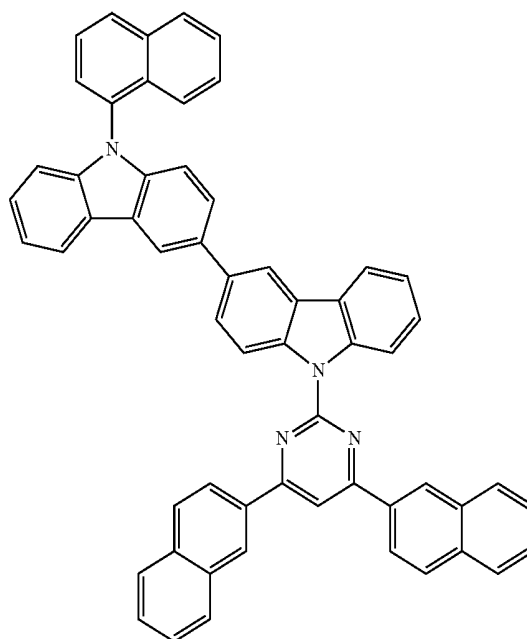
Compound 143



Compound 142



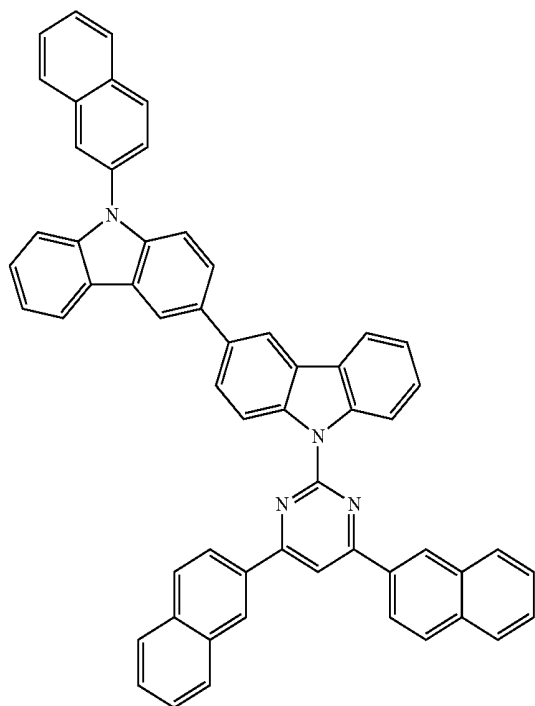
Compound 144



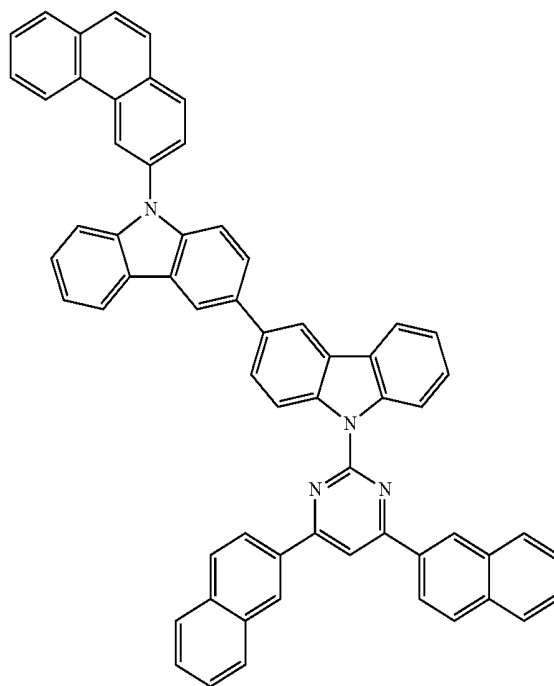
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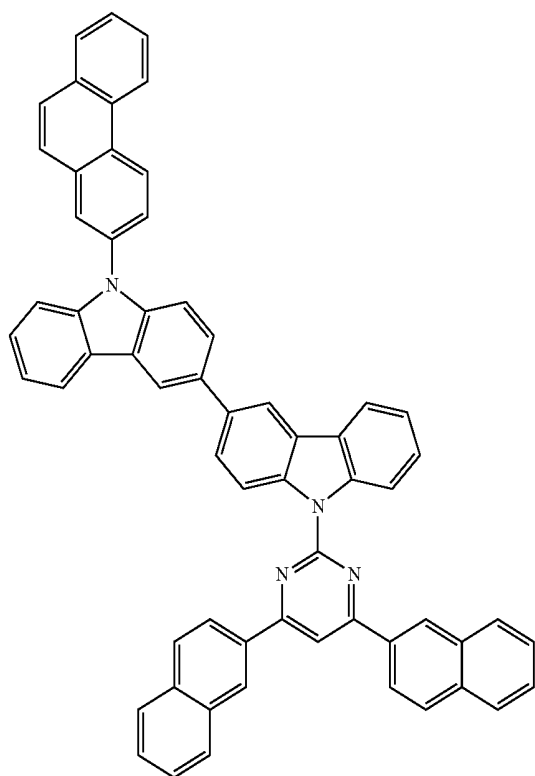
Compound 145



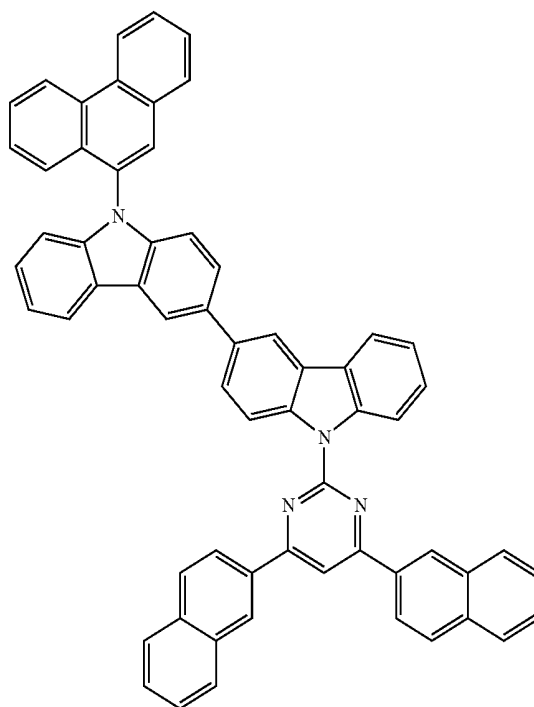
Compound 147



Compound 146



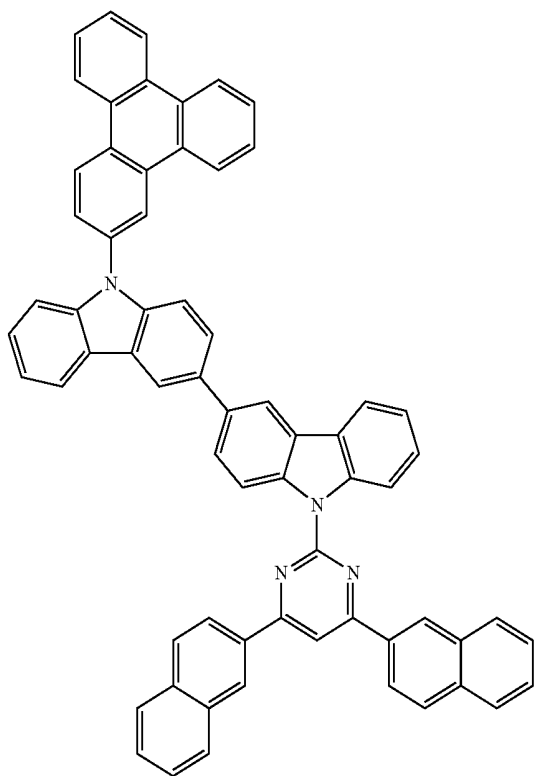
Compound 148



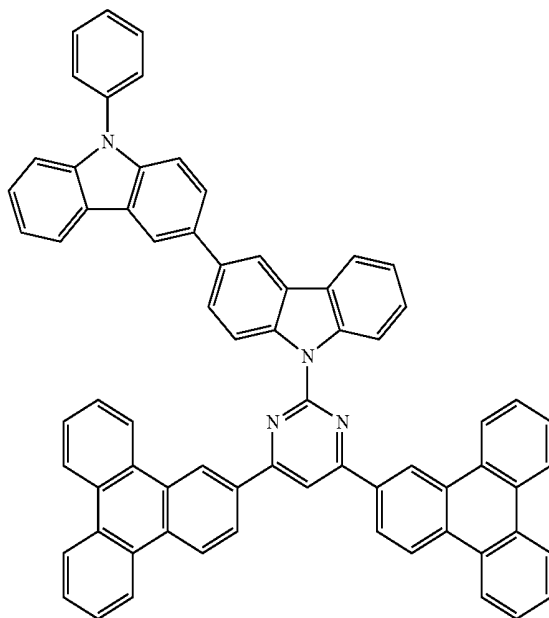
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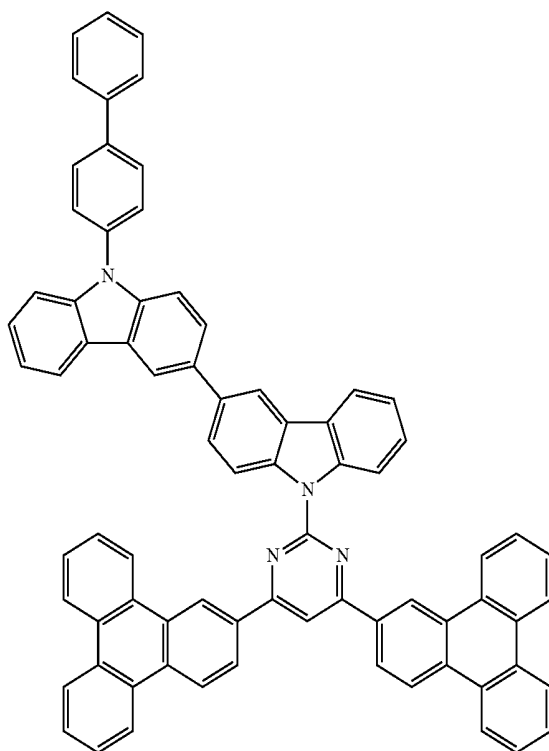
Compound 149



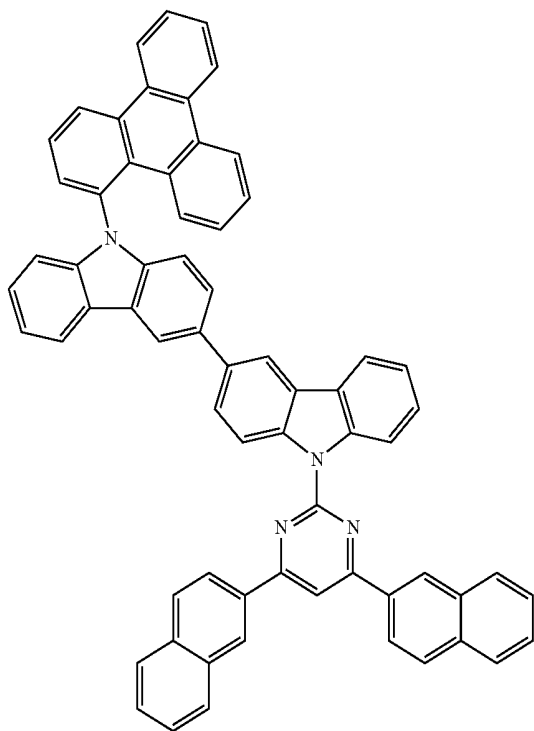
Compound 151



Compound 152



Compound 150

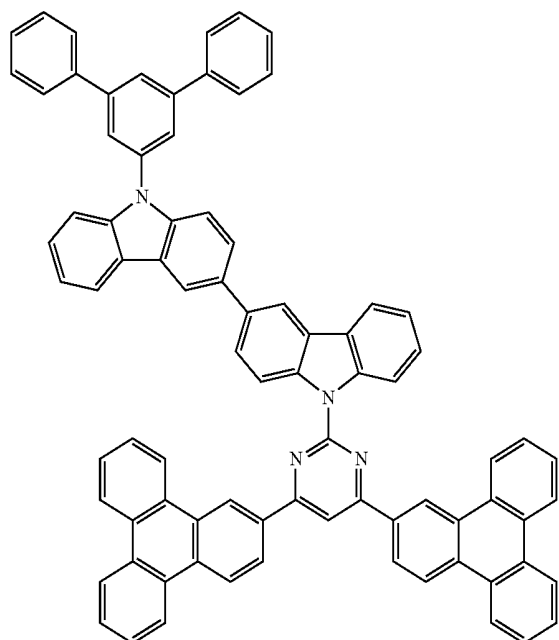




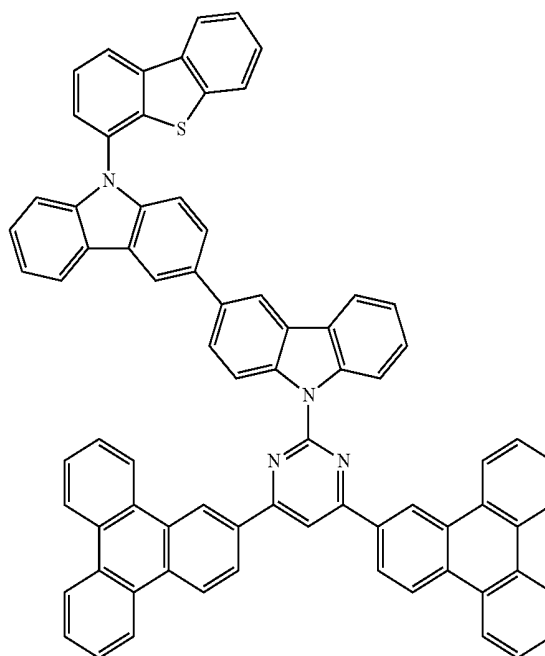
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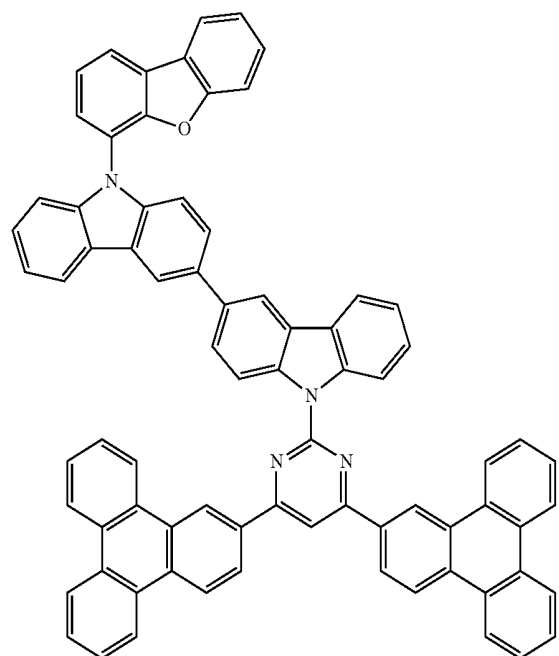
Compound 153



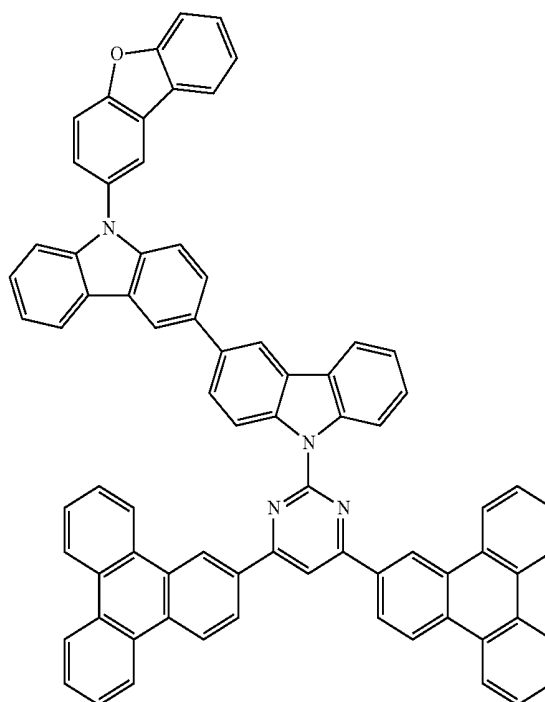
Compound 155



Compound 154



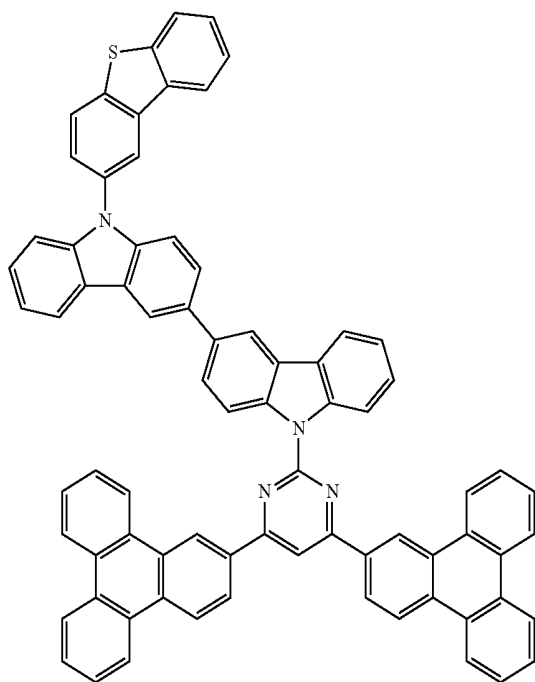
Compound 156



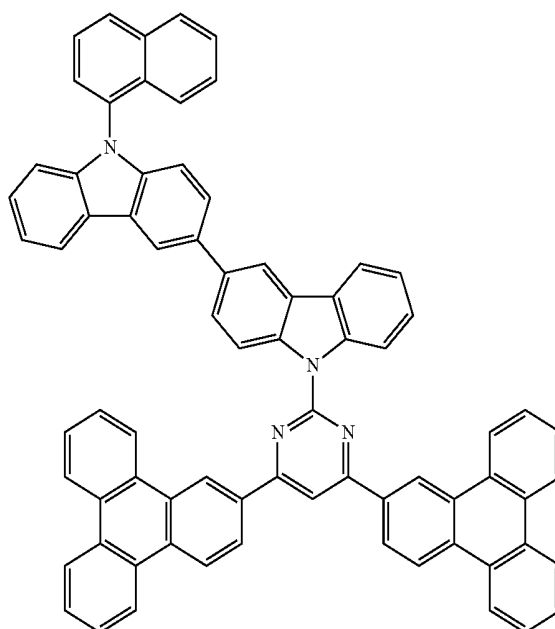
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Compound 157

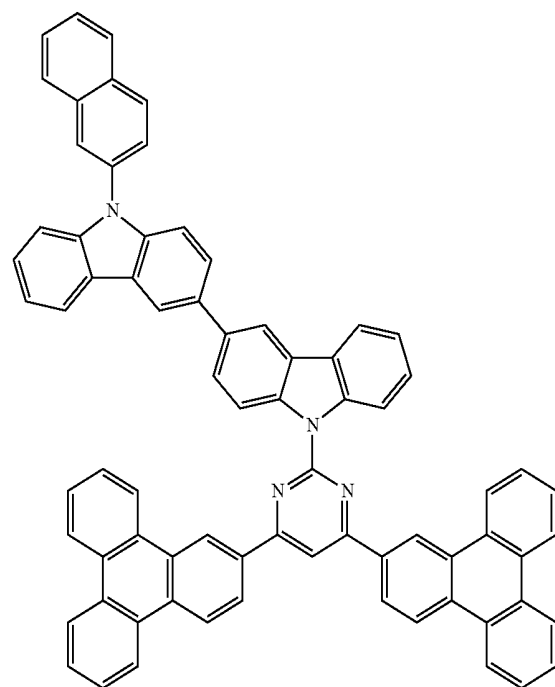
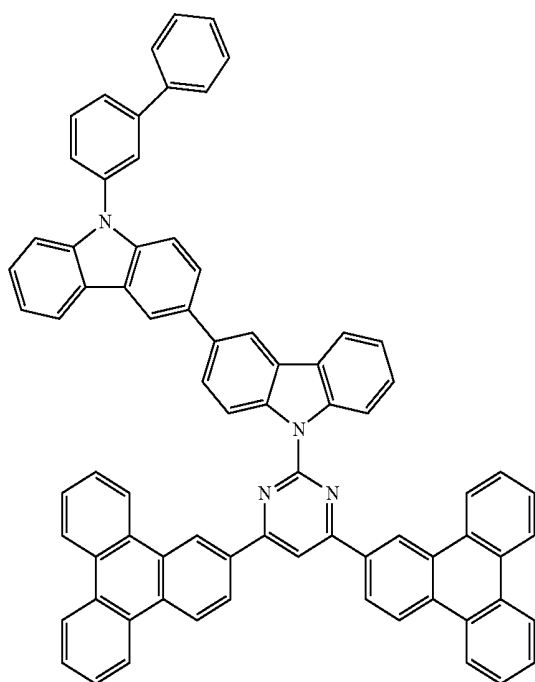


Compound 159



Compound 160

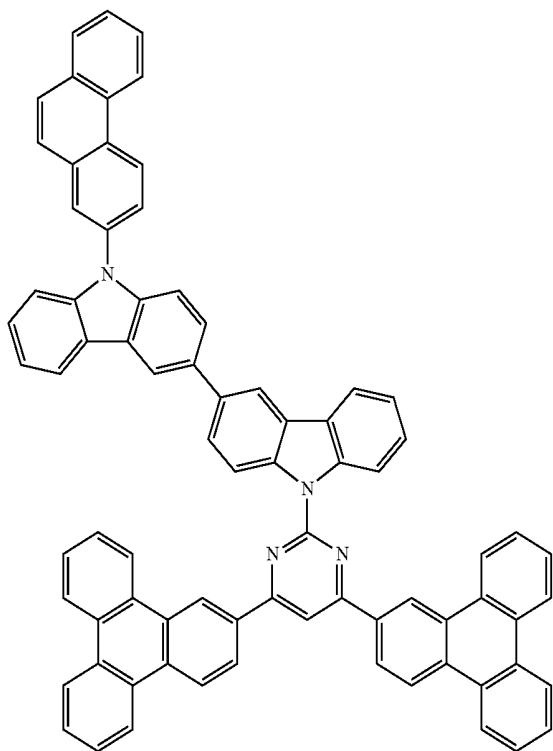
Compound 158



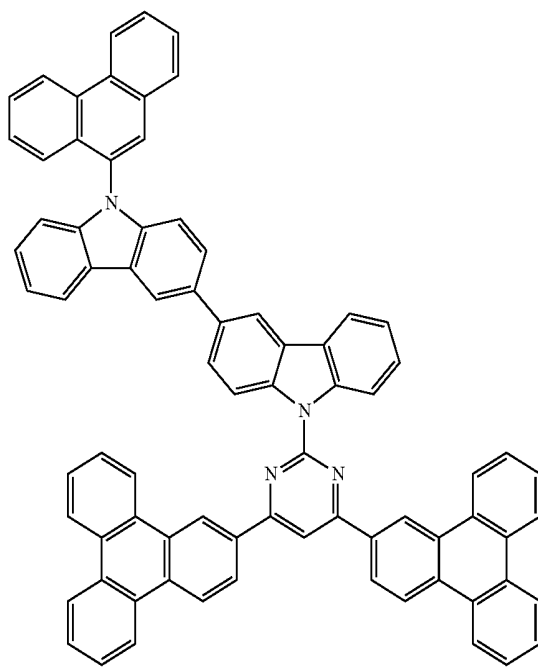
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Compound 161

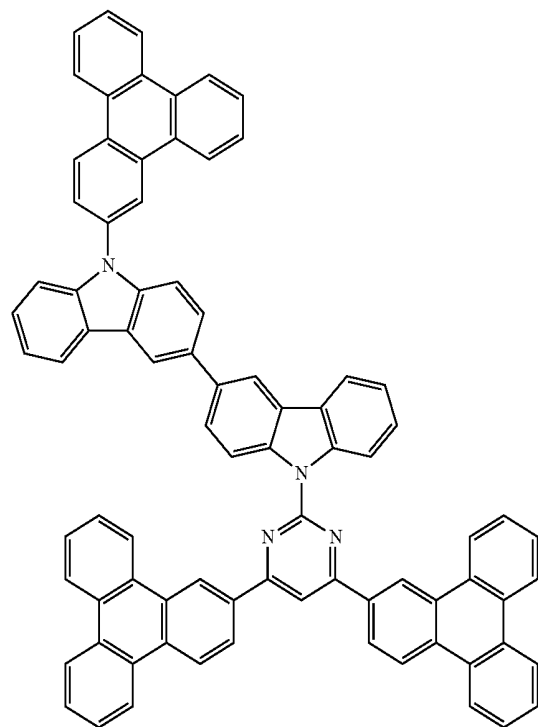
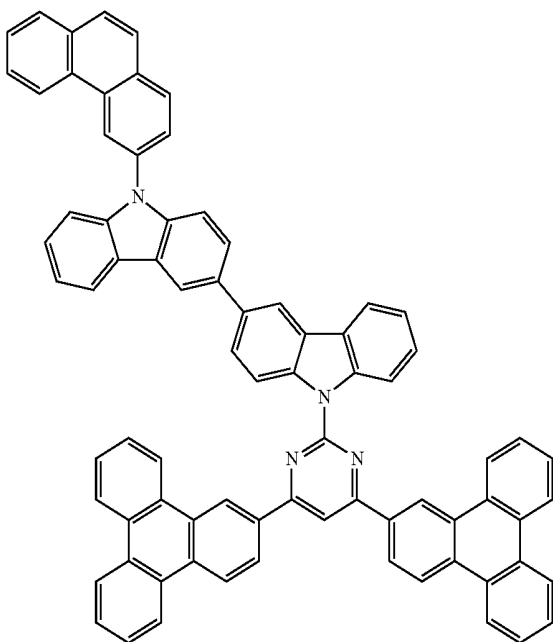


Compound 163



Compound 164

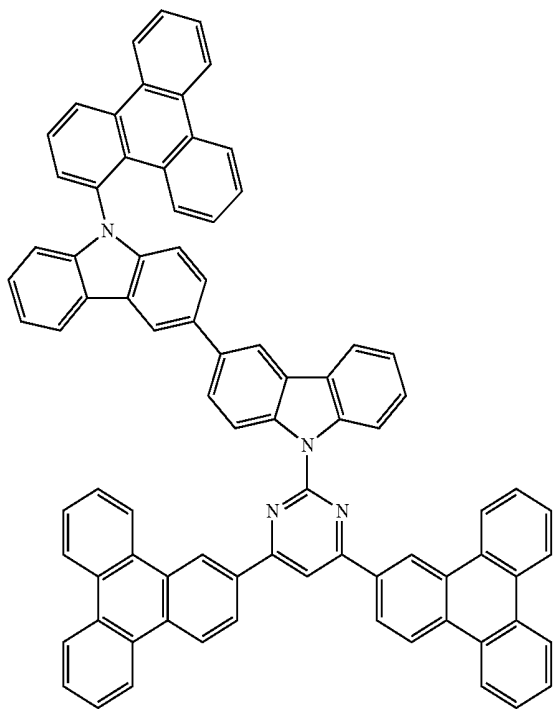
Compound 162



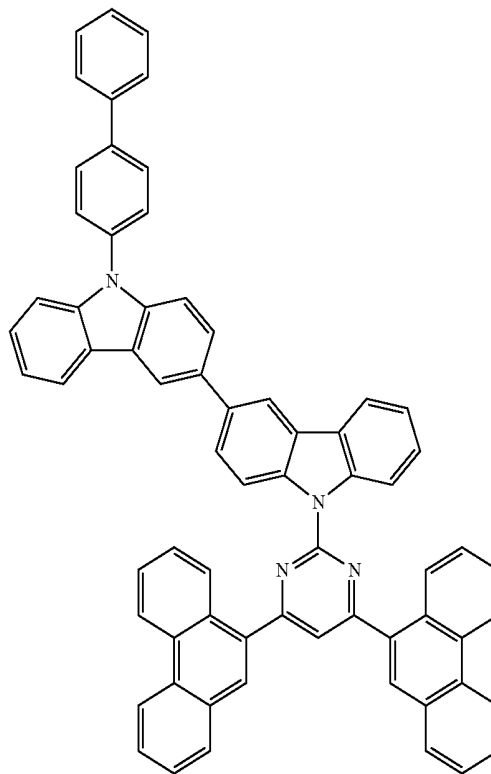
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-continued

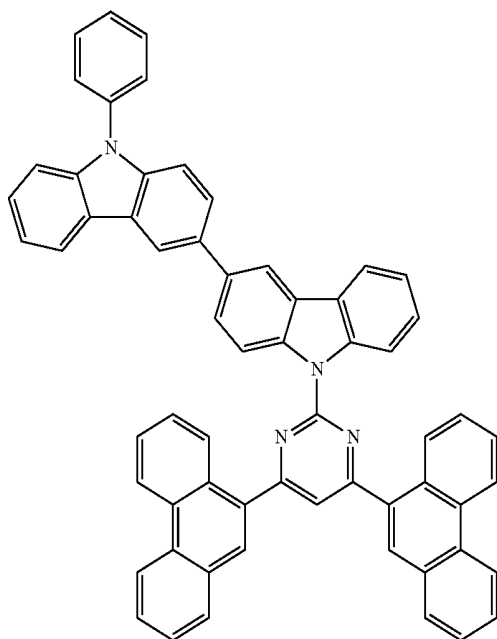
Compound 165



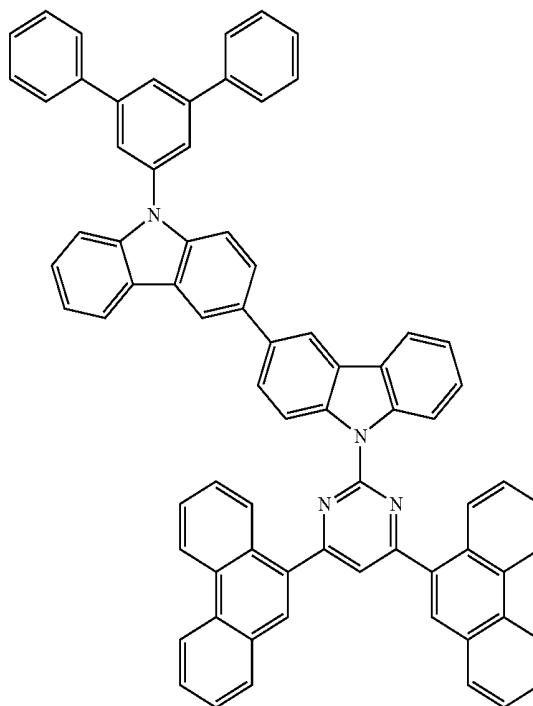
Compound 167



Compound 166

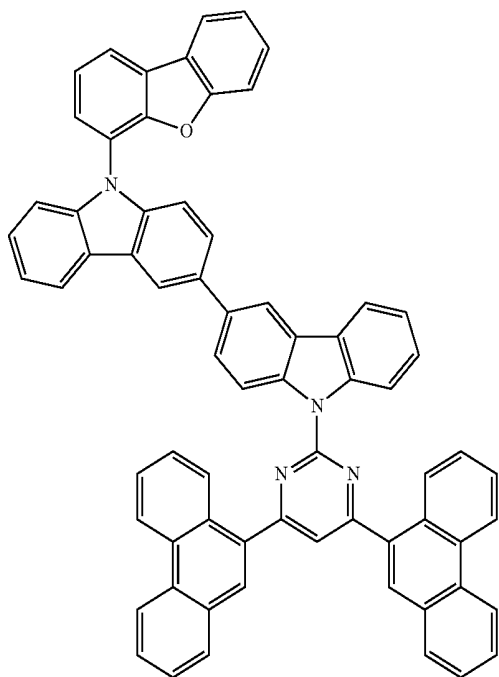


Compound 168



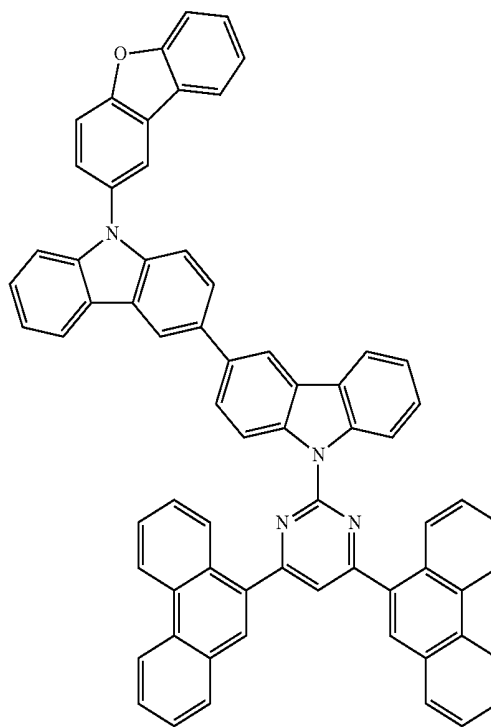
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Compound 169

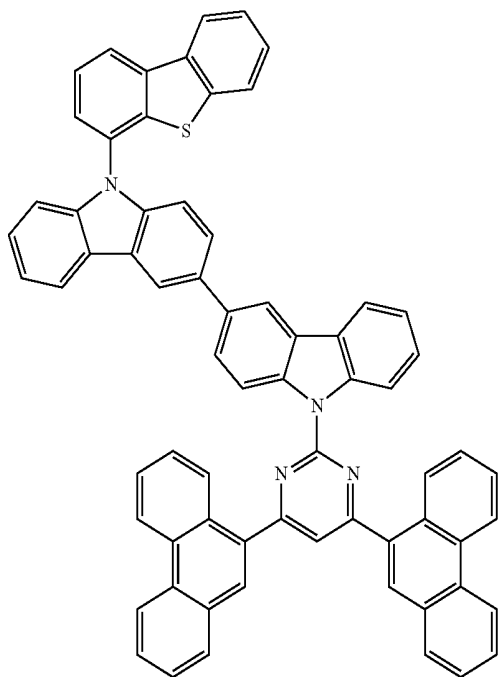


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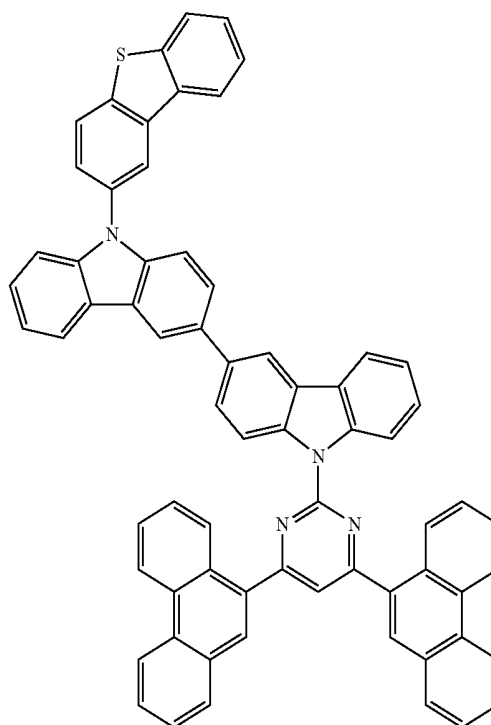
Compound 171



Compound 170



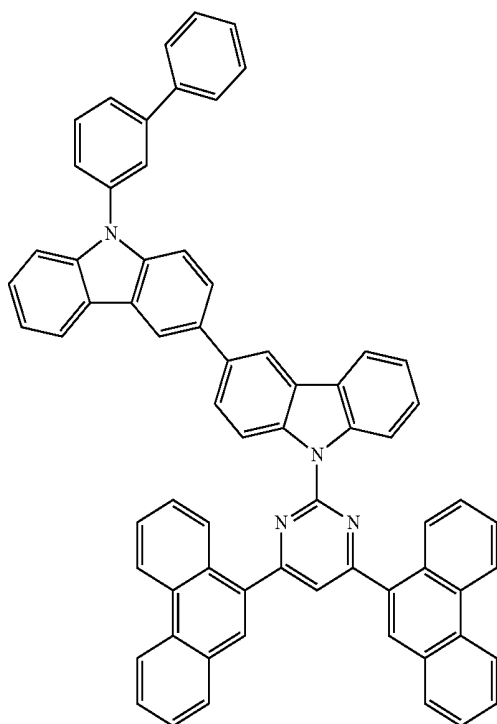
Compound 172



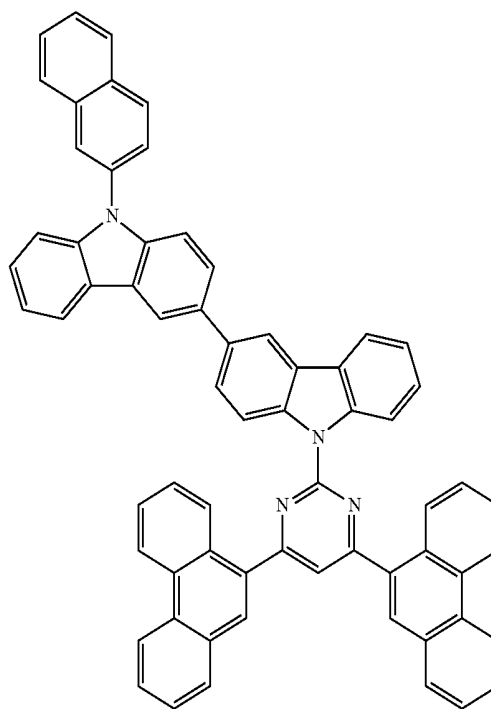
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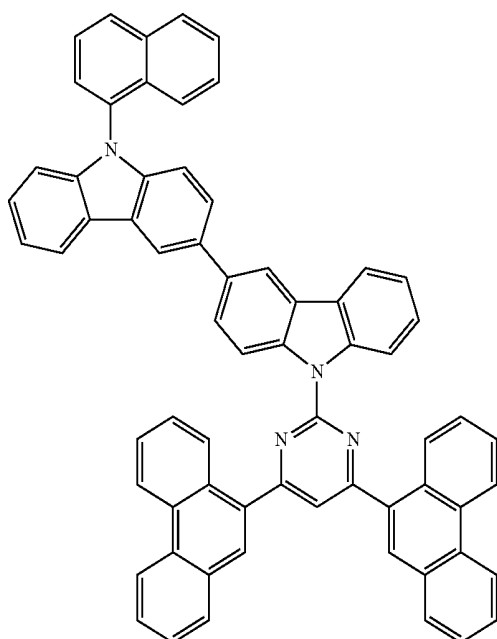
Compound 173



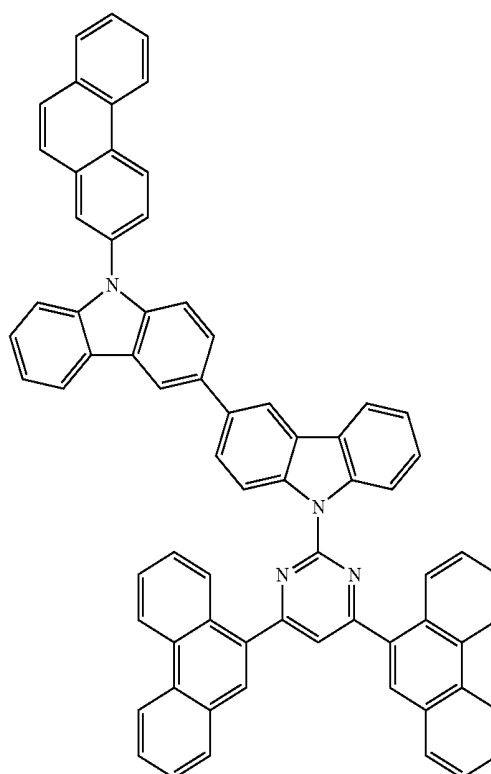
Compound 175



Compound 174



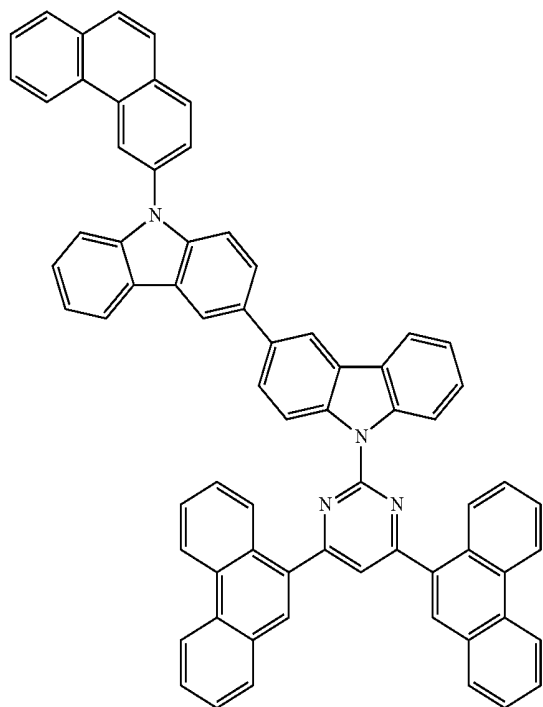
Compound 176



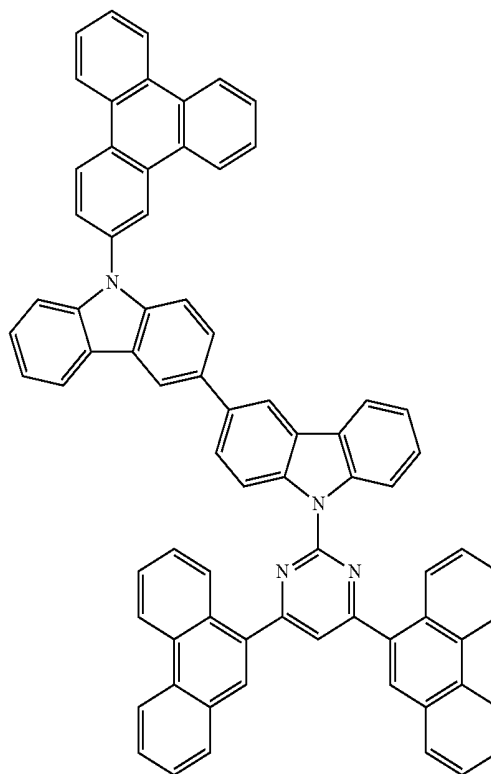
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-continued

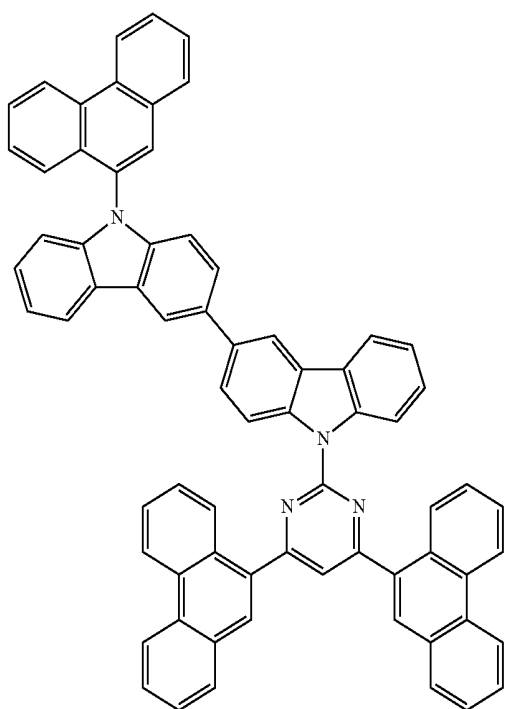
Compound 177



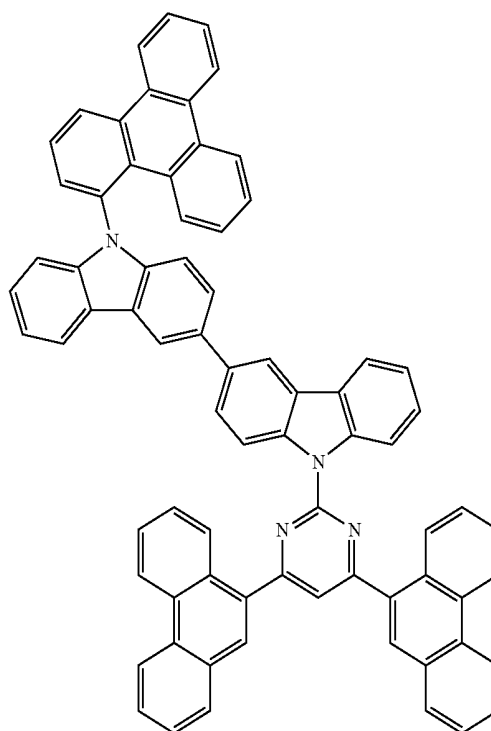
Compound 179



Compound 178

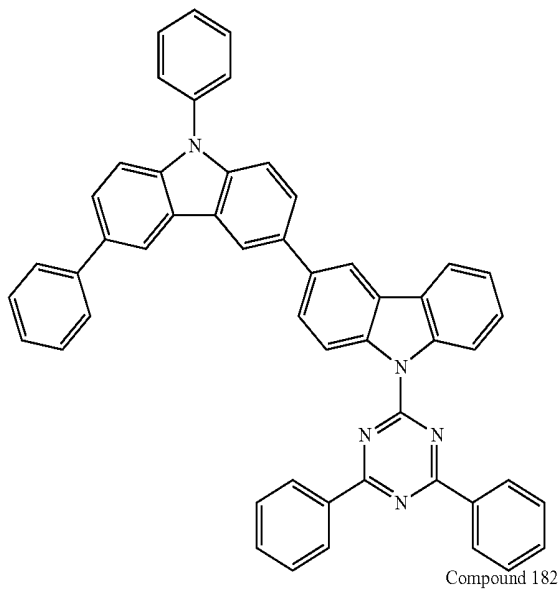


Compound 180

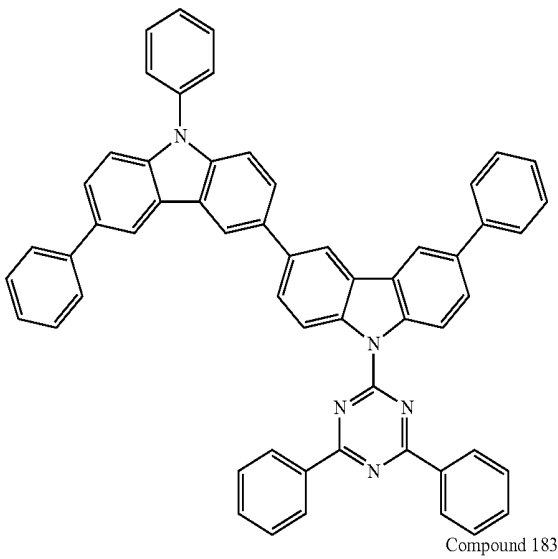


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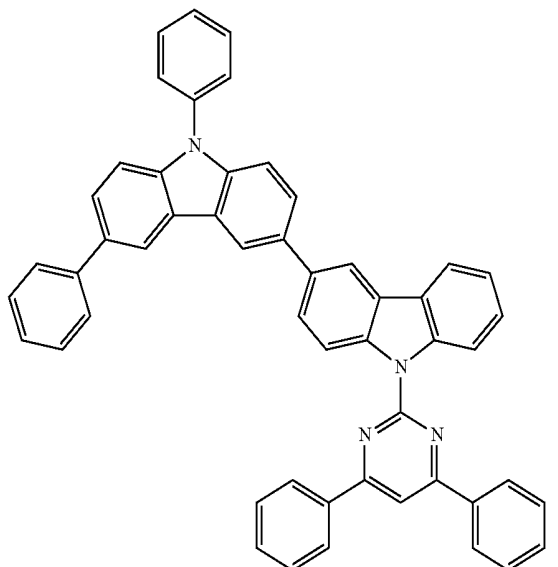
Compound 181



Compound 182

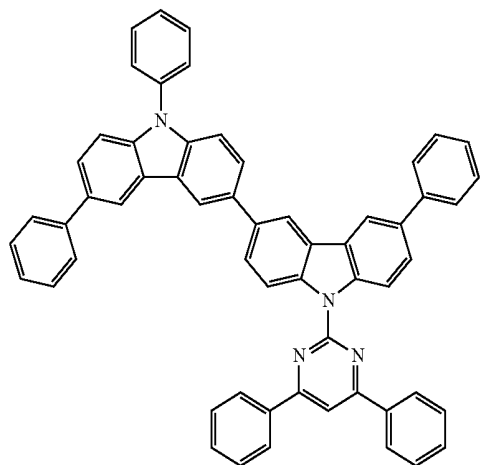


Compound 183



-continued

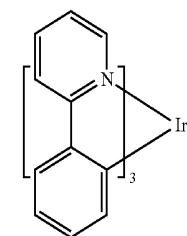
Compound 184



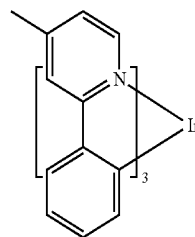
15. The device of claim 8, wherein the organic layer is deposited using solution processing.

16. The device of claim 8, wherein the organic layer is an emissive layer and the compound having Formula I is a host.

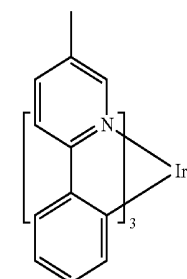
17. The device of claim 16, wherein the organic layer further comprises an emissive dopant having the structure:



D1



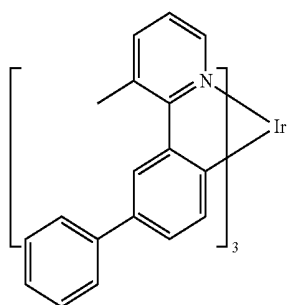
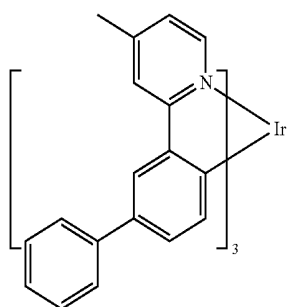
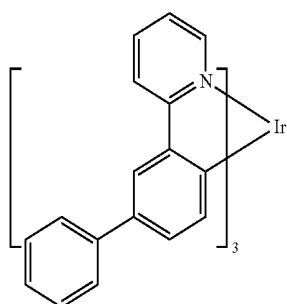
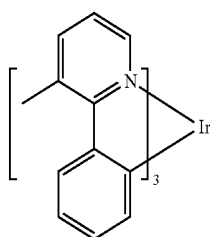
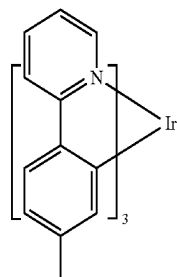
D2



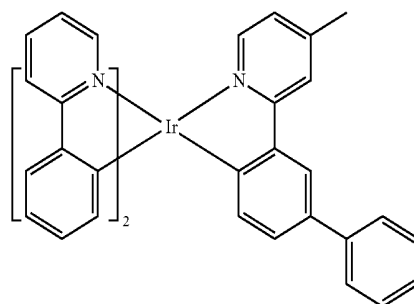
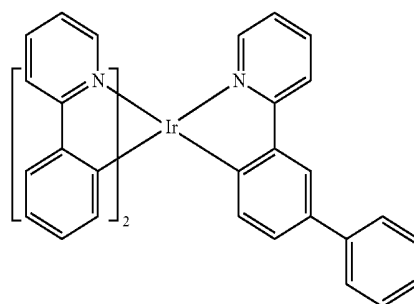
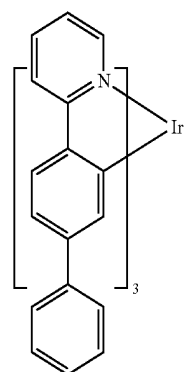
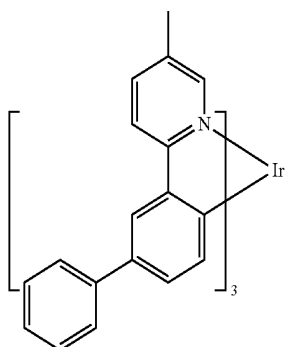
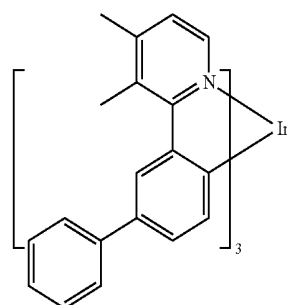
D3



-continued



-continued



D4

D9

D5

D10

D6

D11

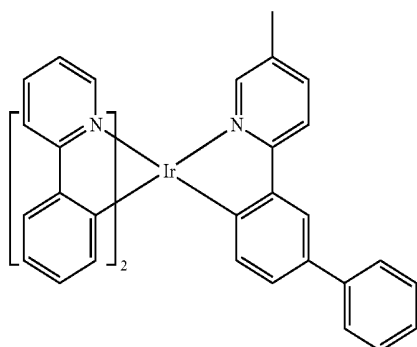
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D12

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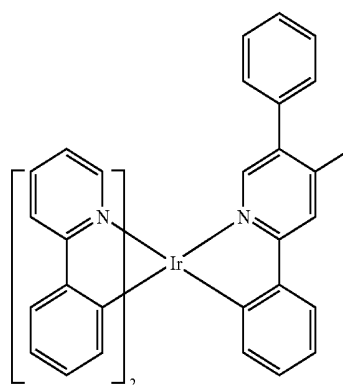
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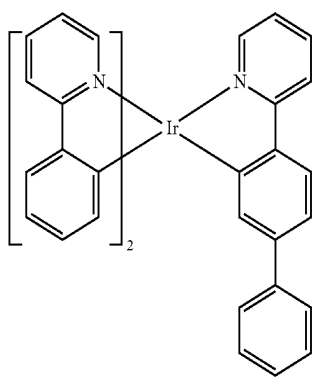


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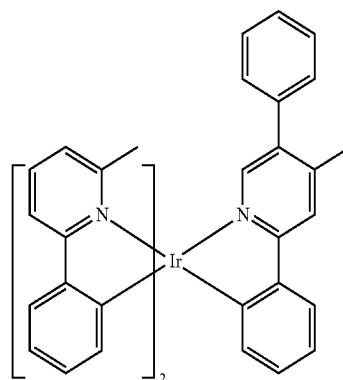
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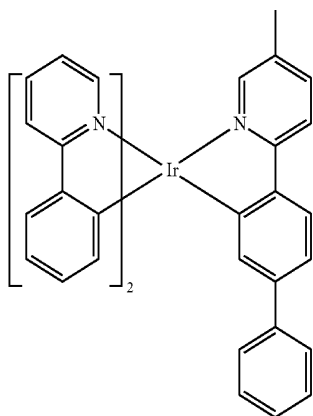
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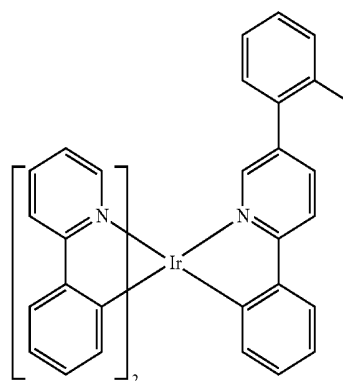
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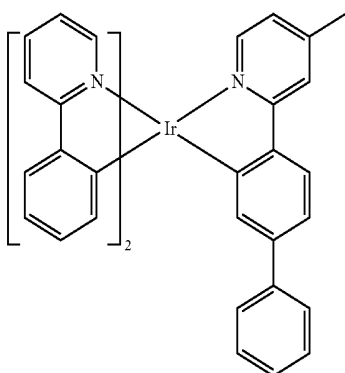
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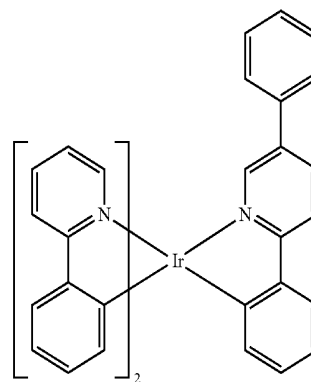
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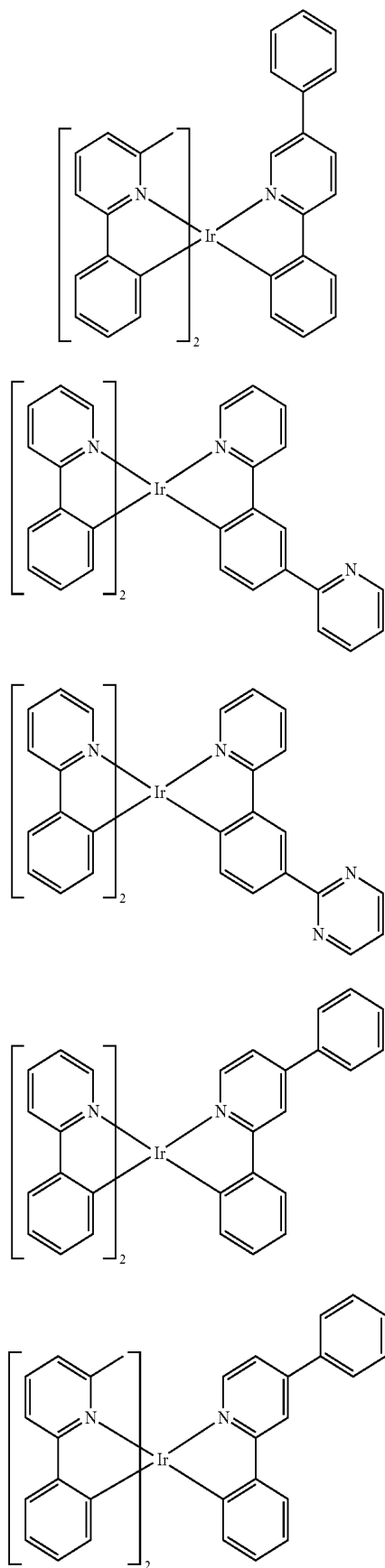


D17



D21

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D22

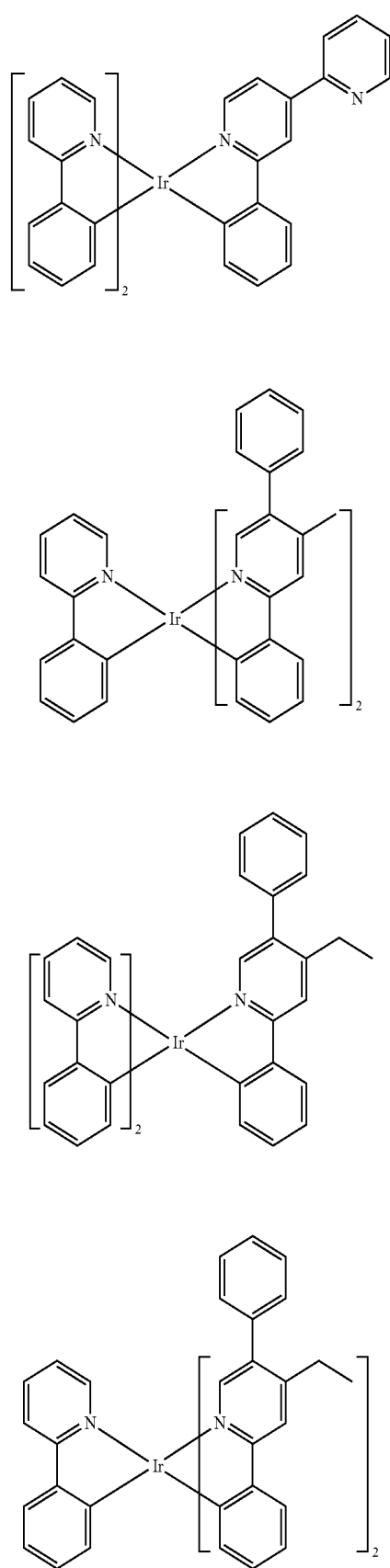
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D24

D25

D26

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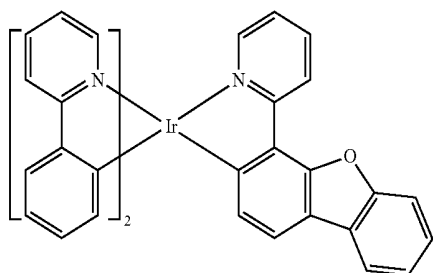
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D28

D29

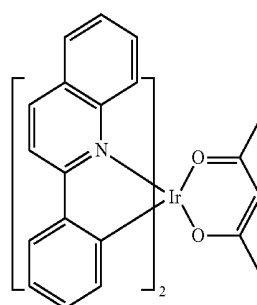
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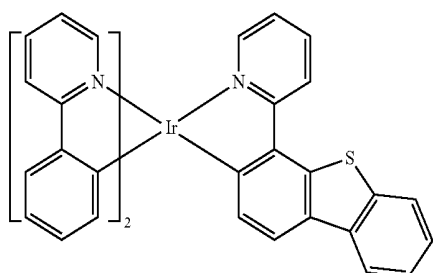


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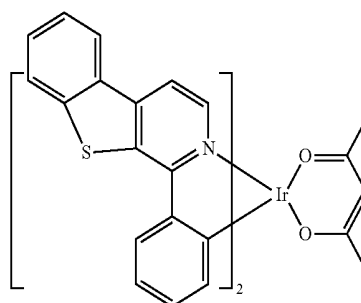
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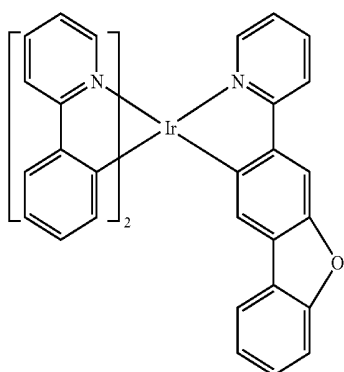
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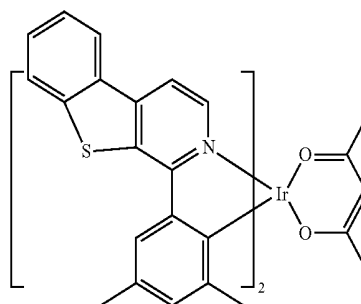
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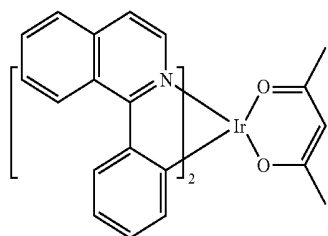
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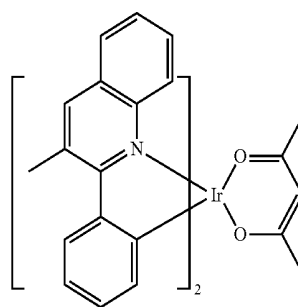
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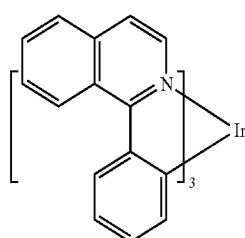
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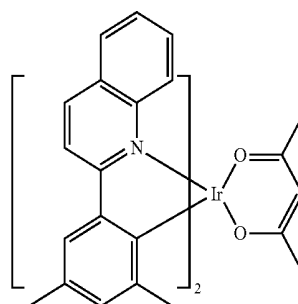
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D39

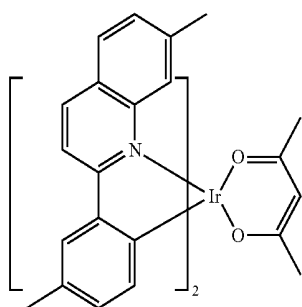


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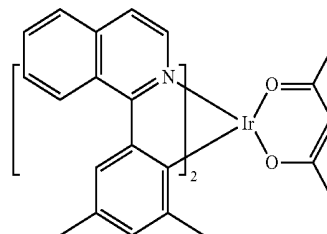
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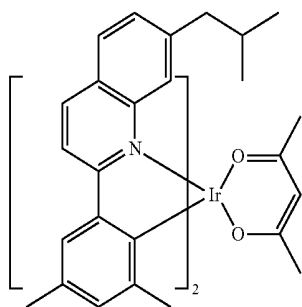


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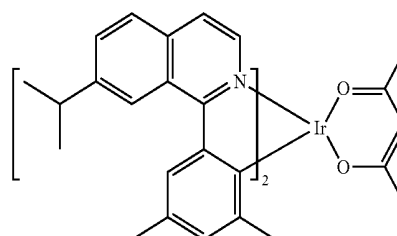
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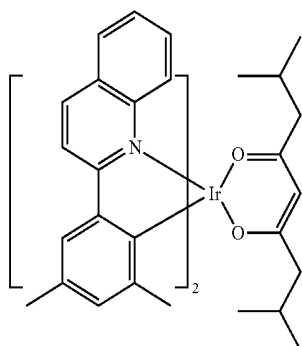
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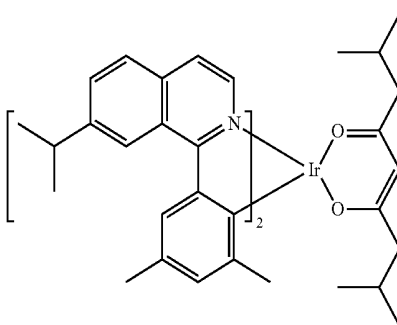
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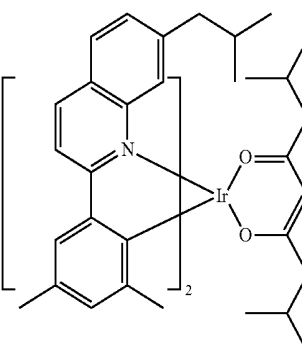
D46



D43



D47



D44

18. The device of claim 8, wherein the first device is a consumer product.

19. The device of claim 8, wherein the first device is an organic light emitting device.

\* \* \* \* \*

|                |  |         |            |
|----------------|--|---------|------------|
| 专利名称(译)        | 用于oleds的双咪唑化合物   |         |            |
| 公开(公告)号        | <a href="#">US20130140549A1</a>  | 公开(公告)日 | 2013-06-06 |
| 申请号            | US13/816407  | 申请日     | 2010-08-20 |
| [标]申请(专利权)人(译) | 夏传军<br>邝RAYMOND<br>黄KEN TSUNG<br>国明城   |         |            |
| 申请(专利权)人(译)    | 夏，川军<br>邝，RAYMOND<br>黄，KEN震<br>KUO，明成  |         |            |
| 当前申请(专利权)人(译)  | 通用显示器公司  |         |            |
| [标]发明人         | XIA CHUANJUN<br>KWONG RAYMOND<br>WONG KEN TSUNG<br>KUO MING CHENG  |         |            |
| 发明人            | XIA, CHUANJUN<br>KWONG, RAYMOND<br>WONG, KEN-TSUNG<br>KUO, MING-CHENG                                      |         |            |
| IPC分类号         | H01L51/00  |         |            |
| CPC分类号         | C07D401/14 C07D403/14 C07D405/14 H01L51/5016 H01L51/0067 H01L51/0072 H01L51/0085<br>C07D409/14 H01L27/3209 |         |            |
| 其他公开文献         | US9954180  |         |            |
| 外部链接           | <a href="#">Espacenet</a> <a href="#">USPTO</a>  |         |            |

#### 摘要(译)

提供了包含双咪唑核的新型有机化合物。特别地，该化合物具有在3位被三嗪或嘧啶取代的3,3'-二咪唑核。该化合物可用于有机发光器件中，以提供具有改进的效率和改善的寿命的器件。

